

Ultra-robust, Highly Proton-conductive Polymer Carbon Dots Membranes through Bioinspired Complexation

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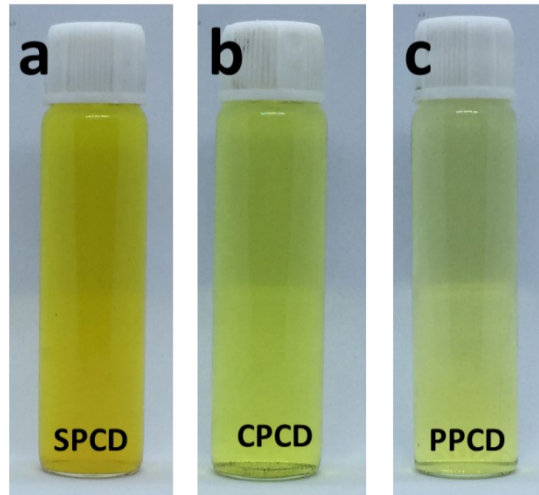


Fig. S1 Photograph of the PCD solution (SPCD(a), CPCD(b), PPCD(c)).

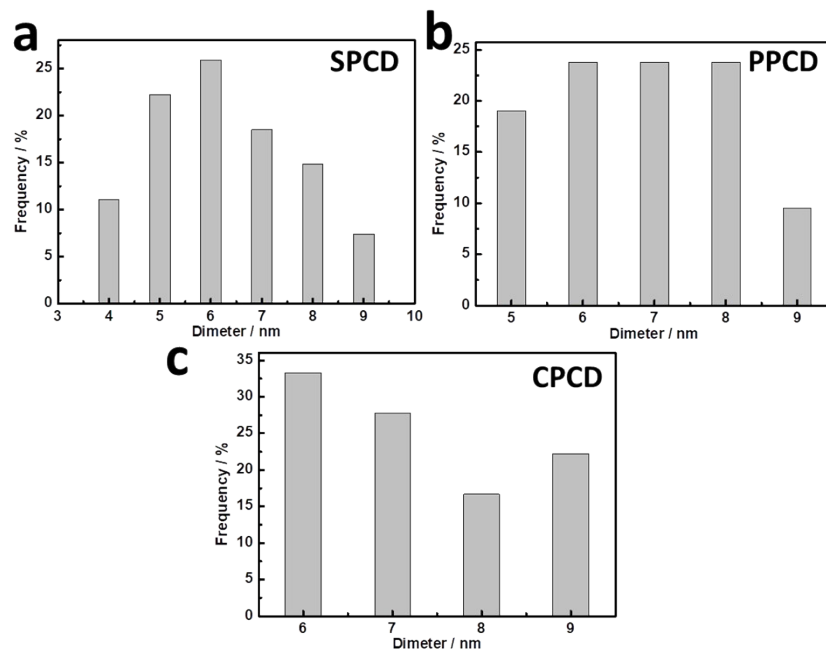


Fig. S2 Frequency distribution histogram of the size of the PCDs(SPCD(a), CPCD(b), PPCD(c)).

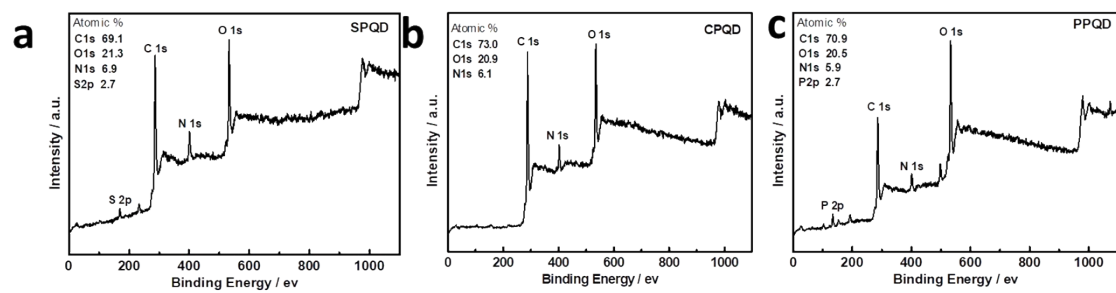


Fig. S3 XPS spectra and the element content of the PCDs (SPQD(a), CPQD(b), PPQD(c)).

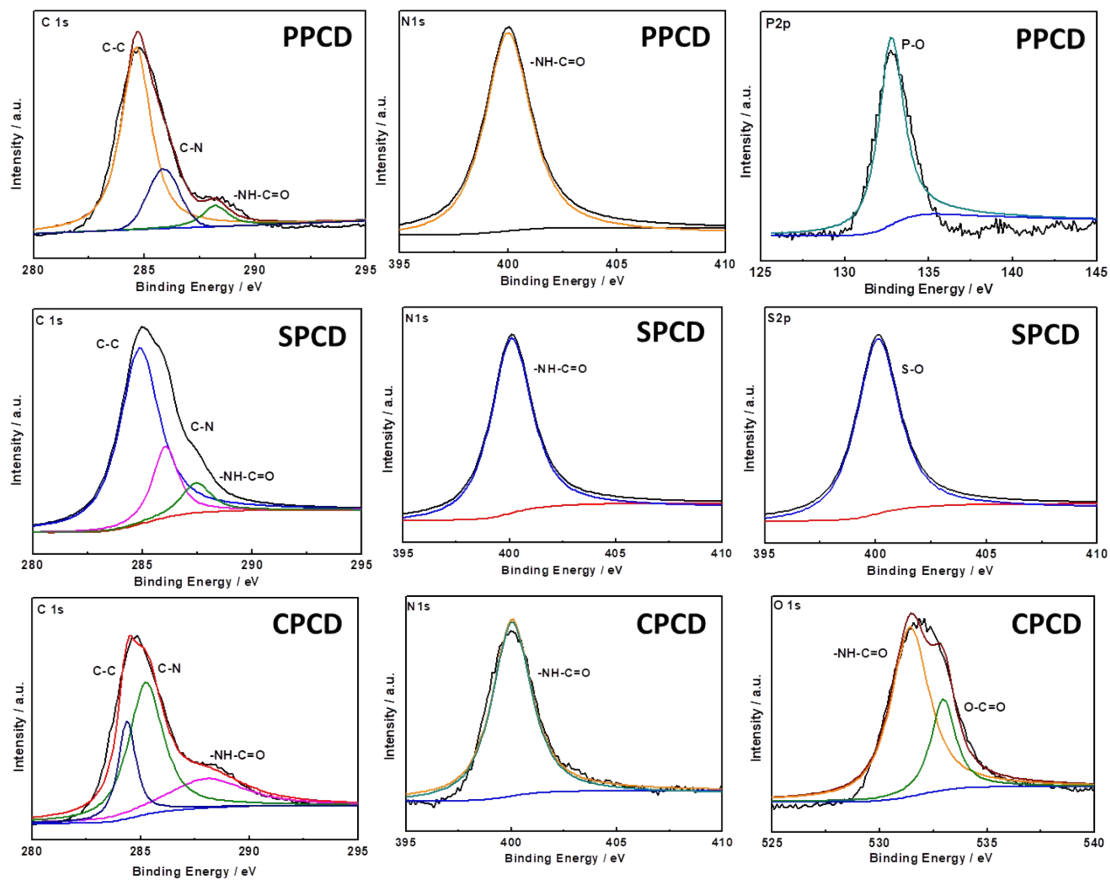


Fig.S4 High-resolution XPS spectra of PCDs

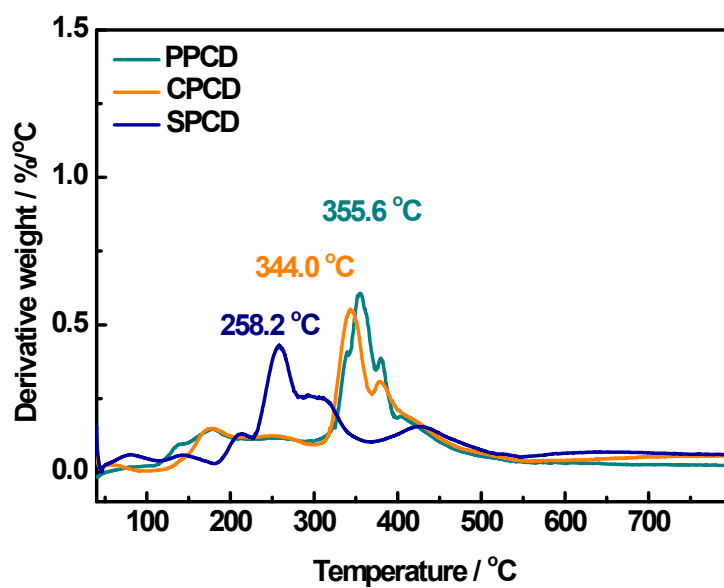


Fig. S5 DTG curves of the PCDs.

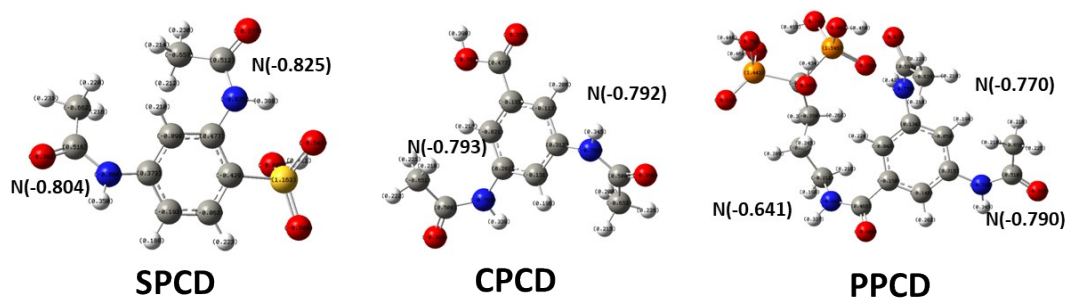


Fig. S6 The charge distribution of N atom in PCD.

The molecular electrostatic potential (MEP) values of the functional groups were calculated in the GAUSSIAN 09W package at the DFT/B3LYP level. The electron density on N atom was related to the charge distribution and higher negative charge indicates higher electron density, and vice versa.

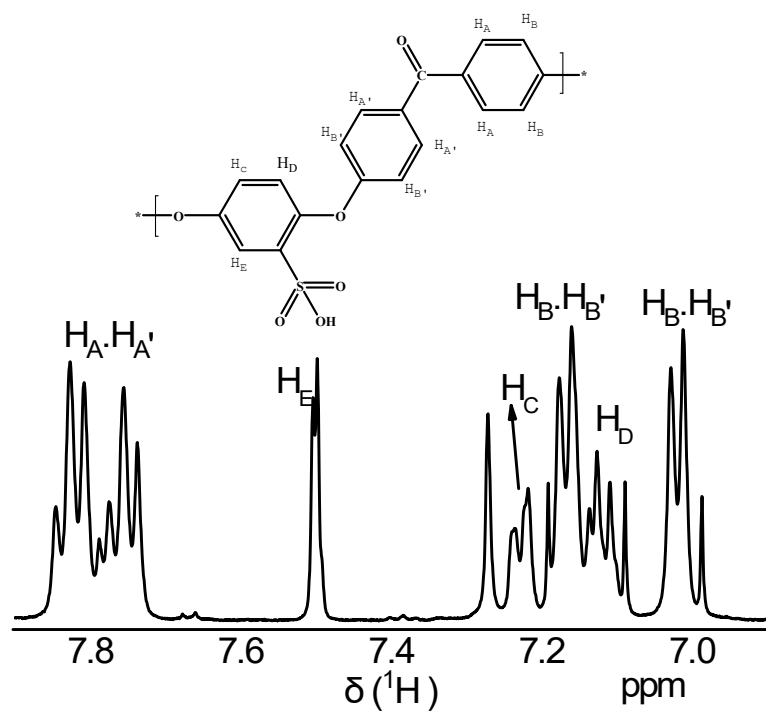


Fig. S7 ¹H-NMR spectrum of the SPEEK.

Table S1 The weight loss of PCDs in PCDMs.

Sample	Weight loss (wt.%)
SPCDM-50	1.5
PPCDM-50	1.9
CPCDM-50	2.4
SPCDM-55	3.7
PPCDM-55	4.2
CPCDM-55	4.7

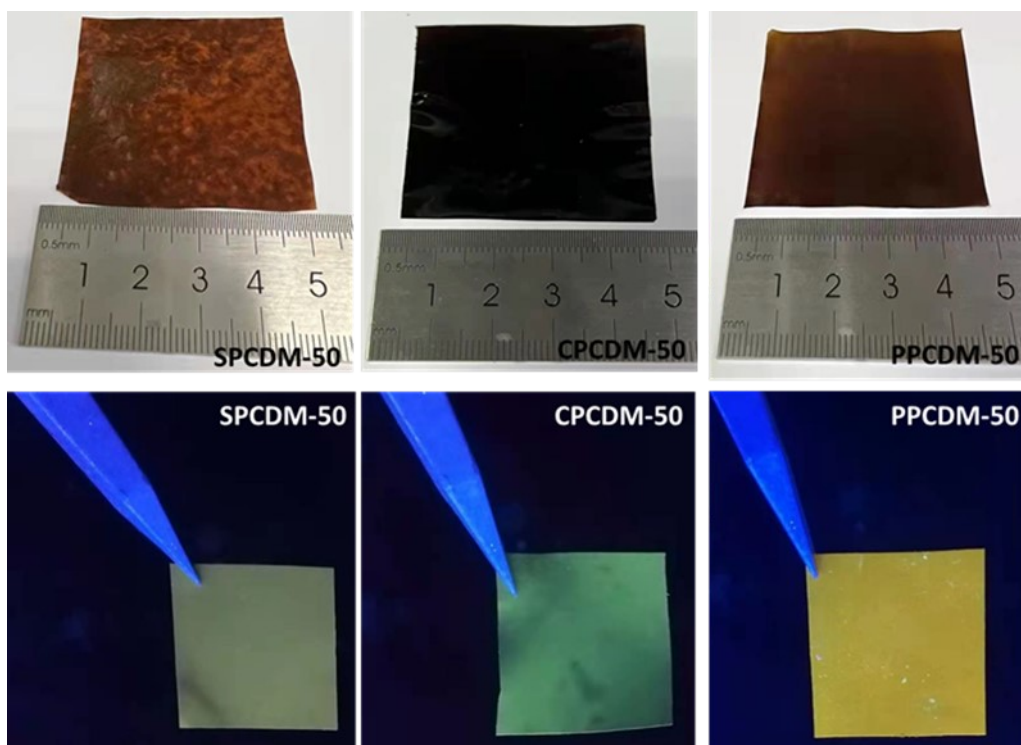


Fig. S8 Fluorescence phenomenon of the PCDMs under UV(365 nm).

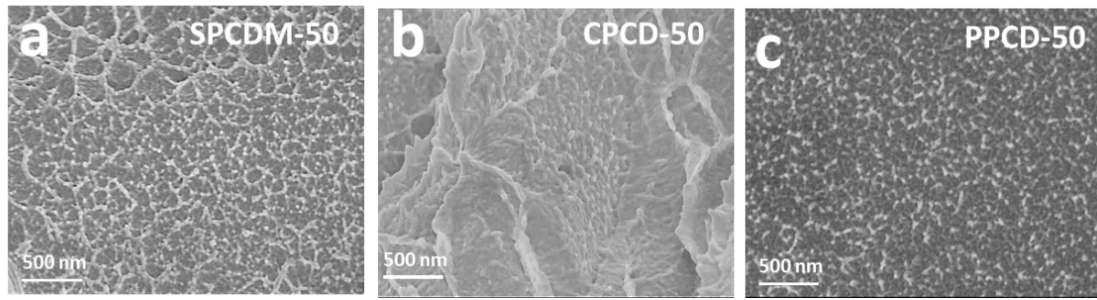


Fig. S9 SEM images of the cross section of the PCDMs.

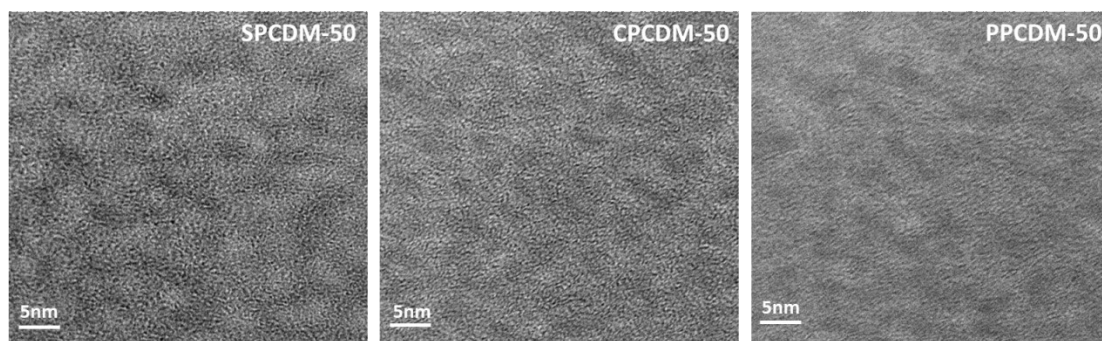


Fig. S10 TEM images of the PCDM.

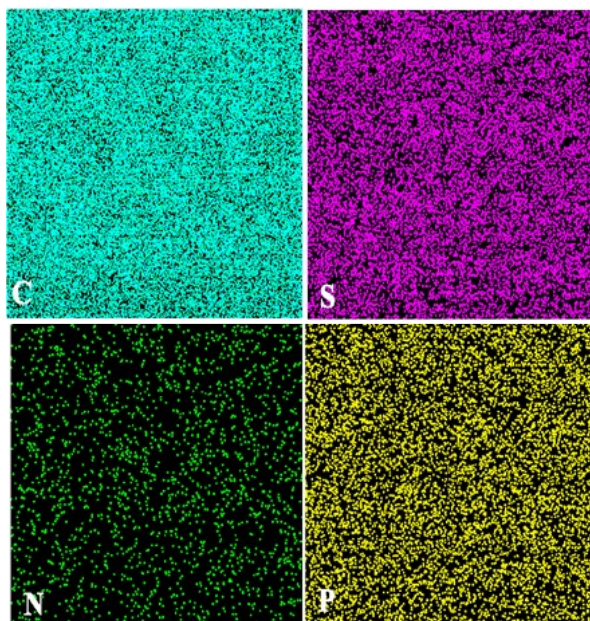


Fig. S11 Element distribution in PCDMs.

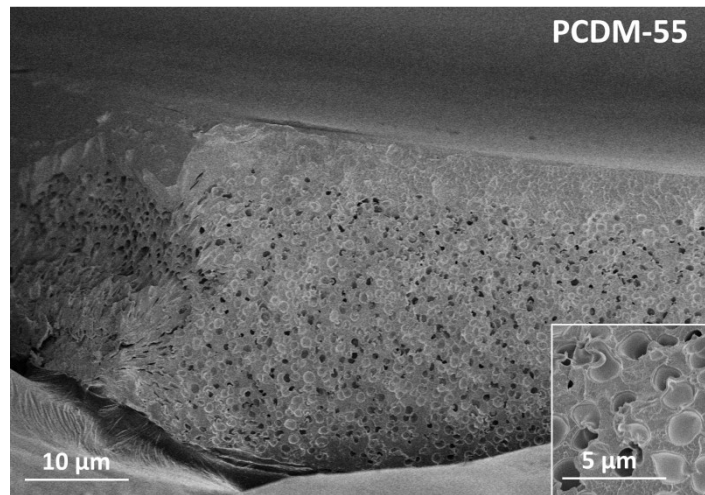


Fig. S12 SEM image of cross section the PCDM-55.

Table S2. Proton conductivity, IEC values, and mechanical stability of PCDMs and membranes in literature

Membrane	IEC values (mmol g ⁻¹)	Stress (MPa)	Proton conductivity(mS cm ⁻¹ (Temperature/ °C)	Reference
PTFE48/SPEEK65	1.44	27.2	183 (90)	1
Aq-PSUT (30%)	1.02	20	180(80)	2
Su-CNTs/Nafion	0.98	20.6	216 (99)	3
SPPEK-SF-7.5	1.62	31.4	226.7(90)	4
Nafion/CeO2-TiC(0.5 wt%)	0.87	46.9	90(90)	5
PTPBSH-90	2.44	51.2	119(90)	6
Nafion-ZrGdNR-1.5	0.909	22.5	164(80)	7
M-BC	1.75	25	90.2 (80)	8
GO-g-SPEEK/Nafion-33	1.45	8.6	219(90)	9
Me-m-SPEEK	1.78	55.4	173(100)	10
SPP-QP	2.6	34	220 (80)	11
SPEEK/S-UiO-66@GO-10	1.7	53.5	268(70)	12
SPAFT5	1.77	44.3	105(80)	13
SF-Nafion-1	0.932	13	130(80)	14
MIV	0.56	6.25	32(80)	15
SPAES-8-33	1.66	20.51	331(80)	16
6F-PAEK-SP22	1.77	29.5	148(100)	17
SPI-5-POSS	1.69	40.8	132(80)	18
C-SPAES-40	2.12	71.5	150.4(80)	19
PEEK-PEMs	3.08	14	431(80)	20
SPEKEBI-4	2.14	11	233(80)	21
SPAEK-100	1.49	45.34	159(100)	22
SPAES50	1.84	54.86	211.6 (80)	23
SPEEK/PANTs-3#-10	1.869	57.25	330(70)	24
poly(SHS-ddm)	3.13	32.49	154 (80)	25
SPAES 50	2.01	9.2	181(80)	26
SPES-3	1.51	27	131.4(100)	27
sPBT-PE57.5	2.55	49.2	130(80)	28
SPP-co-PAEK(5/1)	2.38	46	233(60)	29
PEM-OH	1.6	48	100(80)	30
Am-SPEEK-T	1.53	53.8	140 (80)	31
2-SPAES-80	1.73	53	258(80)	32
SPAEK	1.73	49	130(120)	33
PPCDM-50	2.53	56	264(80)	This work

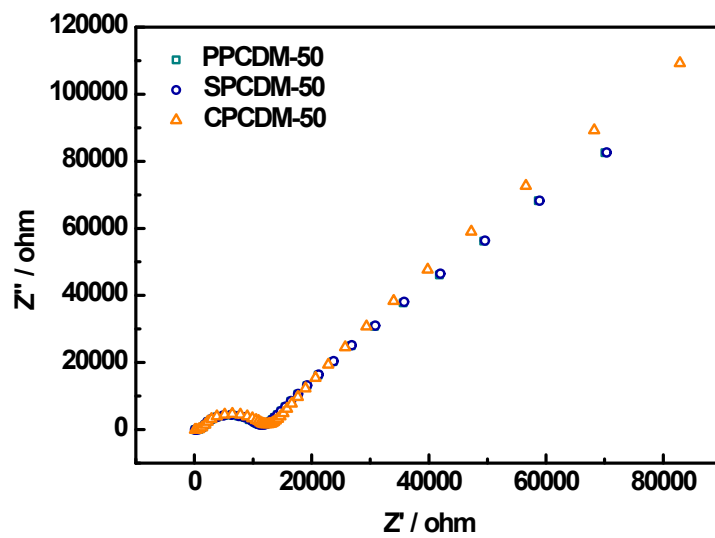


Fig.S13 Nyquist plot of the PCDMs at 30 °C under 100% RH.

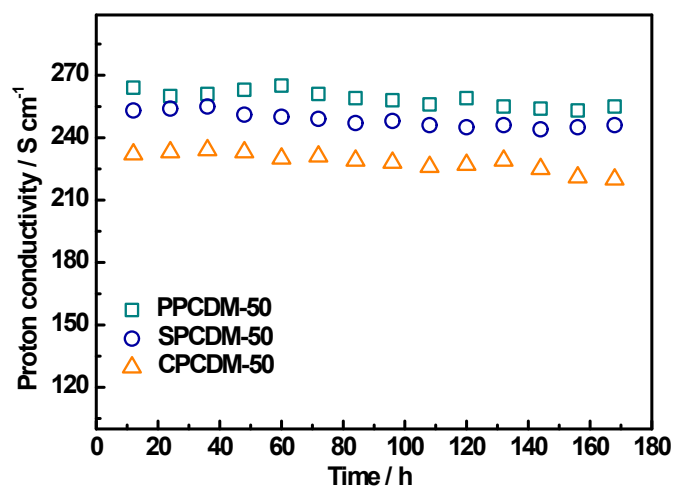


Fig. S14 Time dependent proton conductivity of PCDMs at 80 °C under 100% RH

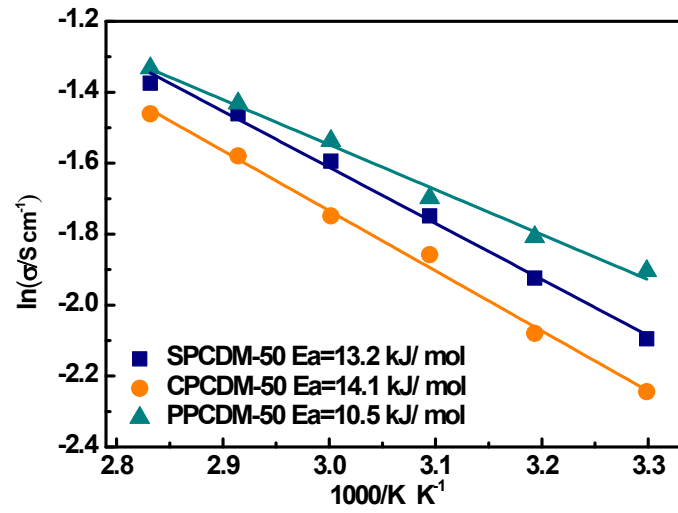


Fig. S15 Arrhenius plots of the proton conductivities of the membranes.

Table S3 H₂ permeability of the PCDM before and after water treatment

Sample	H ₂ permeability / Barrier (without water treatment)	H ₂ permeability / Barrier (water treatment)
SPCDM-50	22.33	9.67
CPCDM-50	59.97	22.2
PPCDM-50	35.43	13.11
SPCDM-55	1557.7	37.88

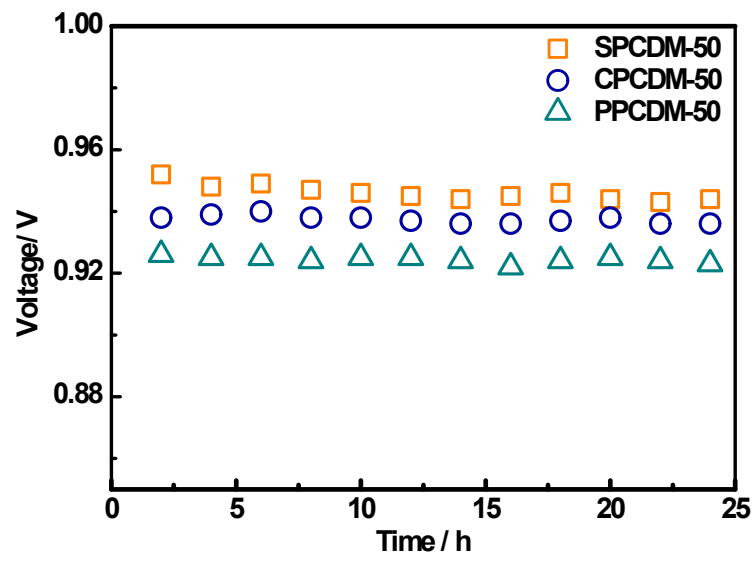


Fig. S16 Durability of the MEAs at 40 mA/cm².

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