

Evaluation of polymorphism and charge transport in a BaO–CaO–Ta₂O₅ perovskite phase diagram using TOF-neutron and synchrotron X-ray diffraction, the bond-valence method and impedance spectroscopy

Kalpana Singh,^{‡b} Hua Yang,^{‡ac} Zheyu Zhang,^{‡b} Maxim Avdeev,^{dg} Ashfia Huq,^e

Dong-Ying Wu,^f Jey-Jau Lee,^f Wang Hay Kan ^{*ac} and Venkataraman Thangadurai ^{*b}

^aDongguan Neutron Science Center, No. 1 Zhongziyuan Road, Dongguan, Guangdong, 523803, China.

^bDepartment of Chemistry, University of Calgary, Calgary, Alberta, T2N 1N4, Canada.

^cInstitute of High Energy Physics, Chinese Academy of Sciences, Beijing 100049, China

^dAustralian Nuclear Science and Technology Organisation (ANSTO), New Illawarra Road, Lucas Heights, New South Wales 2234, Australia

^eNeutron Scattering Science Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA

^fNational Synchrotron Radiation Research Center, No. 101 Hsin-Ann Rd, Hsinchu 300, Taiwan

^gSchool of Chemistry, University of Sydney, Sydney, NSW 2006, Australia

Electronic Supporting Information (ESI)

Table S1. Rietveld refinement of Ba₃Ca_{1+x}Ta_{2-x}O_{9-3x/2} (0 ≤ x ≤ 0.36) prepared at 1000 °C, using monochromatic synchrotron X-ray diffraction data (λ = 1.321 Å).

Ba₃CaTa₂O₉ (<i>P</i>-3<i>m</i>1), 1000 °C, 12 h, wRp = 2.84%, Rp = 2.2%, χ² = 1.299, a = b = 5.917(2) Å, c = 7.2482(2) Å				
Atom	Position	Wyckoff position	Occupancy	100*Uiso
Ba	0, 0, 0	1 <i>a</i>	1	0.373(4)
Ba	0.3333, 0.6667, 0.6672(1)	2 <i>d</i>	1	0.373(4)
Ta	0.3333, 0.6667, 0.1723(2)	2 <i>d</i>	0.9132(4)	0.683(8)
Ca	0.3333, 0.6667, 0.1723(2)	2 <i>d</i>	0.0868(4)	0.683(8)
Ta	0, 0, 0.5	1 <i>b</i>	0.1736(3)	0.683(7)
Ca	0, 0, 0.5	1 <i>b</i>	0.8264(3)	0.683(7)
O	0.5, 0, 0	3 <i>e</i>	1	0.521(2)
O	0.1354(3), 0.8646(3), 0.3322(3)	6 <i>i</i>	1	0.521(2)
Ba₃Ca_{1.09}Ta_{1.91}O_{8.865} (<i>P</i>-3<i>m</i>1), 1000 °C, 12 h, wRp = 2.2%, Rp = 1.81%, χ² = 0.4629, a = b = 5.922(3) Å, c = 7.2563(3) Å				
Ba	0, 0, 0	1 <i>a</i>	1	0.785(8)
Ba	0.3333, 0.6667, 0.663(2)	2 <i>d</i>	1	0.785(8)
Ta	0.3333, 0.6667, 0.1794(3)	2 <i>d</i>	0.7671(2)	0.873(4)
Ca	0.3333, 0.6667, 0.1794(5)	2 <i>d</i>	0.2329(2)	0.873(4)
Ta	0, 0, 0.5	1 <i>b</i>	0.3757(1)	1.376(6)
Ca	0, 0, 0.5	1 <i>b</i>	0.6243(1)	1.376(6)

O	0.5, 0, 0	3e	0.985	1.02 (2)
O	0.1507(7), 0.8493(7), 0.2942(7)	6i	0.985	1.02(2)
Ba₃Ca_{1.18}Ta_{1.82}O_{8.73} (<i>P-3m1</i>), 1000 °C, 12 h, wRp = 4.77%, Rp = 3.91%, χ^2 = 3.153, a = b = 5.919(8) Å, c = 7.2520(8) Å				
Ba	0, 0, 0	1a	1	0.651(7)
Ba	0.3333, 0.6667, 0.663(3)	2d	1	0.651(7)
Ta	0.3333, 0.6667, 0.1693(4)	2d	0.8065(5)	0.332(4)
Ca	0.3333, 0.6667, 0.1693(4)	2d	0.1935(5)	0.332(4)
Ta	0, 0, 0.5	1b	0.2071(3)	0.303(3)
Ca	0, 0, 0.5	1b	0.7929(3)	0.303(3)
O	0.5, 0, 0	3e	0.97	0.231(8)
O	0.1685(6), 0.8315(6), 0.3433(6)	6i	0.97	0.231(8)
Ba₃Ca_{1.27}Ta_{1.73}O_{8.595} (<i>P-3m1</i>), 1000 °C, 12 h, wRp = 3.98%, Rp = 2.66% %, χ^2 = 1.665, a = b = 5.909(7) Å, c = 7.2218(7) Å				
Ba	0, 0, 0	1a	1	0.913(7)
Ba	0.3333, 0.6667, 0.663(2)	2d	1	0.913(7)
Ta	0.3333, 0.6667, 0.1703(4)	2d	0.8004(5)	0.785(9)
Ca	0.3333, 0.6667, 0.1703(4)	2d	0.1996(5)	0.785(9)
Ta	0, 0, 0.5	1b	0.1292(3)	0.578(6)
Ca	0, 0, 0.5	1b	0.8708(3)	0.578(6)
O	0.5, 0, 0	3e	0.955	0.879(8)
O	0.1324(6), 0.8675(6), 0.3681(6)	6i	0.955	0.879(8)
Ba₃Ca_{1.36}Ta_{1.64}O_{8.415} (<i>P-3m1</i>), 1000 °C, 12 h, wRp = 5.90%, Rp = 3.67% %, χ^2 = 6.702, a = b = 5.925(9) Å, c = 7.2554(9) Å				
Ba	0, 0, 0	1a	1	0.865(8)
Ba	0.3333, 0.6667, 0.663(9)	2d	1	0.865(8)
Ta	0.3333, 0.6667, 0.1703(8)	2d	0.7245(9)	0.779(4)
Ca	0.3333, 0.6667, 0.1703(8)	2d	0.2755(9)	0.779(4)
Ta	0, 0, 0.5	1b	0.2309(4)	0.497(3)
Ca	0, 0, 0.5	1b	0.7691(4)	0.497(3)
O	0.5, 0, 0	3e	0.935	0.971(7)
O	0.1223(6), 0.8777(6), 0.4281(6)	6i	0.935	0.971(7)

Table S2. Rietveld refinement of $\text{Ba}_3\text{Ca}_{1+x}\text{Ta}_{2-x}\text{O}_{9-3x/2}$ ($0 \leq x \leq 0.36$) prepared at 1100 °C, using TOF-Neutron diffraction data.

$\text{Ba}_3\text{CaTa}_2\text{O}_9$ (<i>P-3m1</i>), 1100 °C, 12 h, wRp = 2.97%, Rp = 9.34%, $\chi^2 = 1.275$, a = b = 5.90141(8) Å, c = 7.2628(2) Å				
Atom	Position	Wyckoff position	Occupancy	100*Uiso
Ba	0, 0, 0	1a	1	0.734 (6)
Ba	0.3333, 0.6667, 0.6672(3)	2d	1	0.734(6)
Ta	0.3333, 0.6667, 0.1684(6)	2d	0.9132	0.255(6)
Ca	0.3333, 0.6667, 0.1684(6)	2d	0.0868	0.255(6)
Ca	0, 0, 0.5	1b	0.8264	0.807(5)
Ta	0, 0, 0.5	1b	0.1736	0.807(5)
O	0.5, 0, 0	3e	1	1.385(2)
O	0.1748(3), 0.8252(6), 0.3159(4)	6i	1	1.385(2)
$\text{Ba}_3\text{Ca}_{1.09}\text{Ta}_{1.91}\text{O}_{8.865}$ (<i>P-3m1</i>), 1100 °C, 12 h, wRp = 2.94%, Rp = 5.65%, $\chi^2 = 4.599$, a = b = 5.926(5) Å, c = 7.3028(7) Å				
Ba	0, 0, 0	1a	1	1.309(8)
Ba	0.3333, 0.6667, 0.663(2)	2d	1	1.309(8)
Ta	0.3333, 0.6667, 0.1740(3)	2d	0.955(4)	0.368(4)
Ca	0.3333, 0.6667, 0.1740(3)	2d	0.045(4)	0.368(4)
Ca	0, 0, 0.5	1b	1	0.598(6)
O	0.5, 0, 0	3e	0.985	0.781(2)
O	0.1571(4), 0.8429(2), 0.3491(7)	6i	0.985	0.781(2)
$\text{Ba}_3\text{Ca}_{1.18}\text{Ta}_{1.82}\text{O}_{8.73}$ (<i>P-3m1</i>), 1100 °C, 12 h, wRp = 2.78%, Rp = 4.51%, $\chi^2 = 5.207$, a = b = 5.929(5) Å, c = 7.2505(4) Å				
Ba	0, 0, 0	1a	1	1.019(8)
Ba	0.3333, 0.6667, 0.663(8)	2d	1	1.019(8)
Ta	0.3333, 0.6667, 0.1740(9)	2d	0.91(4)	0.638(7)
Ca	0.3333, 0.6667, 0.1740(9)	2d	0.09(4)	0.638(7)
Ca	0, 0, 0.5	1b	1	0.252(3)
O	0.5, 0, 0	3e	0.97	1.71(2)
O	0.1601(5), 0.8398(8), 0.3410(9)	6i	0.97	1.71(2)
$\text{Ba}_3\text{Ca}_{1.27}\text{Ta}_{1.73}\text{O}_{8.595}$ (<i>P-3m1</i>), 1100 °C, 12 h, wRp = 2.51%, Rp = 4.58%, $\chi^2 = 4.532$, a = b = 5.930(3) Å, c = 7.257(6) Å				
Ba	0, 0, 0	1a	1	1.232(1)
Ba	0.3333, 0.6667, 0.663(8)	2d	1	1.232 (1)
Ta	0.3333, 0.6667, 0.1740(9)	2d	0.865(5)	0.638(7)
Ca	0.3333, 0.6667, 0.1740(9)	2d	0.135(5)	0.638(7)
Ca	0, 0, 0.5	1b	1	0.243(6)
O	0.5, 0, 0	3e	0.955	0.389(4)
O	0.1580(3), 0.8420(7), 0.3422(9)	6i	0.955	0.389(4)
$\text{Ba}_3\text{Ca}_{1.09}\text{Ta}_{1.91}\text{O}_{8.865}$ (<i>Fm-3m</i>), 1100 °C, 12 h, wRp = 2.94%, Rp = 5.65%, $\chi^2 = 4.599$, a = b = c = 8.3946(2) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.859(7)
Ta	0.5, 0.5, 0.5	4b	0.8788(6)	0.579(4)
Ca	0.5, 0.5, 0.5	4b	0.1212(6)	0.579(4)
Ta	0, 0, 0	4a	0.3942(5)	0.331(2)

Ca	0, 0, 0	4a	0.6058(5)	0.331(2)
O	0.2544(5), 0, 0	24e	0.985	1.229(5)
Ba₃Ca_{1.18}Ta_{1.82}O_{8.73} (<i>Fm-3m</i>), 1100 °C, 12 h, wRp = 2.78%, Rp = 4.51%, χ^2 = 5.207, a = b = c = 8.444(6) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.502(8)
Ta	0.5, 0.5, 0.5	4b	1	0.674(4)
Ta	0, 0, 0	4a	0.213(2)	0.583(3)
Ca	0, 0, 0	4a	0.787(2)	0.583(3)
O	0.2724(5), 0, 0	24e	0.98	0.424(8)
Ba₃Ca_{1.27}Ta_{1.73}O_{8.595} (<i>Fm-3m</i>), 1100 °C, 12 h, wRp = 2.51%, Rp = 4.58%, χ^2 = 4.532, a = b = c = 8.5105(2) Å				
Ba	0.25, 0.25, 0.25	8c	1	1.204(7)
Ta	0.5, 0.5, 0.5	4b	0.153(4)	1.56(9)
Ca	0.5, 0.5, 0.5	4b	0.847(4)	1.56(9)
Ta	0, 0, 0	4a	1	0.434(5)
O	0.2294(5), 0, 0	24e	0.955	0.280(4)

Table S3. Rietveld refinement of $\text{Ba}_3\text{Ca}_{1+x}\text{Ta}_{2-x}\text{O}_{9-3x/2}$ ($0 \leq x \leq 0.36$) prepared at 1400 °C, using monochromatic synchrotron X-ray diffraction data ($\lambda = 1.321 \text{ \AA}$).

$\text{Ba}_3\text{CaTa}_2\text{O}_9$ (<i>P-3m1</i>), 1400 °C, 12 h, wRp = 5.49%, Rp = 3.83%, $\chi^2 = 0.4524$, a = b = 5.919(3) Å, c = 7.292(3) Å				
Atom	Position	Wyckoff position	Occupancy	100*Uiso
Ba	0, 0, 0	1a	1	0.443(6)
Ba	0.3333, 0.6667, 0.6672(3)	2d	1	0.443(6)
Ta	0.3333, 0.6667, 0.1784(2)	2d	1	1.19(4)
Ca	0, 0, 0.5	1b	1	1.19(4)
O	0.5, 0, 0	3e	1	0.427(2)
O	0.1354(3), 0.8646(3), 0.3322(3)	6i	1	0.427(2)
$\text{Ba}_3\text{Ca}_{1.09}\text{Ta}_{1.91}\text{O}_{8.865}$ (<i>P-3m1</i>), 1400 °C, 12 h, wRp = 5.5%, Rp = 3.5%, $\chi^2 = 0.6333$, a = b = 5.904(3) Å, c = 7.288(5) Å				
Ba	0, 0, 0	1a	1	0.307(8)
Ba	0.3333, 0.6667, 0.663(2)	2d	1	0.307(8)
Ta	0.3333, 0.6667, 0.1827(3)	2d	0.955(3)	0.451(4)
Ca	0.3333, 0.6667, 0.1827(3)	2d	0.045(3)	0.451(4)
Ca	0, 0, 0.5	1b	1	0.501(6)
O	0.5, 0, 0	3e	0.985	1.15 (2)
O	0.1461(2), 0.8539(8), 0.3281(2)	6i	0.985	1.15(2)
$\text{Ba}_3\text{Ca}_{1.18}\text{Ta}_{1.82}\text{O}_{8.73}$ (<i>P-3m1</i>), 1400 °C, 12 h, wRp = 4.94%, Rp = 3.81%, $\chi^2 = 3.322$, a = b = 5.942(5) Å, c = 7.2597(4) Å				
Ba	0, 0, 0	1a	1	0.471(7)
Ba	0.3333, 0.6667, 0.663(3)	2d	1	0.471(7)
Ta	0.3333, 0.6667, 0.1751(4)	2d	0.91	0.547(7)
Ca	0.3333, 0.6667, 0.1751(4)	2d	0.09	0.547(7)
Ca	0, 0, 0.5	1b	1	0.398(3)
O	0.5, 0, 0	3e	0.97	1.71(2)
O	0.1176(6), 0.8823(4), 0.3507(3)	6i	0.97	1.71(2)
$\text{Ba}_3\text{Ca}_{1.27}\text{Ta}_{1.73}\text{O}_{8.595}$ (<i>P-3m1</i>), 1400 °C, 12 h, wRp = 8.32%, Rp = 4.87%, $\chi^2 = 4.222$, a = b = 5.955(6) Å, c = 7.276(1) Å				
Ba	0, 0, 0	1a	1	0.227(1)
Ba	0.3333, 0.6667, 0.663(8)	2d	1	0.227(1)
Ta	0.3333(4), 0.6667(4), 0.1703(4)	2d	0.865(5)	1.502(9)
Ca	0.3333(4), 0.6667(4), 0.1703(4)	2d	0.135(5)	1.502(9)
Ca	0, 0, 0.5	1b	1	0.289(6)
O	0.5, 0, 0	3e	0.955	0.879(8)
O	0.1359(3), 0.8641(7), 0.2975(2)	6i	0.955	0.879(8)
$\text{Ba}_3\text{Ca}_{1.36}\text{Ta}_{1.64}\text{O}_{8.415}$ (<i>P-3m1</i>), 1400 °C, 12 h, wRp = 5.99%, Rp = 3.86%, $\chi^2 = 2.413$, a = b = 5.917(3) Å, c = 7.2346(2) Å				
Ba	0, 0, 0	1a	1	0.433(3)
Ba	0.3333, 0.6667, 0.663(9)	2d	1	0.433(3)
Ta	0.3333, 0.6667, 0.1740(8)	2d	0.72 (2)	0.513(6)
Ca	0.3333, 0.6667, 0.1740(8)	2d	0.18(2)	0.513(6)
Ca	0, 0, 0.5	1b	1	0.235(3)
O	0.5, 0, 0	3e	0.935	0.476(7)
O	0.1481(3), 0.8519(8), 0.3160(9)	6i	0.935	0.476(7)

Ba₃Ca_{1.09}Ta_{1.91}O_{8.865} (<i>Fm-3m</i>), 1400 °C, 12 h, wRp = 5.5%, Rp = 3.5%, $\chi^2 = 0.6333$, a = b = c = 8.403(3) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.785(8)
Ta	0.5, 0.5, 0.5	4b	1	0.873(4)
Ta	0, 0, 0	4a	0.273(6)	1.001(4)
Ca	0, 0, 0	4a	0.727(6)	1.001(6)
O	0.2202(8), 0, 0	24e	0.985	0.579(6)
Ba₃Ca_{1.18}Ta_{1.82}O_{8.73} (<i>Fm-3m</i>), 1400 °C, 12 h, wRp = 4.94%, Rp = 3.81%, $\chi^2 = 3.322$, a = b = c = 8.459(2) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.502(8)
Ta	0.5, 0.5, 0.5	4b	1	0.726(4)
Ta	0, 0, 0	4a	0.213(2)	0.611 (3)
Ca	0, 0, 0	4a	0.787(2)	0.611(3)
O	0.2586(3), 0, 0	24e	0.97	0.726(8)
Ba₃Ca_{1.27}Ta_{1.73}O_{8.595} (<i>Fm-3m</i>), 1400 °C, 12 h, wRp = 8.32%, Rp = 4.87%, $\chi^2 = 4.222$, a = b = c = 8.444(3) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.815(7)
Ta	0.5, 0.5, 0.5	4b	0.153(4)	1.56(9)
Ca	0.5, 0.5, 0.5	4b	0.847(4)	1.56(9)
Ta	0, 0, 0	4a	1	1.42(8)
O	0.2697(5), 0, 0	24e	0.955	0.770(8)
Ba₃Ca_{1.36}Ta_{1.64}O_{8.415} (<i>Fm-3m</i>), 1400 °C, 12 h, wRp = 5.99%, Rp = 3.86%, $\chi^2 = 2.413$, a = b = c = 8.4316(1) Å				
Ba	0, 0, 0	8c	1	0.572(3)
Ta	0.5, 0.5, 0.5	4b	1	0.82(4)
Ca	0, 0, 0	4b	0.91(2)	1.36(4)
Ta	0, 0, 0	4a	0.09(2)	1.36(4)
O	0.2482(1), 0, 0	24e	0.935	1.05(4)

Table S4. Rietveld refinement of $\text{Ba}_3\text{Ca}_{1+x}\text{Ta}_{2-x}\text{O}_{9-3x/2}$ ($0 \leq x \leq 0.36$) prepared at 1550 °C, using TOF-Neutron diffraction data.

$\text{Ba}_3\text{CaTa}_2\text{O}_9$ (<i>P-3m1</i>), 1550 °C, 12 h, wRp = 2.54%, Rp = 6.98%, $\chi^2 = 5.174$, a = b = 5.90263(3) Å, c = 7.27439(8) Å				
Ba	0, 0, 0	1a	1	0.670(3)
Ba	0.3333, 0.6667, 0.6671(2)	2d	1	0.670(3)
Ta	0.3333, 0.6667, 0.1684(2)	2d	1	0.265(6)
Ca	0, 0, 0.5	1b	1	0.265(6)
O	0.5, 0, 0	3e	1	1.133(7)
O	0.1757(5), 0.8243(5), 0.3140(2)	6i	1	1.133(7)
$\text{Ba}_3\text{Ca}_{1.09}\text{Ta}_{1.91}\text{O}_{8.865}$ (<i>Fm-3m</i>), 1550 °C, 12 h, wRp = 4.62%, Rp = 8.37%, $\chi^2 = 14.32$, a = b = c = 8.39421(5) Å				
Ba	0.25, 0.25, 0.25	8c	1	0.847(3)
Ta	0.5, 0.5, 0.5	4b	0.3201(8)	0.462(4)
Ca	0.5, 0.5, 0.5	4b	0.6799(8)	0.462(4)
Ta	0, 0, 0	4a	0.9529(8)	0.462(4)
Ca	0, 0, 0	4a	0.0471(8)	0.462(4)
O	0.2395(6), 0, 0	24e	0.985	1.213(8)
$\text{Ba}_3\text{Ca}_{1.18}\text{Ta}_{1.82}\text{O}_{8.73}$ (<i>Fm-3m</i>), 1550 °C, 12 h, wRp = 3.97%, Rp = 8.39%, $\chi^2 = 10.35$, a = b = c = 8.4236(1) Å				
Ba	0.25, 0.25, 0.25	8c	1	1.06(3)
Ta	0.5, 0.5, 0.5	4b	1	0.451(9)
Ca	0, 0, 0	4a	0.787(7)	0.533(9)
Ta	0, 0, 0	4a	0.213(7)	0.533(8)
O	0.2613(3), 0, 0	24e	0.97	1.226(2)
$\text{Ba}_3\text{Ca}_{1.27}\text{Ta}_{1.73}\text{O}_{8.595}$ (<i>Fm-3m</i>), 1550 °C, 12 h, wRp = 4.73%, Rp = 8.67%, $\chi^2 = 17.13$, a = b = c = 8.43722(4) Å				
Ba	0, 0, 0	8c	1	0.572(3)
Ta	0.5, 0.5, 0.5	4b	0.153	0.781(4)
Ca	0.5, 0.5, 0.5	4b	0.847	0.781(4)
Ta	0, 0, 0	4a	1	0.501(6)
O	0.2373(6), 0, 0	24e	0.955	1.242(7)

Table S5. Relative density of structural dependent BCT samples prepared at 1550 °C.

Composition	Relative Density (%)	Phase
BCT00	70.1	<i>P-3m1</i>
BCT09	72.9	<i>Fm-3m</i>
BCT18	78.8	<i>Fm-3m</i>
BCT27	85.7	<i>Fm-3m</i>
BCT36	89.9	<i>Fm-3m</i>
BCN18	92.5	<i>Fm-3m</i>

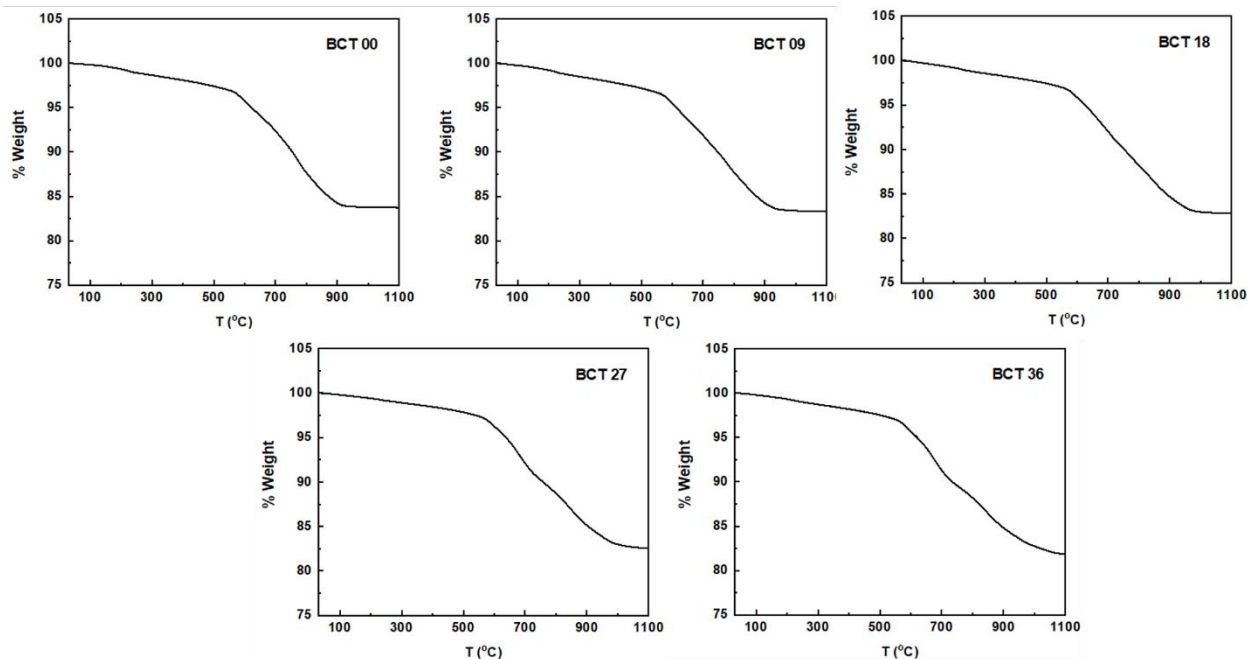


Figure S1. Thermogravimetric analysis (TGA) of $\text{Ba}_3\text{Ca}_{1+x}\text{Ta}_{2-x}\text{O}_{9-3x/2}$ ($0 \leq x \leq 0.36$) precursors heated in air from room temperature to 1100 °C.

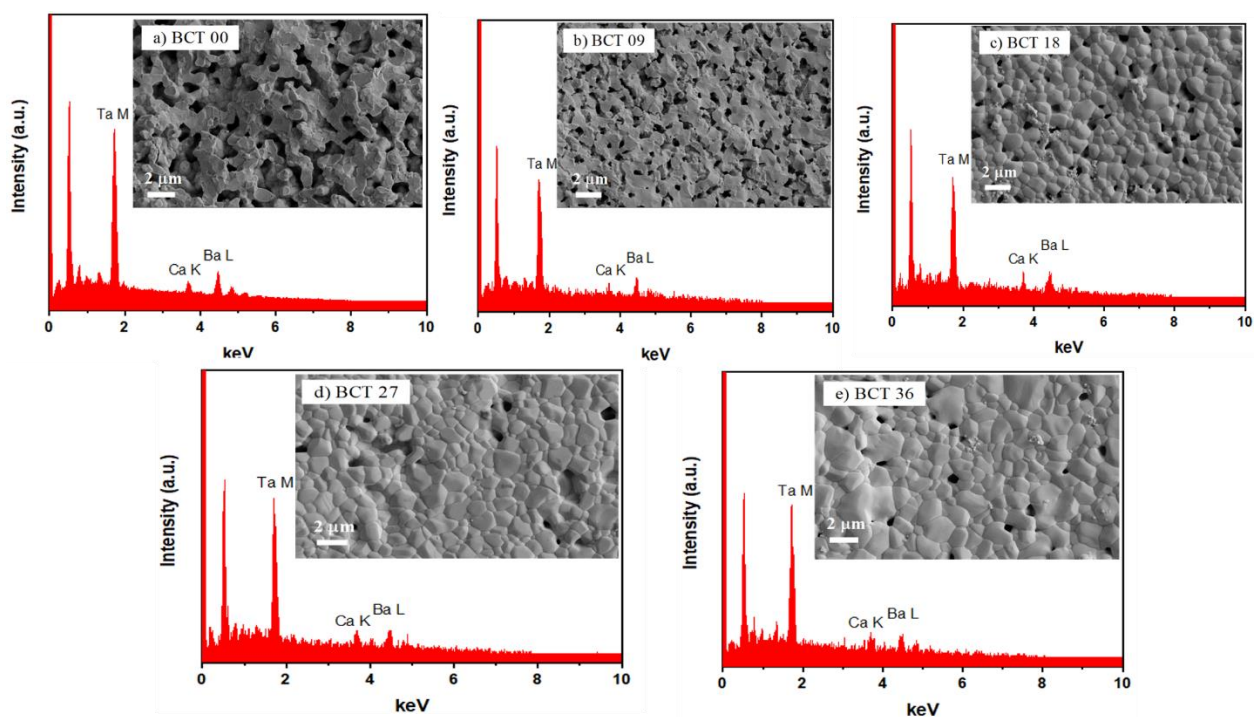


Figure S2. SEM and EDX images of $\text{Ba}_3\text{Ca}_{1+x}\text{Ta}_{2-x}\text{O}_{9-3x/2}$ ($0 \leq x \leq 0.36$) prepared at 1550 °C.

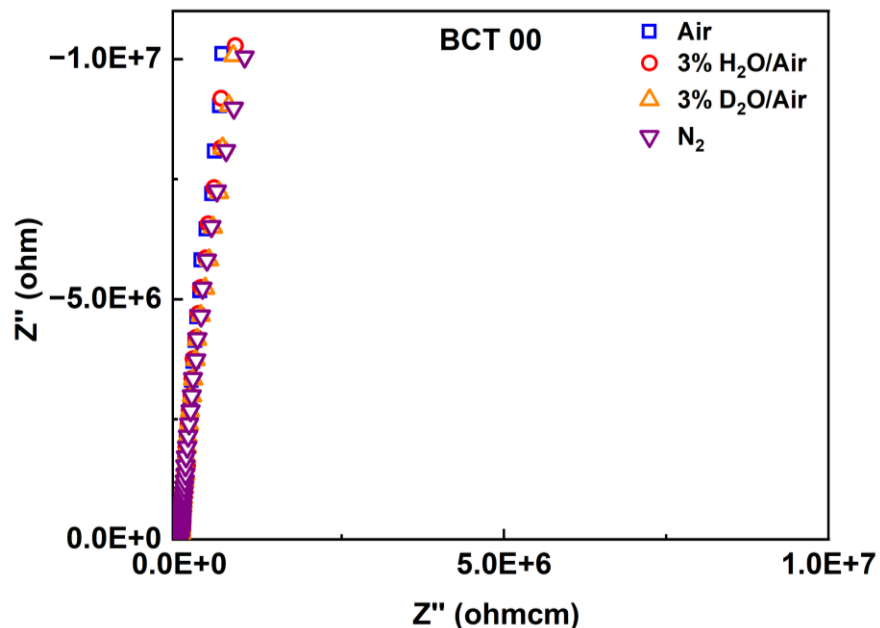


Figure S3. Nyquist plots at temperature 500 °C for $Ba_3CaTa_2O_9$ (BCT00) in air, 3% H_2O /air, 3% D_2O /air, and N_2 . The samples for electrical conductivity measurements were prepared at 1550 °C for 24 h in air.

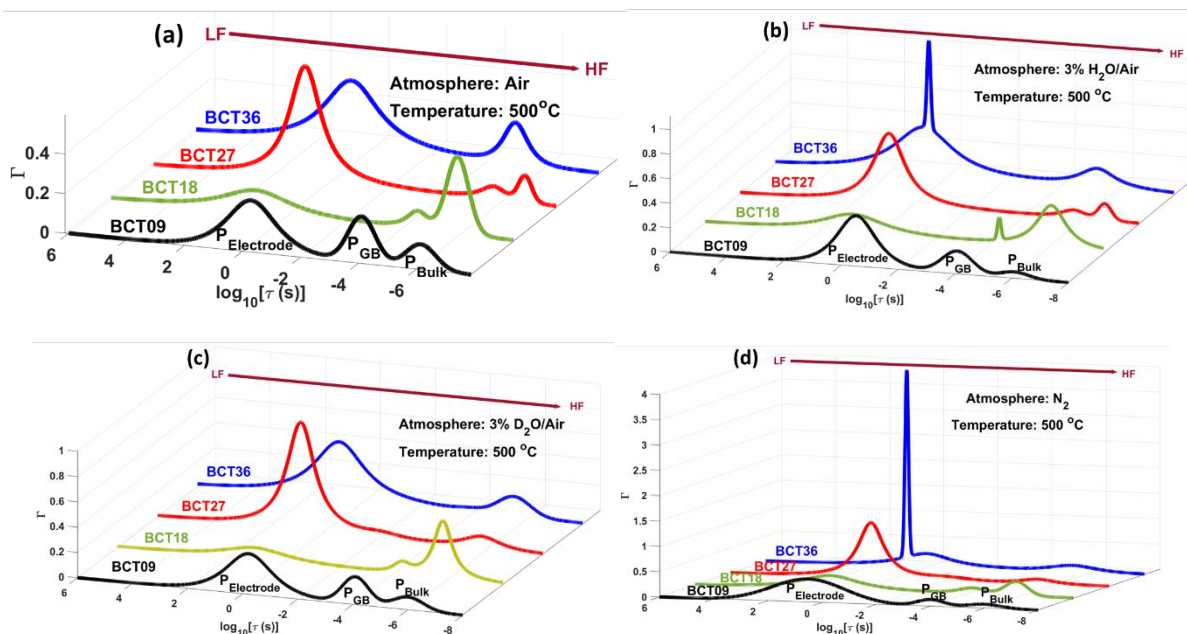


Figure S4. DFRT plots for $Ba_3Ca_{1+x}Ta_{2-x}O_{9-3x/2}$ ($0 \leq x \leq 0.36$) under (a) air, (b) 3% H_2O /Air, (c) 3% D_2O /air, and (d) N_2 at 500 °C showing presence of three peaks due to bulk, grain-boundary (GB), and electrode transport processes.