Electronic Supplementary Material (ESI) for Sustainable Energy & Fuels. This journal is © The Royal Society of Chemistry 2022

Electronic Supplementary Information

Performance improvement strategies of cobalt based electrode for the

electrooxidation of sodium borohydride

Ya Li, Tong Sun, Youzhi Liu, Weizhou Jiao, Jing Gao, Dongming Zhang*

Shanxi Province Key Laboratory of Higee-Oriented Chemical Engineering,

North University of China, Taiyuan, 030051, P.R.China.

E-mails: <u>zhangdongming06@outlook.com</u>; Fax: +86 351 3921497; Tel: +86 351

3921986

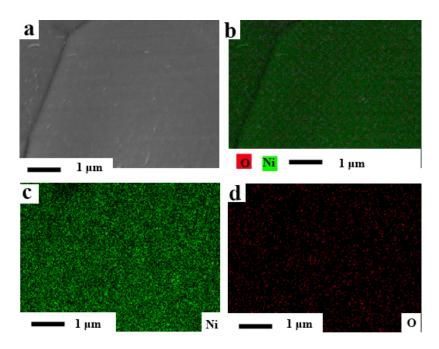


Fig. S1 SEM image (a) of the NF electrode and corresponding EDS images (b-d).

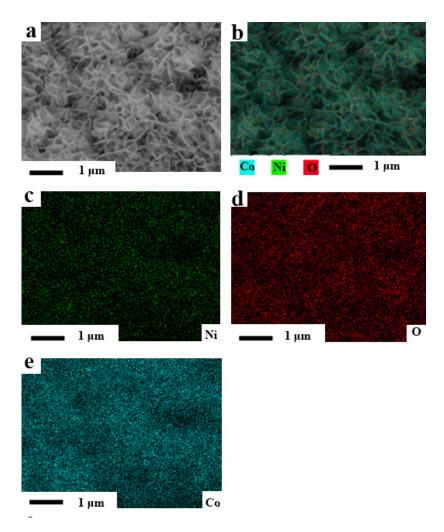


Fig. S2 SEM image (a) of the Co/NF electrode and corresponding EDS images

(b-e).

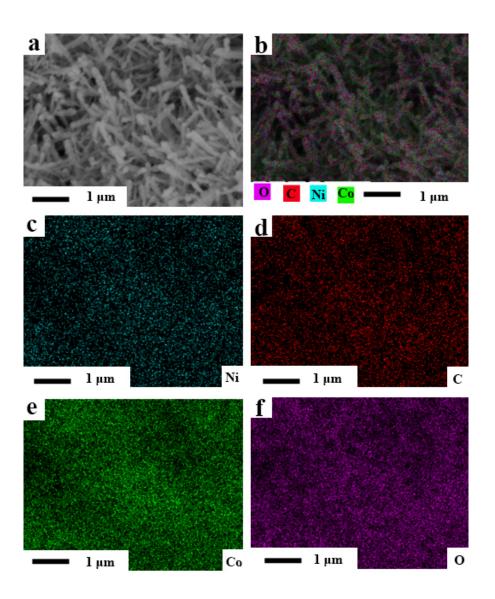


Fig. S3 SEM image of the $Co/NF-CoC_2O_4$ electrode (a) and corresponding EDS

images (b-f).

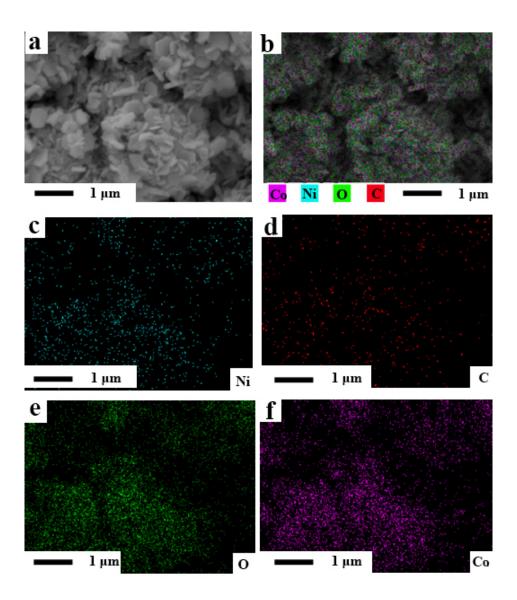


Fig. S4 SEM image of Co/NF-CoC₂O₄-Co electrode after the CV test in 1.5 mol

dm⁻³ NaOH (a) and corresponding EDS images (b-f).

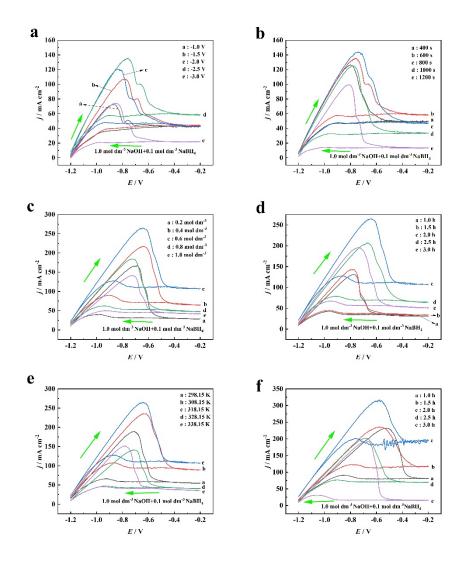


Fig. S5 CV curves of Co/NF electrode prepared under different deposition conditions: deposition potential (-1.0~-3.0 V) (a), deposition time (400~1200 s) (b); CV curves of Co/NF-CoC₂O₄ electrode prepared under different etching conditions: concentration of H₂C₂O₄ (0.2~1.0 mol dm⁻³) (c), time of H₂C₂O₄ etching (1~3 h) (d) and temperature of H₂C₂O₄ etching (298.15~338.15 K) (e); CV curves of Co/NF-CoC₂O₄-Co electrode prepared under different reduction times (1~3 h) (f). All CV

tests were performed in 1.0 mol dm⁻³ NaOH+0.1 mol dm⁻³ NaBH₄ solution at the scan

rate of 10 mV s⁻¹.