

Supplementary Information

One-step Construction of Y, C, and O Tridoped g-C₃N₄ as Bifunctional Photocatalyst for H₂ Evolution and Organic Pollutants Degradation under Visible Light Irradiation

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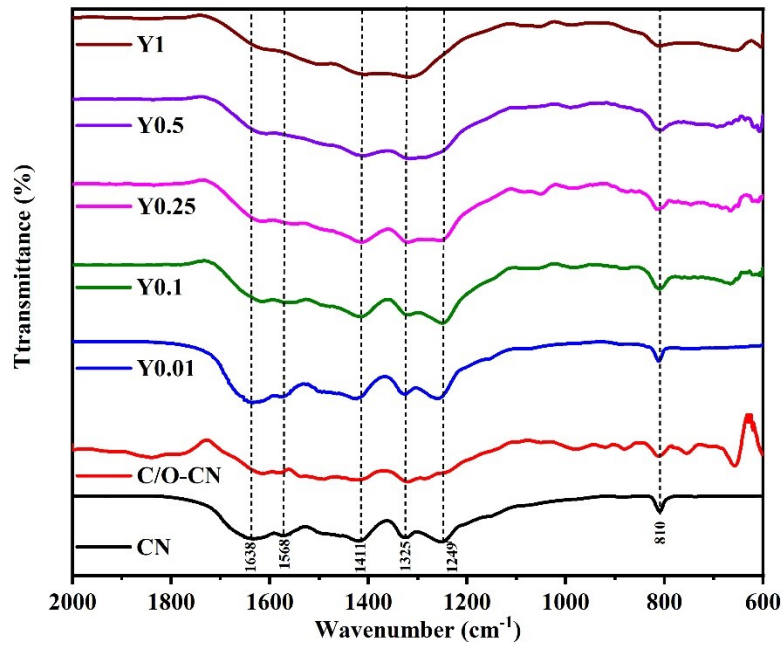


Fig. S1. FT-IR spectra in the range of 600 cm^{-1} to 2000 cm^{-1} of CN and doped CN samples.

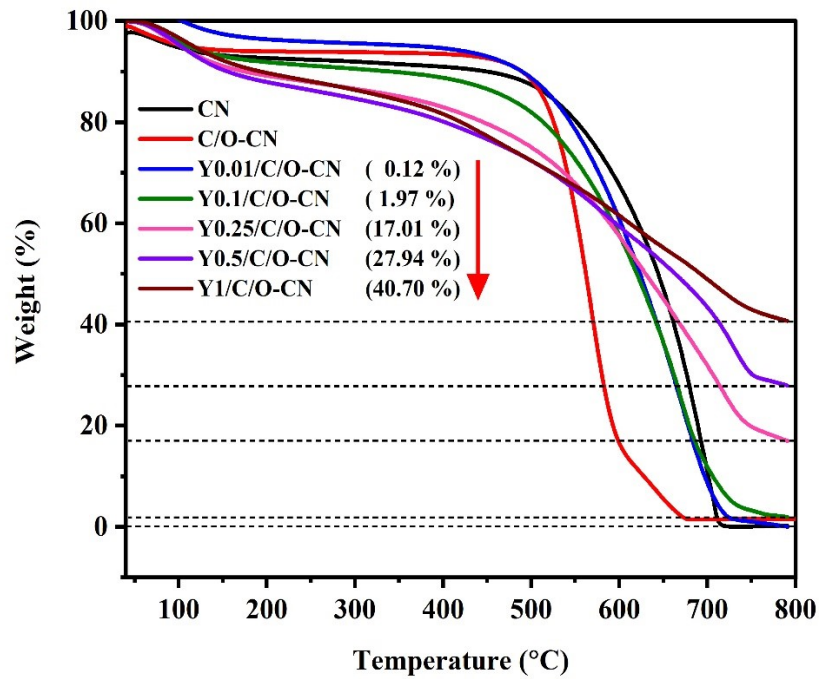
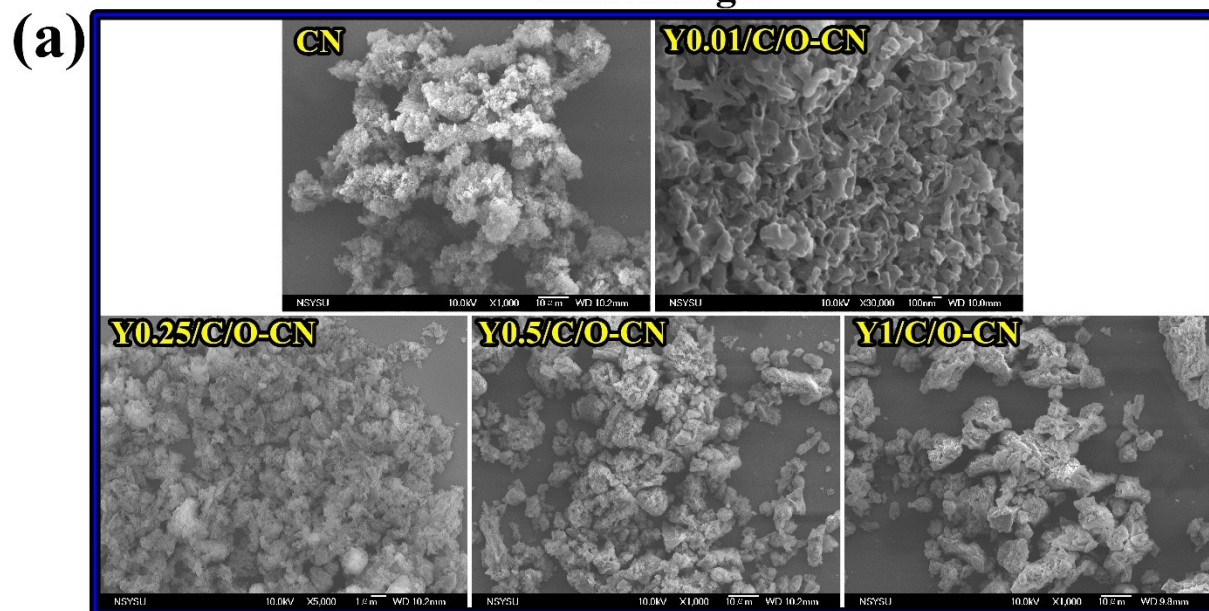


Fig. S2. TGA curves of bulk-CN, C/O-CN, and $\text{Y}_x/\text{C}/\text{O}$ tridoped CN samples.

SEM Images



TEM Images

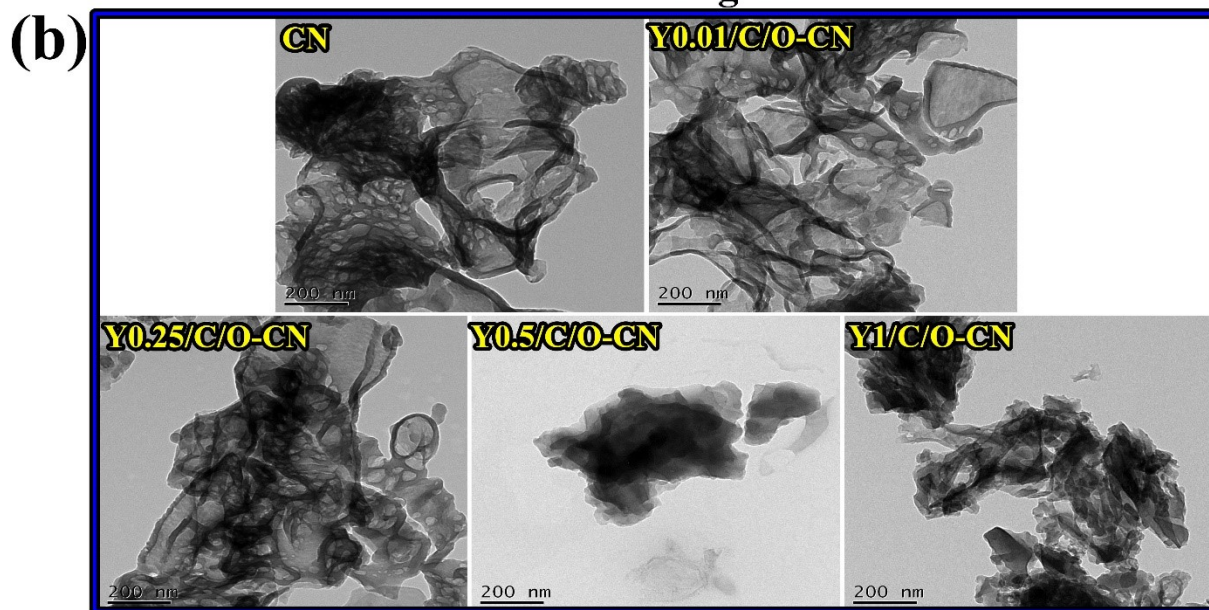


Fig. S3. (a) SEM images and (b) TEM images of bulk-CN and Y_x/C/O tridoped CN samples.

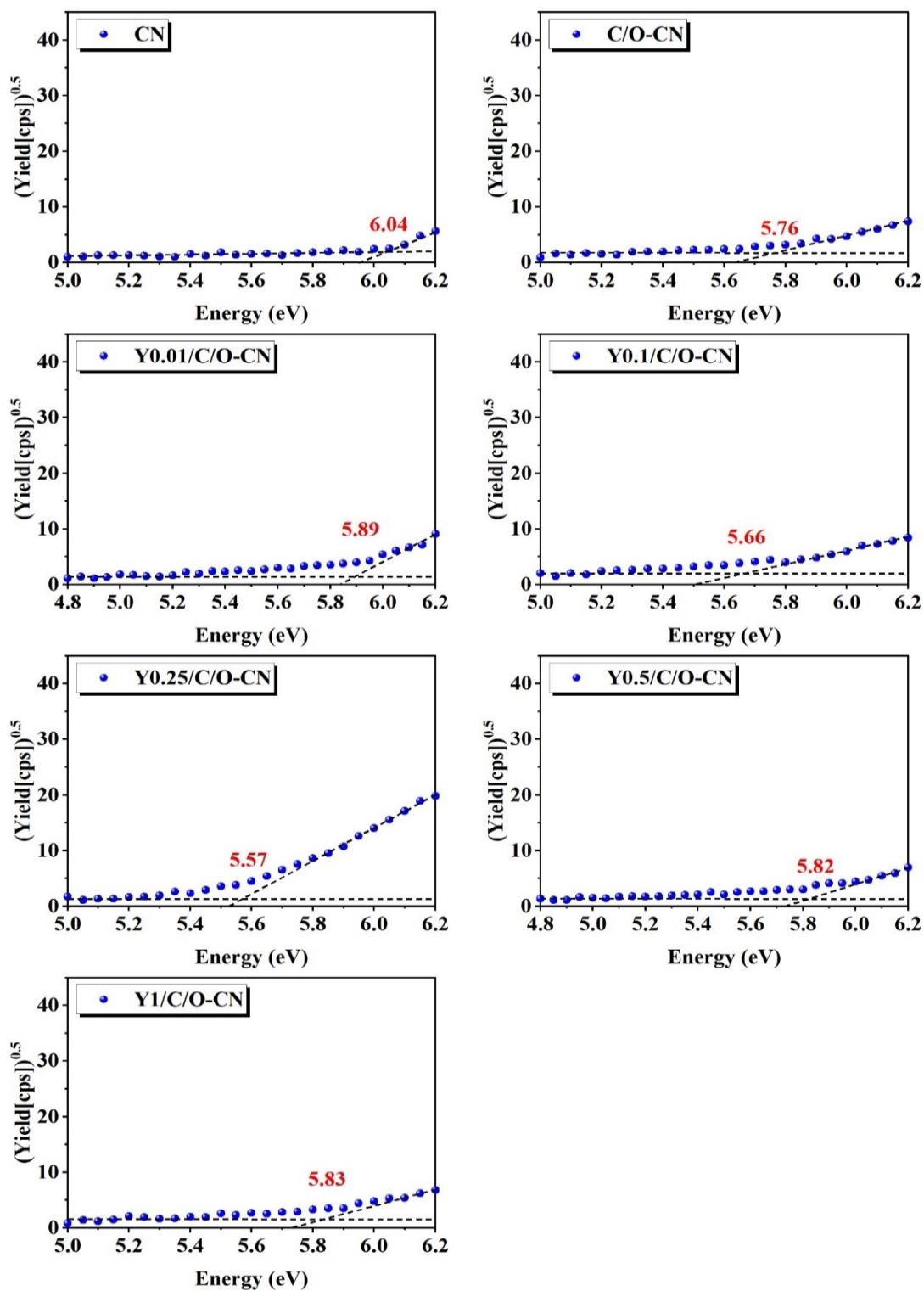


Fig. S4. Calculated valence band (VB) edges of CN and doped CN samples by photoelectron spectroscopy (model: AC-2).

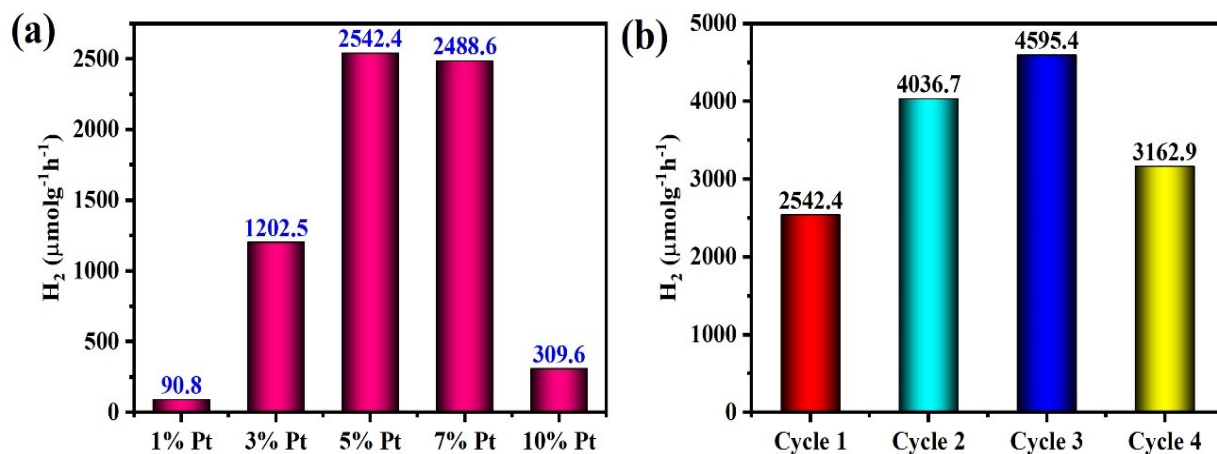


Fig. S5. (a) Effect of Pt co-catalyst concentration on photocatalytic H₂ evolution with Y0.1/C/O-CN and (b) stability test of Y0.1/C/O-CN for H₂ evolution.

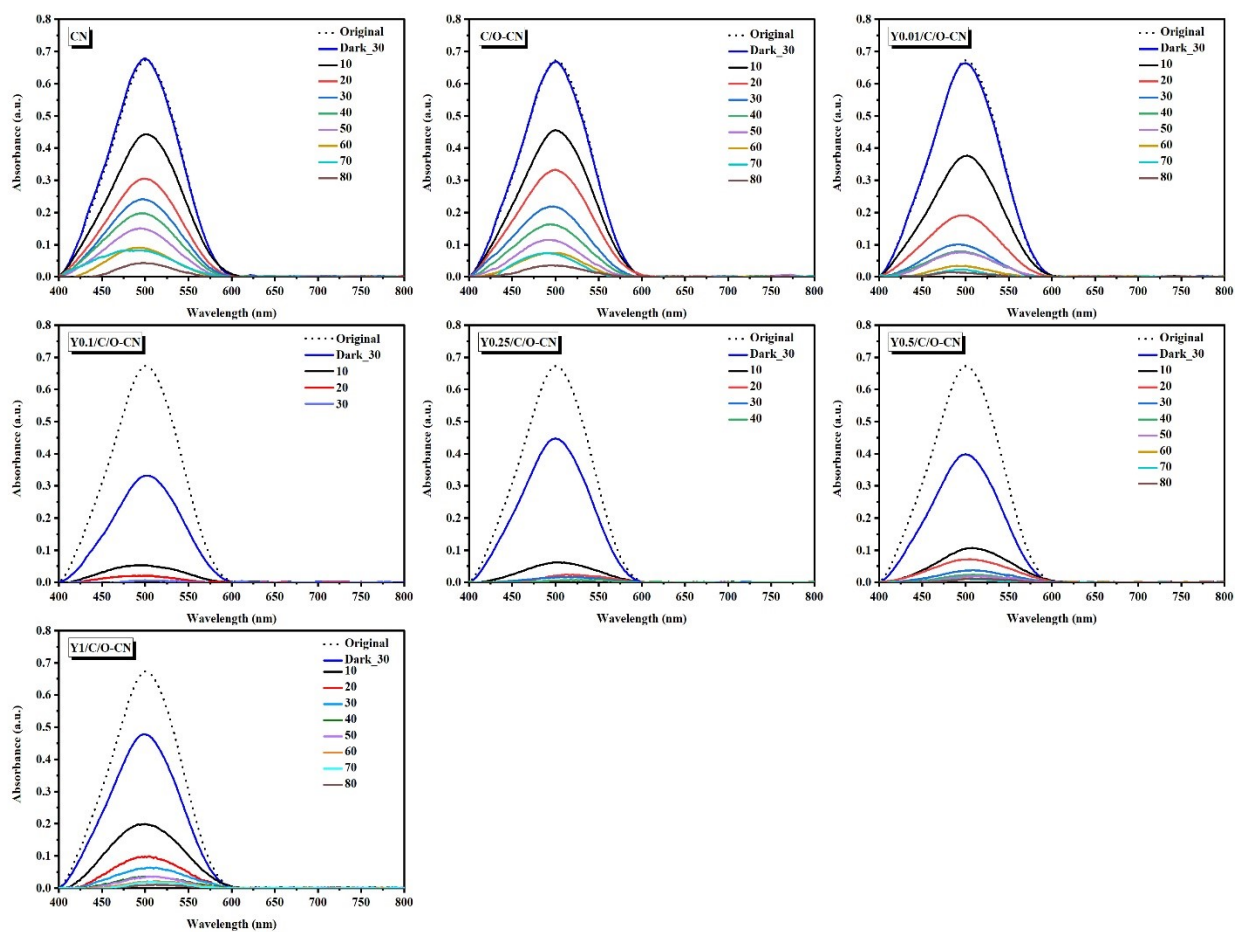


Fig. S6. The time-dependent absorption spectra of photocatalytic degradation of congo red (CR) dye under visible light irradiation with different photocatalysts.

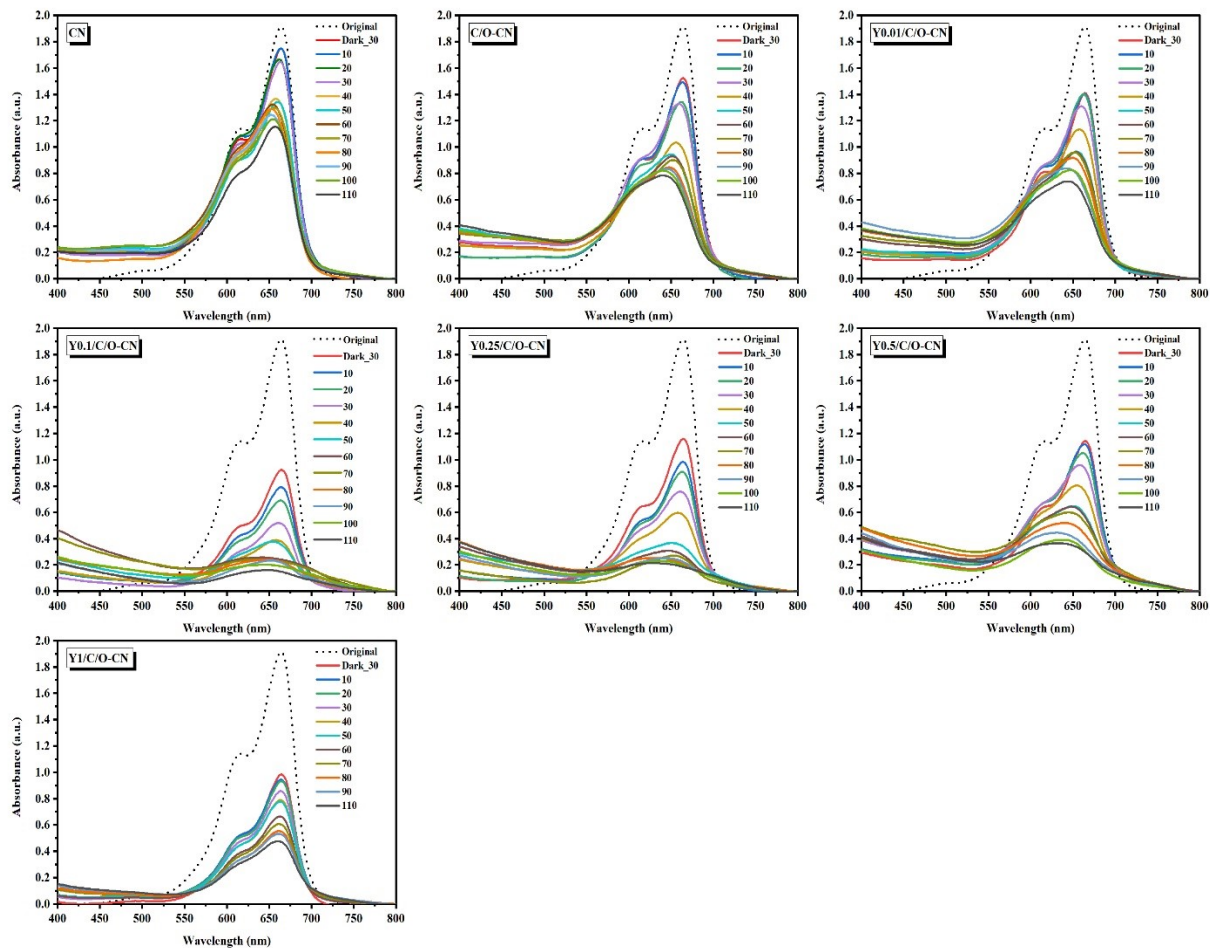


Fig. S7. The time-dependent absorption spectra of photocatalytic degradation of methylene blue (MB) dye under visible light irradiation with different photocatalysts.

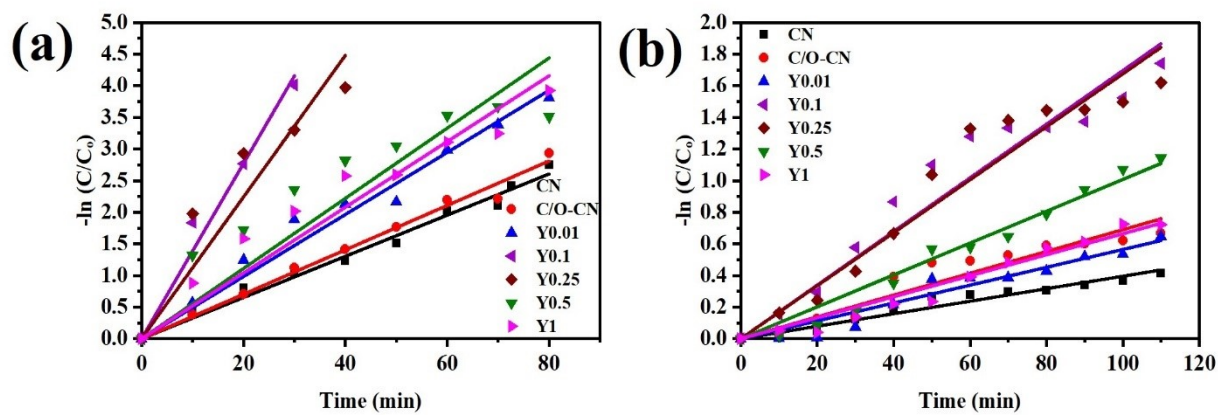


Fig. S8. $-\ln(C/C_0)$ versus reaction time plots for (a) CR and (b) MB degradation over bulk-CN, C/O-CN, and Yx/C/O tridoped CN, respectively.

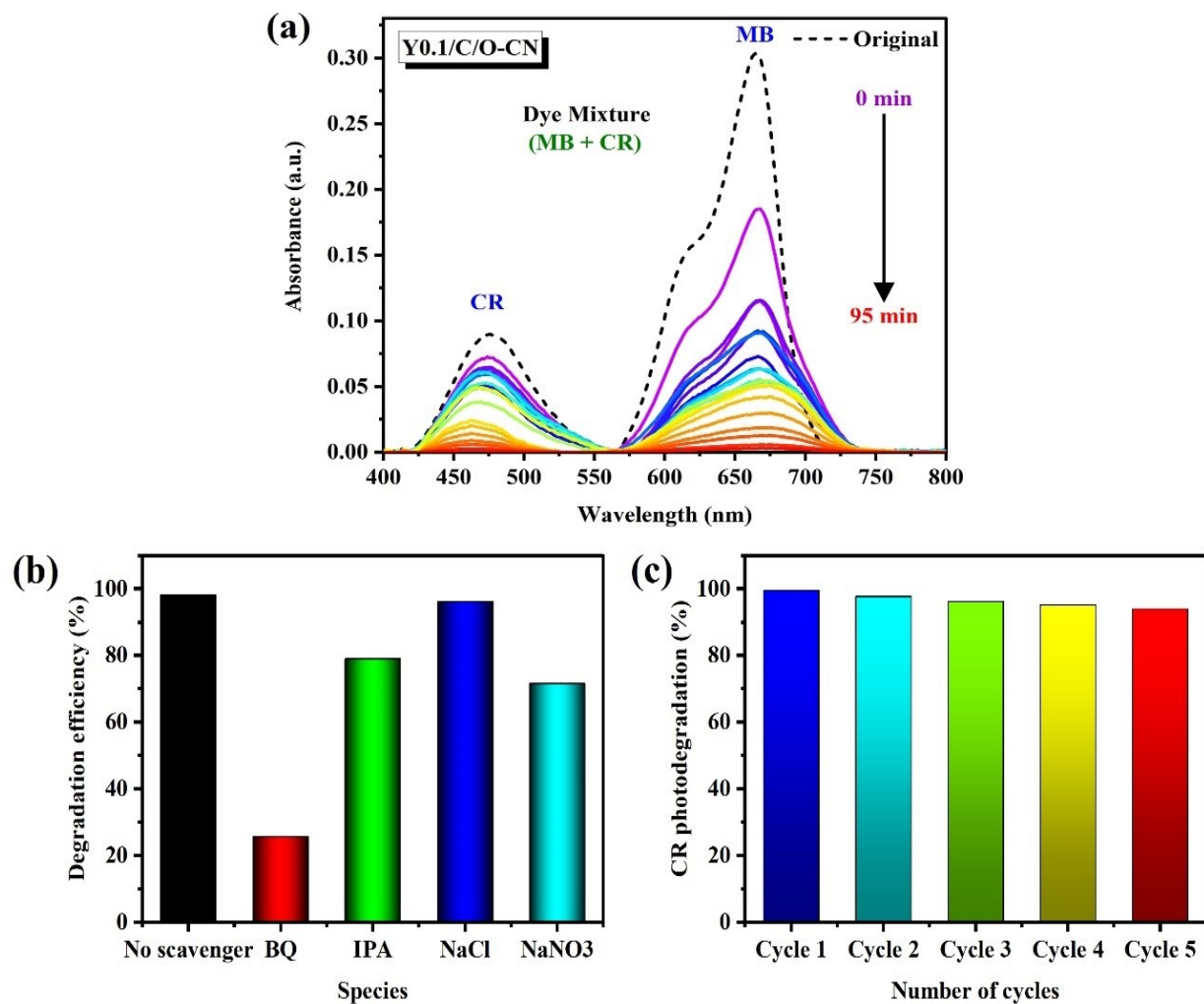


Fig. S9. (a) UV-visible absorbance of the model dye mixture (CR and MB) with the Y0.1/C/O-CN photocatalyst. (b) The effect of different scavengers on the CR photodegradation by Y0.1/C/O-CN, and (c) the reusability of Y0.1/C/O-CN photocatalyst for the degradation of CR.

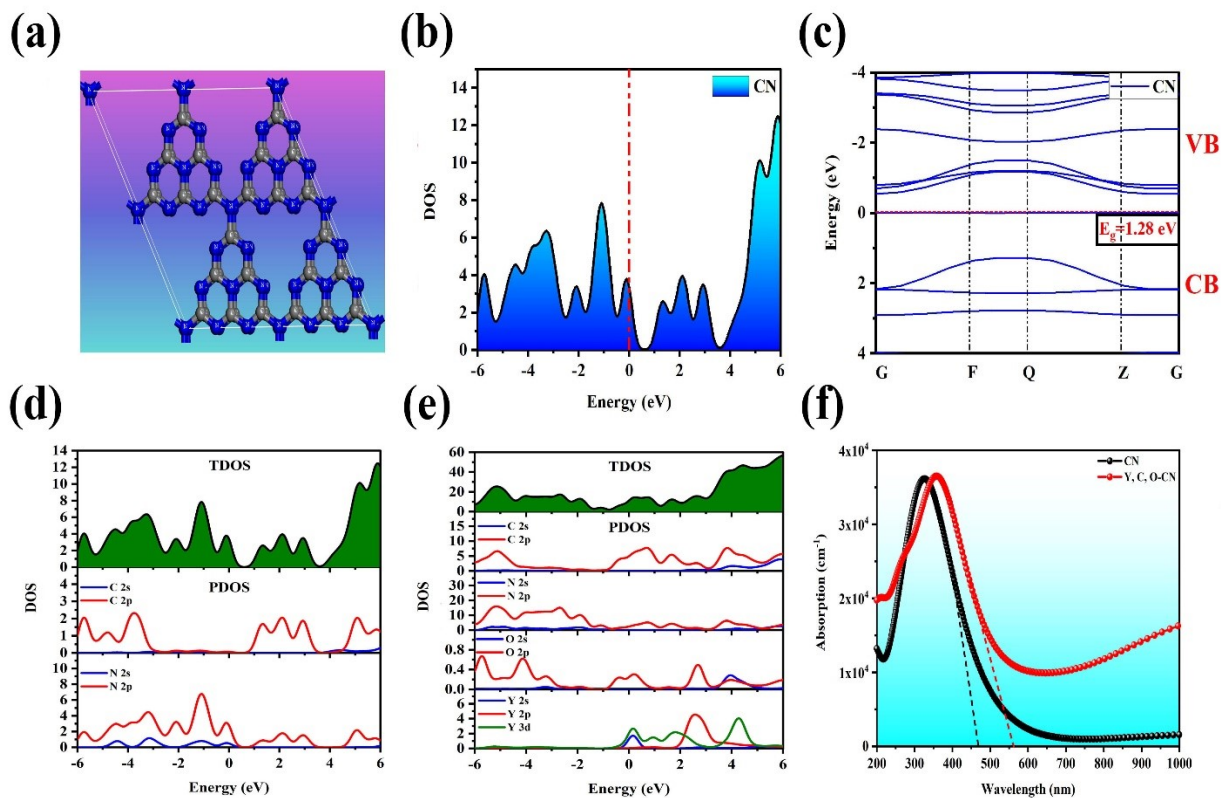


Fig. S10. (a) Optimized $2 \times 2 \times 1$ supercell structure used in the DFT calculations, (b) calculated DOS diagram, and (c) corresponding band structure of pure CN, respectively. (d,e) TDOS and PDOS of pure CN and Y, C, O tridoped CN, respectively. (f) Optical absorption curve of pure CN and Y, C, O tridoped CN.

Table S1: XRD data information of CN and doped CN samples.

Samples	Average crystallite size (nm)	Dislocation density ($\times 10^{-19} \text{ m}^{-2}$)	Lattice strain ($\times 10^{-2}$)
CN	2.569	1.529	1.353
C/O-CN	1.876	2.841	1.848
Y0.01/C/O-CN	2.053	2.372	1.688
Y0.1/C/O-CN	2.409	1.726	1.440
Y0.25/C/O-CN	2.059	2.402	1.694
Y0.5/C/O-CN	1.879	3.107	1.903
Y1/C/O-CN	1.557	4.375	2.271

Table S2: Surface elemental composition determined by XPS of pure-CN, C-O/CN, and Y_x/C/O-CN samples.

Sample	Atomic (%)				C/N ratio
	C	N	O	Y	
g-C₃N₄ (CN)	44.13	53.91	1.970	-	0.819
C/O-CN	44.73	53.02	2.252	-	0.844
Y0.01/C/O-CN	44.06	52.51	3.351	0.081	0.839
Y0.1/C/O-CN	43.10	51.23	5.030	0.640	0.841
Y0.25/C/O-CN	41.92	46.64	9.662	1.782	0.899
Y0.5/C/O-CN	41.48	42.83	12.66	3.020	0.968
Y1/C/O-CN	39.49	32.59	22.39	5.531	1.212

Table S3: XPS C 1s spectra results of CN, C/O-CN, and Y0.1/ C/O-CN.

Sample	(N=C-(N) ₂)	(C-C)	(C-O)	(N=C-(N) ₂) / (C-C) ratio
Pure-CN	88.4	10.2	1.4	8.65
C/O-CN	85.3	12.1	2.6	7.08
Y0.1/C/O-CN	85.4	12.7	1.9	6.71

Table S4: XPS N 1s spectra results of CN, C/O-CN, and Y0.1/ C/O-CN.

Sample	(C=N-C)	(N-(C) ₃)	(-NH _x)
Pure-CN	59.6	29.2	11.2
C/O-CN	52.4	36.4	11.2
Y0.1/C/O-CN	48.9	39.4	11.7

Table S5: XPS O 1s spectra results of CN, C/O-CN, and Y0.1/ C/O-CN.

Sample	(H ₂ O)	(O ₂)	(C-O)
Pure-CN	82.5	11.6	5.9
C/O-CN	58.4	25.1	16.5
Y0.1/C/O-CN	57.7	27.6	14.7

Table S6: BET surface area, pore size distribution, and bandgap energy (E_g) of pure-CN, C/O-CN, C/O-CN, and Y_x /C/O-CN samples.

Sample	BET surface area (m^2/g)	Pore volume (cm^3/g)	Bandgap (eV)
g-C₃N₄ (CN)	62.07	0.054	2.75
C/O-CN	36.91	0.034	1.86
Y.01/C/O-CN	43.47	0.051	1.88
Y0.1/C/O-CN	80.39	0.112	1.90
Y0.25/C/O-CN	73.90	0.125	2.11
Y0.5/C/O-CN	46.35	0.123	2.15
Y1/C/O-CN	28.76	0.119	2.27

Table S7: Comparison of photocatalytic activity of Y_x/C/O tridoped g-C₃N₄ (x= 0.01, 0.1, 0.25, 0.5, and 1 g) for H₂ evolution and the corresponding AQY with other g-C₃N₄ based photocatalysts.

Photocatalyst	Experimental details	Light source	HER ($\mu\text{mol h}^{-1} \text{g}^{-1}$)	AQY (%) (420nm)	Ref.
g-C₃N₄-Based Heterojunction Photocatalysts					
N-doped MoS ₂ / S-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	659	-	1
CdS QDs/ P-doped g-C ₃ N ₄	Water/ Na ₂ S/ Na ₂ SO ₃	300 W Xe lamp (>420 nm)	1579	-	2
B-doped g-C ₃ N ₄ / ZnO	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	357	3.29 (240 nm)	3
Monodoped g-C₃N₄ Photocatalysts					
S-doped g-C ₃ N ₄ nanosheets	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	982	-	4
Br-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	960	-	5
Cl-doped g-C ₃ N ₄	Water/ lactic acid/ Pt	300 W Xe lamp (>420 nm)	537	13.7	6
P-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	1596	3.6	7
C-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>400 nm)	178	-	8
F-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	745	-	9
O-doped porous network g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	1204	7.8	10
O-doped porous g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	1430	-	11
O-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	772	8.6	12

K-doped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	1028	-	13
Codoped g-C₃N₄ Photocatalysts					
Mo-S codoped porous g-C ₃ N ₄ nanosheets	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	294	0.24	14
P-Na codoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>400 nm)	2032	6.8	15
Fe-P codoped g-C ₃ N ₄	Water/ methanol / Pt	250 W Na lamp (>400 nm)	122	6.9	16
C-O codoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	1101	-	This work
Tridoped g-C₃N₄ Photocatalysts					
S-P-O tridoped g-C ₃ N ₄ nanosheet	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	2480	-	17
Y0.01-C-O tridoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	2212	-	This work
Y0.1-C-O tridoped g-C₃N₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	2543	6	This work
Y0.25-C-O tridoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	958	-	This work
Y0.5-C-O tridoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	335	-	This work
Y1-C-O tridoped g-C ₃ N ₄	Water/ TEOA/ Pt	300 W Xe lamp (>420 nm)	59	-	This work

Table S8: Comparison of photocatalytic activity of Yx/C/O tridoped g-C₃N₄ (x= 0.01, 0.1, 0.25, 0.5, and 1 g) for pollutant degradation with other g-C₃N₄ based photocatalyst.

Photocatalyst	Catalyst dose (g/L)	Dye Used	Dye Conc. (ppm)	Light source	Degradation (%)	Time (min)	Ref.
CR Photodegradation							
S-doped g-C ₃ N ₄ /TiO ₂	0.2	CR	50	300 W Xe lamp	95	60	18
g-C ₃ N ₄ /RGO/Bi ₂ Fe ₄ O ₉	0.5	CR	10	30 W LED lamp (>420 nm)	87.7	60	19
CoFe ₂ O ₄ /g-C ₃ N ₄	0.5	CR	25	150 W Xe lamp simulated solar light	~100%	90	20
<i>(In the presence of 30%H₂O₂)</i> Exfoliated g-C ₃ N ₄	1	CR	100	Solar irradiation	95.1	60	21
g-C ₃ N ₄ /precipitated silica	2	CR	60	300 W Xe lamp (>400 nm)	88	360	22
Acid-etched template free g-C ₃ N ₄	0.05	CR	10	Visible light	90.08	150	23
C-O codoped g-C ₃ N ₄	0.2	CR	50	18 W LED lamp (>420 nm)	94.7	80	This work
Y0.01-C-O tridoped g-C ₃ N ₄	0.2	CR	50	18 W LED lamp (>420 nm)	97.8	80	This work
Y0.1-C-O tridoped g-C₃N₄	0.2	CR	50	18 W LED lamp (>420 nm)	98.5	30	This work
Y0.25-C-O tridoped g-C ₃ N ₄	0.2	CR	50	18 W LED lamp (>420 nm)	98.1	40	This work
Y0.5-C-O tridoped g-C ₃ N ₄	0.2	CR	50	18 W LED lamp (>420 nm)	97.0	80	This work
Y1-C-O tridoped g-C ₃ N ₄	0.2	CR	50	18 W LED lamp (>420 nm)	98.0	80	This work
MB Photodegradation							
Li-doped g-C ₃ N ₄	0.2	MB	30	300 W Xe lamp (>470 nm)	90	180	24
Ca-doped NaNbO ₃ /g-C ₃ N ₄	0.4	MB	1	85 W CFL lamp	100	240	25
ZnO/Au/g-C ₃ N ₄	0.5	MB	10	300 W Xe lamp	99	120	26
g-C ₃ N ₄ /H-ZSM-5	0.1	MB	5	None	92	120	27

g-C ₃ N ₄ /RGO/Bi ₂ Fe ₄ O ₉	0.5	MB	10	30 W LED (>420 nm)	87.7	60	19
UiO-66/g-C ₃ N ₄	0.25	MB	10	350 W Xe lamp (>420 nm)	~100	240	28
Sm ₂ O ₃ /S-doped g-C ₃ N ₄	0.7	MB	8	300 W Halogen lamp (>400 nm)	93	150	29
Porous CeO ₂ /S-doped g-C ₃ N ₄	-	MB	7	300 W Halogen lamp (>400 nm)	91	150	30
ZnS/g-C ₃ N ₄	0.15	MB	6	100 W Halogen lamp	90	100	31
g-C ₃ N ₄ /Cu ₂ O	0.5	MB	3.1	400 W Na lamp	81	120	32
C-O codoped g-C ₃ N ₄	0.2	MB	10	18 W LED lamp (>420 nm)	48.6	110	This work
Y0.01-C-O tridoped g-C ₃ N ₄	0.2	MB	10	18 W LED lamp (>420 nm)	47.6	110	This work
Y0.1-C-O tridoped g-C₃N₄	0.2	MB	10	18 W LED lamp (>420 nm)	82.4	110	This work
Y0.25-C-O tridoped g-C ₃ N ₄	0.2	MB	10	18 W LED lamp (>420 nm)	80.2	110	This work
Y0.5-C-O tridoped g-C ₃ N ₄	0.2	MB	10	18 W LED lamp (>420 nm)	68.1	110	This work
Y1-C-O tridoped g-C ₃ N ₄	0.2	MB	10	18 W LED lamp (>420 nm)	51.5	110	This work
Mixed Dyes Photodegradation							
Y0.1-C-O tridoped g-C₃N₄	0.2	MB, CR	10, 25	18 W LED lamp (>420 nm)	~100	95	This work
CoFe ₂ O ₄ /g-C ₃ N ₄	0.5	MB, CR, MO	25, 25, 10	150 W Xe lamp simulated solar light	~100	180	20
<i>(In the presence of 30%H₂O₂)</i>							

Reference

1. Y. Chen, F. Su, H. Xie, R. Wang, C. Ding, J. Huang, Y. Xu and L. Ye, *Chemical Engineering Journal*, 2021, **404**, 126498.
2. Q. Liang, C. Zhang, S. Xu, M. Zhou, Y. Zhou and Z. Li, *Journal of Colloid and Interface Science*, 2020, **577**, 1-11.
3. D. Kim and K. Yong, *Applied Catalysis B: Environmental*, 2021, **282**, 119538.
4. C. Feng, L. Tang, Y. Deng, J. Wang, Y. Liu, X. Ouyang, H. Yang, J. Yu and J. Wang, *Applied Catalysis B: Environmental*, 2021, **281**, 119539.
5. Z.-A. Lan, G. Zhang and X. Wang, *Applied Catalysis B: Environmental*, 2016, **192**, 116-125.
6. C. Liu, Y. Zhang, F. Dong, A. Reshak, L. Ye, N. Pinna, C. Zeng, T. Zhang and H. Huang, *Applied Catalysis B: Environmental*, 2017, **203**, 465-474.
7. J. Ran, T. Y. Ma, G. Gao, X.-W. Du and S. Z. Qiao, *Energy & Environmental Science*, 2015, **8**, 3708-3717.
8. J. Cao, H. Fan, C. Wang, J. Ma, G. Dong and M. Zhang, *Ceramics International*, 2020, **46**, 7888-7895.
9. Y. Luo, J. Wang, S. Yu, Y. Cao, K. Ma, Y. Pu, W. Zou, C. Tang, F. Gao and L. Dong, *Journal of Materials Research*, 2018, **33**, 1268-1278.
10. Z.-F. Huang, J. Song, L. Pan, Z. Wang, X. Zhang, J.-J. Zou, W. Mi, X. Zhang and L. Wang, *Nano Energy*, 2015, **12**, 646-656.
11. L. Yang, J. Huang, L. Shi, L. Cao, Q. Yu, Y. Jie, J. Fei, H. Ouyang and J. Ye, *Applied Catalysis B: Environmental*, 2017, **204**, 335-345.
12. Y. Wang, F. Silveri, M. K. Bayazit, Q. Ruan, Y. Li, J. Xie, C. R. A. Catlow and J. Tang, *Advanced Energy Materials*, 2018, **8**, 1801084.
13. M. Wu, J. M. Yan, X. n. Tang, M. Zhao and Q. Jiang, *ChemSusChem*, 2014, **7**, 2654-2658.
14. Y. Li, S. Zhu, Y. Liang, Z. Li, S. Wu, C. Chang, S. Luo and Z. Cui, *Applied Surface Science*, 2021, **536**, 147743.
15. C. Wang, W. Wang, H. Fan, N. Zhao, J. Ma, M. Zhang and A. K. Yadav, *ACS applied materials & interfaces*, 2019, **12**, 5234-5243.

16. S. Hu, L. Ma, J. You, F. Li, Z. Fan, G. Lu, D. Liu and J. Gui, *Applied surface science*, 2014, **311**, 164-171.
17. Q. Liu, J. Shen, X. Yu, X. Yang, W. Liu, J. Yang, H. Tang, H. Xu, H. Li and Y. Li, *Applied Catalysis B: Environmental*, 2019, **248**, 84-94.
18. J. Wang, G. Wang, B. Cheng, J. Yu and J. Fan, *Chinese Journal of Catalysis*, 2021, **42**, 56-68.
19. M. B. Shekardasht, M. H. Givianrad, P. Gharbani, Z. Mirjafary and A. Mehrizad, *Diamond and Related Materials*, 2020, **109**, 108008.
20. D. Gogoi, P. Makkar and N. N. Ghosh, *ACS omega*, 2021, **6**, 4831-4841.
21. S. P. Pattnaik, A. Behera, R. Acharya and K. Parida, *Journal of Environmental Chemical Engineering*, 2019, **7**, 103456.
22. C. Li, Z. Sun, X. Li, L. Liu and S. Zheng, *Advanced Powder Technology*, 2016, **27**, 2051-2060.
23. A. Kumar, S. Singh and M. Khanuja, *Materials Chemistry and Physics*, 2020, **243**, 122402.
24. M. Chegeni, F. Goudarzi and M. Soleymani, *ChemistrySelect*, 2019, **4**, 13736-13745.
25. A. S. Vig, N. Rani, A. Gupta and O. Pandey, *Solar Energy*, 2019, **185**, 469-479.
26. S. J. Lee, T. Begildayeva, H. J. Jung, R. Koutavarapu, Y. Yu, M. Choi and M. Y. Choi, *Chemosphere*, 2021, **263**, 128262.
27. S. Barman and S. Basu, *Chemosphere*, 2020, **241**, 124981.
28. Y. Zhang, J. Zhou, Q. Feng, X. Chen and Z. Hu, *Chemosphere*, 2018, **212**, 523-532.
29. M. Jourshabani, Z. Shariatinia and A. Badiei, *Applied Surface Science*, 2018, **427**, 375-387.
30. M. Jourshabani, Z. Shariatinia and A. Badiei, *Journal of colloid and interface science*, 2017, **507**, 59-73.
31. W. J. Kim, E. Jang and T. J. Park, *Applied Surface Science*, 2017, **419**, 159-164.
32. G. R. Surikanti, P. Bajaj and M. V. Sunkara, *ACS omega*, 2019, **4**, 17301-17316.