

Supporting Information

Design of High-Performance Antimony / MXene Hybrid Electrodes for Sodium-Ion Batteries

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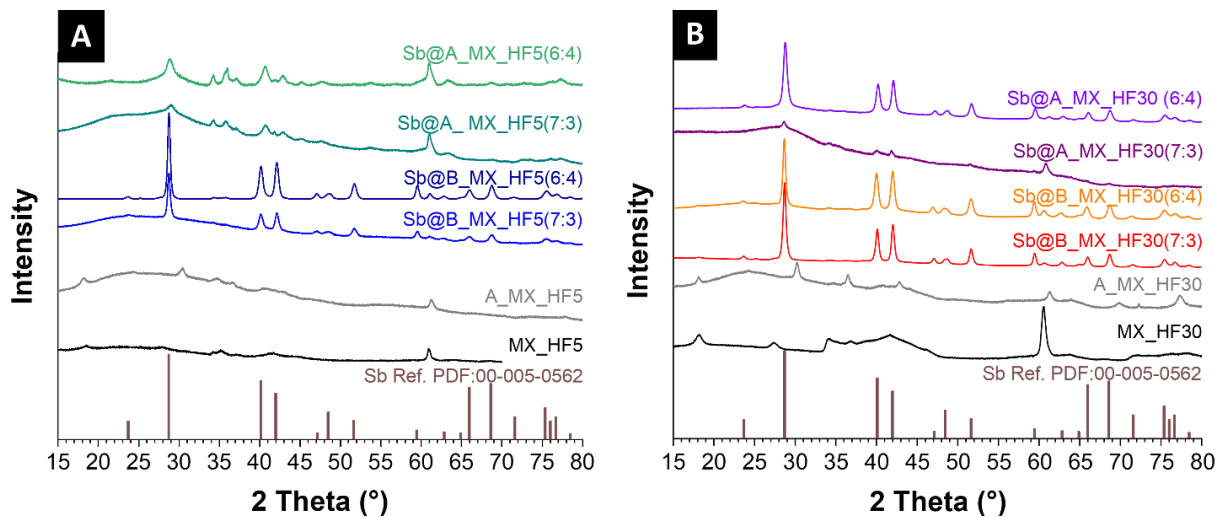


Figure S1: X-ray diffraction patterns of different Sb@Ti₃C₂T_x hybrid materials (+pure Ti₃C₂T_x and expanded Ti₃C₂T_x) by using the MXene etched with 5 mass% HF (A) and 30 mass% HF (B).

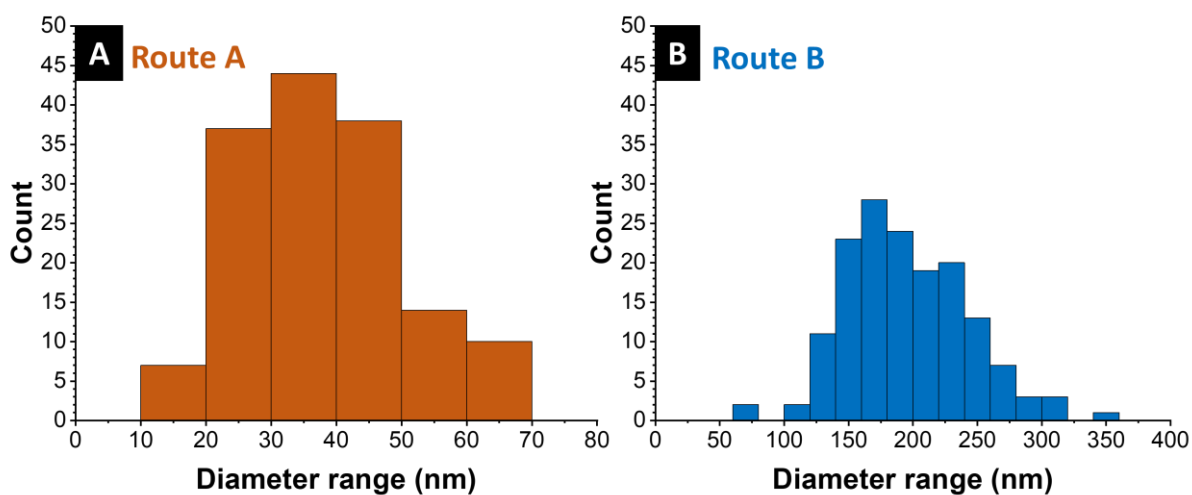


Figure S2: Particle size distribution derived from image analysis of scanning electron micrographs of 150 antimony particles synthesized via synthesis Route A (A) and via synthesis Route B (B).

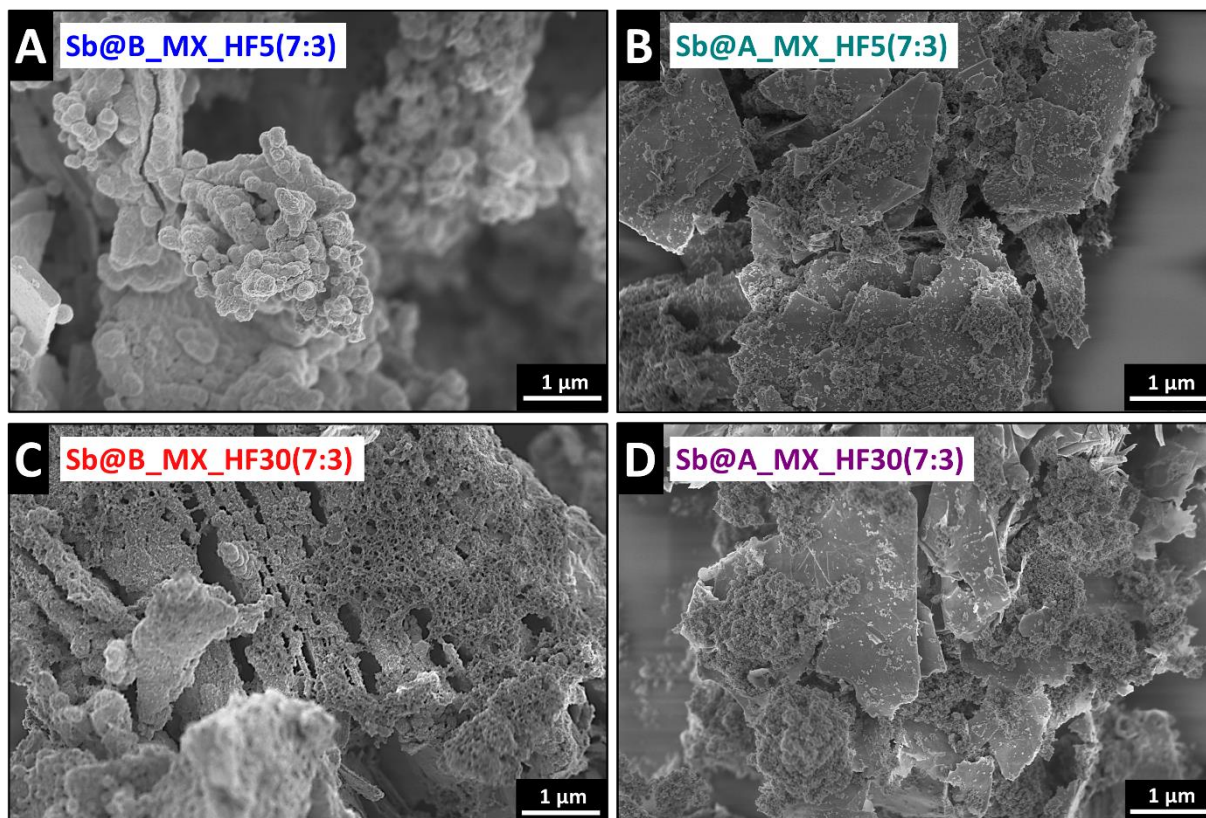


Figure S3: Scanning electron micrographs of (A) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF) hybrid with a composition of 7:3 (B) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF) hybrid with a composition of 7:3 (C) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF) hybrid with a composition of 7:3 (D) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF) hybrid with a composition of 7:3.

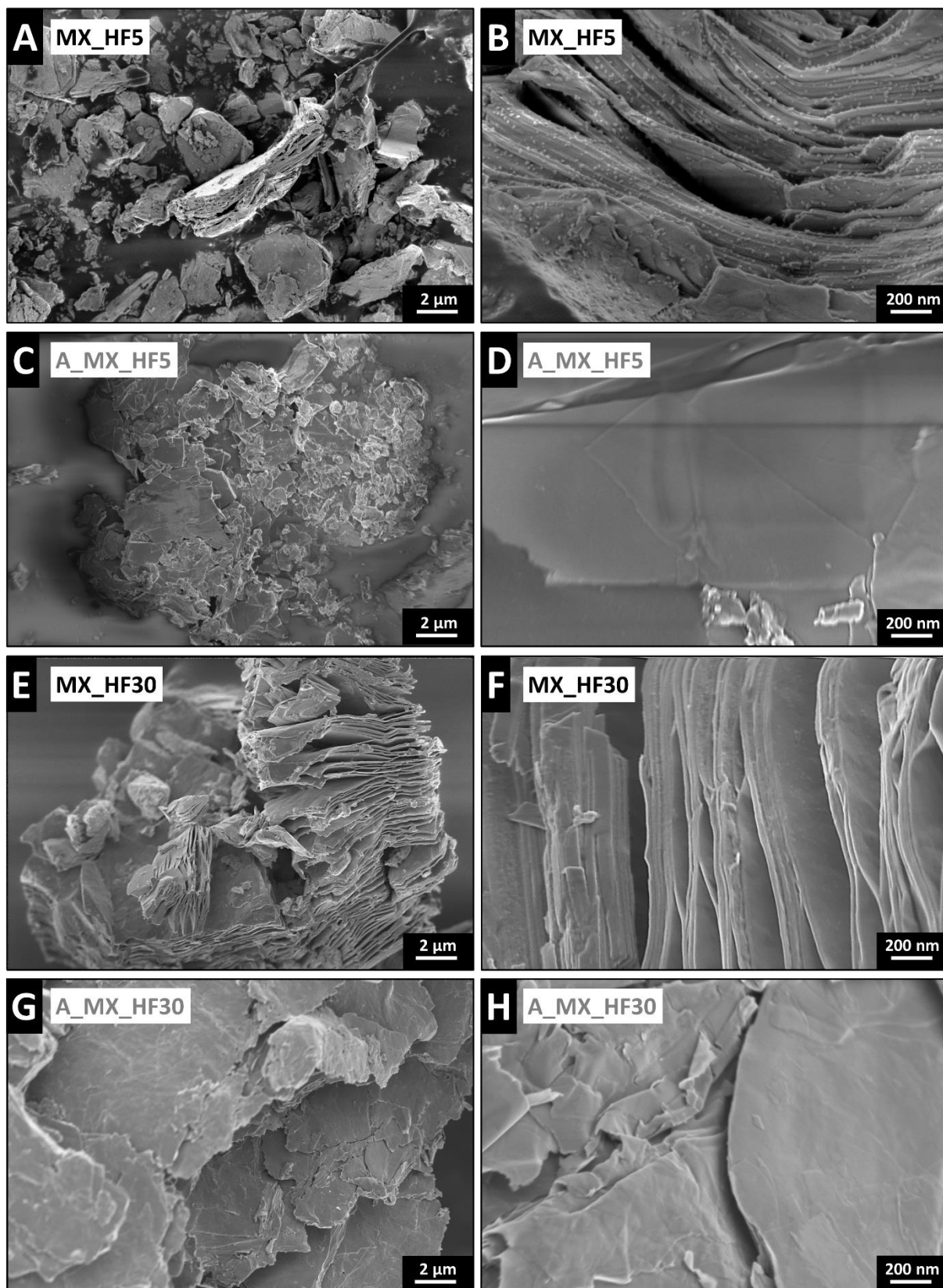


Figure S4: Material characterization of the different MXenes. Scanning electron micrographs of (A-B) $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF), (C-D) with TMAOH expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF), (E-F) $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF), (G-H) with TMAOH expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF).

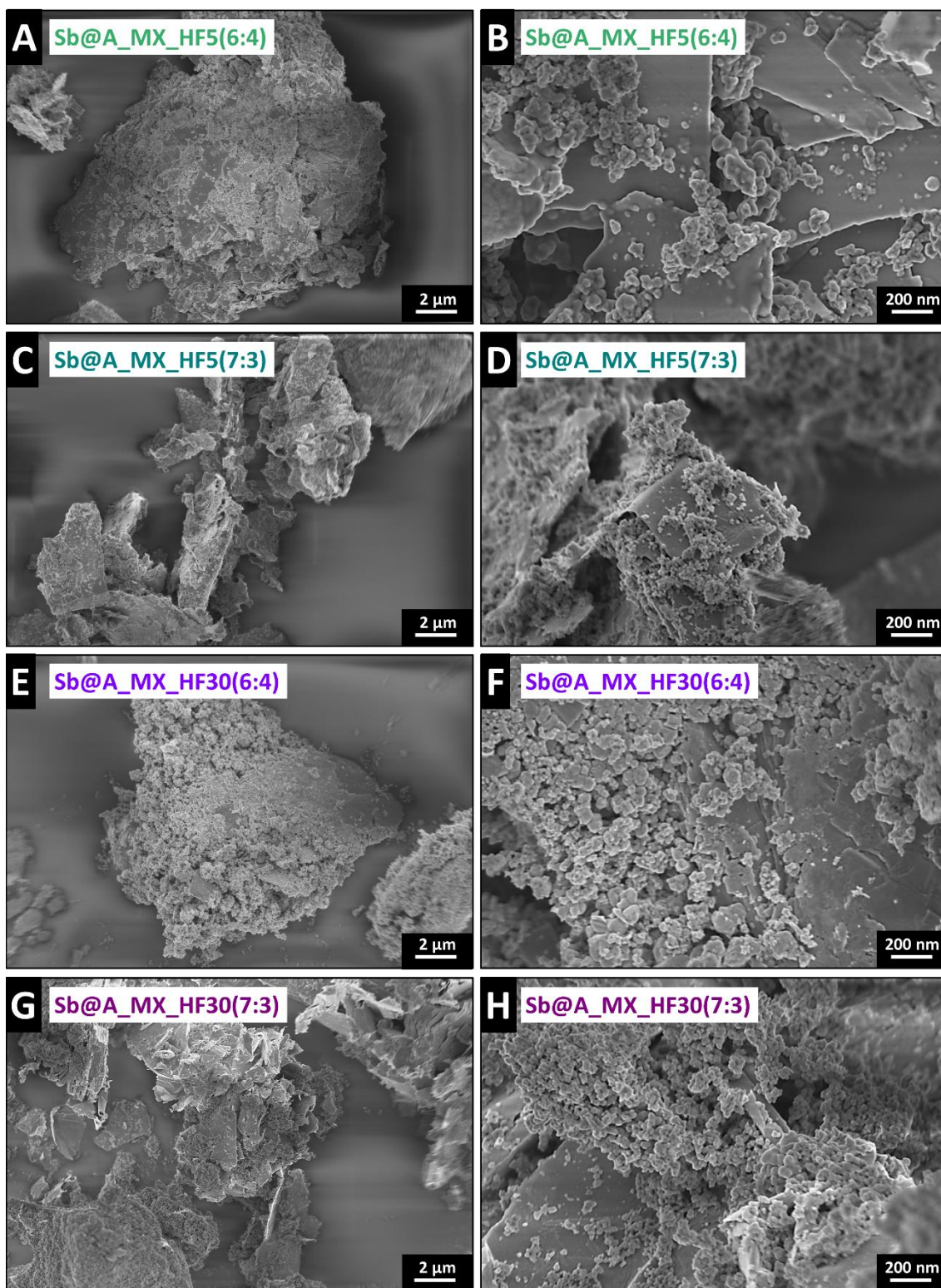


Figure S5: Material characterization of the different with TMAOH expanded MXene antimony hybrids. Scanning electron micrographs of (A-B) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF) hybrid with a composition of 6:4, (C-D) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5 mass% HF) hybrid with a composition of 7:3, (E-F) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF) hybrid with a composition of 6:4, (G-H) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (30 mass% HF) hybrid with a composition of 7:3.

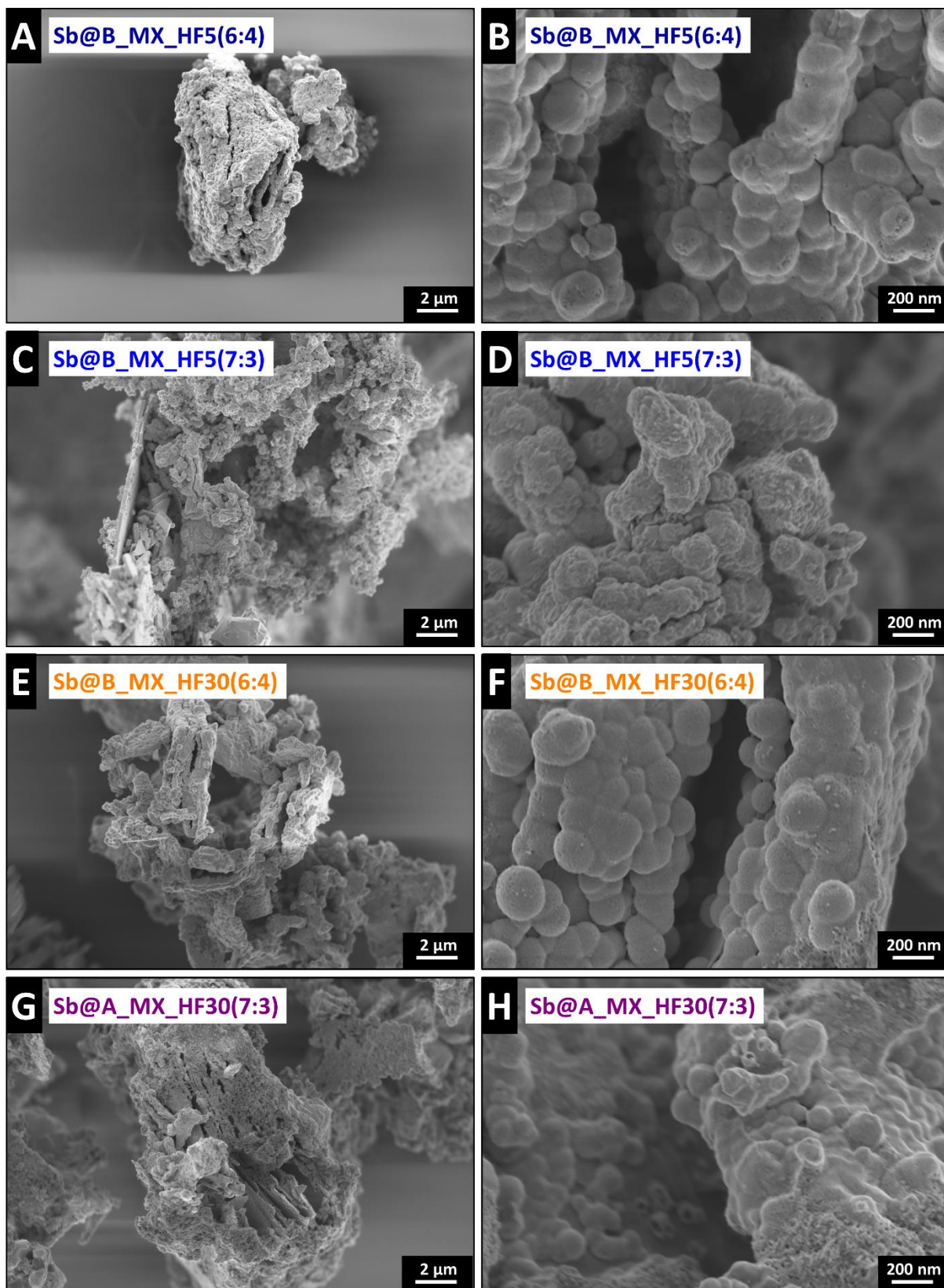


Figure S6: Material characterization of the different MXenes hybrids. Scanning electron micrographs of (A-B) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF) hybrid with a composition of 6:4, (C-D) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF) hybrid with a composition of 7:3, (E-F) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (30% mass% HF) hybrid with a composition of 6:4, (G-H) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (30% mass% HF) hybrid with a composition of 7:3.

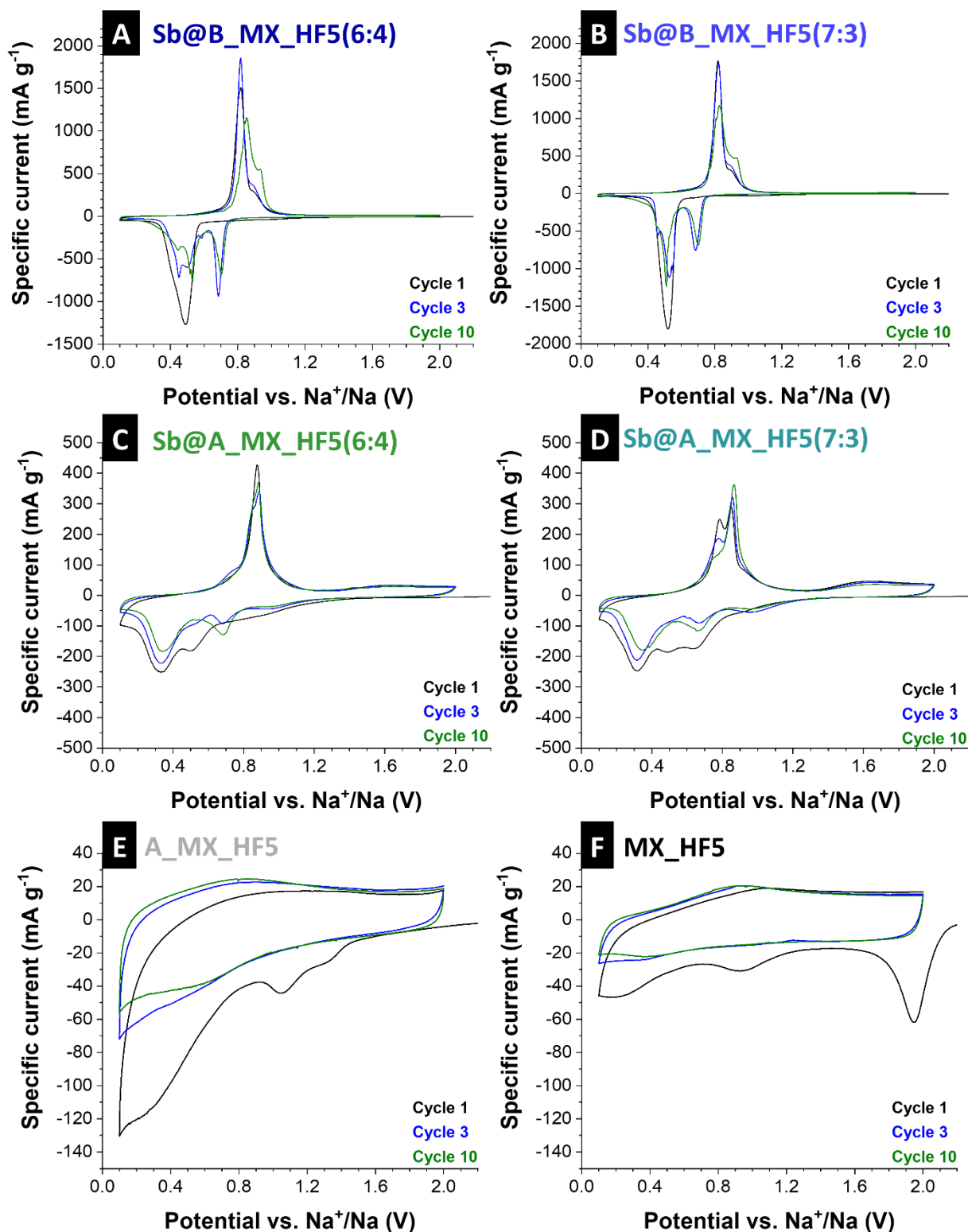


Figure S7: Cyclic voltammograms recorded at 0.1 mV s^{-1} in the potential range of 0.1-2.0 V vs. Na^+/Na for (A) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF) hybrid with a composition of 6:4, (B) antimony $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF) hybrid with a composition of 7:3, (C) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5% HF) hybrid with a composition of 6:4, (D) antimony expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF) hybrid with a composition of 7:3, (E) v $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF), (F) not expanded $\text{Ti}_3\text{C}_2\text{T}_z$ (5% mass% HF).

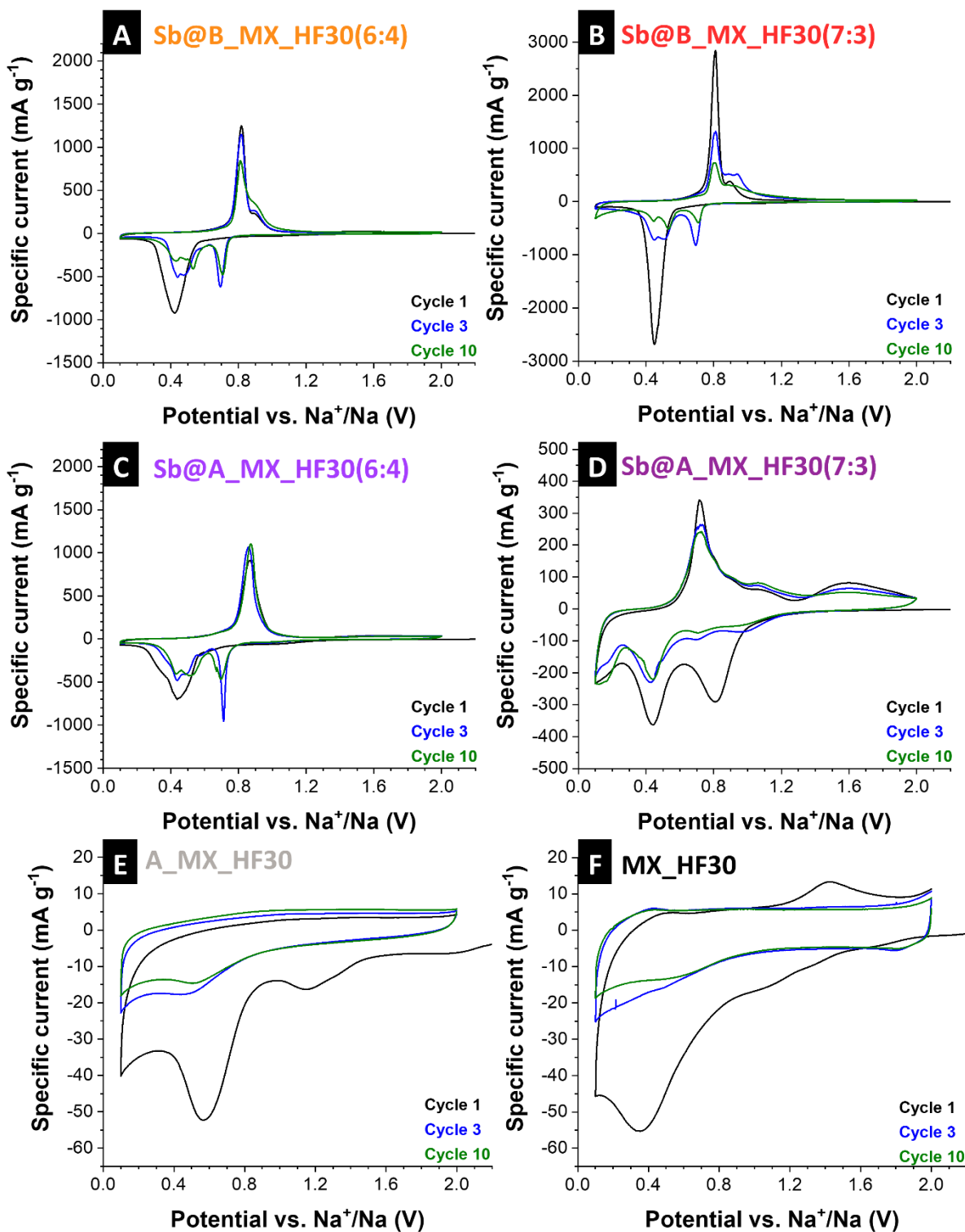


Figure S8: Cyclic voltammograms recorded at 0.1 mV s⁻¹ in the potential range of 0.1-2.0 V vs. Na⁺/Na for (A) antimony Ti₃C₂T_z (30% mass% HF) hybrid with a composition of 6:4, (B) antimony Ti₃C₂T_z (30% mass% HF) hybrid with a composition of 7:3, (C) antimony expanded Ti₃C₂T_z (30% mass% HF) hybrid with a composition of 6:4, (D) antimony expanded Ti₃C₂T_z (30% mass% HF) hybrid with a composition of 7:3, (E) expanded Ti₃C₂T_z (30% mass% HF), (F) not expanded Ti₃C₂T_z (30% mass% HF).

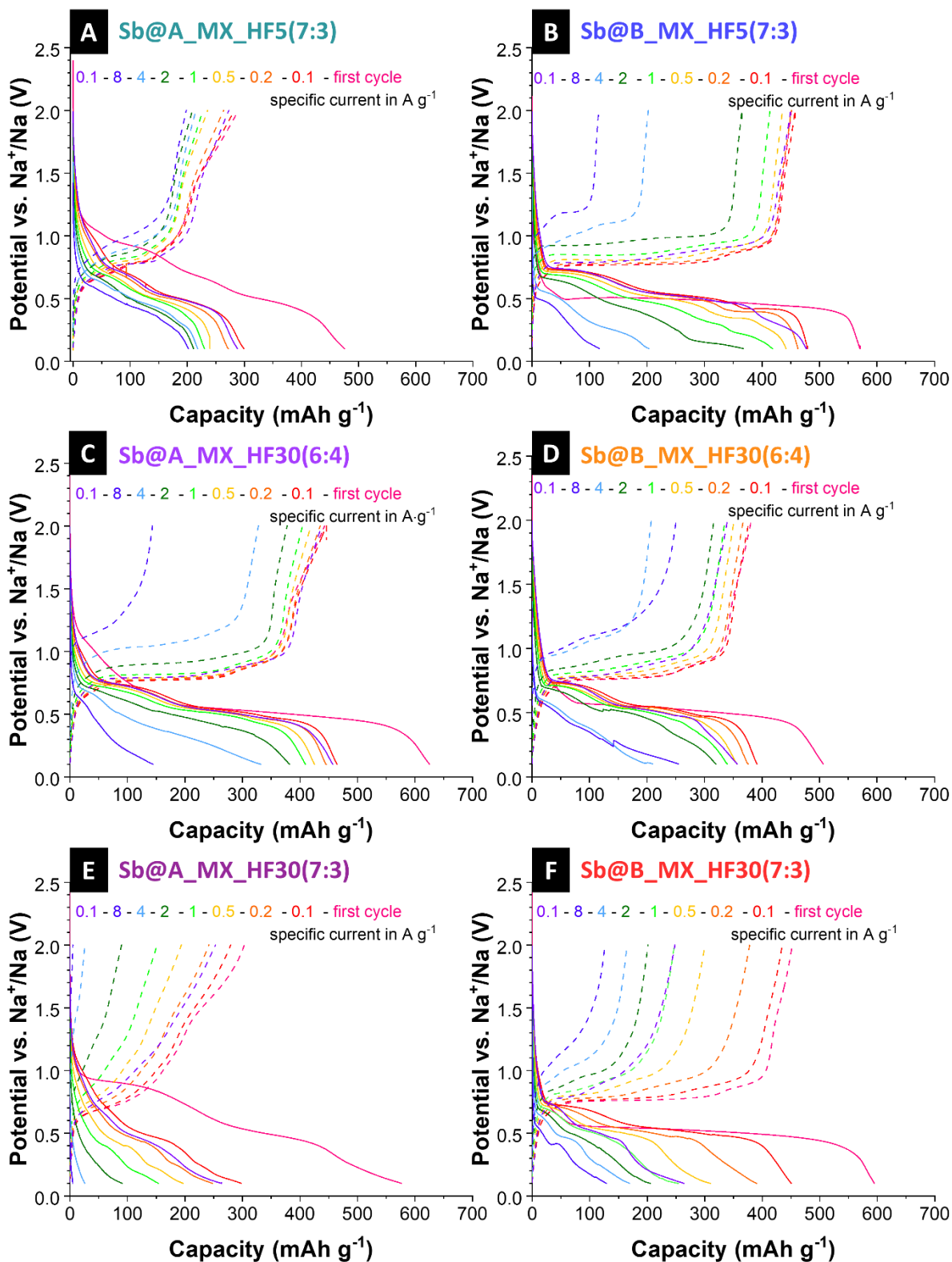


Figure S9: Galvanostatic charge and discharge profiles at different applied specific currents of 0.1-8 A g⁻¹ between 1.0 V and 3.0 V vs. Na⁺/Na of (A) Sb@A_MX_HF5(7:3), (B) Sb@B_MX_HF5(7:3), (C) Sb@A_MX_HF30(6:4), (D) Sb@B_MX_HF30(6:4), (E) Sb@A_MX_HF30(7:3), and (F) Sb@B_MX_HF30(7:3).

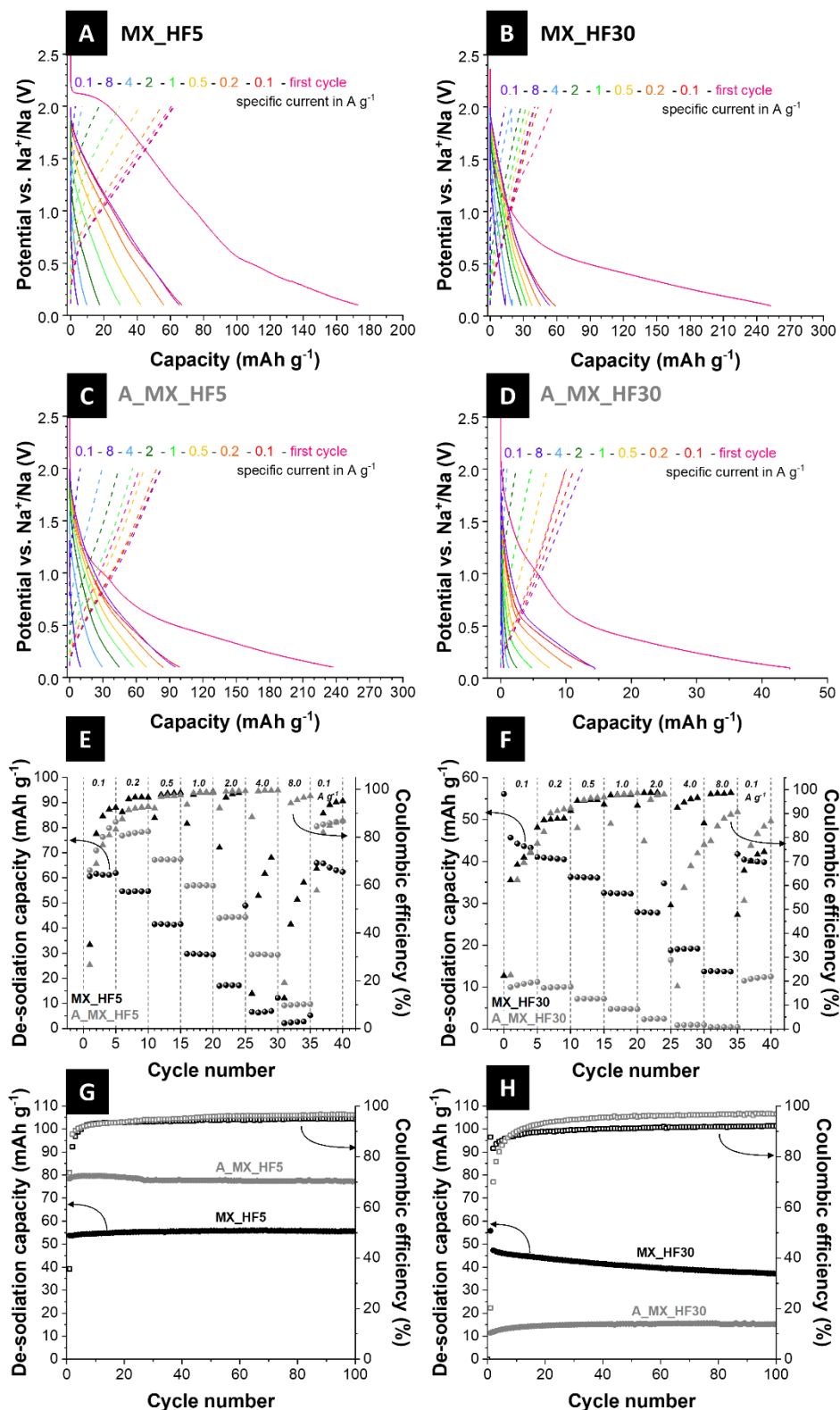


Figure S10: Electrochemical characterization of the non-expanded and expanded $\text{Ti}_3\text{C}_2\text{T}_x$, etched with 5 mass% HF. Galvanostatic charge and discharge profiles at rates of 0.1-8 A g^{-1} (1.0-3.0 V vs. Na^+/Na) of (A) MX_HF5, (B) MX_HF30, (C) A_MX_HF5, and (D) A_MX_HF30. Rate performance using galvanostatic charge/discharge cycling and Coulombic efficiency values at different rates for (E) MX_HF5 and A_MX_HF5, (F) MX_HF30 and A_MX_HF30. Galvanostatic charge/discharge cycling stability and Coulombic efficiency values at a specific current of 0.1 A g^{-1} for (G) MX_HF5 and A_MX_HF5, (H) MX_HF30 and A_MX_HF30.

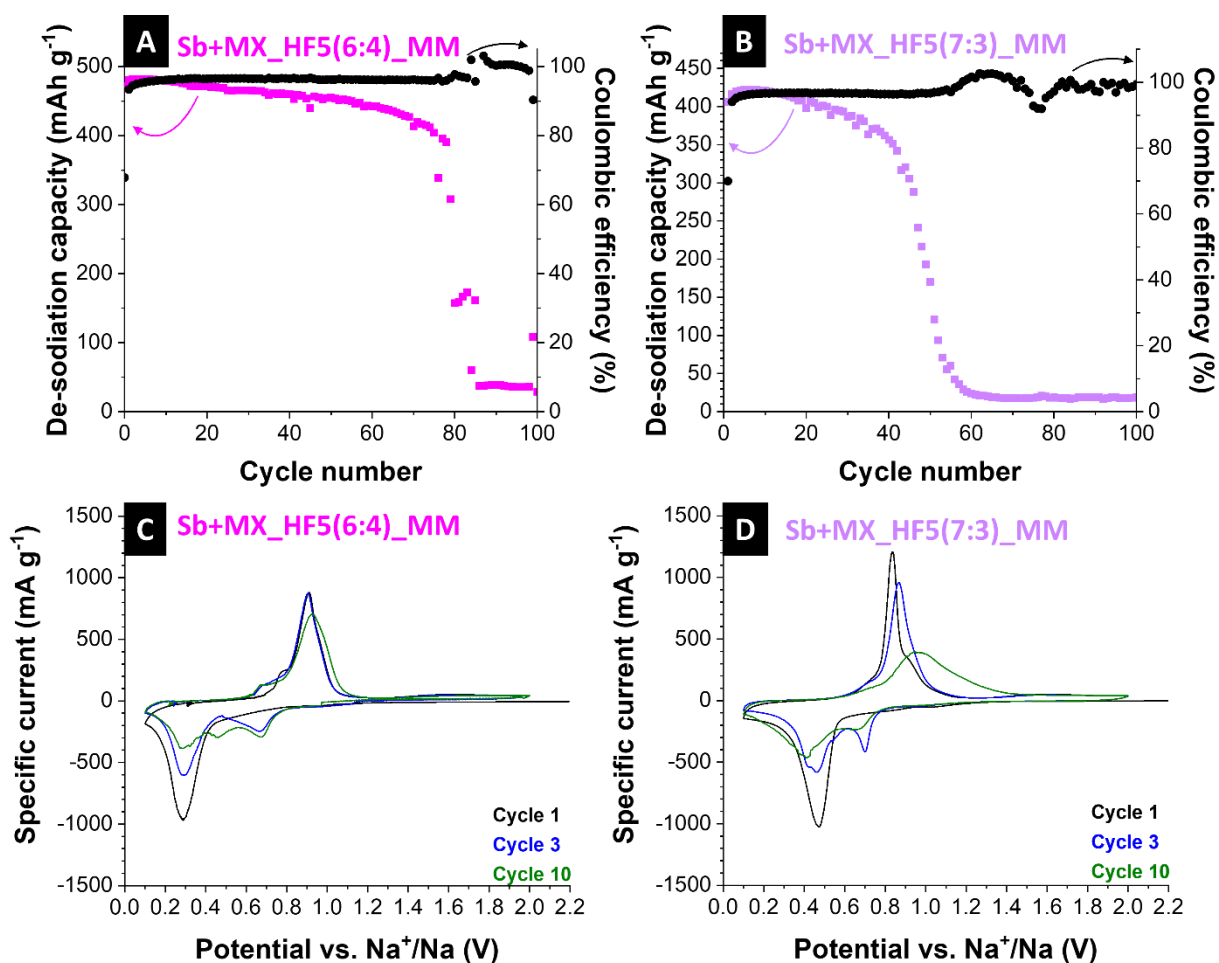


Figure S11: Galvanostatic charge/discharge cycling performance electrochemical stability with corresponding Coulombic efficiency values at a specific current of 0.1 A g^{-1} for (A) Sb+MX_HF5(6:4)_MM, and (B) Sb+MX_HF5(7:3)_MM. Cyclic voltammograms recorded at a rate of 0.1 mV s^{-1} in the potential range of 0.1-2.0 V vs. Na⁺/Na for (C) Sb+MX_HF5(6:4)_MM, and (D) Sb+MX_HF5(7:3)_MM.

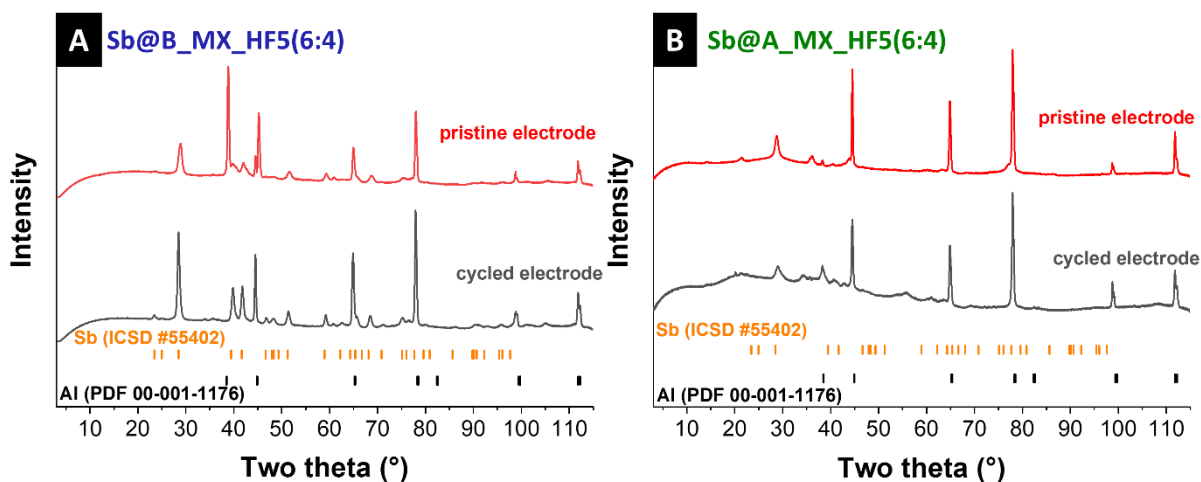


Figure S12: X-ray diffraction pattern of (A) pristine and post mortem Sb@B_MX_HF5(6:4) electrode (B) pristine and cycled Sb@A_MX_HF5(6:4) electrode. The diffraction patterns were recorded with the setup XRD-2 (see Experimental).

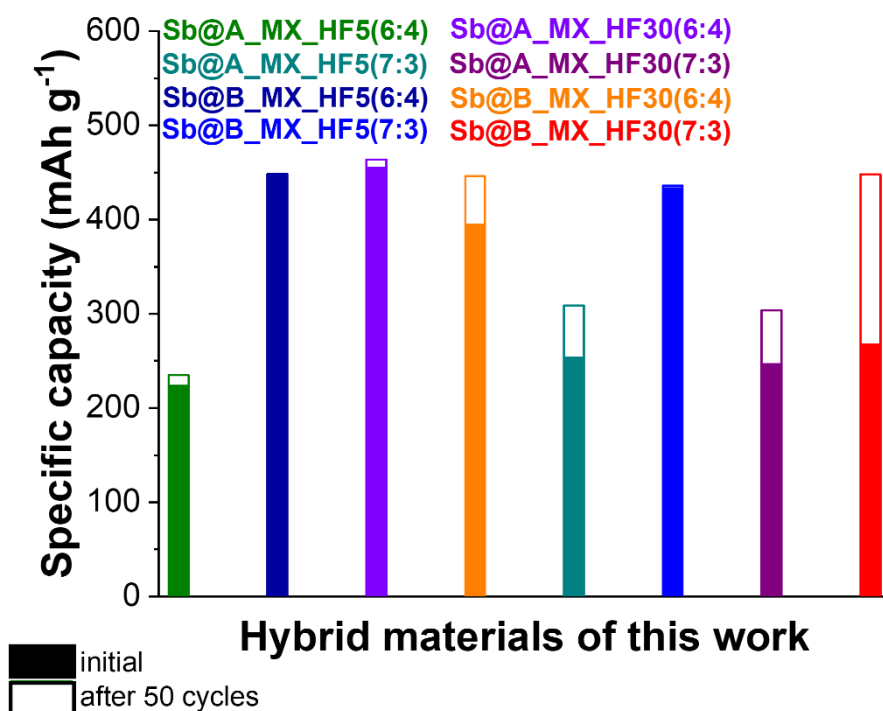


Figure S13: Graphical illustration and overview of initial specific capacities as well as values after 50 cycles of cycling at a specific current of 0.1 A g⁻¹ of all hybrid antimony MXene hybrid materials produced in this work.