

Ionic liquid as modulate interface for high-efficient and stable perovskite solar cells

Xiang Chen,^a Lixin Song,^{*a} Ningxia Gu,^a Pengyun Zhang,^a Lei Ning,^a Pingfan Du,^a Fengfeng Chen,^b and Jie Xiong^{*a,b}

a College of Textile Science and Engineering, b School of Materials Science & Engineering, Zhejiang Sci-Tech University, Hangzhou, 310018, China. E-mail: jxiong@zstu.edu.cn, lxsong@zstu.edu.cn

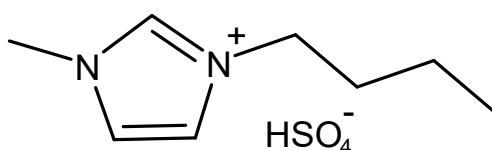


Fig.S1 Chemical structure of BMIMHSO₄ molecule.

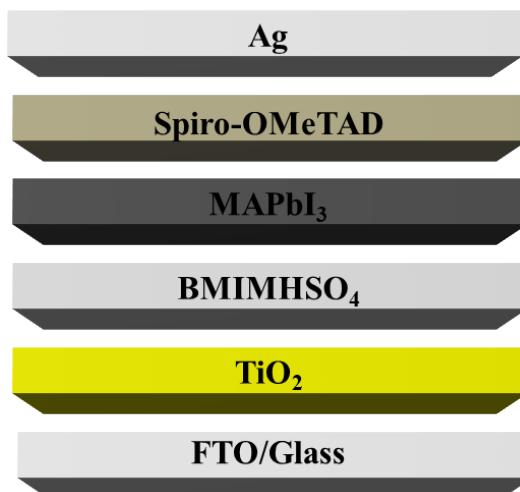


Fig.S2 Device structure of the PSCs.

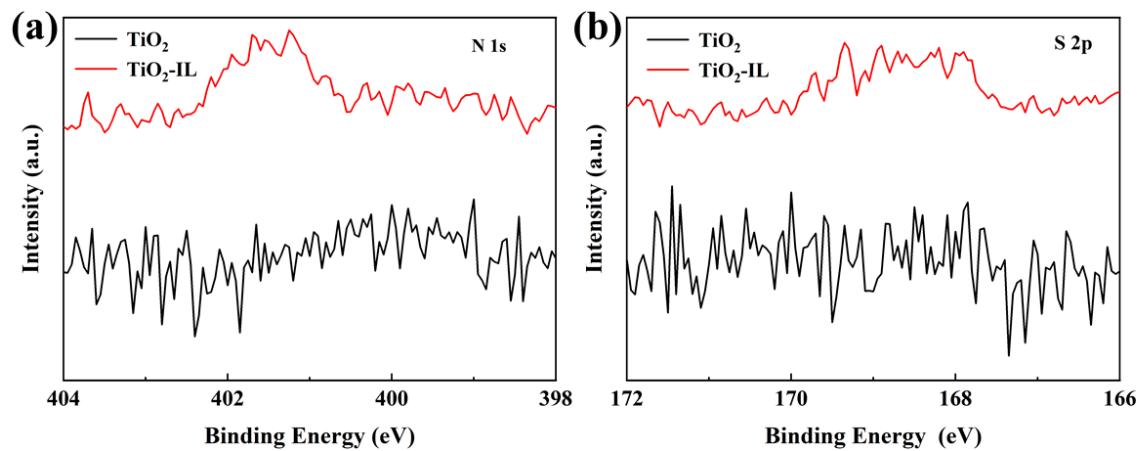


Fig.S3 The high-resolution XPS spectra of (a) N and (b) S element.

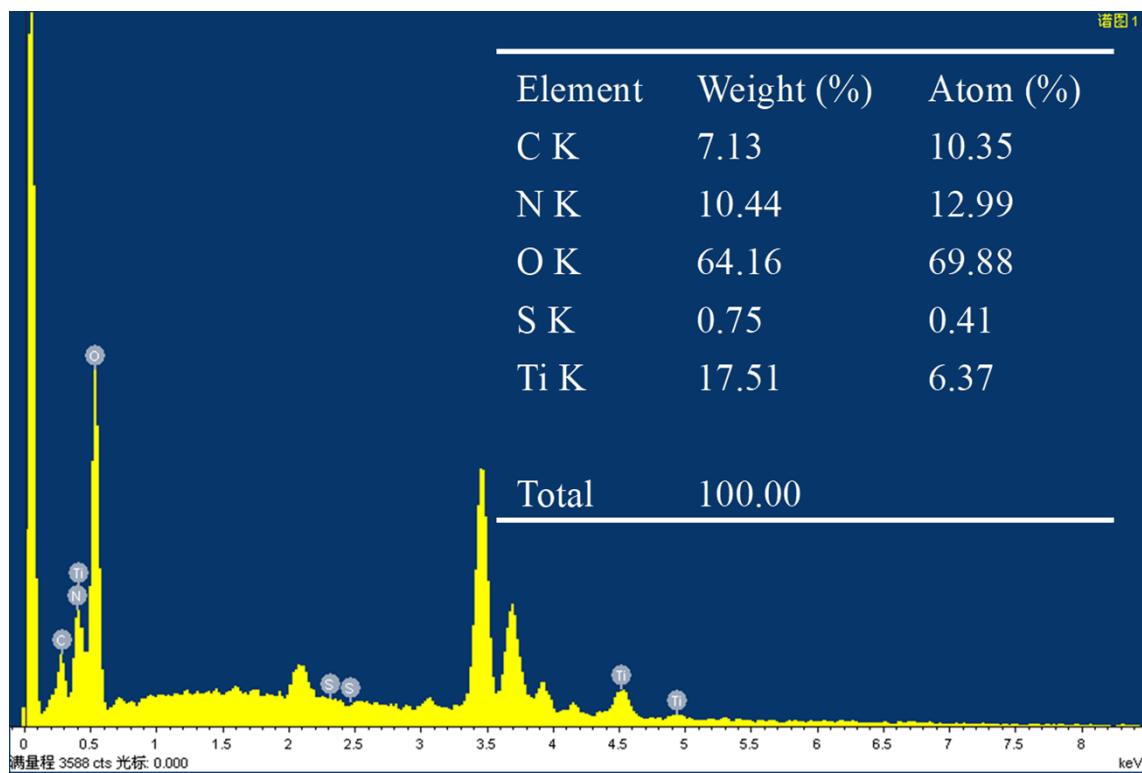


Fig.S4 The EDS measurement of $\text{TiO}_2\text{-IL}$ sample with the top-right showing the detail element analysis.

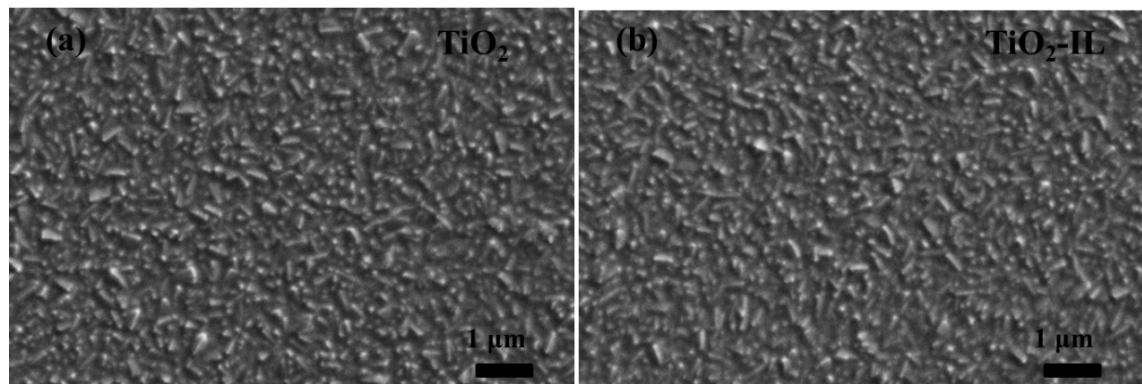


Fig.S5 The SEM images of (a) TiO₂ and (b) TiO₂-IL film.

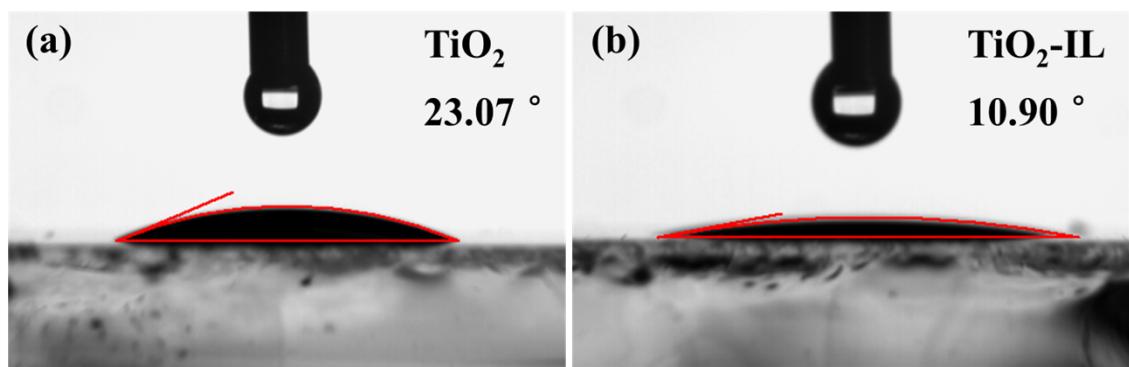


Fig.S6 Contact angle performance using DMSO solvent of (a) TiO₂ and (b) TiO₂ film.

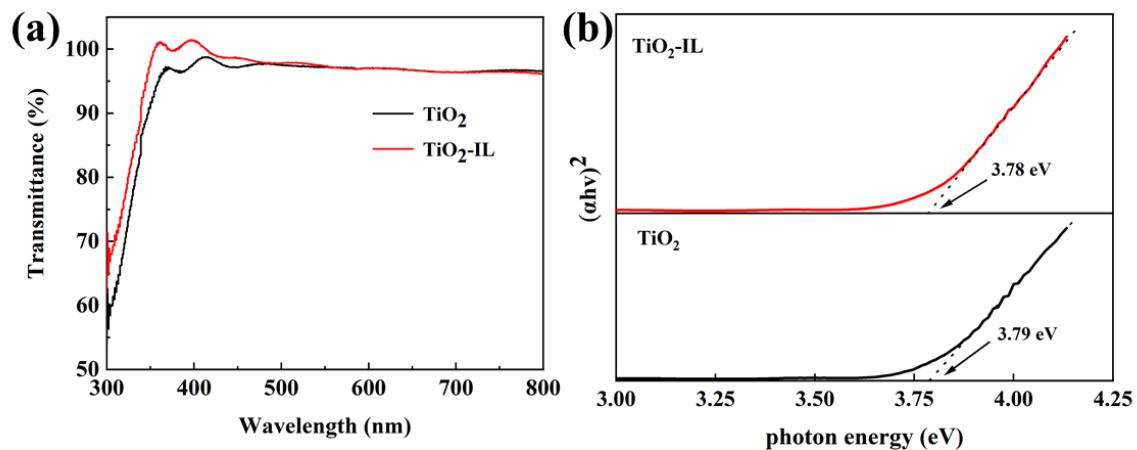


Fig.S7 (a)Transmission spectra and (b) Tauc plot of absorption spectra of TiO₂ and TiO₂-IL film.

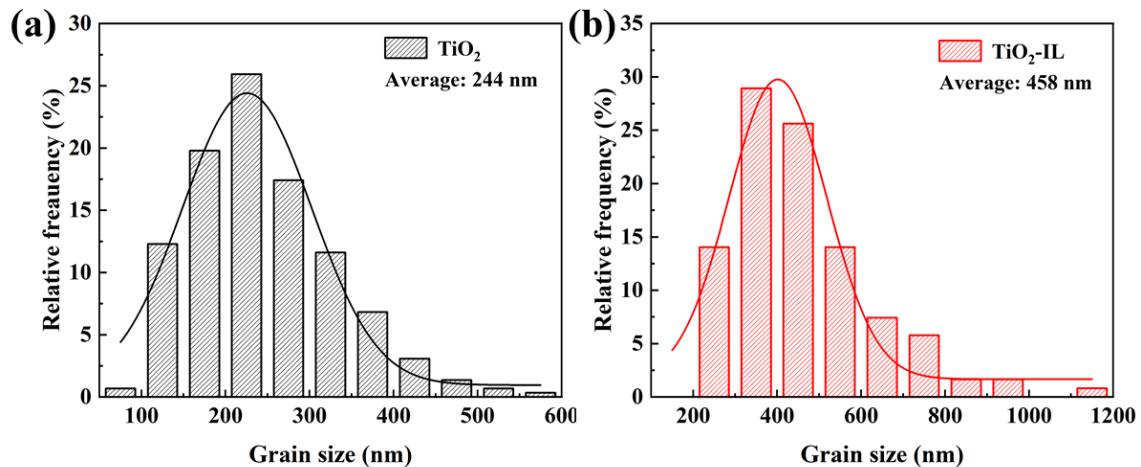


Fig.S8 The grain size distribution histograms of (a) TiO₂-based and (b) TiO₂-IL-based perovskite film.

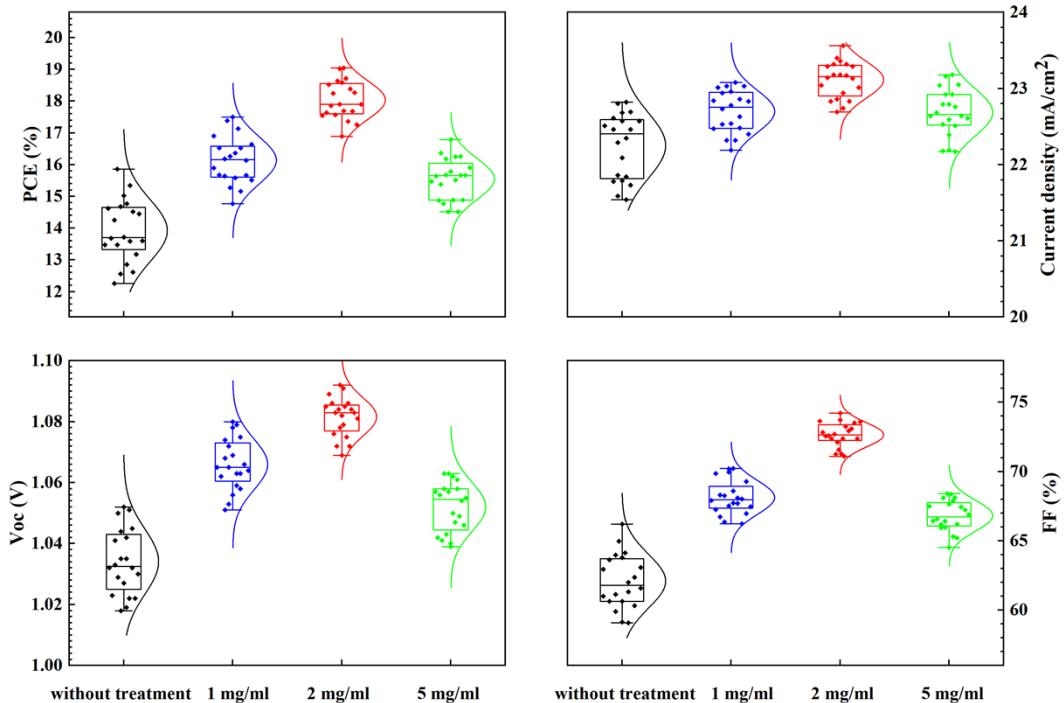


Fig.S9 Statistics on photovoltaic parameters for studied PSCs at various IL concentrations. J-V curves collected at 1 sun irradiation are utilized to derive parameters. A box is used to illustrate the standard error. Twenty devices with a cell active area of 0.06 cm² were fabricated for each PSC structure.

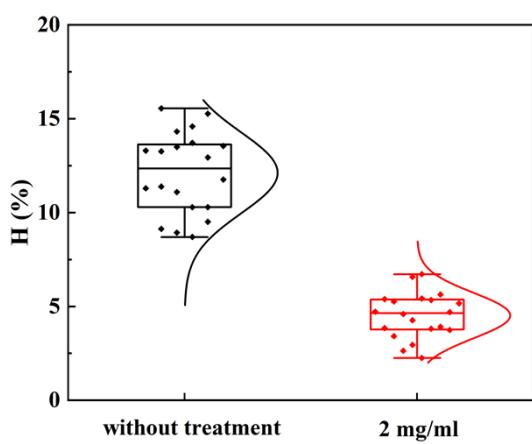


Fig.S10 Hysteresis index of twenty devices.

Table S1 The electron mobility and conductivity parameters based on two ETL.

Samples	Mobility ($\text{cm}^2 \text{ V}^{-1} \text{ S}^{-1}$)	Conductivity (mS cm^{-1})
TiO ₂	1.43×10^{-7}	1.27×10^{-3}
TiO ₂ -IL	4.88×10^{-7}	1.93×10^{-3}

Table S2 Photovoltaic parameters of PSCs based on TiO₂-IL with different concentrations.

Samples	Jsc (mA cm^{-2})	Voc (V)	FF (%)	PCE (%)
Without treatment	22.82	1.05	66.20	15.92
1 mg/ml	23.08	1.08	70.23	17.56
2 mg/ml	23.56	1.09	74.21	19.13
5 mg/ml	23.18	1.06	68.36	16.83

Table S3 Photovoltaic parameters measured at different scan directions of the champion PSCs based on the 0 mg/ml and 2 mg/ml IL concentration devices.

Samples	Scan direction	Jsc (mA cm^{-2})	Voc (V)	FF (%)	PCE (%)	H (%)
Without treatment	Reverse	22.82	1.05	66.20	15.92	8.70
	Forward	22.55	1.04	61.76	14.53	
2 mg/ml	Reverse	23.56	1.09	74.21	19.13	2.40
	Forward	23.45	1.09	72.86	18.67	

Table S4. The fitting parameters of TRPL of perovskite film for growing on TiO₂ and TiO₂-IL substrates.

Samples	A ₁	τ_1 (ns)	A ₂	τ_2 (ns)	τ_{avg} (ns)
TiO ₂	1.12	6.53	0.25	212.79	187.86
TiO ₂ -IL	1.89	2.64	0.57	23.55	17.88

Table S5 the electron trap density was calculated by SCLC measures of the deposited on TiO₂ film and TiO₂-IL film devices.

Samples	V _{TFL} (V)	N _{trap} (cm ⁻³)
TiO ₂	0.683	1.20×10^{14}
TiO ₂ -IL	0.475	0.83×10^{14}

Table S6 EIS parameters of PSCs based on TiO₂ and TiO₂-IL devices.

Samples	R _s (Ω)	R _{rct} (Ω)
TiO ₂	30.48	12388.32
TiO ₂ -IL	22.31	17704.03

Table S7 summarized the stability research about BMIMHSO₄ IL with other ILs.

Ionic liquid	Perovskite	PCE (%)	Decrease in stability (%) after hours	Ref

MAAc	CsPbIBr ₂	8.85	Air(room temperature), remains 82% of initial PCE after 30 days.	1
BMIMPF ₆	CsPbI ₂ Br	13.19	Air(20 °C, ~20% RH), remains 91% of initial PCE after 60 days.	2
FBABF ₄	(FAPbI ₃) _{1-x} (MAPbBr ₃) _x	23.07	Air(35±5% RH), remains 85% of initial PCE after 3000h.	3
N(CH ₃) ₄ OH (TMAH)	FA _{0.75} MA _{0.25} PbI _{2.5} Br _{0.5}	20.28	Desiccator(~15% RH), remains 97% of initial PCE after 360h.	4
BMIMBF ₄	MAPbI ₃	19.62	Air(AM 1.5G illumination), remains 85% of initial PCE after 240 min.	5
BMIMBF ₄	FA _{0.83} MA _{0.17} Pb(I _{0.83} Br _{0.17}) ₃	20.80	Air(85 °C, dark, under 45% RH), remains less than 50% of initial PCE after 800h.	6
EMIMI	MAPbI ₃	14.59	Dry air(RH<5%), remains 93% of initial PCE after 360h; air (30-40% RH), remains 51% of initial PCE after 360h.	7
[EMIM]PF ₆	MAPbI ₃	13.50	Air, remains 93.5% of initial PCE after 45 days.	8
MAAc	MAPbI ₃	21.08	Light, N ₂ -filled glovebox, remains 86% of initial PCE after 400h.	9
BMIMHSO ₄	MAPbI ₃	19.13	Air(~40% RH, 25±5°C, dark), remains 90% of initial PCE after 600h.	This work

Reference

1. L. Shi, H. Yuan, X. Sun, X. Li, W. Zhu, J. Wang, L. Duan, Q. Li, Z. Zhou, Z. Huang, X. Ban and D. Zhang, ACS Appl. Energy Mater., 2021, **4**, 10584-10592.
2. R. Yin, K.-X. Wang, S. Cui, B.-B. Fan, J.-W. Liu, Y.-K. Gao, T.-T. You and P.-G. Yin, ACS Appl. Energy Mater., 2021, **4**, 9294-9303.
3. D. Gao, L. Yang, X. Ma, X. Shang, C. Wang, M. Li, X. Zhuang, B. Zhang, H. Song, J. Chen and C. Chen, J. Energy Chem., 2022, **69**, 659-666.
4. C. Huang, P. Lin, N. Fu, K. Sun, M. Ye, C. Liu, X. Zhou, L. Shu, X. Hao, B. Xu, X. Zeng, Y. Wang and S. Ke, J. Mater. Chem. A, 2018, **6**, 22086-22095.
5. D. Yang, X. Zhou, R. Yang, Z. Yang, W. Yu, X. Wang, C. Li, S. Liu and R. P. H. Chang, Energy Environ. Sci., 2016, **9**, 3071-3078.
6. N. K. Noel, S. N. Habisreutinger, A. Pellaroque, F. Pulvirenti, B. Wenger, F. Zhang, Y.-H. Lin, O. G. Reid, J. Leisen, Y. Zhang, S. Barlow, S. R. Marder, A. Kahn, H. J. Snaith, C. B. Arnold and B. P. Rand, Energy Environ. Sci., 2019, **12**, 3063-3073.
7. J. Xu, X. Shi, J. Chen, J. Luan and J. Yao, J Solid State Chem, 2019, **276**, 302-308.
8. W. Zhang, Z. Ren, Y. Guo, X. He and X. Li, Electrochim. Acta, 2018, **268**, 539-545.
9. D. Li, L. Chao, C. Chen, X. Ran, Y. Wang, T. Niu, S. Lv, H. Wu, Y. Xia, C. Ran, L. Song, S. Chen, Y. Chen and W. Huang, Nano Lett., 2020, **20**, 5799-5806.