

Supporting information

Shear recovery and temperature stability of Ca^{2+} and Ag^+ glycolipid fibrillar metallogels with unusual β -sheet-like domains

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Table S 1 – Literature survey based on the scattering (SAXS or SANS) data reported in each cited article. Articles cited under *SAFiN with disordered fibers* report a typical scattering profile of the fiber alone, with or without a structure peak. Articles cited under *SAFiN with suprafibrillar assembly* report scattering profiles where the fiber's form factor is superimposed with the structure factor associated to the 3D organization of the fibers. Articles cited under *β -sheet-like gel (“nano-fishnet”)* report those hydrogels characterized by entanglement and β -sheet or β -sheet-like structure.

SAFiN with disordered fibers	
Not gelled	1–4
Gels	5,6,15–17,7–14
SAFiN with suprafibrillar assembly	
Hexagonal bundles/Columnar hexagonal	18–23
Raft-like/lamellar (solution, not gelled)	24,25
β-sheet-like gel (“nano-fishnet”)	
Biopolymers (fibroin, actin)	26–33
SAFiN	This work

The typical SAXS profiles of self-assembled fibrillar hydrogels from amphiphiles generally show an intense low-q scattering, of which the slope is related to the morphology (-1 for thin fibers and -2 for belts and ribbons⁴) and either a broad oscillation (form factor)³ or a broad structural peak.³⁴ We are not aware of SAXS fingerprints of fibrillar hydrogels from low-molecular weight amphiphiles with a series of diffraction peaks in a lamellar order.

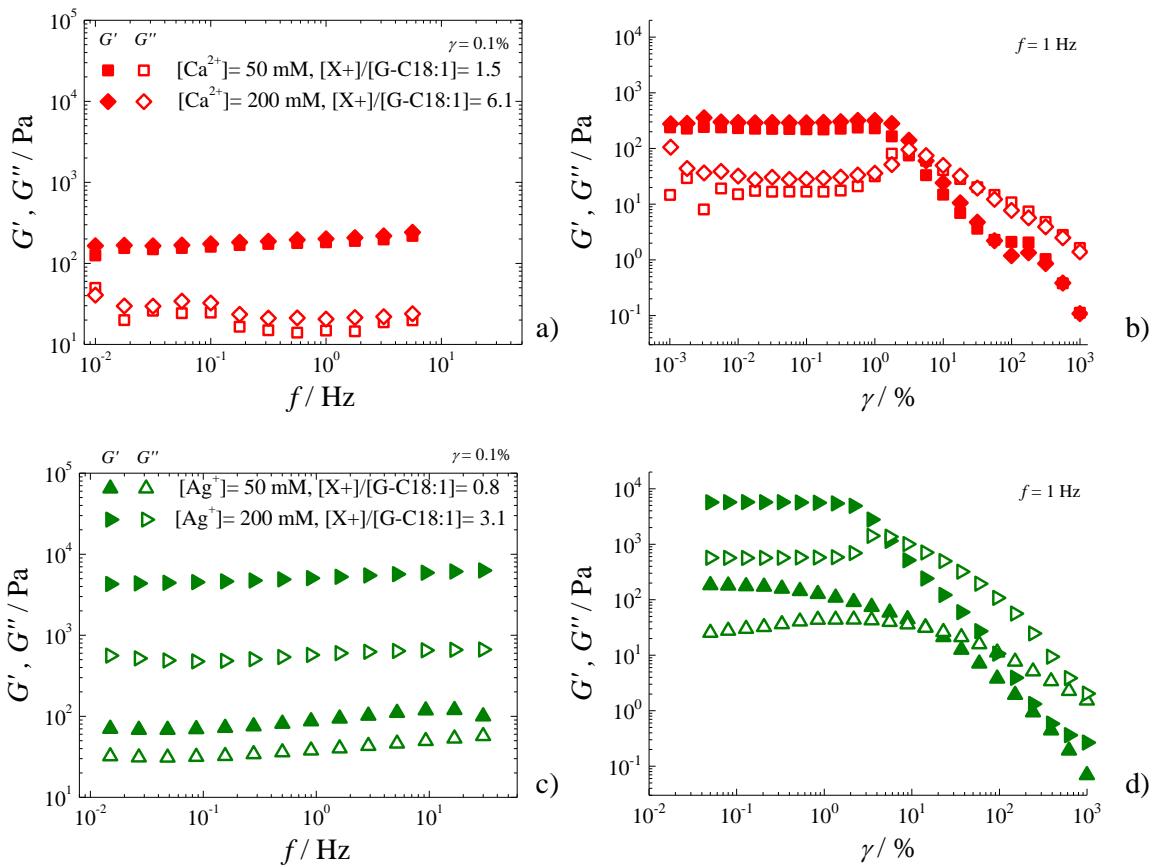


Figure S 1 - Elastic and viscous moduli, G' and G'' , of a,b) $\{\text{Ca}^{2+}\}\text{G-C18:1}$ (red square and diamonds) or c,d) $\{\text{Ag}^+\}\text{G-C18:1}$ (green triangle) hydrogels ($C_{\text{G-C18:1}} = 3 \text{ wt\%}$, pH 10). Ion concentrations in the lipid solution are 50 mM and 200 mM.

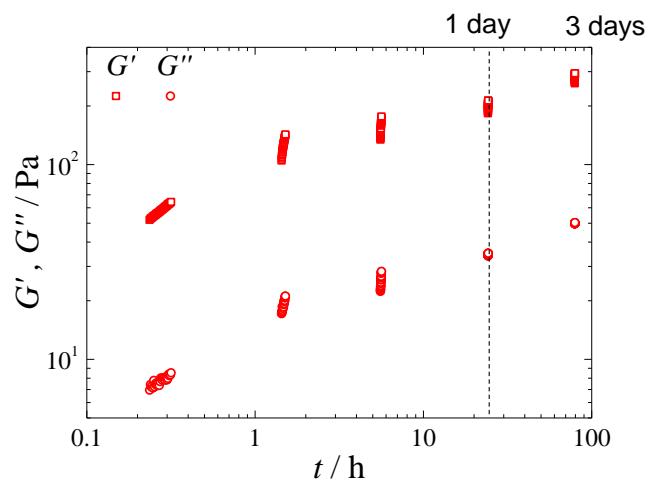


Figure S 2 – Time-dependency of the gelation ($\gamma = 0.3\%$, $f = 1 \text{ Hz}$) of $\{\text{Ca}^{2+}\}\text{G-C18:1}$ hydrogels ($C = 3 \text{ wt\%}$, pH 10). The charge ratio is $[\text{Ca}^{2+}]/[\text{G-C18:1}] = 0.61$. Hydrogel is left in a closed sample-holder and sampled at regular intervals, and not left under air on the rheometer geometry.

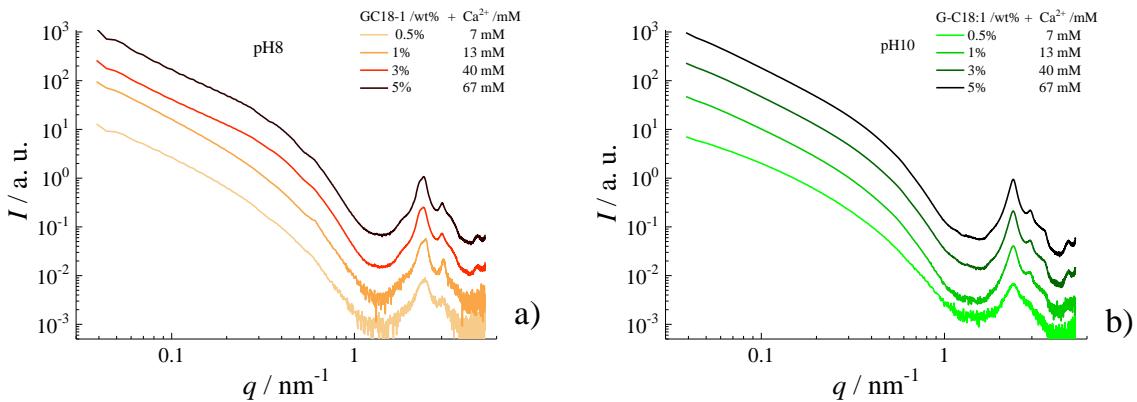


Figure S 3 – SAXS spectra of a $\{\text{Ca}^{2+}\}$ G-C18:1 hydrogels at a molar ratio 0.61 for different surfactant concentrations, at a) pH8, b) pH10. The charge ratio is $[\text{Ca}^{2+}]/[\text{G-C18:1}] = 0.61$. Data are shifted by a factor of about 5.

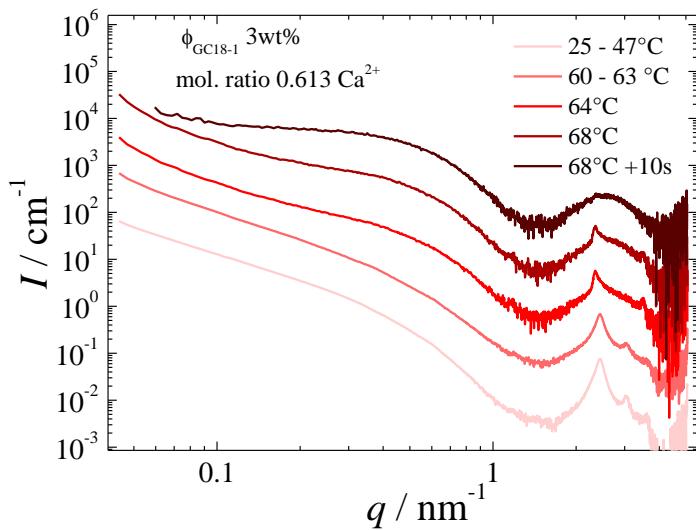


Figure S 4 - SAXS profiles recorded during temperature increase from 25°C to 70°C for a $\{\text{Ca}^{2+}\}\text{G-C18:1}$ gel at 3 wt%, basic pH and $[\text{Ca}^{2+}]/[\text{G-C18:1}] = 0.61$. Please note that the experimental setup in the SAXS-coupled rheometer (here, ID02, ESRF) is not suitable for a precise control over temperature. Data are shifted by a factor of about 15.

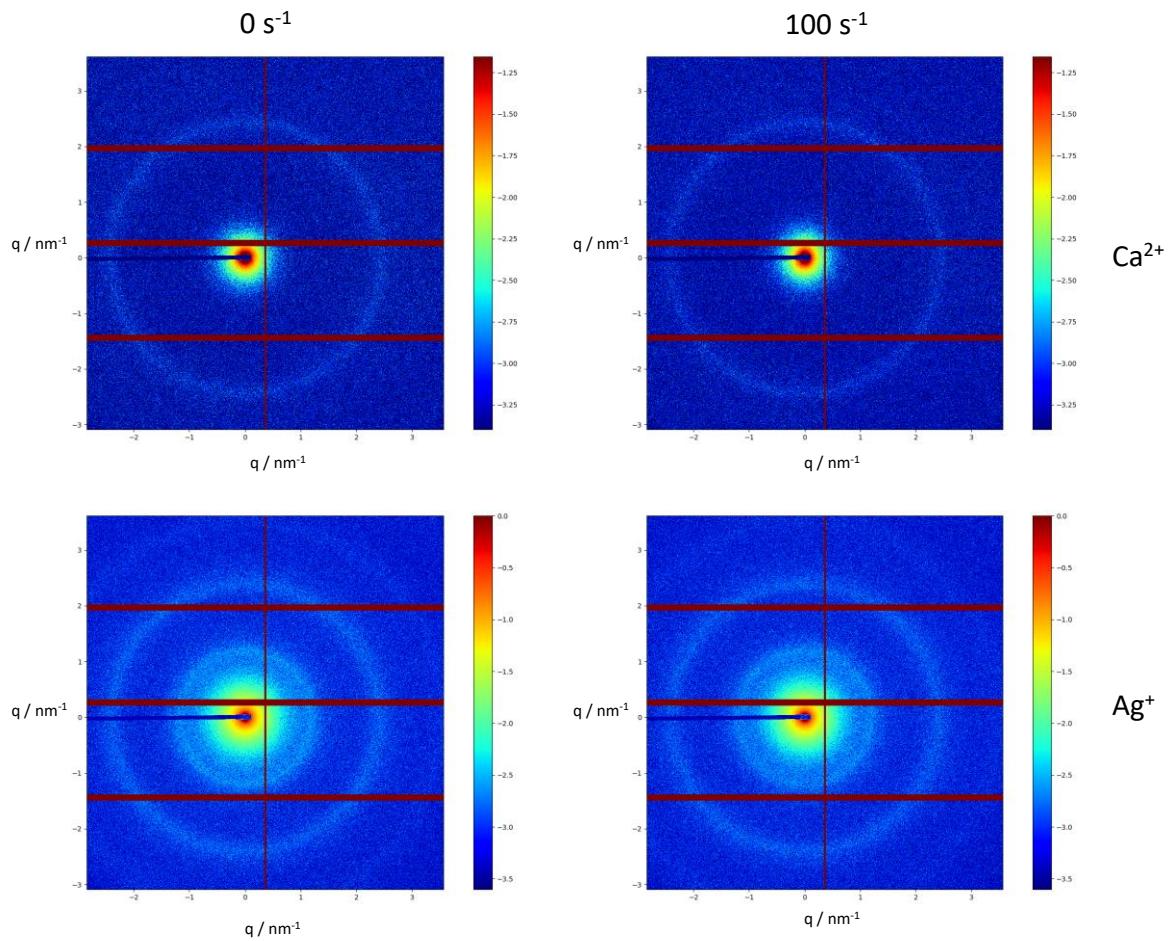


Figure S 5 – Rheo-SAXS experiment. 2D SAXS spectra of $\{\text{Ca}^{2+}\}$ (molar ratio 0.61) or $\{\text{Ag}^+\}$ (molar ratio 1.0) G-C18:1 hydrogel ($C_{\text{G-C18:1}} = 3\%$) at pH 10 sheared at 0 s^{-1} or 100 s^{-1}

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