

Electronic Supplementary Information for

**Phosphorus-triggered activation of PdPb nanoflowers for enhanced oxygen reduction
electrocatalysis**

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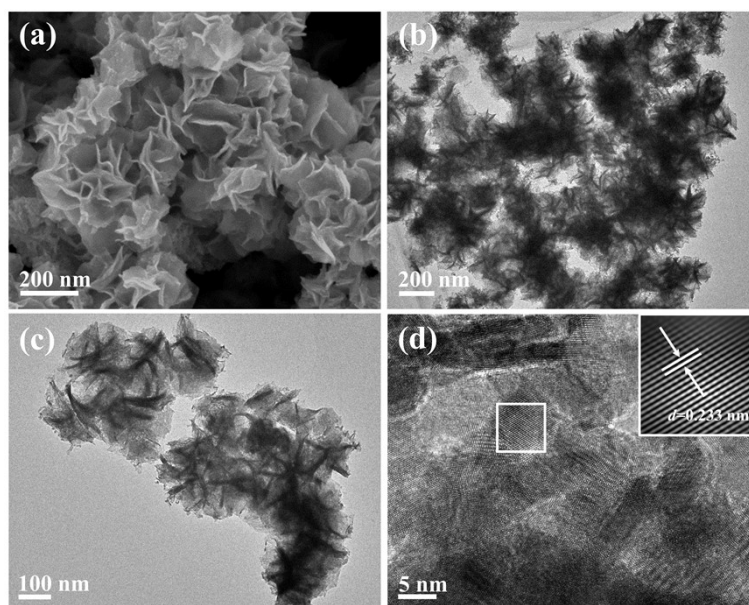


Fig. S1 (a) SEM and (b, c) TEM images of PdPb NFs. (d) HRTEM images of PdPb NFs and corresponding Fourier filtered lattice fringes of the square area.

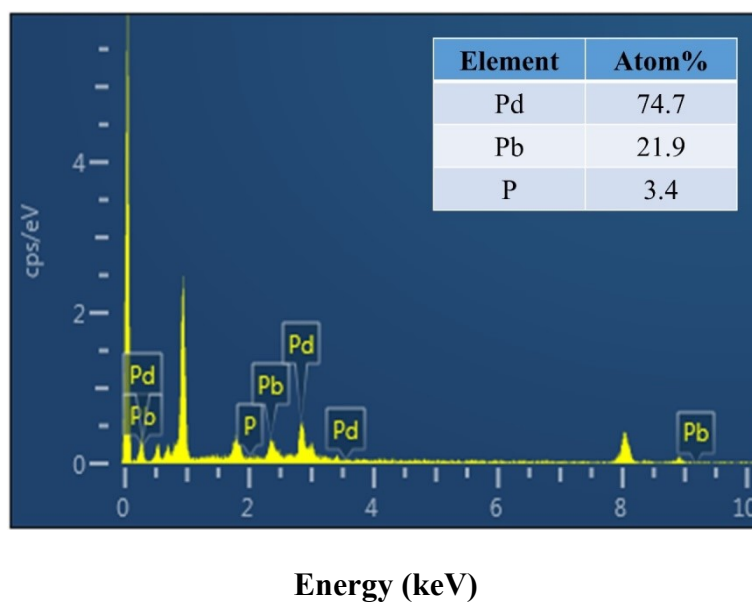


Fig. S2 EDX spectrum and element composition of P-PdPb NFs.

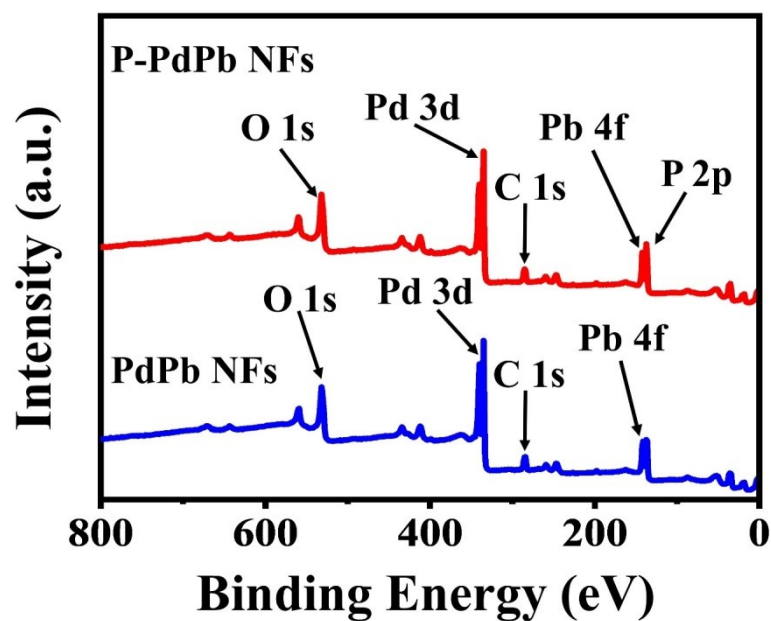


Fig. S3 XPS survey spectra of P-PdPb NFs and PdPb NFs.

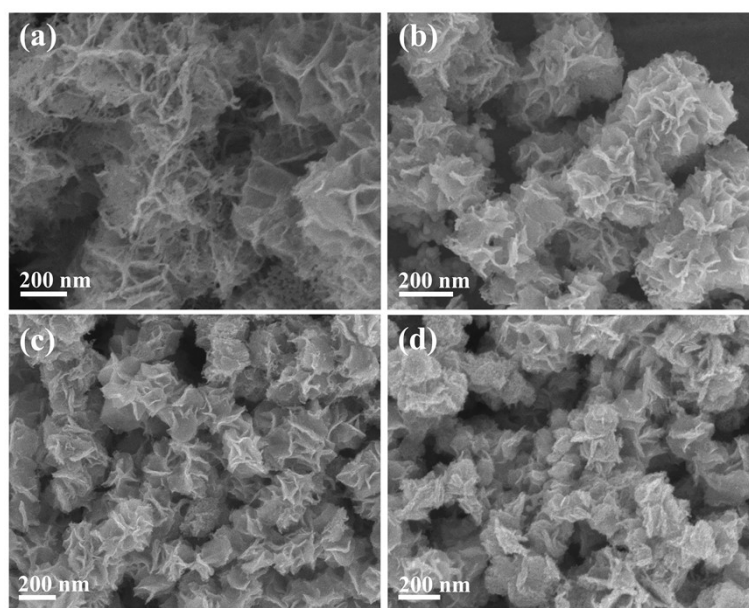


Fig. S4 SEM images of the P-PdPb NFs samples prepared with different contents of $W(CO)_6$ under the identical synthesis conditions: (a) 5 mg, (b) 30 mg, (c) 60 mg, and (d) 100 mg.

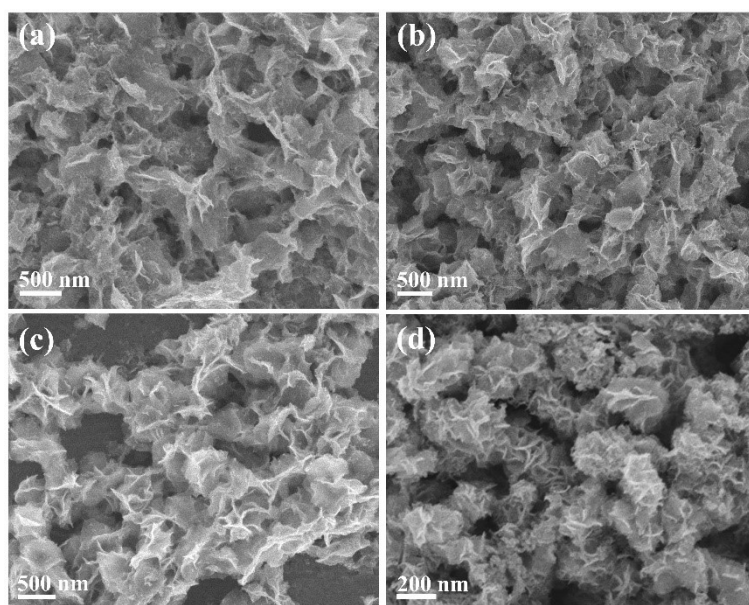


Fig. S5 SEM images of P-PdPb NFs synthesized at different reaction times: (a) 5 min, (b) 30 min, (c) 1 h, and (d) 2 h.

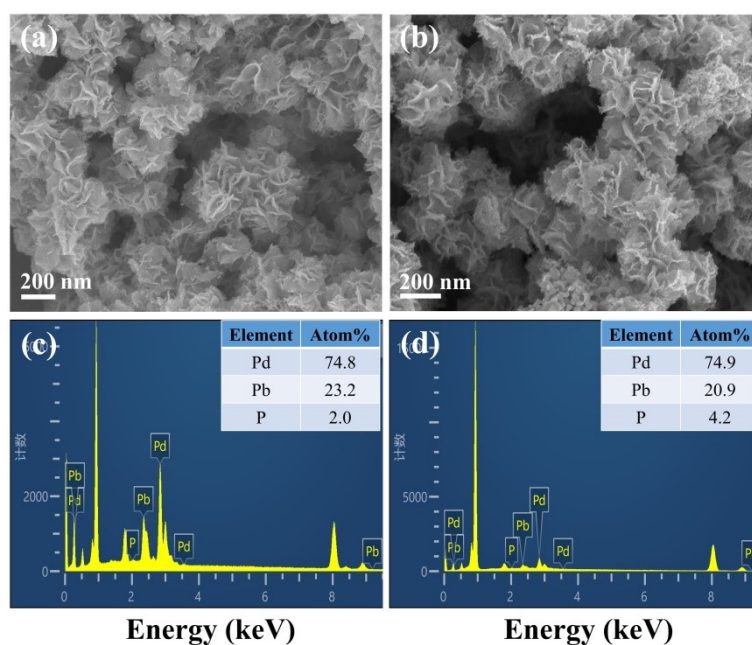


Fig. S6 SEM images of (a) $P_{0.5}$ -PdPb NFs and (b) $P_{2.0}$ -PdPb NFs, and corresponding (c, d) EDX spectra and element composition.

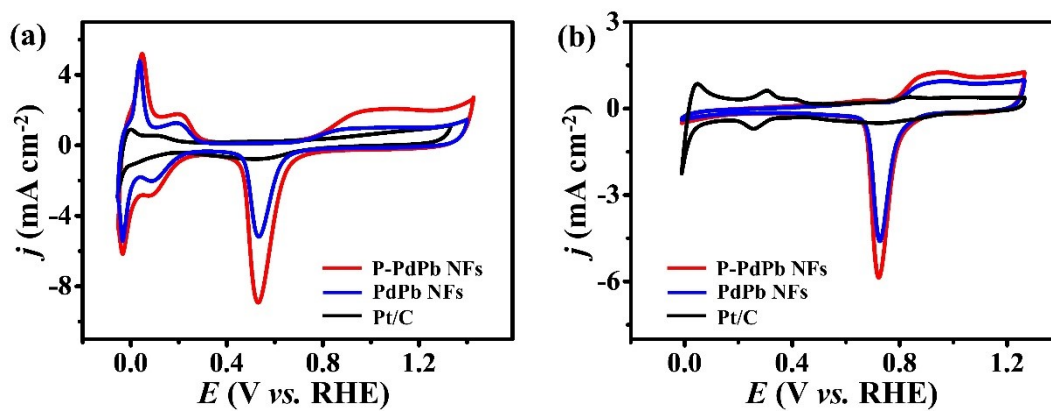


Fig. S7 CV curves of three catalysts in (a) 0.1 M HClO_4 and (b) 0.1 M KOH .

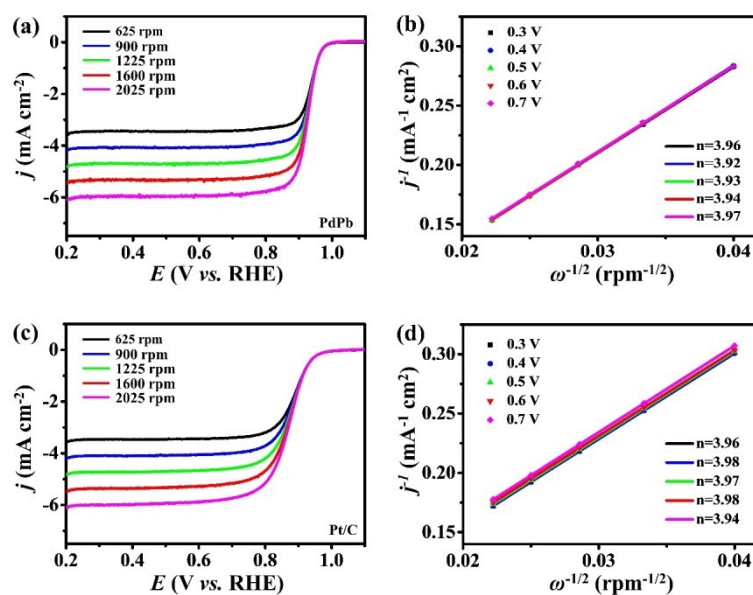


Fig. S8 ORR polarization curves at different rotating speeds for (a) PdPb NFs and (c) Pt/C. The electron transfer numbers at different potentials for (b) PdPb NFs and (d) Pt/C.

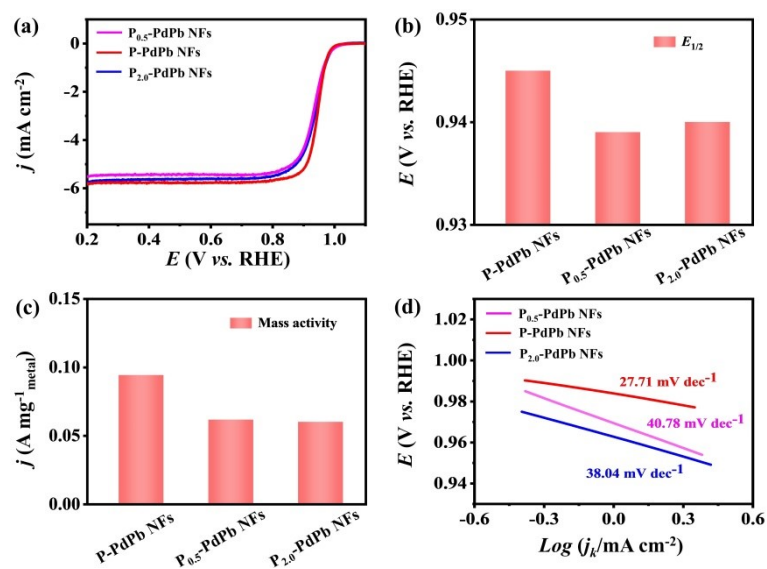


Fig. S9 (a) ORR polarization curves of samples, and corresponding (b) $E_{1/2}$, (c) mass activity at 0.95 V and (d) Tafel plots.

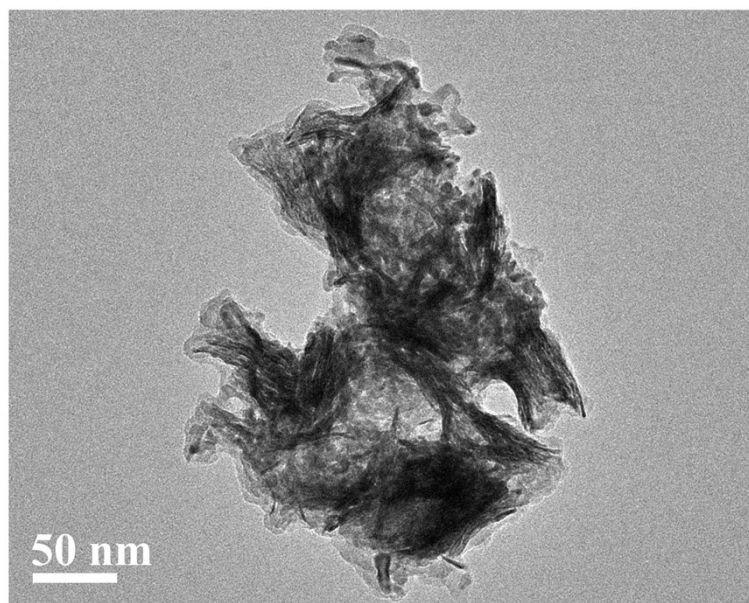


Fig. S10 TEM image of P-PdPb NFs after stability test.

Table S1. The comparison of ORR performance of P-PdPb NFs with some reported Pd-based electrocatalysts under alkaline conditions.

Catalyst	Electrolyte	E_{onset}	$E_{1/2}$	Ref
P-PdPb NFs	0.1 M KOH	1.06	0.95	This work
AuPd@TA LCs	0.1 M KOH	1.01	0.90	1
PdCu MNs	0.1 M KOH	0.94	0.89	2
Pd Metallene	0.1 M KOH	1.02	0.90	3
Pd ₅₀ Ru ₅₀ /CNs	0.1 M KOH	0.90	0.80	4
PdNi nanocorals	0.1 M KOH	0.95	0.85	5
Pd ₃ Pb SNP	0.1 M KOH	/	0.89	6
Pd ₂ N NCs	0.1 M KOH	/	0.92	7
Pd@Pt NCs	0.1 M KOH	0.88	/	8
Pd/PNCNF	0.1 M KOH	0.92	/	9

References

1. S. Yin, S. Liu, S. Jiao, H. Zhang, Y. Xu, Z. Wang, X. Li, L. Wang and H. Wang, *J. Mater. Chem. A*, 2021, **9**, 15678-15683.
2. M. Iqbal, Y. Kim, C. Li, B. Jiang, T. Takei, J. Lin, B. Yulianto, Y. Bando, J. Henzie and Y. Yamauchi, *ACS Appl. Mater. Interfaces*, 2019, **11**, 36544-36552.
3. H. Yu, T. Zhou, Z. Wang, Y. Xu, X. Li, L. Wang and H. Wang, *Angew. Chem., Int. Ed.*, 2021, **60**, 12027-12031.
4. J. Tian, W. Wu, Z. Tang, Y. Wu, R. Burns, B. Tichnell, Z. Liu and S. Chen, *Catalysts*, 2018, **8**, 329.
5. Z. Liu, X. Yang, L. Cui, Z. Shi, B. Lu, X. Guo, J. Zhang, L. Xu, Y. Tang and Y. Xiang, *Part. Part. Syst. Character.*, 2018, **35**, 1700366.
6. K. Wang, Y. Qin, F. Lv, M. Li, Q. Liu, F. Lin, J. Feng, C. Yang, P. Gao and S. Guo, *Small Methods*, 2018, **2**, 1700331.
7. R. Guo, K. Zhang, Y. Liu, Y. He, C. Wu and M. Jin, *J. Mater. Chem. A*, 2021, **9**, 6196-6204.
8. K. Qi, W. Zheng and X. Cui, *Nanoscale*, 2016, **8**, 1698-1703.
9. J. Wang, L. Liu, S. Chou, H. Liu and J. Wang, *J. Mater. Chem. A*, 2017, **5**, 1462-1471.