

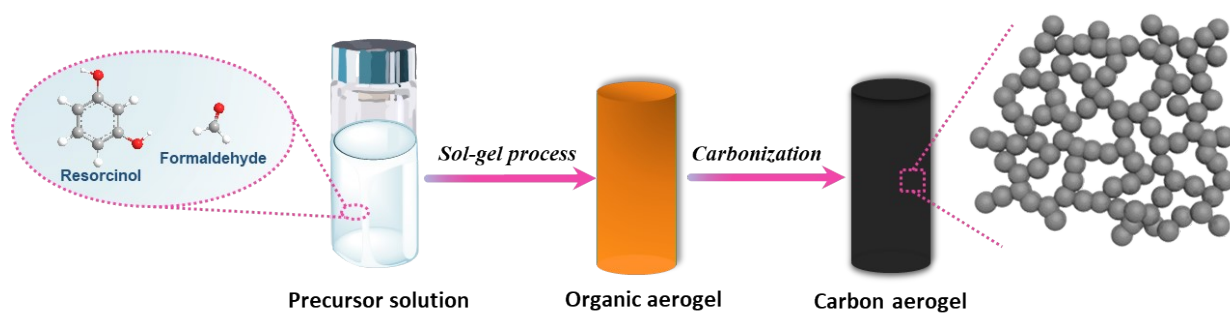
## Supplementary Materials

### Keys to Intimately Coupling Metal Chalcogenides with Carbon Nanonetwork for Potassium-Ion Storage

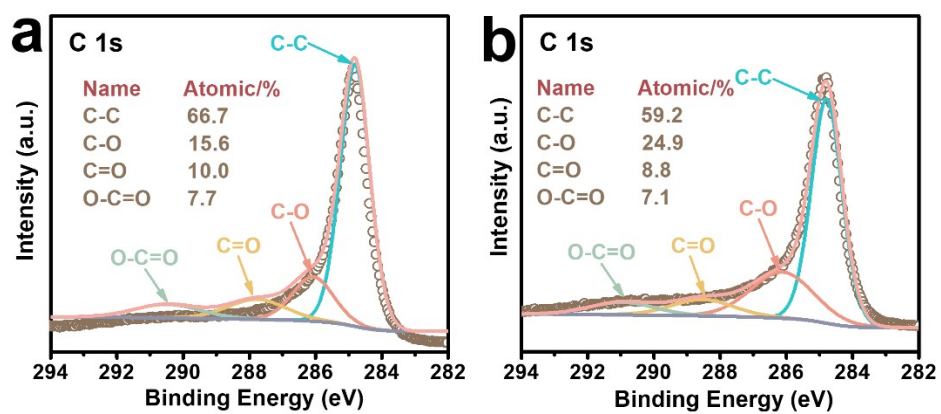
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## Supplementary Figures



**Figure S1.** Schematic illustration for preparing CA.



**Figure S2.** XPS spectra of C 1s spectrums in (a) CA and (b) O-CA.

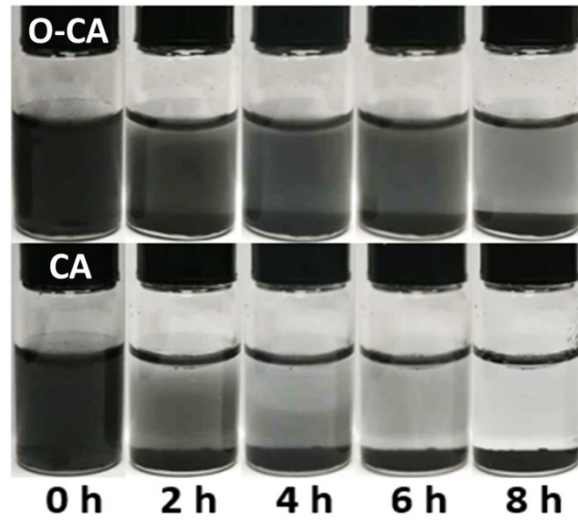


Figure S3. Digital photographs of CA and O-CA dispersed in deionized water.

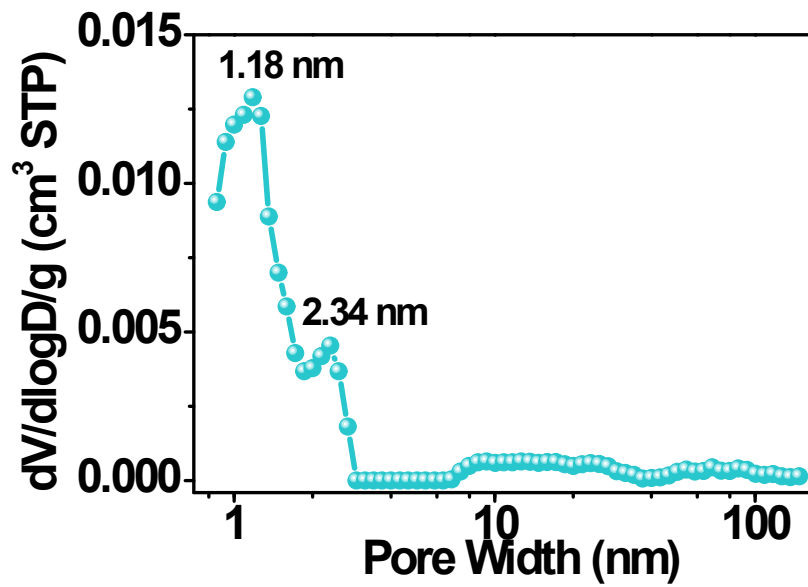


Figure S4. The pore size distribution of CA@MoSe<sub>2</sub>.

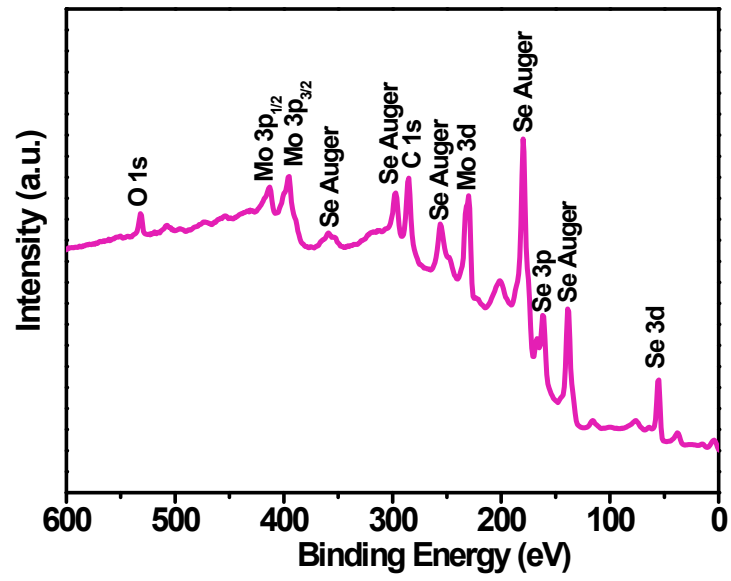


Figure S5. The XPS survey spectrum of CA@MoSe<sub>2</sub>.

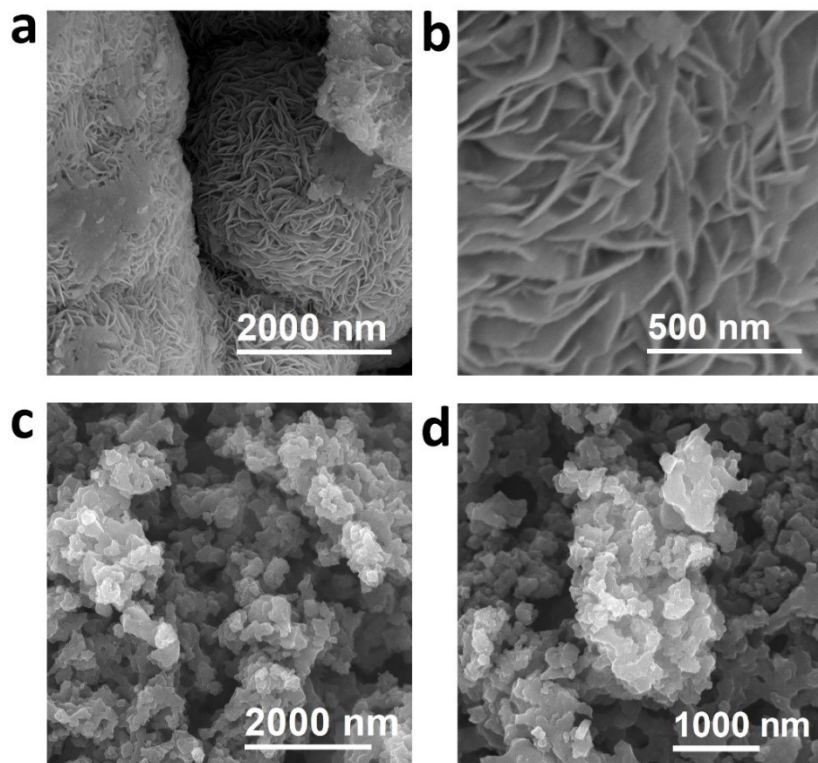
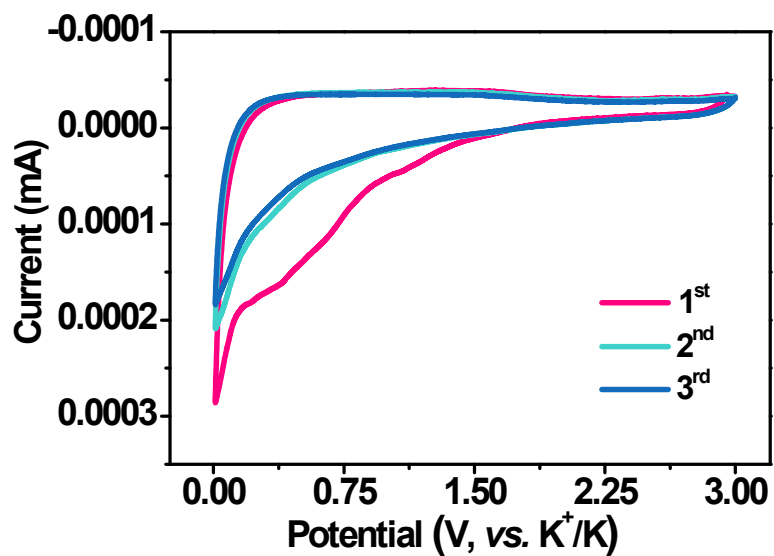
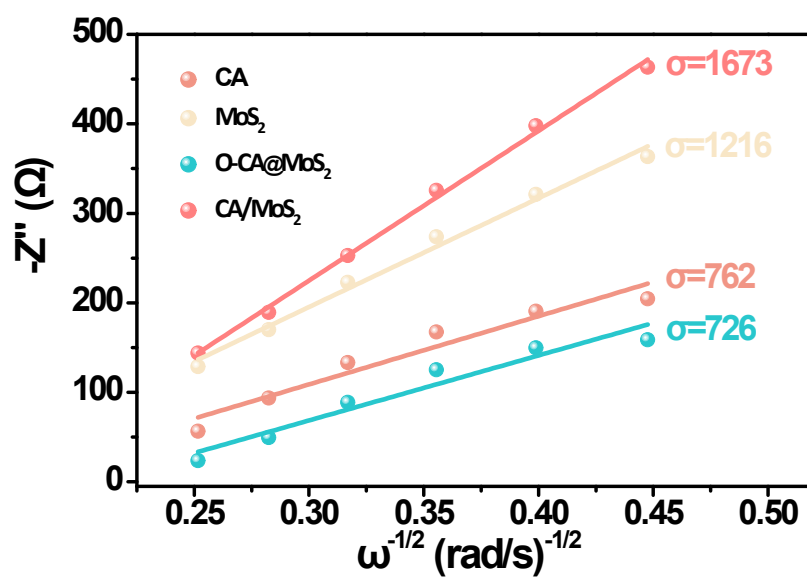


Figure S6. SEM images of (a, b) MoS<sub>2</sub> and (c, d) CA/MoS<sub>2</sub>.



**Figure S7.** CV curves of CA@MoS<sub>2</sub> anode cycled at the 1<sup>st</sup>, 2<sup>nd</sup>, and 3<sup>rd</sup> between 0.01 and 3 V (vs. K<sup>+</sup>/K) at a scan rate of 0.1 mV s<sup>-1</sup>.



**Figure S8.** Plots of  $\omega^{-1/2}$  versus  $-Z''$  of different anodes in KIBs.

## Supplementary Tables

**Table S1.** Chemical structure of CA and O-CA characterized by XPS.

<b>Sample</b>	<b>C 1s (at. %)</b>	<b>O 1s (at. %)</b>	<b>C/O ratio (at. %)</b>
CA	95.2	3.4	28.0
O-CA	92.8	5.7	16.3

**Table S2.** Binding energy and net electron transfer of graphene-Mo and graphene oxide-Mo.

<b>Systems</b>	<b><math>E_{\text{total}}</math> (eV)</b>	<b><math>E_{\text{substrate}}</math> (eV)</b>	<b><math>E_{\text{Mo}}</math> (eV)</b>	<b><math>E_{\text{b}}</math> (eV)</b>	<b><math>\Delta q</math> (e)</b>
Carbon-Mo	-465.78	-461.07	-0.43	-4.28	0.38
Oxidized carbon-Mo	-583.00	-575.64	-0.43	-6.93	1.02

**Table S3.** Comparisons of the discharge capacity of different anodes for KIBs.

<b>Anodes</b>	<b>Current density (mA g<sup>-1</sup>)</b>	<b>Discharge capacity (mAh g<sup>-1</sup>)</b>	<b>References</b>
Mesoporous carbon	50	286	[1]
Graphitic carbon nanocage	14	137	[2]
Porous hard carbon microspheres	50	227	[3]
CoS@graphene	50	311	[4]
CoS/N-doped carbon	200	303	[5]
Fe-Mo selenide@N-doped carbon	200	298	[6]
Iron sulfide/carbon hybrid cluster	100	226	[7]
FeCl <sub>3</sub> -intercalated expanded graphite	50	270	[8]
FeS <sub>2</sub> @rGO	50	264	[9]
Sandwich-like MoS <sub>2</sub> @SnO <sub>2</sub> @C	50	312	[10]
P@nanotube-backboned mesoporous carbon	50	244	[11]
Phosphorus/carbon composite	50	324	[12]
ReS <sub>2</sub> /N-doped carbon	50	253	[13]
Sb@N, P co-doped mesoporous carbon	50	266	[14]
3D hierarchically porous carbon/Sn	50	276	[15]
KTi <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> @C	20	293	[16]
Metal-organic framework-MIL-125 (Ti)	10	208	[17]
Sn-interspersed MoS <sub>2</sub> /C nanosheets	100	290	[18]
MoSe <sub>2</sub> /N-Doped Carbon	100	258	[19]
Pistachio-shuck-like MoSe <sub>2</sub> /C	200	322	[20]
<b>CA@MoS<sub>2</sub></b>	<b>100</b>	<b>389</b>	<b>This work</b>



**Table S4.** The kinetic parameters of different anodes.

<b>Parameters</b>	<b>CA@MoS<sub>2</sub></b>	<b>CA</b>	<b>CA/MoS<sub>2</sub></b>	<b>MoS<sub>2</sub></b>
R <sub>s</sub> (Ω)	2.3	3.6	4.0	30.5
R <sub>ct</sub> (Ω)	1619.1	1287.4	1923.0	1740.5
σ (Ω Hz <sup>1/2</sup> )	726.3	762.3	1673.1	1215.9
D (10 <sup>-20</sup> cm <sup>2</sup> s <sup>-1</sup> )	8.1	7.4	1.5	2.9

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