Supporting Information

Solvent-Free Molten Co-Assembly to Ordered Mesoporous Carbon for Efficiently Supported Adsorption and Separation of SO₂

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Figure S1 ¹H NMR spectra of mPEI.



Figure S2 Wide-angle XRD patterns of OMC and various OMC@mPEI-x.



Figure S3 (a) N_2 adsorption-desorption isotherms at -196 °C and (b) pore size distributions of OMC, OMS and TPA-OMC.



Figure S4 Low-angle XRD pattern of OMS.



Figure S5 Photographs of OMC and various OMC@mPEI-x.



Figure S6 FT-IR spectra of mPEI.



Figure S7 SO₂ adsorption capacities of OMC@mPEI-x samples at 25 °C and 1 bar.



Figure S8 mPEI efficiencies of (A) OMC@mPEI-1, (B) OMC@mPEI-1.5, (C) OMC@mPEI-2 and (D) OMC@mPEI-2.5 for SO₂ adsorption at 25 °C and different pressures: (a) 1 bar and (b) 0.002 bar.

Note: the mPEI utilization efficiency of OMC@mPEI-x was calculated from the following equations of S1 and S2:

$$M = \frac{x}{1+x}Mi$$
(S1)

$$n = \frac{m}{M}$$
(S2)

where M is the molar of tertiary amine group per g catalyst; x represents the mass ratio of mPEI to OMC; n is the utilization efficiency of mPEI; m is the adsorption capacity of SO₂ (mmol/g); (Mi = 16.6 mmol/g for the molar of tertiary amine group per g PEI).



Figure S9 SO₂ adsorption isotherm of UiO-66-NH₂, OMC@mPEI-2, ZSM-5@mPEI-

2, SBA-15@mPEI-2, and TPA-OMC@mPEI-1.5 at 25 °C and 1 bar.



Figure S10 IAST selectivities of OMC@mPEI-x for (a) SO₂/CO₂ and (b) SO₂/N₂ at 50

°C.



Figure S11 N_2 adsorption-desorption isotherms at -196 °C of fresh OMC@mPEI-1.5 and used OMC@mPEI-1.5 in SO₂ adsorption.



Figure S12 Breakthrough curves of SBA-15@mPEI-2 for SO₂/CO₂/N₂ gas mixture (20 mL/min) at 25 °C



Figure S13 Breakthrough curves of OMC@mPEI-2 for SO $_2$ /CO $_2$ /N $_2$ /H $_2$ O gas mixture

(20 mL/min) at 25 °C

| Samples | Condition | SO ₂ capacities | Ref |
|------------------------|------------------|----------------------------|-----------|
| Sumples Condition | | (mmol/g) | Net. |
| OMC@mPEI-2 | 25 °C, 1.0 bar | 16.1 | This work |
| OMC@mPEI-2 | 25 °C, 0.002 bar | 3.9 | This work |
| UiO-66-NH ₂ | 25 °C, 1.0 bar | 7.9 | This work |
| ZSM-5@mPEI-2 | 25 °C, 1.0 bar | 10.1 | This work |
| SBA-15@mPEI-2 | 25 °C, 1.0 bar | 11.5 | This work |
| TPA-OMC@mPEI-1.5 | 25 °C, 1.0 bar | 12.5 | This work |
| P(Ph-4MVIm-Br) | 25 °C, 0.002 bar | 1.6 | (S1) |
| SIFSIX-2-Cu-i | 25 °C, 0.002 bar | 2.3 | (S2) |
| DT-NPC-3 | 25 °C, 0.1 bar | 5.2 | (S3) |
| ACT | 25 °C, 1.0 bar | 7.9 | (S4) |
| CFT-CSU38 | 25 °C, 0.1 bar | 7.9 | (S5) |
| MIL-160 | 25 °C, 0.97 bar | 7.3 | (S6) |
| Mg-MOF-74 | 25 °C, 1.0 bar | 8.6 | (S7) |
| MFM-600 | 25 °C, 1.0 bar | 5.0 | (S8) |
| MFM-300(In) | 25 °C, 1.0 bar | 8.3 | (S9) |
| MFM-300(Al) | 25 °C, 1.0 bar | 7.1 | (S10) |
| MFM-202a | 25 °C, 1.0 bar | 10.2 | (S11) |
| GU-1 | 25 °C, 1.0 bar | 7.6 | (S12) |
| HNIP-TBMB-1 | 25 °C, 1.0 bar | 7.2 | (\$13) |

Table S1 SO_2 capacities of different porous solid adsorbents.

| Samples | Condition | Temperature | IAST selectivity | Ref. | |
|---------------|----------------------------------|-------------|------------------|-----------|--|
| OMC@mPEI-2 | SO ₂ /CO ₂ | 25 °C | 1789 | This work | |
| | (0.1:0.9 v/v, 1 bar) | 25 0 | 1705 | | |
| | SO ₂ /CO ₂ | | 07.1 | (S2) | |
| SIFSIX-2-Cu-I | (0.1:0.9 v/v, 1 bar) | 25 C | 87.1 | | |
| | SO ₂ /CO ₂ | 25.00 | 30 | (S2) | |
| ELM-12 | (0.1:0.9 v/v, 1 bar) | 25 C | | | |
| | SO ₂ /CO ₂ | 25.40 | 124 | (S6) | |
| MIL-160 | (0.1:0.9 v/v, 1 bar) | 25 °C | | | |
| | SO ₂ /CO ₂ | | 67.5 | (S8) | |
| MFM-601 | (0.1:0.9 v/v, 1 bar) | 25 °C | | | |
| MFM-300(In) | SO ₂ /CO ₂ | 25 °C | 33.6 | (S9) | |

 Table S2 IAST selectivities of different porous solid adsorbents.

| | (0.1:0.9 v/v, 1 bar) | | | |
|------------|----------------------------------|-------|------|--------|
| Mg gallata | SO ₂ /CO ₂ | | 224 | (S14) |
| Mg-gallate | (0.1:0.9 v/v, 1 bar) | 25 C | 321 | |
| Co gallato | SO ₂ /CO ₂ | 25 °C | | (S14) |
| Co-gallate | (0.1:0.9 v/v, 1 bar) | 25 C | 145 | |
| MEN4 170 | SO ₂ /CO ₂ | 25 °C | 32 | (S15) |
| | (0.1:0.9 v/v, 1 bar) | 25 C | | |
| NKU-100 | SO ₂ /CO ₂ | 25 °C | 940 | (S16) |
| NK0-100 | (0.5:0.5 v/v, 100000 Pa) | 25 C | | |
| | SO ₂ /CO ₂ | 25 °C | 160 | (S17) |
| | (0.1:0.9 v/v, 1 bar) | 25 C | 109 | |
| FCUT-111 | SO ₂ /CO ₂ | 25 °C | 25.2 | (\$18) |
| | (1:99 v/v, 1 bar) | 25 C | 23.2 | (510) |
| HBU-20 | SO ₂ /CO ₂ | 25 °C | 44.3 | (S19) |

| | (0.1:0.9 v/v, 1 bar) | | | |
|----------------|--|-------|-----|-------|
| 2.0mPEI@g-C3N4 | SO ₂ /CO ₂ (0.1:0.9 v/v, 1 bar) | 25 °C | 630 | (S20) |



Scheme S1 Diagram for dual-chamber equipment used for measurements of SO₂ adsorption isotherms (1: gas cylinder; 2: storage chamber; 3: adsorption chamber; 4: water bath; 5~6: pressure transducers; 7: digital displayer; 8~10: needle valve; 11: temperature probe; 12: vacuum pump; 13: elevator platform; 14: magnetic stirrer).

Measuring procedures:

The equipment has two chambers made of 316 L stainless steel: one is used for the storage of SO₂, and the other is used for the adsorption of SO₂. The temperature of two chambers is controlled by a water bath. The pressures in two chambers are monitored by two Wideplus-8 transducers respectively. In a typical run, a certain amount of sample was placed in the adsorption chamber; the equipment was completely evacuated by a pump, and a certain amount of SO₂ was introduced to the storage chamber from the cylinder; after stabilization, the pressure in storage chamber was recorded as P_0 ; then a certain amount of SO₂ was introduced to the adsorption chamber from the storage chamber; after stabilization, the pressure in storage chamber from the storage chamber; after stabilization, the pressures in storage chamber and adsorption chamber were recorded as P_1 and P_2 respectively; the amount of SO₂ adsorbed at the equilibrium pressure of P_2 was calculated using the following equation:

$$n = \rho_0 V_1 - \rho_1 V_1 - \rho_2 V_2 \tag{S3}$$

where *n* is the amount of SO₂ adsorbed; ρ_i is the density of SO₂ at P_i (*i*=0~2); V_j is the free volume of chamber (*j*=1 for storage chamber and 2 for adsorption chamber). The values of ρ_i

were obtained from the NIST Chemistry WebBook.^{S21} The values of V_j were measured using He as the probing gas. The amounts of SO₂ adsorbed at elevated pressures were measured by loading more SO₂ to the adsorption chamber. The amounts of SO₂ adsorbed were finally transformed to mmol of SO₂ per gram of sample.

Density functional theory (DFT) calculation

Density functional theory (DFT) calculations were conducted using Gaussian 16 program.^{S21} Geometry optimizations and frequency calculations was performed with the wb97XD hybrid exchange-correlation functional with the def2-SVP basis set Single-point energies were further calculated with the wb97XD hybrid exchange-correlation functional with a larger basis set def2-TZVP basis set.^{S22,S23} The energies reported here refer to the single-point energies.

N,*N*-dimethylethanamine, *N*-ethyl-*N*-methylethanamine and triethylamine were used to represent the three kinds of tertiary amine species, and for each model, the adsorption energy (ΔE) was calculated according to the equation below:

 $\Delta E = E(\text{amine} \text{SO}_2) - E(\text{amine}) - E(\text{SO}_2)$

Where $E(\text{amine } SO_2)$, E(amine) and $E(SO_2)$ are the energies of optimized model of adsorbed SO₂, amine and SO₂, respectively.

Bond distances are in angstrom.

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Single-point energies (a.u.) and Cartesian coordinates

 SO_2

E = -548.6516585

| S | 0.000000000 | 0.370418000 | 0.000000000 |
|---|--------------|--------------|-------------|
| 0 | 1.246194000 | -0.370418000 | 0.000000000 |
| 0 | -1.246194000 | -0.370418000 | 0.000000000 |

N,N-dimethylethanamine

E = -213.8001251

| N | 0.456881000 | -0.003335000 | -0.296842000 |
|---|--------------|--------------|--------------|
| С | 0.627537000 | 1.377046000 | 0.091788000 |
| Н | 1.504121000 | 1.803717000 | -0.418623000 |
| Н | -0.244401000 | 1.979355000 | -0.199879000 |
| Η | 0.778952000 | 1.505021000 | 1.189216000 |
| С | 1.657358000 | -0.772797000 | -0.071292000 |
| Η | 1.531870000 | -1.796614000 | -0.455727000 |
| Η | 2.506177000 | -0.316792000 | -0.603618000 |
| Η | 1.935711000 | -0.847302000 | 1.005652000 |
| С | -0.708860000 | -0.632221000 | 0.296150000 |
| Η | -0.690476000 | -1.695587000 | 0.006835000 |
| Η | -0.654297000 | -0.616266000 | 1.411545000 |
| С | -2.032036000 | -0.030014000 | -0.156267000 |
| Η | -2.871278000 | -0.638619000 | 0.211171000 |
| Η | -2.179857000 | 0.991588000 | 0.224090000 |
| Η | -2.078681000 | 0.002767000 | -1.255040000 |

N,N-dimethylethanamine-SO₂

E = -762.472943

| N | -0.686793000 | 0.568604000 | 0.151174000 |
|---|--------------|-------------|--------------|
| С | -1.069716000 | 1.341533000 | -1.017454000 |
| Η | -0.234861000 | 2.001586000 | -1.291837000 |
| Η | -1.268682000 | 0.678621000 | -1.870117000 |

| Η | -1.969237000 | 1.958656000 | -0.830501000 |
|---|--------------|--------------|--------------|
| С | -0.238232000 | 1.411568000 | 1.247683000 |
| Η | 0.141416000 | 0.774509000 | 2.059479000 |
| Η | 0.588139000 | 2.047924000 | 0.899816000 |
| Η | -1.055457000 | 2.048168000 | 1.634346000 |
| С | -1.654039000 | -0.433836000 | 0.599763000 |
| Н | -1.158156000 | -0.996698000 | 1.406370000 |
| Η | -2.532545000 | 0.071745000 | 1.049823000 |
| С | -2.110504000 | -1.401388000 | -0.480314000 |
| Η | -2.720697000 | -2.192281000 | -0.022258000 |
| Η | -2.727337000 | -0.915843000 | -1.250205000 |
| Η | -1.254098000 | -1.883874000 | -0.974024000 |
| S | 1.316167000 | -0.555454000 | -0.397392000 |
| 0 | 2.087566000 | 0.669444000 | -0.613302000 |
| 0 | 1.459351000 | -1.193535000 | 0.912190000 |

N-ethyl-N-methylethanamine

E = -253.1167972

| Ν | 0.002783000 | 0.371870000 | -0.190165000 |
|---|--------------|-------------|--------------|
| С | -0.561972000 | 1.699928000 | -0.115224000 |
| Η | 0.045678000 | 2.403462000 | -0.703925000 |
| Η | -1.577286000 | 1.717320000 | -0.535726000 |
| Η | -0.622379000 | 2.090253000 | 0.927099000 |
| С | 1.390436000 | 0.350657000 | 0.242281000 |
| Η | 1.872508000 | 1.256571000 | -0.158228000 |
| Η | 1.473100000 | 0.428574000 | 1.351901000 |

| С | -0.818196000 | -0.620172000 | 0.482837000 |
|---|--------------|--------------|--------------|
| Η | -0.242427000 | -1.555028000 | 0.543937000 |
| Η | -1.025299000 | -0.318828000 | 1.537224000 |
| С | -2.128789000 | -0.920155000 | -0.233179000 |
| Η | -2.652058000 | -1.751725000 | 0.261713000 |
| Η | -2.813821000 | -0.059459000 | -0.234622000 |
| Η | -1.933166000 | -1.201742000 | -1.278390000 |
| С | 2.171585000 | -0.860793000 | -0.250352000 |
| Η | 1.811484000 | -1.803132000 | 0.188293000 |
| Η | 2.091707000 | -0.942965000 | -1.344278000 |
| Н | 3.234087000 | -0.763178000 | 0.017978000 |

N-ethyl-N-methylethanamine-SO₂

| E = -801.7888257 | |
|------------------|--|
| E = -801.7888257 | |

| N | 0.717785000 | 0.018308000 | 0.389995000 |
|---|--------------|--------------|--------------|
| С | 1.615197000 | -0.684312000 | -0.531773000 |
| Н | 1.514425000 | -0.204713000 | -1.516714000 |
| Н | 2.666754000 | -0.532549000 | -0.210627000 |
| С | 0.545725000 | -0.642794000 | 1.673691000 |
| Н | -0.110930000 | -0.026620000 | 2.303491000 |
| Н | 0.052328000 | -1.614352000 | 1.539461000 |
| Н | 1.511103000 | -0.784964000 | 2.195349000 |
| С | 1.051877000 | 1.434010000 | 0.573091000 |
| Н | 0.239107000 | 1.868167000 | 1.175460000 |
| Н | 1.983239000 | 1.514890000 | 1.169978000 |
| С | 1.189921000 | 2.238541000 | -0.709526000 |

| Η | 1.322200000 | 3.298366000 | -0.450283000 |
|---|--------------|--------------|--------------|
| Η | 2.057475000 | 1.937949000 | -1.313924000 |
| Η | 0.282661000 | 2.163100000 | -1.325986000 |
| S | -1.496912000 | -0.011028000 | -0.512775000 |
| 0 | -1.893403000 | -1.298142000 | 0.061823000 |
| 0 | -1.953421000 | 1.180294000 | 0.204111000 |
| С | 1.329169000 | -2.171932000 | -0.678978000 |
| Η | 0.268517000 | -2.356244000 | -0.900251000 |
| Η | 1.932717000 | -2.575029000 | -1.504399000 |
| Н | 1.589765000 | -2.738019000 | 0.226378000 |

triethylamine

E = -292.4337314

| Ν | 0.000403000 | 0.000036000 | 0.059334000 |
|---|--------------|--------------|--------------|
| С | -0.314619000 | -1.364258000 | 0.451843000 |
| Н | 0.611525000 | -1.956787000 | 0.407994000 |
| Н | -0.650401000 | -1.411189000 | 1.513091000 |
| С | -1.023108000 | 0.954992000 | 0.453450000 |
| Н | -1.999160000 | 0.448499000 | 0.414125000 |
| Н | -0.892320000 | 1.271397000 | 1.513569000 |
| С | 1.339400000 | 0.408929000 | 0.452545000 |
| Н | 1.389075000 | 1.507402000 | 0.411405000 |
| Н | 1.548127000 | 0.139090000 | 1.513111000 |
| С | 2.429186000 | -0.149890000 | -0.452859000 |
| Н | 3.419626000 | 0.210982000 | -0.136739000 |
| Н | 2.458762000 | -1.249874000 | -0.434698000 |

| Η | 2.254613000 | 0.163675000 | -1.492455000 |
|---|--------------|--------------|--------------|
| С | -1.085495000 | 2.177483000 | -0.452797000 |
| Η | -0.147234000 | 2.752464000 | -0.437196000 |
| Η | -1.272126000 | 1.868739000 | -1.491738000 |
| Η | -1.892160000 | 2.855575000 | -0.135576000 |
| С | -1.345602000 | -2.027270000 | -0.452119000 |
| Η | -1.528164000 | -3.065681000 | -0.136661000 |
| Η | -2.312714000 | -1.502440000 | -0.431509000 |
| Η | -0.988841000 | -2.032020000 | -1.492441000 |

triethylamine-SO₂

E = -841.1047165

| N | 0.712741000 | -0.003634000 | 0.233580000 |
|---|-------------|--------------|--------------|
| С | 1.667104000 | -0.672224000 | -0.660061000 |
| Η | 1.632891000 | -0.148720000 | -1.626397000 |
| Η | 2.694967000 | -0.541353000 | -0.263385000 |
| С | 0.473456000 | -0.760452000 | 1.470192000 |
| Η | 0.001397000 | -1.709290000 | 1.178801000 |
| Η | 1.448578000 | -1.004528000 | 1.939767000 |
| С | 1.047421000 | 1.403212000 | 0.481646000 |
| Η | 0.181177000 | 1.855789000 | 0.982969000 |
| Η | 1.903070000 | 1.455688000 | 1.185531000 |
| С | 1.351301000 | 2.227432000 | -0.760315000 |
| Η | 1.459423000 | 3.280256000 | -0.463909000 |
| Η | 2.283672000 | 1.930059000 | -1.260403000 |
| Н | 0.524409000 | 2.176197000 | -1.483066000 |

| S | -1.369669000 | 0.003504000 | -0.909714000 |
|---|--------------|--------------|--------------|
| 0 | -1.890975000 | -1.228379000 | -0.311443000 |
| 0 | -1.875081000 | 1.256604000 | -0.343376000 |
| С | 1.393600000 | -2.150820000 | -0.901699000 |
| Η | 0.347316000 | -2.326519000 | -1.189503000 |
| Η | 2.040195000 | -2.504480000 | -1.717195000 |
| Η | 1.610466000 | -2.767983000 | -0.019007000 |
| С | -0.439335000 | -0.077836000 | 2.479102000 |
| Η | 0.023777000 | 0.801161000 | 2.948946000 |
| Η | -1.386573000 | 0.226348000 | 2.013713000 |
| Н | -0.672083000 | -0.794929000 | 3.278872000 |

CO_2

E = -188.6055421

| С | 0.000000000 | 0.000000000 | 0.000000000 |
|---|-------------|-------------|--------------|
| 0 | 0.000000000 | 0.000000000 | 1.158564000 |
| 0 | 0.000000000 | 0.000000000 | -1.158564000 |

N,N-dimethylethanamine-CO₂

E = -402.4128763

| N | -0.622630000 | -0.523658000 | -0.107019000 |
|---|--------------|--------------|--------------|
| С | -0.950914000 | -1.136575000 | 1.162145000 |
| Η | -0.192169000 | -1.891976000 | 1.413045000 |
| Η | -0.948242000 | -0.391742000 | 1.969756000 |
| Η | -1.946832000 | -1.633574000 | 1.154708000 |
| С | -0.479591000 | -1.515824000 | -1.150809000 |

| Η | -0.148655000 | -1.036490000 | -2.084888000 |
|---|--------------|--------------|--------------|
| Η | 0.279128000 | -2.258842000 | -0.861682000 |
| Η | -1.426649000 | -2.059823000 | -1.364103000 |
| С | -1.533368000 | 0.539757000 | -0.502142000 |
| Η | -1.212524000 | 0.887406000 | -1.497099000 |
| Η | -2.569359000 | 0.146708000 | -0.627024000 |
| С | -1.551849000 | 1.731598000 | 0.443507000 |
| Η | -2.175390000 | 2.532597000 | 0.020486000 |
| Η | -1.970097000 | 1.480992000 | 1.429424000 |
| Η | -0.536013000 | 2.128534000 | 0.584533000 |
| 0 | 2.274733000 | -0.539431000 | 0.716153000 |
| 0 | 1.778985000 | 1.278647000 | -0.634358000 |
| С | 1.978300000 | 0.355726000 | 0.040234000 |

N-ethyl-N-methylethanamine-CO₂

E = -441.7295949

| Ν | 0.623480000 | 0.165891000 | 0.392572000 |
|---|--------------|--------------|--------------|
| С | 1.559368000 | -0.437127000 | -0.545599000 |
| Η | 1.333812000 | -0.041476000 | -1.545874000 |
| Η | 2.604996000 | -0.126768000 | -0.312110000 |
| С | 0.822824000 | -0.278566000 | 1.756736000 |
| Η | 0.067118000 | 0.182458000 | 2.408718000 |
| Η | 0.702009000 | -1.367366000 | 1.837453000 |
| Η | 1.829164000 | -0.013969000 | 2.151638000 |
| С | 0.615598000 | 1.621826000 | 0.329861000 |
| Н | -0.109345000 | 1.969981000 | 1.083047000 |

| Η | 1.604036000 | 2.036287000 | 0.636839000 |
|---|--------------|--------------|--------------|
| С | 0.227722000 | 2.203289000 | -1.023377000 |
| Η | 0.062713000 | 3.286198000 | -0.926452000 |
| Η | 1.007858000 | 2.062006000 | -1.785113000 |
| Η | -0.699433000 | 1.747779000 | -1.402083000 |
| 0 | -1.833435000 | -0.858845000 | -1.181768000 |
| 0 | -2.322166000 | -0.125660000 | 0.962664000 |
| С | 1.482988000 | -1.955934000 | -0.615801000 |
| Η | 0.450428000 | -2.280623000 | -0.810432000 |
| Η | 2.115452000 | -2.321043000 | -1.438019000 |
| Η | 1.835838000 | -2.442189000 | 0.305278000 |
| С | -2.029200000 | -0.482899000 | -0.101497000 |

triethylamine- CO_2

E = -481.0449936

| Ν | 0.646150000 | 0.013029000 | 0.150437000 |
|---|-------------|--------------|--------------|
| С | 1.504982000 | -0.663615000 | -0.816467000 |
| Н | 1.438850000 | -0.113543000 | -1.765426000 |
| Н | 2.572439000 | -0.604757000 | -0.497201000 |
| С | 0.622152000 | -0.689941000 | 1.428241000 |
| Н | 0.266311000 | -1.712369000 | 1.239317000 |
| Н | 1.653775000 | -0.786701000 | 1.841497000 |
| С | 1.006392000 | 1.418269000 | 0.306052000 |
| Н | 0.279985000 | 1.875041000 | 0.992297000 |
| Н | 2.005351000 | 1.514814000 | 0.793120000 |
| С | 0.995494000 | 2.230773000 | -0.982083000 |

| Η | 1.086419000 | 3.299611000 | -0.739795000 |
|---|--------------|--------------|--------------|
| Η | 1.827773000 | 1.979612000 | -1.655068000 |
| Η | 0.048739000 | 2.086228000 | -1.522277000 |
| 0 | -2.136410000 | -1.204477000 | -0.572610000 |
| 0 | -2.188778000 | 1.112774000 | -0.571073000 |
| С | 1.144324000 | -2.117368000 | -1.094444000 |
| Η | 0.072181000 | -2.218486000 | -1.315127000 |
| Η | 1.713052000 | -2.474684000 | -1.965359000 |
| Η | 1.386382000 | -2.784587000 | -0.254802000 |
| С | -0.277559000 | -0.073224000 | 2.491459000 |
| Η | 0.114840000 | 0.874003000 | 2.887619000 |
| Η | -1.290073000 | 0.115256000 | 2.103726000 |
| Η | -0.368201000 | -0.767382000 | 3.339491000 |
| С | -2.117346000 | -0.044835000 | -0.563693000 |

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