

Supporting Information

COF in situ growing on CdS nanorods as core-shell heterojunction for improving the charge-separation efficiency

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Part S1

Characterization

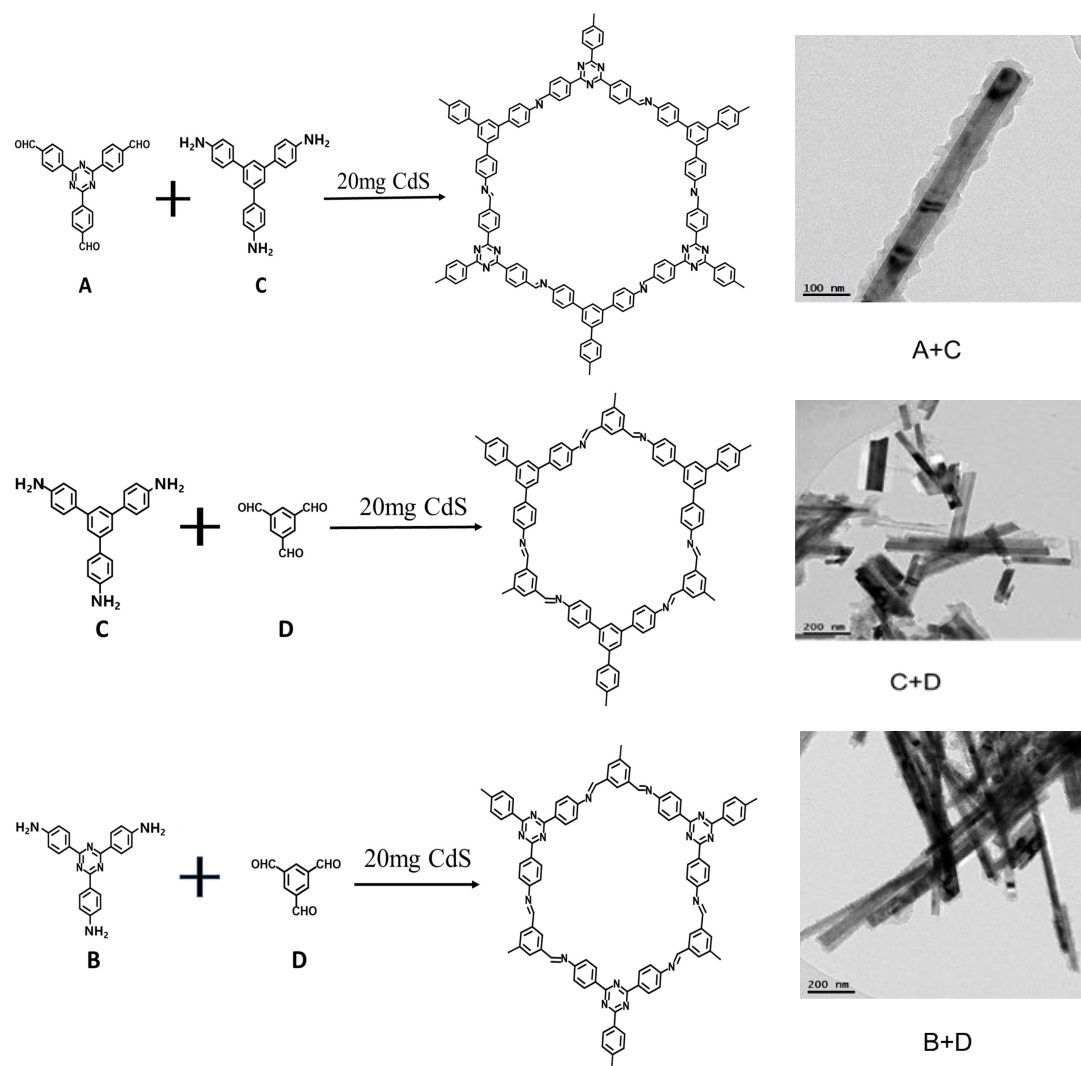


Fig. S1 Synthetic process and TEM images of CdS@COF_{A+C}, CdS@COF_{C+D} and CdS@COF_{B+D}, respectively.

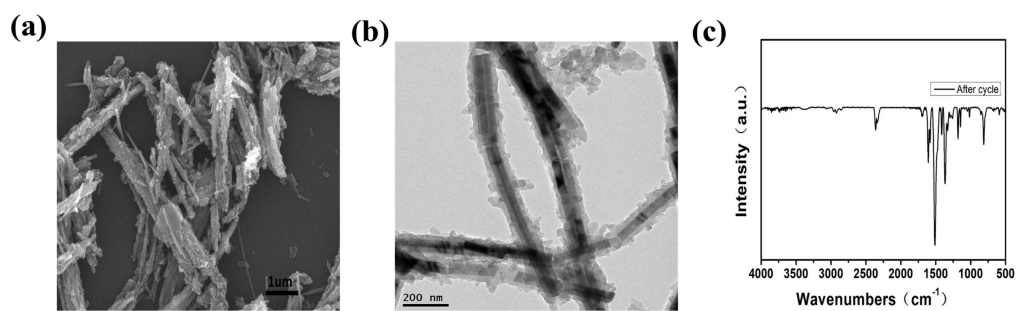


Fig. S2 SEM, TEM and FT-IR after hydrogen evolution cycling reaction of CdS@TTI-COF-60

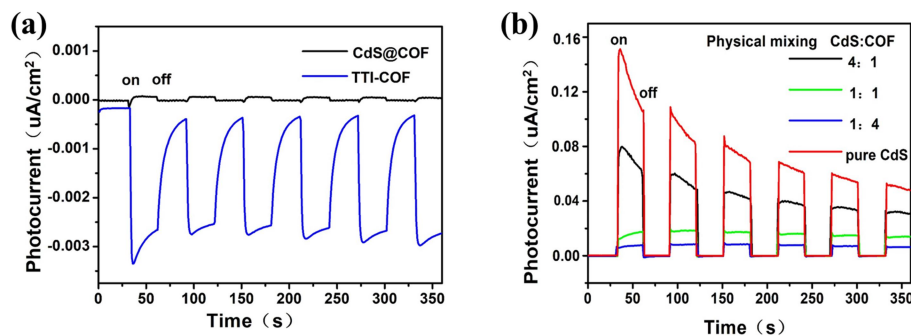


Fig. S3 Photocurrent response of (a)TTI-COF and CdS@TTI-COF-60 (b)pure CdS and physical mixing of CdS and TTI-COF

Part S2

Calculation of apparent quantum yield

The apparent quantum yield (AQY) of CdS@TTI-COF-60 was calculated based on the H₂ production under simulated solar irradiation by a Xe lamp equipped with a 420nm monochromatic optical filter, the ILT 950 spectroradiometer was used to measure the intensity of incident monochromatic illumination . The AQY is calculated by using the following equation

$$AQY = \frac{2MN_Ahc}{AIt\lambda} \times 100\%$$

Where M is molar amounts of H₂ during irradiation 1 h ($M=4.25 \times 10^{-6}$ mol/h), N_A is Avogadro constant (6.022×10^{23} mol⁻¹). h is Planck constant (6.626×10^{-34} J·s). c is the light velocity (3×10^8 m/s). A is the irradiation area (1.6×10^{-3} m²). I is the intensity of irradiation light (281.46 W/m²). t is the reaction time (3600 s). λ is the wavelength of the monochromatic light ($\lambda= 420$ nm). Finally, AQY=0.15%

Part S3

Carrier density calculation

The carrier density (N_D) is calculated using this formula: ¹⁻³

$$N_D = (2/e\epsilon\epsilon_0)[d(U_{FL})/d(1/C^2)]$$

where $e = 1.6 \times 10^{-19}$ C, $\epsilon_0 = 8.86 \times 10^{-12}$ F m⁻¹, ϵ is the dielectric constant, and C is the

capacitance. The dielectric constants of TTI-COF and CdS are about 3.67 and 6.56, respectively. The slopes of TTI-COF and CdS are 6.4×10^{10} and 3×10^{10} . Therefore, the N_D values of TTI-COF and CdS are about $5.99 \times 10^{18} \text{ cm}^{-3}$ and $7.16 \times 10^{18} \text{ cm}^{-3}$, respectively.

References

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- 3 G. Wang, Q. Wang, W. Lu, J.H. Li, *J. Phys. Chem. B*, 2006, **110**, 22029-22034.