

## Supporting Information

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## 1. Synthesis

The reactions were partly done in an inert gas atmosphere performing standard Schlenk techniques, Toluene was pre-dried with sodium, pentane was dried with CaH<sub>2</sub>. <sup>n</sup>Pr<sub>3</sub>PAuCl, Et<sub>3</sub>PAuCl and <sup>n</sup>Pr<sub>2</sub><sup>n</sup>Bu were synthesized as described earlier by our group.

### 1.1 Synthesis of GaCp

GaCp is generated in situ before each reduction and then used directly. 196 mg GaI (1 mmol) and 89 mg NaCp (1 mmol) are suspended in 20 ml toluene and stirred for one hour. The green solution turns gray after a few minutes. After one hour, stirring is stopped to allow the metathesis salt to settle. The GaCp dissolved in toluene is then filtered off and can be used for reduction.

### 1.2 Synthesis of Au<sub>6</sub>(GaCl<sub>2</sub>)<sub>4</sub>(PEt<sub>3</sub>)<sub>6</sub> **1**:

351 mg of Et<sub>3</sub>PAuCl (1 mmol) is dissolved in 20 ml of toluene. After one hour, a solution of 1 mmol GaCp dissolved in toluene is added. The dark red solution is stirred overnight before all volatiles are removed in vacuo. The crude product is washed with pentane, dissolved in toluene and stored at -30 °C. **1** crystallizes after one week.

Yield in respect to gold: 5 mg, 2 µmol, 1.3 %

<sup>31</sup>P-NMR (C<sub>6</sub>D<sub>6</sub>): 45.2 (2P), 56.3 (4P)

### 1.3 Synthesis of Au<sub>7</sub>(GaCl<sub>2</sub>)<sub>3</sub>(P<sup>n</sup>Pr<sub>3</sub>)<sub>6</sub> **2**:

393 mg of <sup>n</sup>Pr<sub>3</sub>PAuCl (1 mmol) is dissolved in 20 ml of toluene. After one hour, a solution of 1 mmol GaCp dissolved in toluene is added. The dark red solution is stirred overnight before all volatiles are removed in vacuo. The crude product is dissolved in pentane and stored at -18 °C. **2** crystallizes after a few days.

Yield in respect to gold: 55 mg, 19.9 µmol, 13.9 %

<sup>31</sup>P-NMR (C<sub>6</sub>D<sub>6</sub>): 47.5 (2P), 60.8 (1P), 66.3 (1P), 76.7 (1P), 104.5 (1P)

#### 1.4 Synthesis of Au<sub>13</sub>(GaCl<sub>2</sub>)<sub>5</sub>(GaCl)(P<sup>n</sup>Pr<sub>2</sub><sup>n</sup>Bu)<sub>9</sub> **3**:

407 mg of <sup>n</sup>Pr<sub>2</sub><sup>n</sup>BuPAuCl (1 mmol) is dissolved in 20 ml of toluene. After one hour, a solution of 1 mmol GaCp dissolved in toluene is added. The dark violet solution is stirred overnight before all volatiles are removed in vacuo. The crude product is dissolved in pentane and stored at -30 °C. **3** crystallizes after a few weeks.

Yield in respect to gold: 20 mg, 4.1 µmol, 5.2 %

<sup>31</sup>P-NMR (C<sub>6</sub>D<sub>6</sub>): 34.8 pm (2P), 33.6, 35, 37.8, 39 (2P), 45.8 (2P), 60.9 (1P), 69 (2P)

#### 1.5 Synthesis of Au<sub>13</sub>(GaCl<sub>2</sub>)<sub>5</sub>(P<sup>n</sup>Pr<sub>3</sub>)<sub>9</sub> **4**:

393 mg of <sup>n</sup>Pr<sub>3</sub>PAuCl (1 mmol) is dissolved in 20 ml of toluene. After one hour, a solution of 1 mmol GaCp dissolved in toluene is added. The dark red solution is stirred overnight before all volatiles are removed in vacuo. The crude product is dissolved in pentane and stored at -18 °C. **4** crystallizes after one week.

Yield in respect to gold: 2 mg, 0.4 µmol, 0.55 %

## 2. NMR-Data

Device: Bruker AVIIHD-300

The chemical shifts are given in ppm against the external standard 85% phosphoric acid. C<sub>6</sub>D<sub>6</sub> was dried with 3 Å molecular sieves.

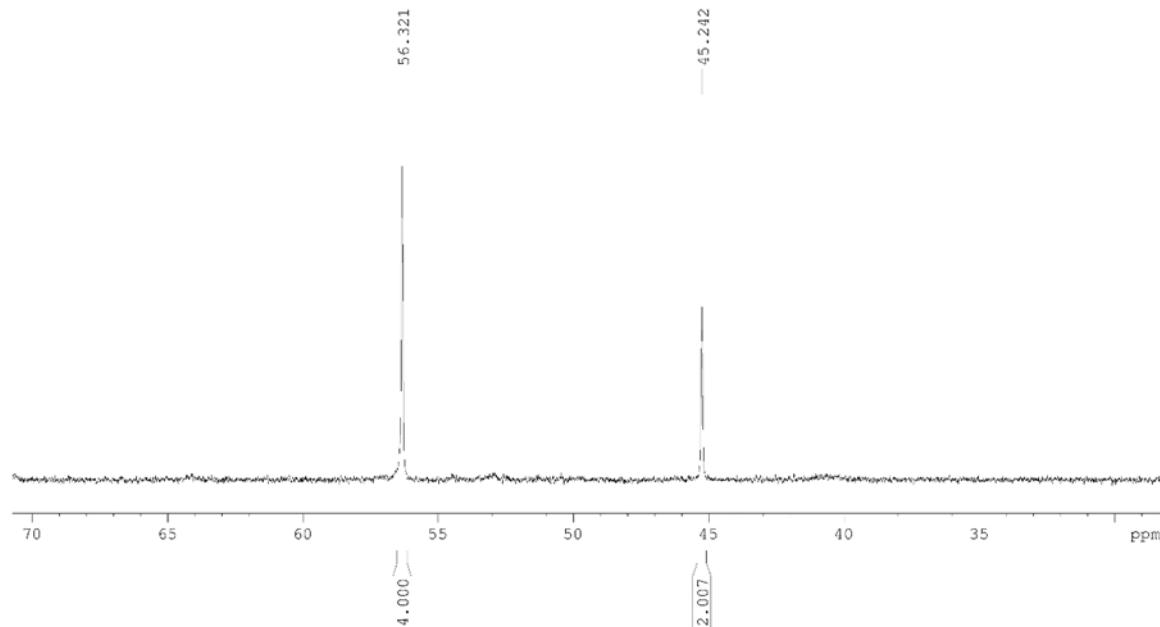


Figure 1: <sup>31</sup>P-NMR spectra of Au<sub>6</sub>(GaCl<sub>2</sub>)<sub>4</sub>(PEt<sub>3</sub>)<sub>6</sub> **1**

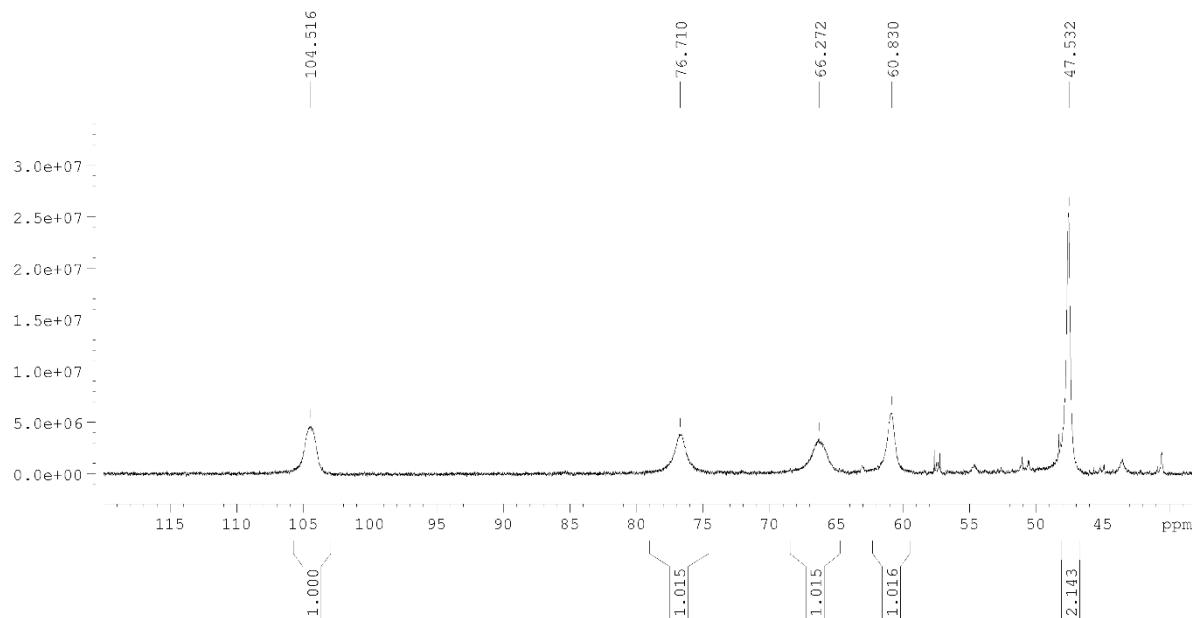


Figure 2:  $^{31}\text{P}$ -NMR spectra of  $\text{Au}_7(\text{GaCl}_2)_3(\text{P}^n\text{Pr}_3)_6$  **2**

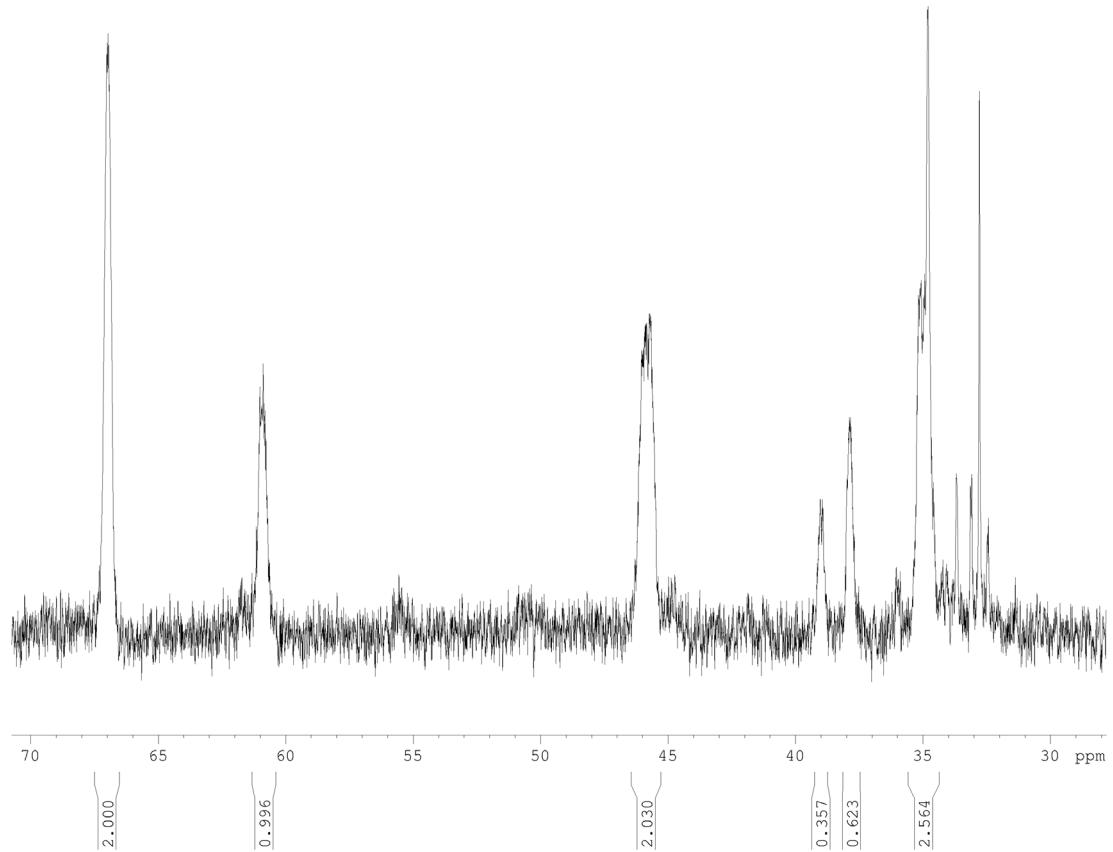


Figure 3:  $^{31}\text{P}$ -NMR spectra of  $\text{Au}_{13}(\text{GaCl}_2)_5(\text{GaCl})(\text{P}^n\text{Pr}_2^n\text{Bu})_9$  **3**

### 3. Crystallographic data

Crystals of **1,2** and **4** were mounted on the diffractometer at 150 K (200 K for **4**). The data were collected on a Bruker APEX II DUO diffractometer equipped with an  $1\mu\text{S}$  microfocus sealed tube and QUAZAR optics for monochromated MoK $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) and equipped with an Oxford Cryosystems cryostat. A semiempirical absorption correction was applied using the program SADABS. The data of **3** were collected on a Rigaku XtaLAB Synergy-S X-ray diffractometer for single crystal X-ray diffraction at 100 K equipped with a PhotonJet-S microfocus sealed tube for monochromated CuK $\alpha$  radiation ( $\lambda = 1.54184 \text{ \AA}$ ) and equipped with an Oxford Cryosystems cryostat. A numerical absorption correction based on gaussian integration over a multifaceted crystal model was applied using the program CrysAlisPro. The structures were solved by direct methods and refined against F2 for all observed reflections. Programs used: SHELXS and SHELXL<sup>1</sup> within the Olex2 program package.<sup>2</sup> The supplementary crystallographic data (for CCDC numbers 2226515 (**1**), 2226516 (**2**), 2226517 (**3**) and 2226518 (**4**)) can be obtained online free of charge at [www.ccdc.cam.ac.uk/conts/retrieving.html](http://www.ccdc.cam.ac.uk/conts/retrieving.html) or from Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB21EZ; Fax: (+44)1223-336-033; or deposit@ccdc.cam.ac.uk.

**Au<sub>6</sub>(GaCl<sub>2</sub>)<sub>4</sub>(PEt<sub>3</sub>)<sub>6</sub> 1**

CCDC Number	2226515
Empirical formula:	C <sub>50</sub> H <sub>106</sub> Au <sub>6</sub> Cl <sub>8</sub> Ga <sub>4</sub> P <sub>6</sub>
Formula weight:	2637.44
Temperature/K:	150.01
Crystal system:	monoclinic
Space group:	C2/m
a/ $\text{\AA}$	16.034(2)
b/ $\text{\AA}$	23.766(3)
c/ $\text{\AA}$	11.388(3)
$\alpha/^\circ$	90
$\beta/^\circ$	108.155(2)
$\gamma/^\circ$	90
Volume/ $\text{\AA}^3$	4123.8(13)
Z	2
$\rho_{\text{calc}}/\text{cm}^3$	2.124
$\mu/\text{mm}^{-1}$	12.315
F(000)	2460.0
Crystal size/ $\text{mm}^3$	0.21 × 0.08 × 0.06
Radiation	MoK $\alpha$ ( $\lambda = 0.71073$ )
2 $\Theta$ range for data collection/ $^\circ$	4.24 to 51.998
Index ranges	-19 ≤ h ≤ 19, -29 ≤ k ≤ 29, -14 ≤ l ≤ 14
Reflections collected	24715
Independent reflections	4168 [R <sub>int</sub> = 0.0773, R <sub>sigma</sub> = 0.0553]
Data/restraints/parameters	4168/164/244
Goodness-of-fit on F <sub>2</sub>	1.043
Final R indexes [ $ I  >= 2\sigma(I)$ ]	R <sub>1</sub> = 0.0391, wR <sub>2</sub> = 0.0882
Final R indexes [all data]	R <sub>1</sub> = 0.0637, wR <sub>2</sub> = 0.1025
Largest diff. peak/hole / e $\text{\AA}^{-3}$	1.10/-1.65

**Au<sub>7</sub>(GaCl<sub>2</sub>)<sub>3</sub>(P<sup>n</sup>Pr<sub>3</sub>)<sub>6</sub> 2**

CCDC Number	2226516
Empirical formula:	C <sub>54</sub> H <sub>126</sub> Au <sub>7</sub> Cl <sub>6</sub> Ga <sub>3</sub> P <sub>6</sub>
Formula weight:	2761.99
Temperature/K :	150.0
Crystal system :	triclinic
Space group:	P-1
a/A	13.0624(10)
b/A	15.2056(11)
c/A	21.0776(16)
α/°	92.022(2)
β/°	98.393(2)
γ/°	91.024(2)
Volume/A <sub>3</sub>	4137.9(5)
Z	2
ρcalcg/cm <sub>3</sub>	2.217
μ/mm-1	13.656
F(000)	2576.0
Crystal size/mm <sup>3</sup>	0.276 × 0.162 × 0.038
Radiation	MoKα ( $\lambda = 0.71073$ )
2θ range for data collection/°	4.09 to 55.998
Index ranges	-17 ≤ h ≤ 17, -20 ≤ k ≤ 20, -27 ≤ l ≤ 27
Reflections collected	237222
Independent reflections	19942 [R <sub>int</sub> = 0.0722, R <sub>sigma</sub> = 0.0348]
Data/restraints/parameters	19942/42/767
Goodness-of-fit on F <sub>2</sub>	1.076
Final R indexes [I>=2σ (I)]	R <sub>1</sub> = 0.0313, wR <sub>2</sub> = 0.0681
Final R indexes [all data]	R <sub>1</sub> = 0.0472, wR <sub>2</sub> = 0.0786
Largest diff. peak/hole / e A <sup>-3</sup>	1.55/-1.08

**Au<sub>13</sub>(GaCl<sub>2</sub>)<sub>5</sub>(GaCl)(P<sup>n</sup>Pr<sub>2</sub><sup>n</sup>Bu)<sub>9</sub> 3**

CCDC Number	2226517
Empirical formula:	C <sub>90</sub> H <sub>207</sub> Au <sub>13</sub> Cl <sub>11</sub> Ga <sub>6</sub> P <sub>9</sub>
Formula weight:	4937.11
Temperature/K :	100.00(13)
Crystal system:	monoclinic
Space group:	P2 <sub>1</sub> /n
a/Å	17.9369(2)
b/Å	29.9431(3)
c/Å	26.0812(3)
α/°	90
β/°	99.9143(12)
γ/°	90
Volume/A <sub>3</sub>	13798.7(3)
Z	4
ρcalcg/cm <sup>3</sup>	2.377
μ/mm <sup>-1</sup>	29.702
F(000)	9128.0
Crystal size/mm <sup>3</sup>	0.146 × 0.08 × 0.023
Radiation	Cu Kα ( $\lambda = 1.54184$ )
2θ range for data collection/°	4.532 to 129.998
Index ranges	-21 ≤ h ≤ 21, -20 ≤ k ≤ 35, -30 ≤ l ≤ 30
Reflections collected	115496
Independent reflections	23236 [R <sub>int</sub> = 0.0711, R <sub>sigma</sub> = 0.0474]
Data/restraints/parameters	23236/549/1147
Goodness-of-fit on F <sub>2</sub>	1.054
Final R indexes [I>=2σ (I)]	R <sub>1</sub> = 0.0526, wR <sub>2</sub> = 0.1270
Final R indexes [all data]	R <sub>1</sub> = 0.0745, wR <sub>2</sub> = 0.1436
Largest diff. peak/hole / e A <sup>-3</sup>	1.74/-1.70

**Au<sub>13</sub>(GaCl<sub>2</sub>)<sub>5</sub>(P<sup>n</sup>Pr<sub>3</sub>)<sub>9</sub> 4**

CCDC Number	2226518
Empirical formula:	C <sub>81</sub> H <sub>189</sub> Au <sub>13</sub> Cl <sub>10</sub> Ga <sub>5</sub> P <sub>9</sub>
Formula weight.	4705.71
Temperature/K:	200
Crystal system:	monoclinic
Space group:	P21/n
a/Å	16.1624(5)
b/Å	27.8938(8)
c/Å	28.6686(10)
α/°	90
β/°	94.5860(10)
γ/°	90
Volume/Å <sup>3</sup>	12883.3(7)
Z	4
ρ <sub>calc</sub> g/cm <sup>3</sup>	2.426
μ/mm <sup>-1</sup>	16.114
F(000)	8648.0
Crystal size/mm <sup>3</sup>	0.093 × 0.081 × 0.044
Radiation	MoKα ( $\lambda = 0.71073$ )
2θ range for data collection/°	4.046 to 49.5
Index ranges	-19 ≤ h ≤ 19, -32 ≤ k ≤ 32, -33 ≤ l ≤ 33
Reflections collected	224751
Independent reflections	22016 [ $R_{int} = 0.1032$ , $R_{sigma} = 0.0519$ ]
Data/restraints/parameters	22016/193/1220
Goodness-of-fit on F <sub>2</sub>	1.080
Final R indexes [I>=2σ (I)]	$R_1 = 0.0338$ , $wR_2 = 0.0637$
Final R indexes [all data]	$R_1 = 0.0619$ , $wR_2 = 0.0771$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.27/-0.98

#### 4. Quantum chemical calculations

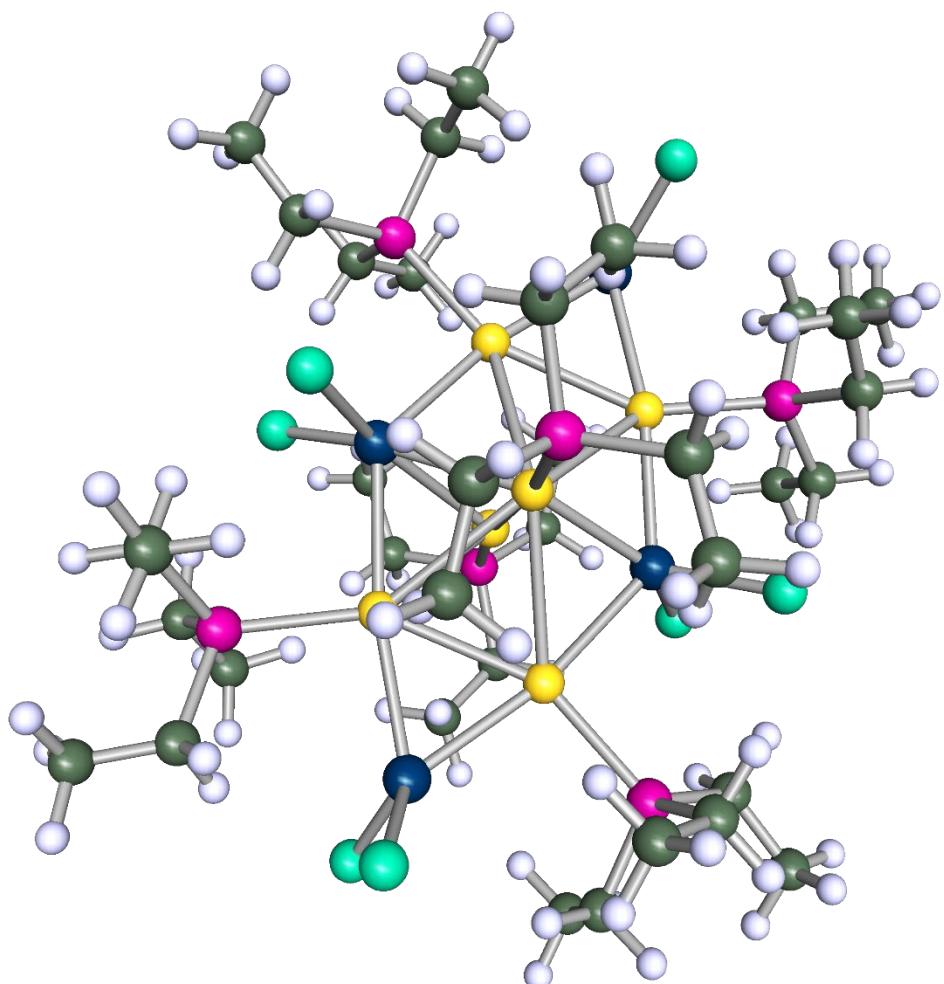


Figure S1: Geometry optimized structure of  $\mathbf{1}^3$

Point group:  $C_1$

Energy: -15669.38746354991 Hartree

HOMO-LUMO-gap: 1.917 eV

Atomic coordinates:

Au 3.660147 14.400983 6.912071

Au 4.335799 11.915599 8.647803

Ga 5.719402 15.722459 5.969620

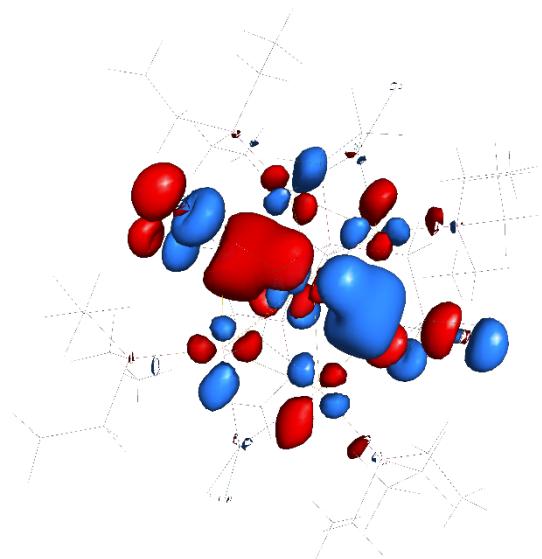
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Au 3.641924 11.284513 6.132132

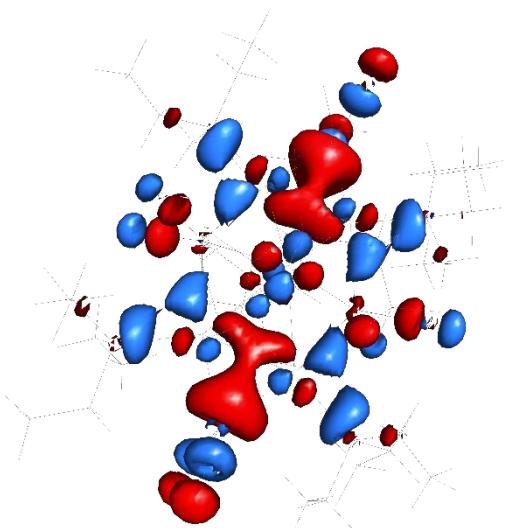
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P 2.026776 16.211033 6.590987	C -0.322779 17.568345 7.613959	H -1.228210 10.105060 11.816621
Ga 6.318041 10.661874 7.375175	C 3.310473 18.344527 7.991432	H -1.448712 11.887706 11.837664
P 8.648117 13.478875 6.239780	C 2.142951 16.066304 3.722845	H 5.232249 16.446137 10.537104
Au 4.517368 8.884398 8.289950	C 10.577366 15.272056 5.025250	H 6.522647 15.231675 10.783476
Au 1.925077 10.016149 8.441495	C 9.187295 14.568889 8.820861	H -0.801753 17.608014 6.611425
P 4.705379 12.654971 10.901700	C 9.014360 12.079172 3.780655	H 0.205721 18.532402 7.779313
P 3.385828 10.534904 3.910571	Cl 2.246422 6.684987 11.063375	H 4.052039 9.837478 12.202520
C 4.474764 14.487977 11.132680	Cl 1.408990 5.923012 7.656678	H 2.732975 10.267055 11.073447
Cl 5.869100 16.369238 3.808186	C 7.596413 6.974730 7.406206	H 10.554817 12.269416 5.335644
Cl 6.733511 17.389765 7.133326	C 5.399469 5.368625 8.530367	H 9.197354 11.175234 5.760721
Cl 1.056221 13.657823 9.885666	C 6.957731 7.169821 10.253805	H -2.444622 10.898264 9.705206
Cl 0.034267 12.956606 6.533060	C -0.950050 8.147165 9.599684	H -1.099301 12.037800 9.365010
C 0.598666 16.351449 7.776948	C -1.431861 9.797645 7.197380	H 10.995796 14.479511 4.367218
C 2.806566 17.905197 6.611251	C -1.348309 11.057484 9.826220	H 11.201284 15.319037 5.944373
C 1.199136 16.101311 4.928874	C 4.915666 12.749916 2.871702	H 8.992927 5.701867 8.558812
Cl 6.902699 9.674119 5.351517	C 3.171436 10.499499 12.070338	H 8.009224 4.797678 7.355625
Cl 8.258569 10.335986 8.621220	C 8.526446 5.762123 7.551348	H 6.547757 13.057113 12.445389
C 9.091541 15.051392 5.340498	C 4.926650 4.930273 7.138710	H 7.120144 12.692497 10.783464
C 9.650136 13.520728 7.802831	C 5.997517 7.106291 11.446710	H 0.584064 15.175429 4.984380
C 9.452869 12.128087 5.248608	C -2.440026 7.910863 9.882591	H 0.496837 16.962867 4.854630
Ga 2.414310 7.492614 8.958072	C -0.862419 8.871186 6.117358	H 2.721210 8.035006 5.571127
P 6.154409 7.073619 8.578843	C -0.945497 11.051243 11.305214	H 1.688625 7.877605 4.107578
P -0.491381 9.768202 8.798529	H 2.845930 15.208517 3.780861	H 9.221531 15.604689 8.420081
C 1.707755 10.759633 3.134456	H 2.760655 16.985022 3.636309	H 8.139574 14.375527 9.137033
C 4.619435 11.254773 2.714247	H 10.714478 13.678435 7.513800	H 1.027151 16.303285 8.800460
C 6.418406 12.364292 11.581266	H 9.565763 12.495671 8.227888	H 0.032539 15.402905 7.654895
C 3.522541 11.986402 12.179269	H 3.691005 8.488951 2.629297	H 7.177801 7.045369 6.379835
C 3.699121 8.711329 3.722322	H 4.741516 8.569084 4.085790	H 8.152730 7.925057 7.560089
C 2.731499 7.800901 4.484113	H 3.946503 12.235879 13.180238	H -0.548403 7.351583 8.932844
C 1.241235 12.206666 2.941958	H 2.603382 12.599391 12.044087	H -0.350739 8.073671 10.533820
C 5.488161 15.372654 10.402225	H 5.288784 12.977789 3.894752	H 0.175622 9.171724 5.856655
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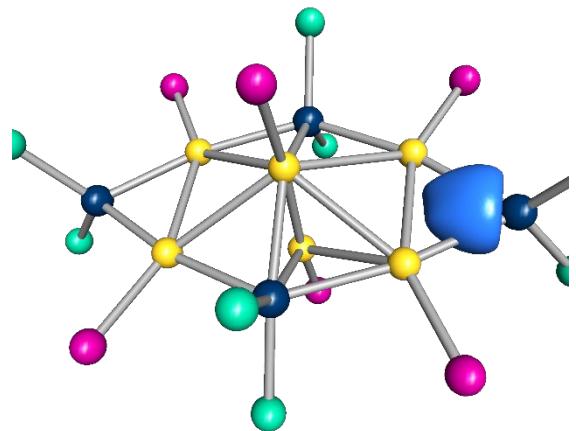
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H -1.393864 10.856679 6.858786	H 5.767865 4.863827 6.414637	H 4.448128 3.928613 7.198163
H 7.812359 10.833820 12.259346	H 4.263756 11.010455 1.685938	H 9.356333 5.844455 6.814359
H 6.549343 10.218702 11.151517	H 5.547979 10.667870 2.893063	H 9.480349 11.204727 3.275843
H 2.051823 18.622492 6.213757	H 6.562821 7.236995 12.396034	H 10.703379 16.242009 4.494841
H 3.646086 17.886643 5.880943	H 5.459467 6.136547 11.505719	H 5.697612 13.069355 2.149109
H 0.996955 10.219805 3.798382	H 4.496320 14.678434 12.231620	H 1.553898 15.966021 2.784137
H 1.730007 10.210717 2.163726	H 3.438461 14.690954 10.781380	H 0.208747 12.218330 2.527785
H -2.862521 8.656398 10.590825	H 7.706115 6.346517 10.309258	H 3.039632 6.737969 4.379327
H -3.054873 7.931168 8.956320	H 7.525117 8.127354 10.238761	H -1.143244 17.511578 8.363751
H 3.807624 19.335906 7.916845	H 1.217495 12.749066 3.911306	H 4.066096 17.634780 8.390379
H 2.482399 18.432297 8.728389	H 1.890438 12.763702 2.233218	H 5.494499 15.169212 9.308210
H 6.158192 4.659795 8.935817	H 7.913472 11.959440 3.697912	H 9.831449 14.531943 9.727377
H 4.548941 5.371946 9.248435	H 9.312409 12.992101 3.220003	H 6.140916 10.598742 12.869382
H 8.479193 15.067873 4.412044	H 0.149016 11.199113 11.420210	H 5.223136 7.901052 11.395144
H 8.695785 15.884709 5.963232	H -2.574413 6.906333 10.341719	H 2.422952 10.220077 12.843623



HOMO-1



HOMO-2



LMO382

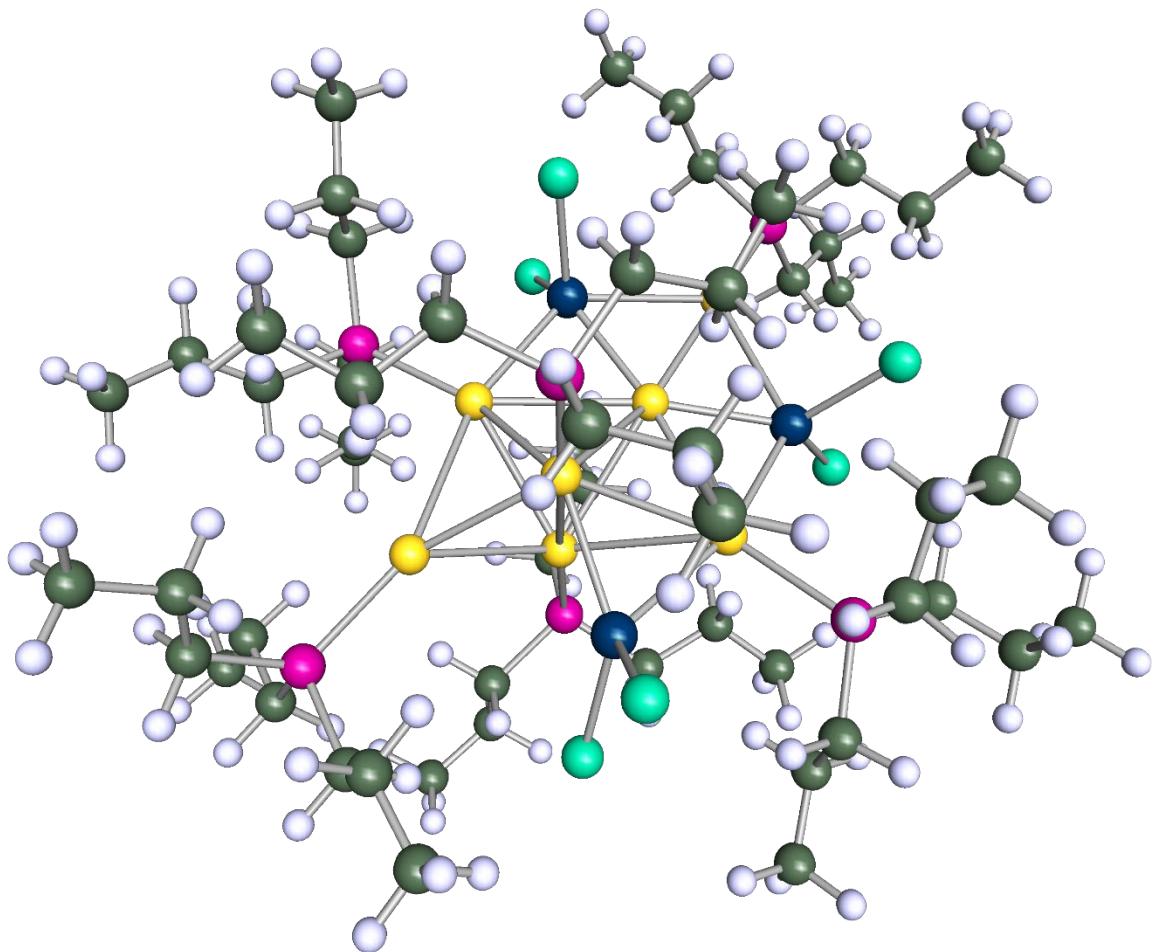


Figure S2: Geometry optimized structure of  $2^3$

Point group:  $C_1$

Energy: -26647.62315677302 Hartree

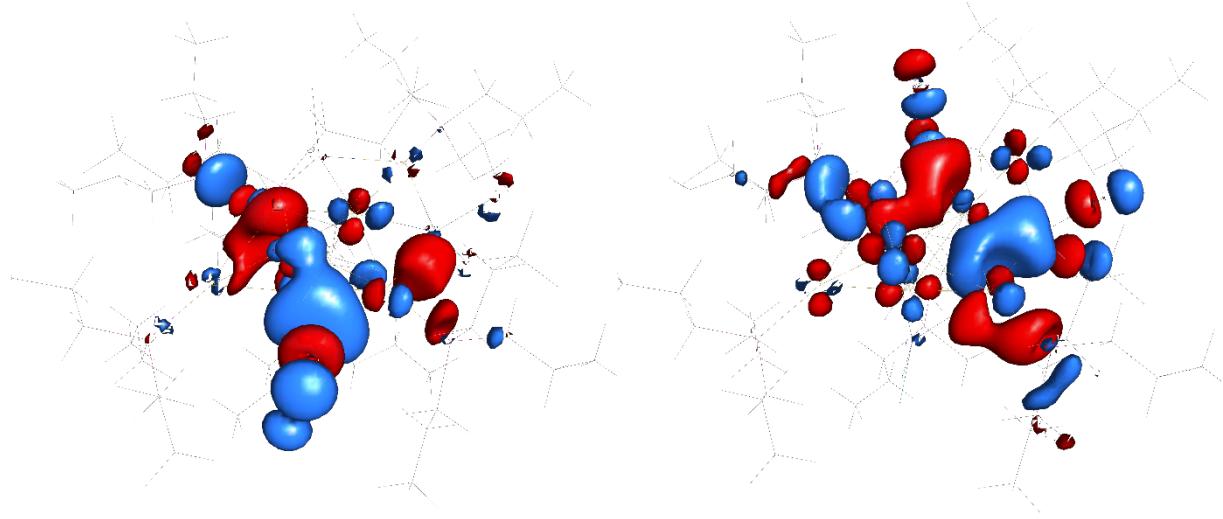
HOMO-LUMO-gap: 1.569 eV

Atomic coordinates:

Au 0.645049 10.211717 16.996844	Ga 1.226665 7.708616 17.563041	Au 4.301291 12.906188 13.682747
Au -1.753719 9.024781 15.941220	P 0.431969 11.812520 18.747515	Ga 5.043102 10.613369 15.252020
Au -0.398380 11.385853 14.647468	P -3.576746 7.445848 16.297020	H 3.146830 12.387093 17.416725
Au 2.617428 11.171687 14.968315	H -2.472890 11.060392 18.166025	P 5.031704 7.636517 17.879576
Au 3.410362 8.970134 16.603412	Ga 1.473590 13.090574 13.863977	H 5.010696 10.366733 19.292337
Au 0.902283 8.728143 14.742373	P -2.248975 12.470648 13.524143	H 4.054515 7.359298 14.144506

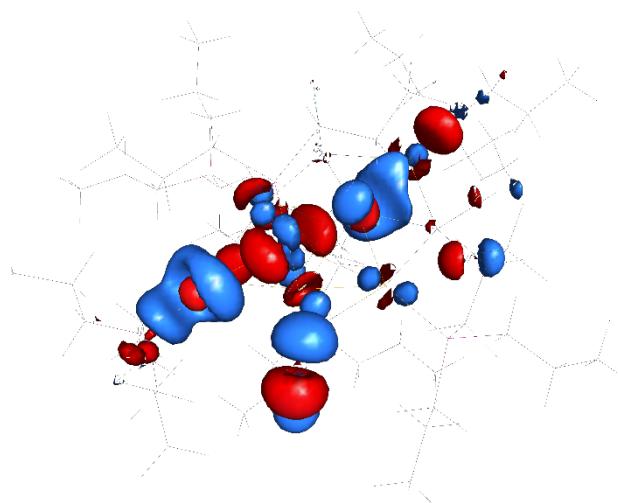
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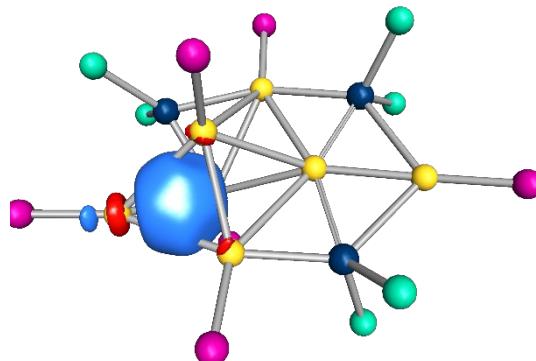


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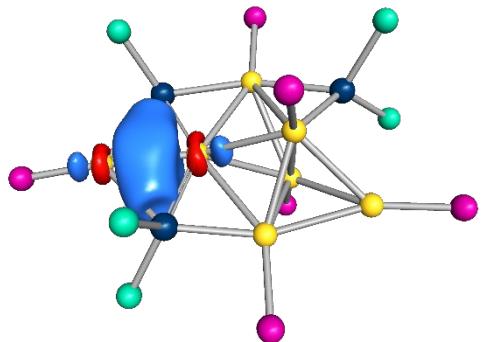
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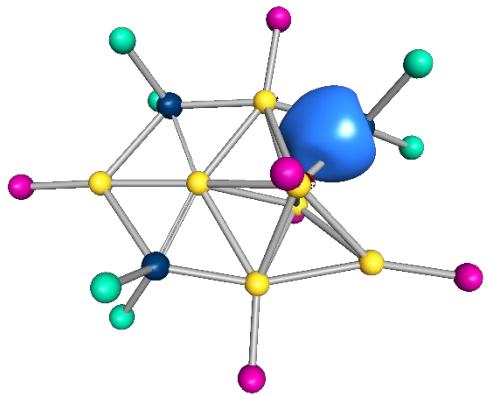
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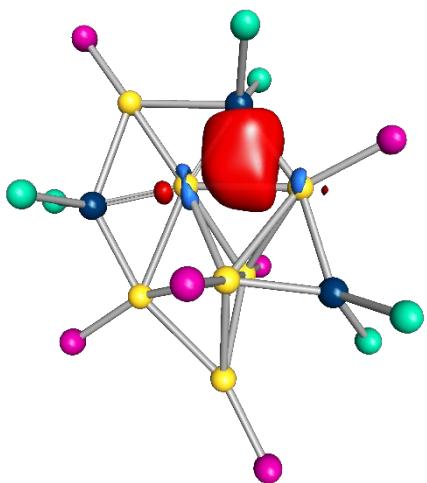
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LMO433



LMO431



LMO 430

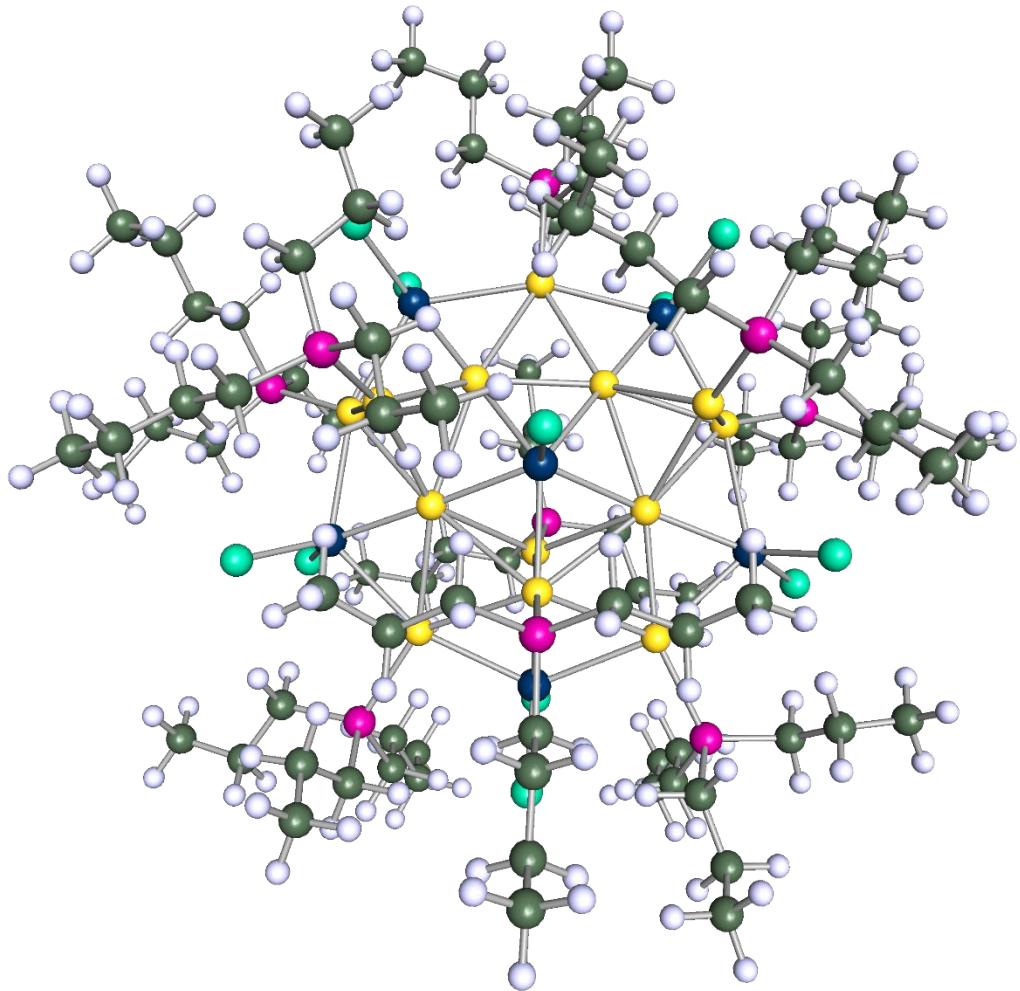


Figure S2: Geometry optimized structure of  $\mathbf{3}^3$

Point group:  $C_1$

Energy: -25001.85851252075 Hartree

HOMO-LUMO-gap: 1.536 eV

Atomic coordinates:

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Au 5.001678 21.926996 14.376869

P 1.726132 18.812065 14.137638

Au 2.716738 23.600045 15.352490

Au 0.885328 23.404498 12.987535

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Ga 1.617283 22.405683 17.393207

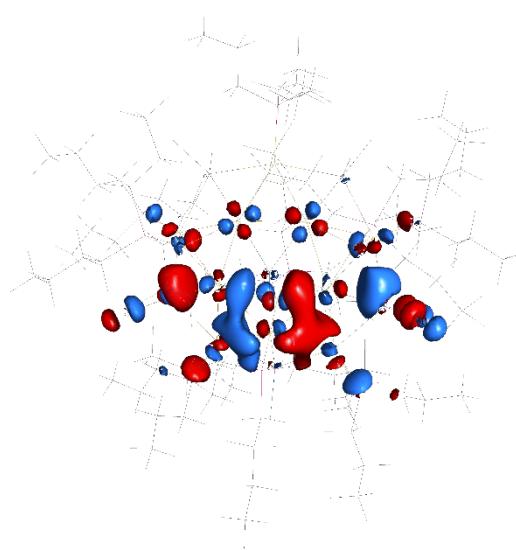
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P -2.112484 23.588186 17.109861	Ga 1.150164 21.728156 8.508584	C 1.702273 16.795081 10.374929
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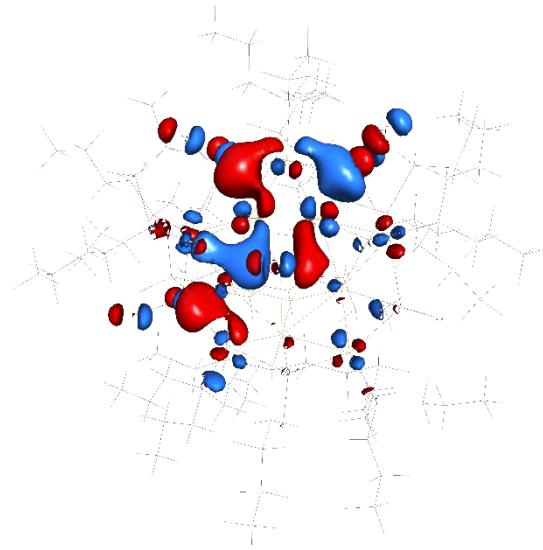
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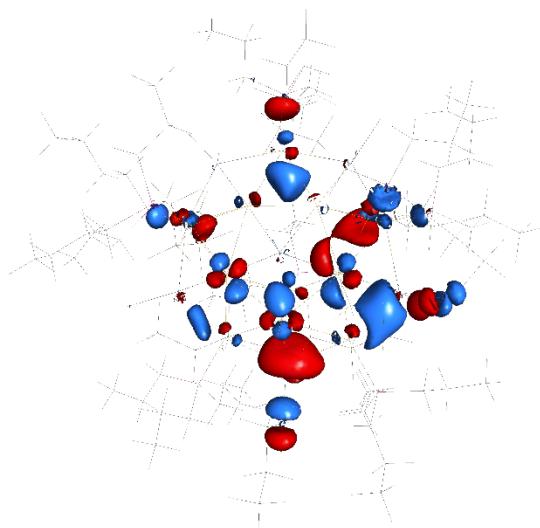
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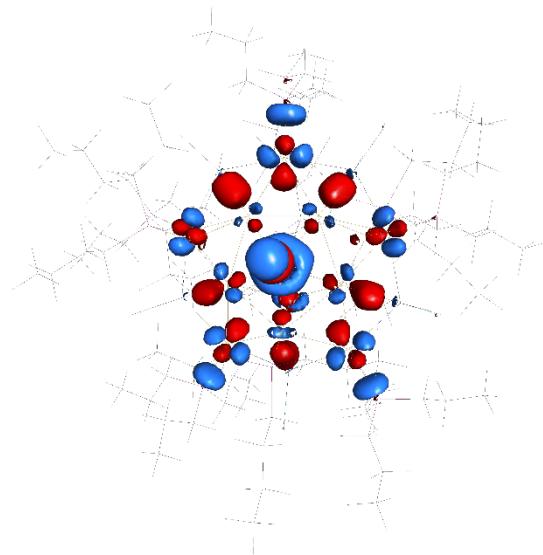
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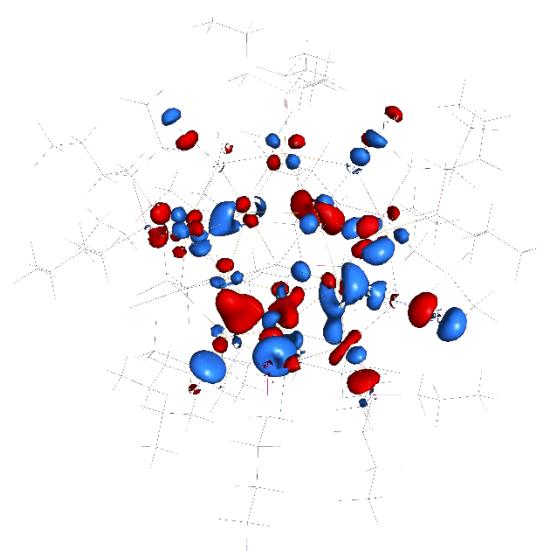
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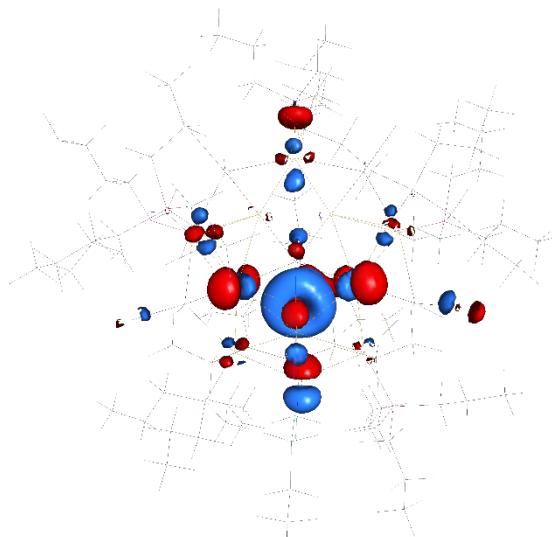
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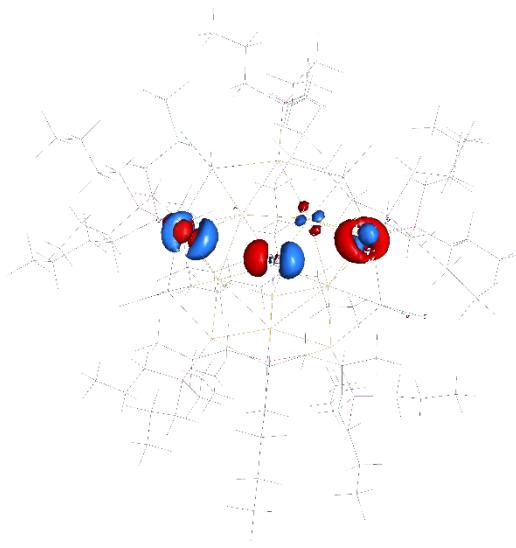
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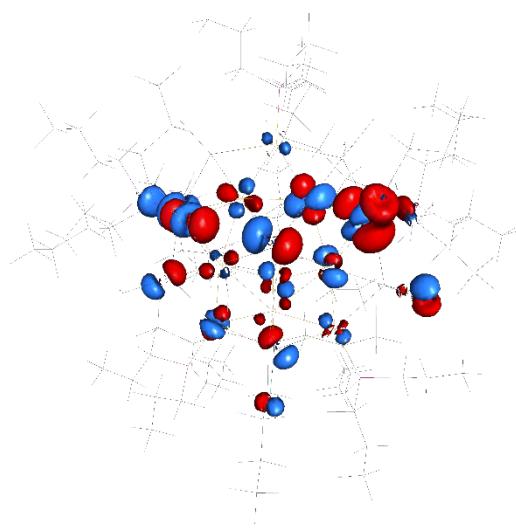
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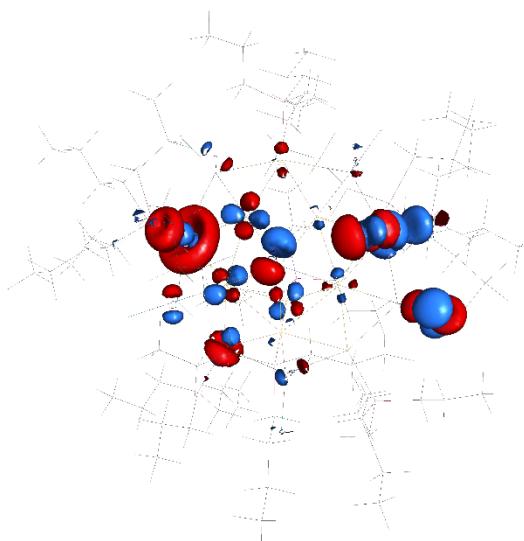
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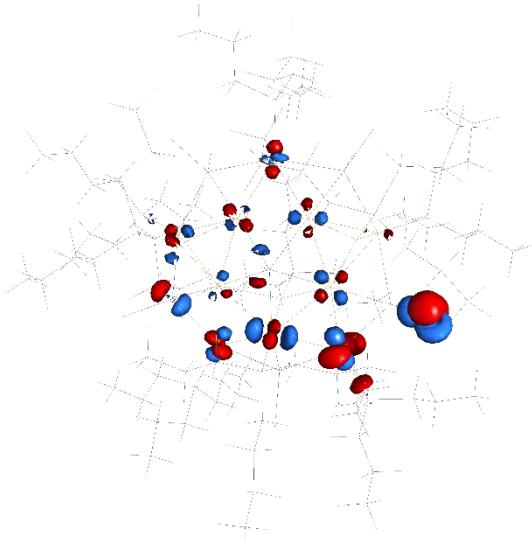
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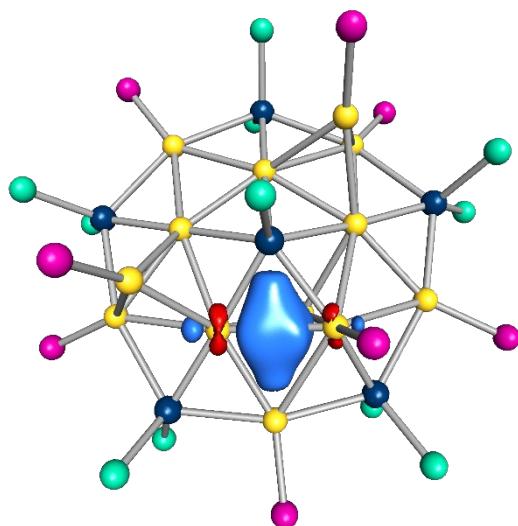
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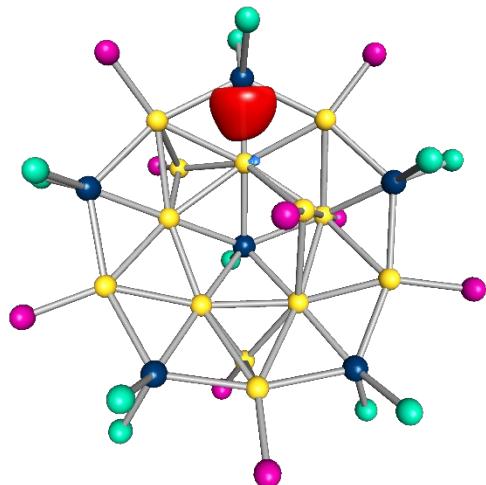
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LMO 748

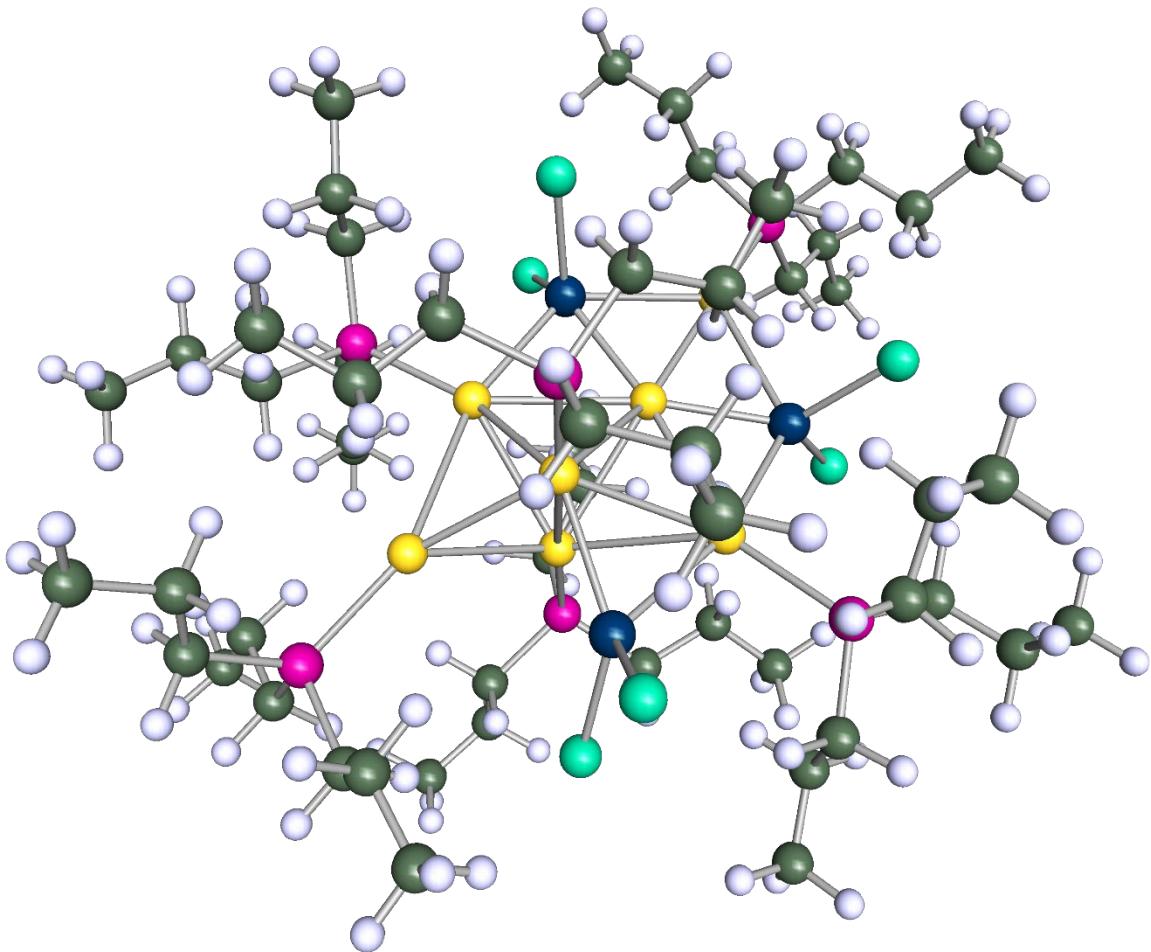


Figure S2: Geometry optimized structure of  $\mathbf{4}^3$

Point group: C<sub>1</sub>

Energy: -22261.99453588454 Hartree

HOMO-LUMO-gap: 1.158 eV

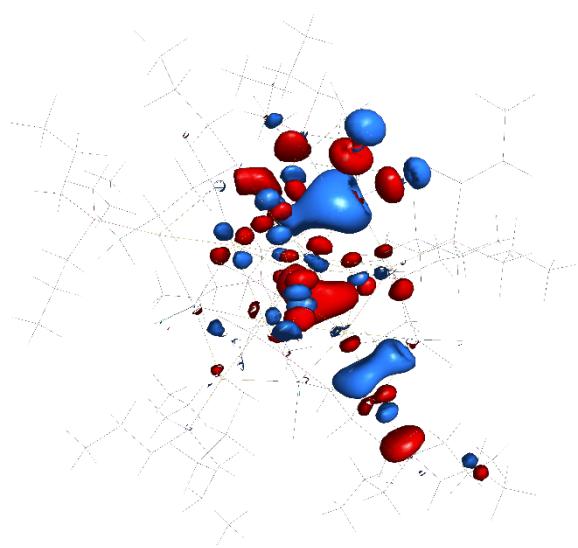
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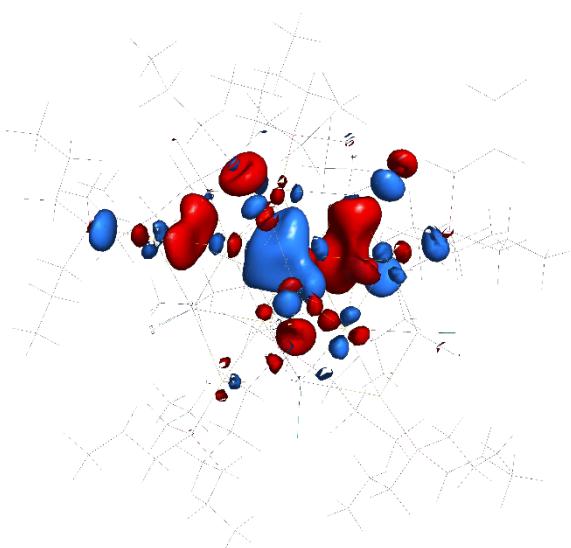
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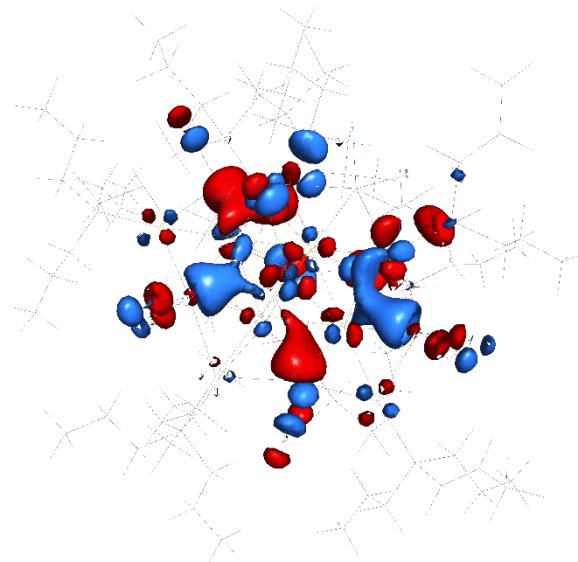
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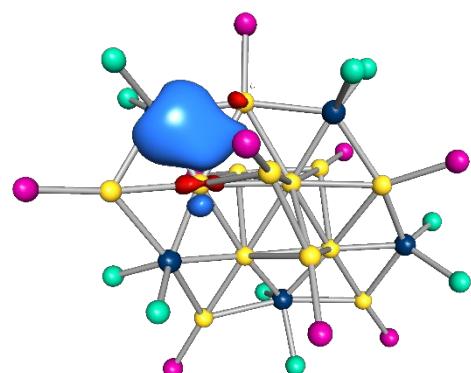
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HOMO-3



HOMO-4



LMO687

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- [2] O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. Puschmann, *J. Appl. Crystallogr.* 2009, 42, 339–341.
- [3] Turbomole: O. Treutler, R. Ahlrichs, *J. Chem. Phys.* 1995, 102, 346 – 354; BP86 functional: J. P. Perdew, *Phys. Rev. B* 1986, 33, 8822 – 88824; A. D. Becke, *Phys. Rev. A* 1988, 38, 3098 – 3100; RI-DFT: K. Eichkorn, O. Treutler, H. Öhm, M. Häser, R. Ahlrichs, *Chem. Phys. Lett.* 1995, 240, 283 – 290; SVP: A. Schäfer, H. Horn, R. Ahlrichs, *J. Chem. Phys.* 1992, 97, 2751 – 2577, TmoleX client: C. Steffen, K. Thomas, U. Huniar, A. Hellweg, O. Rubner, A. Schroer, *J. Comput. Chem.* 2010, 31, 2967 – 2970.