

Supporting Information

Multi-interfacial engineering of coil-like NiS-Ni₂P/Ni hybrid to efficiently boost electrocatalytic hydrogen generation in alkaline and neutral electrolyte

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1. Supplementary Figures

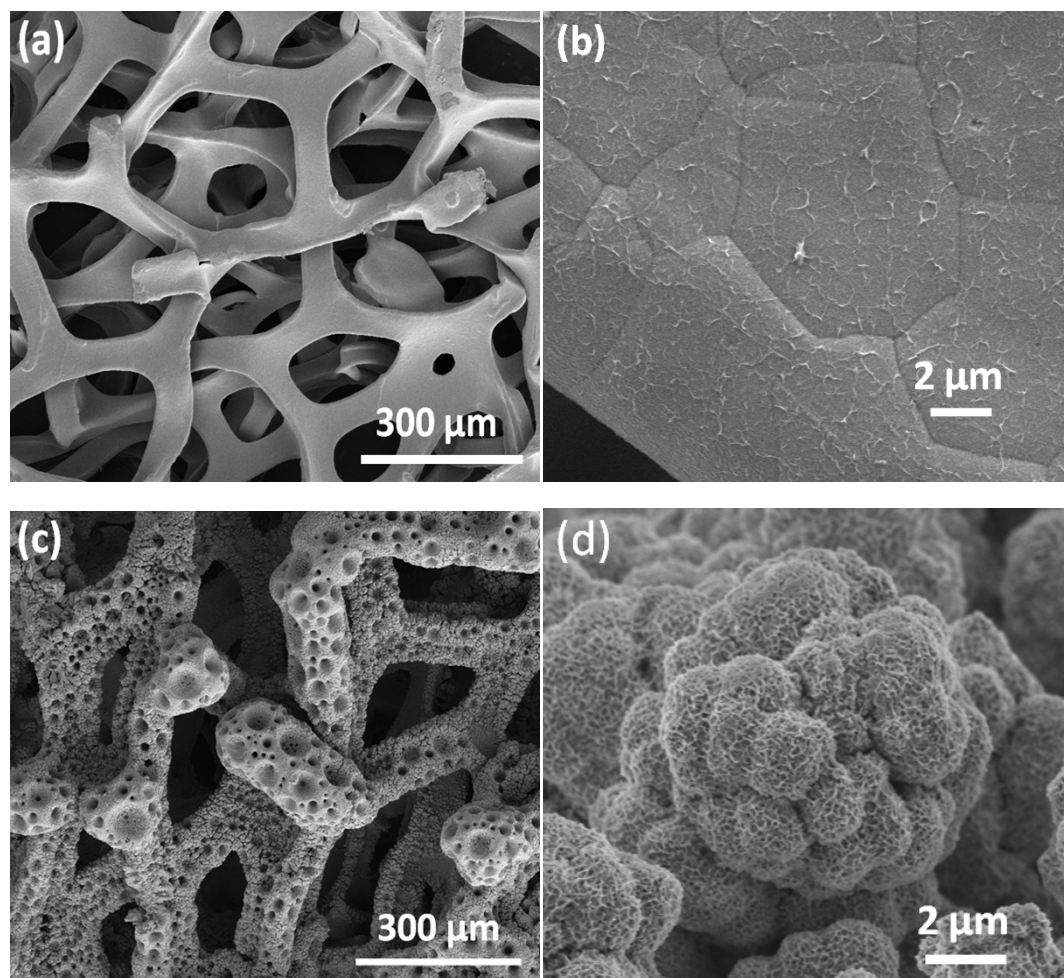


Figure S1. SEM characterization of NF and Ni/NF. (a,b) bare NF and (c,d) electrodeposited Ni/NF.

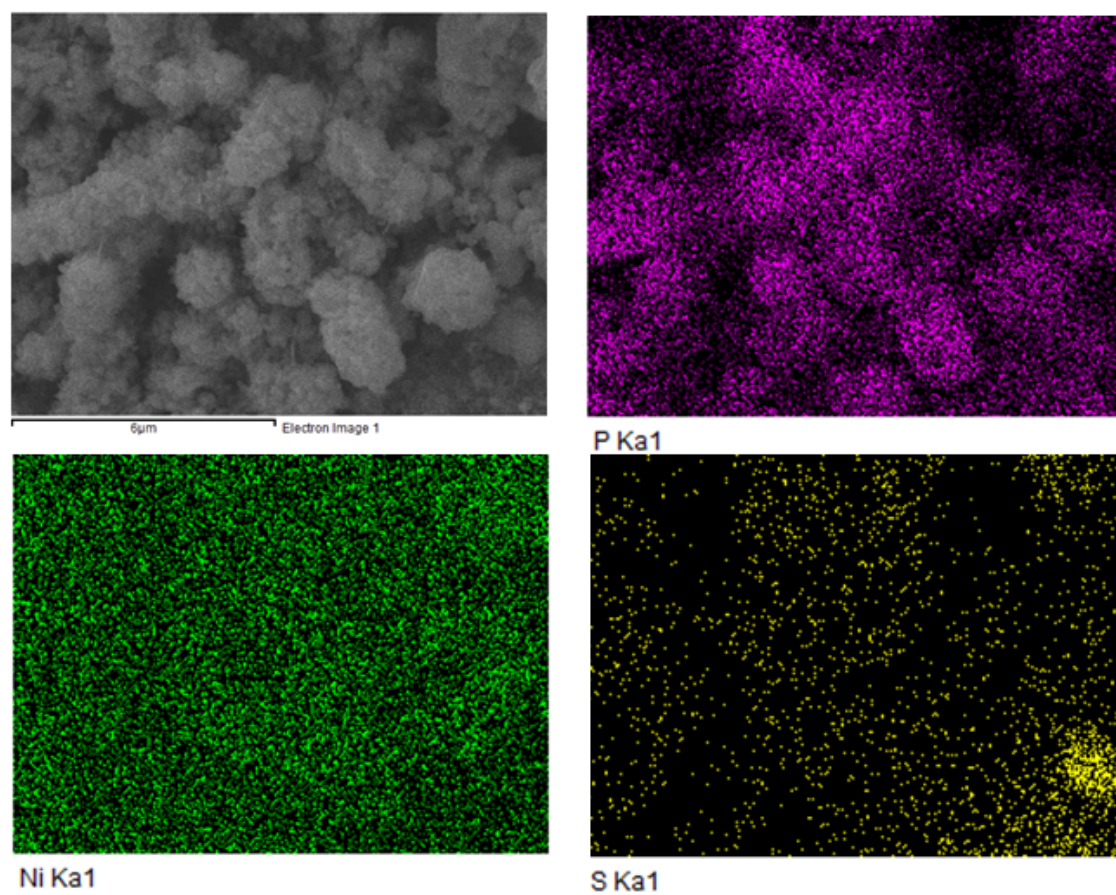


Figure S2. Corresponding elemental mapping images of NiS-Ni₂P/Ni/NF catalyst.

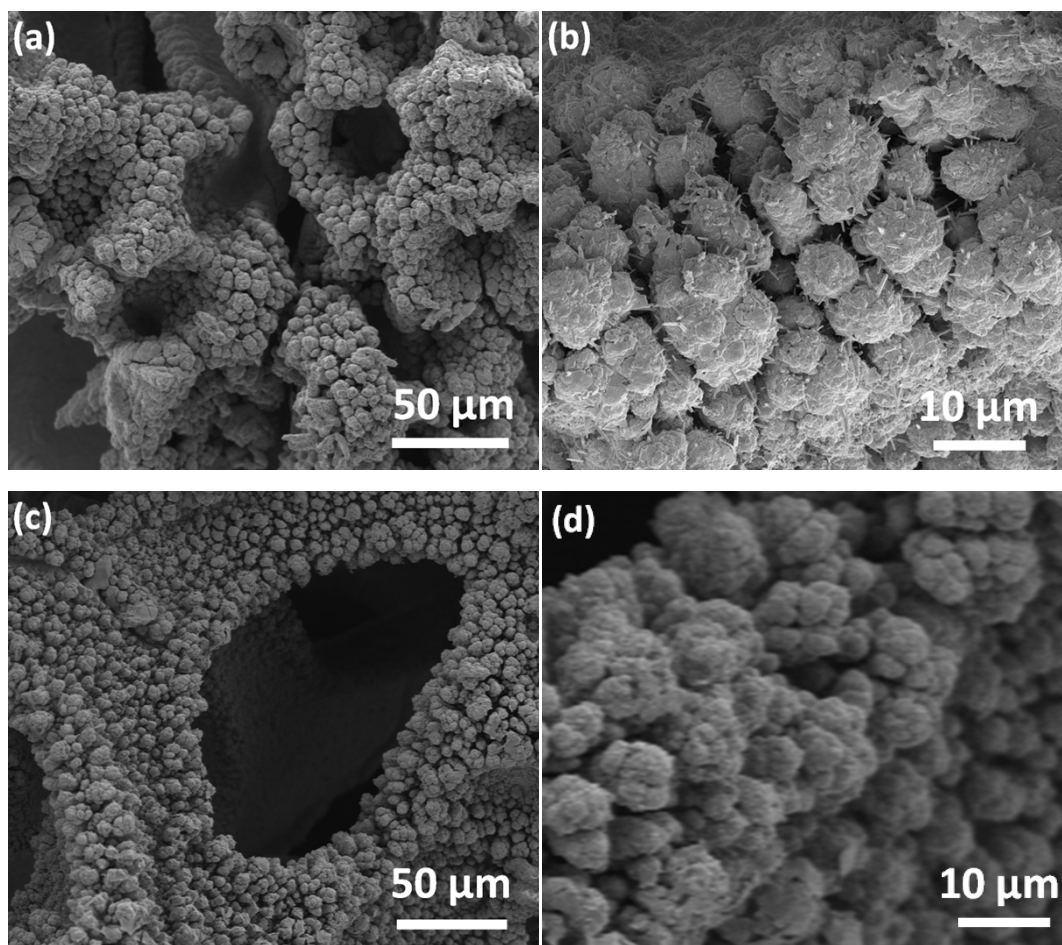


Figure S3. SEM characterization of $\text{Ni}_2\text{P}/\text{Ni}/\text{NF}$ and $\text{NiS}/\text{Ni}/\text{NF}$. (a,b) $\text{Ni}_2\text{P}/\text{Ni}/\text{NF}$ and (c,d) $\text{NiS}/\text{Ni}/\text{NF}$.

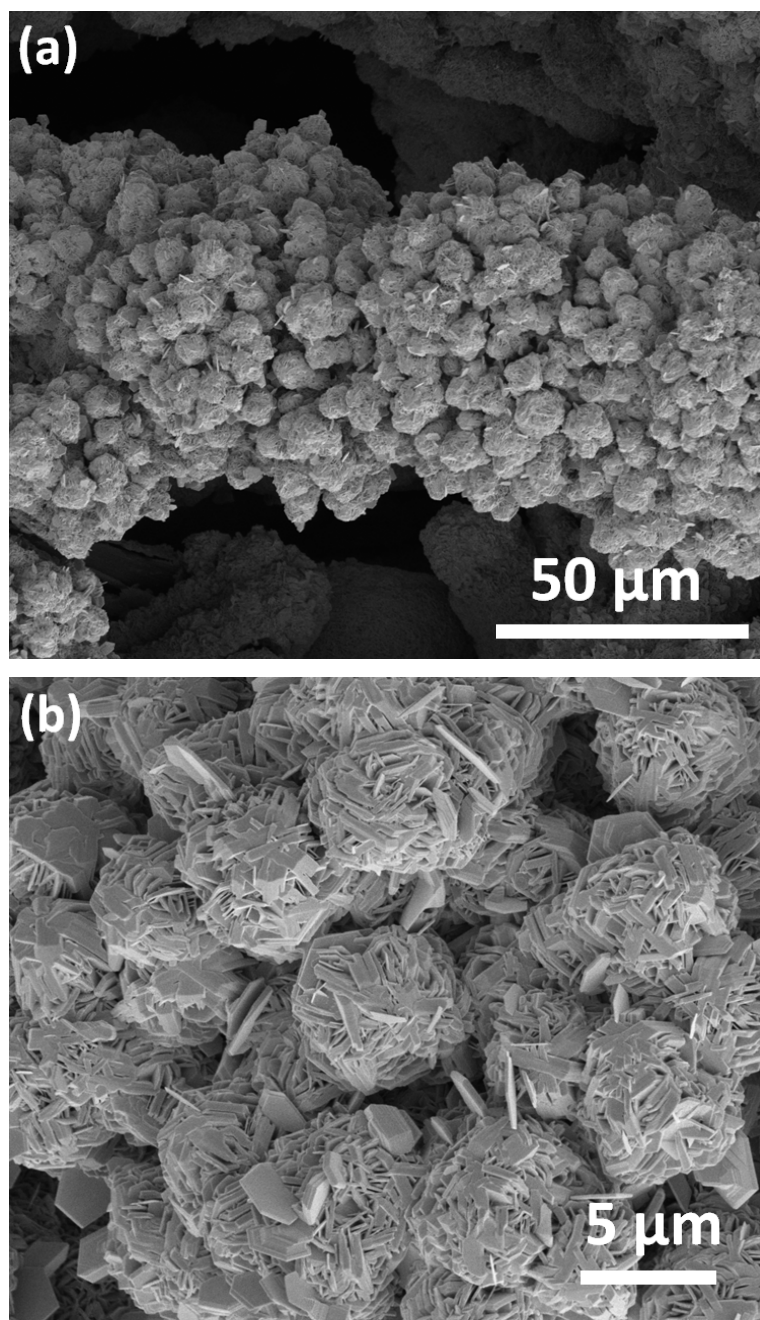


Figure S4. SEM images at different magnifications of the NiS-Ni₂P/Ni/CC with different magnifications.

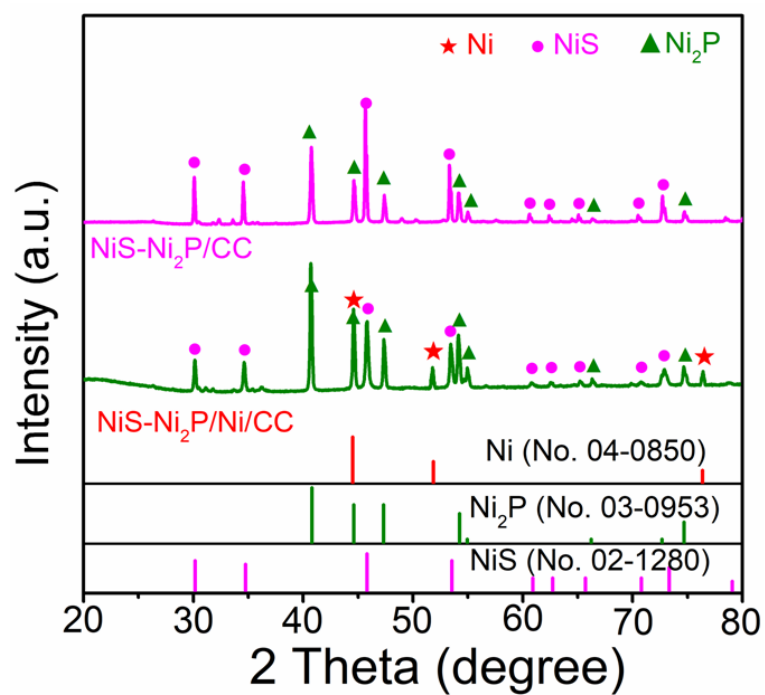


Figure S5. XRD patterns of excessive sulfuration/phosphorization (NiS-Ni₂P/CC) and partial sulfuration/phosphorization (NiS-Ni₂P/Ni/CC).

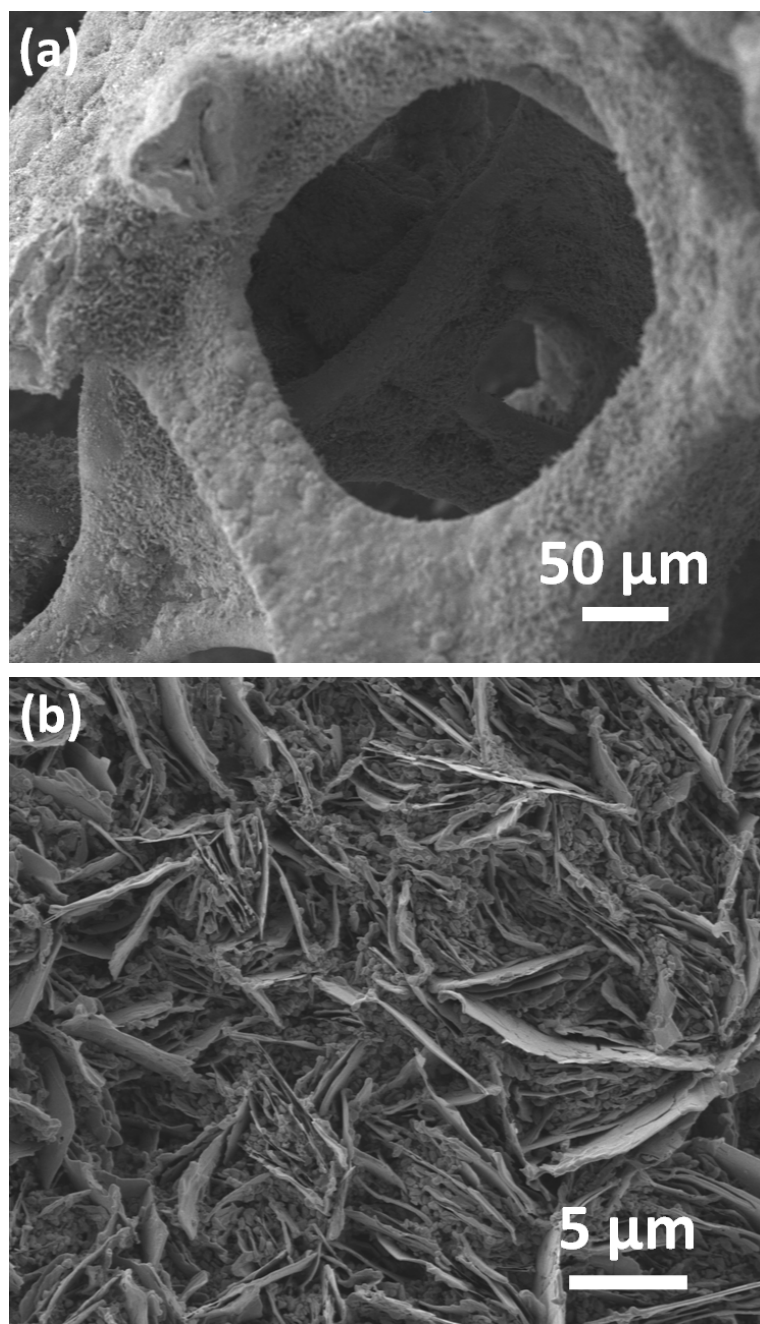


Figure S6. SEM images at different magnifications of the NiS-Ni₂P/NF with different magnifications.

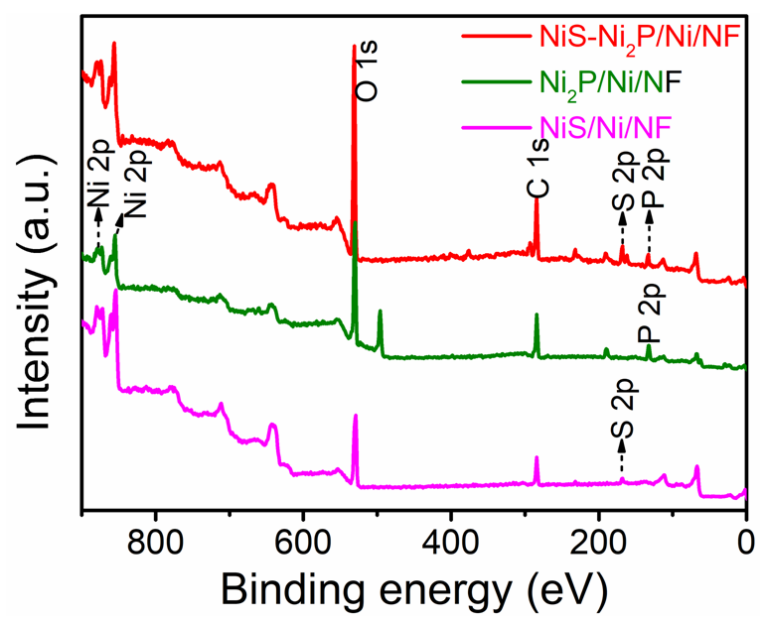


Figure S7. XPS survey spectra of NiS-Ni₂P/Ni/NF, Ni₂P/Ni/NF, and NiS/Ni/NF.

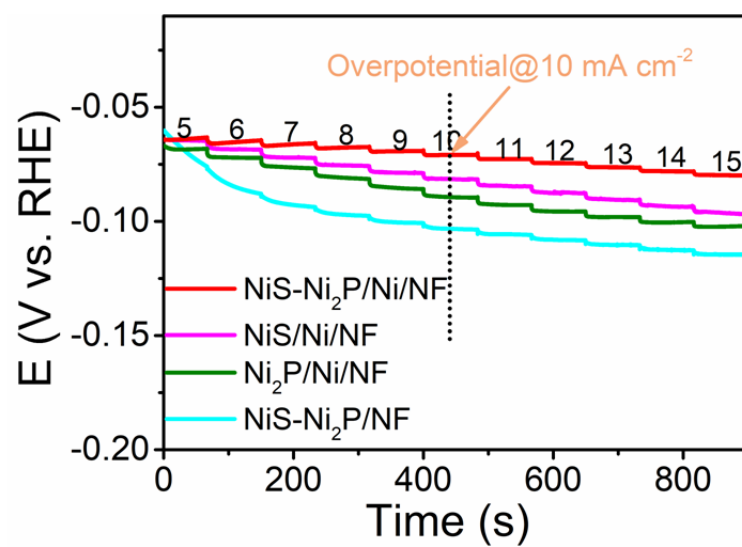


Figure S8. Chronopotentiometric plots of HER of the samples at different current densities in 1 M KOH.

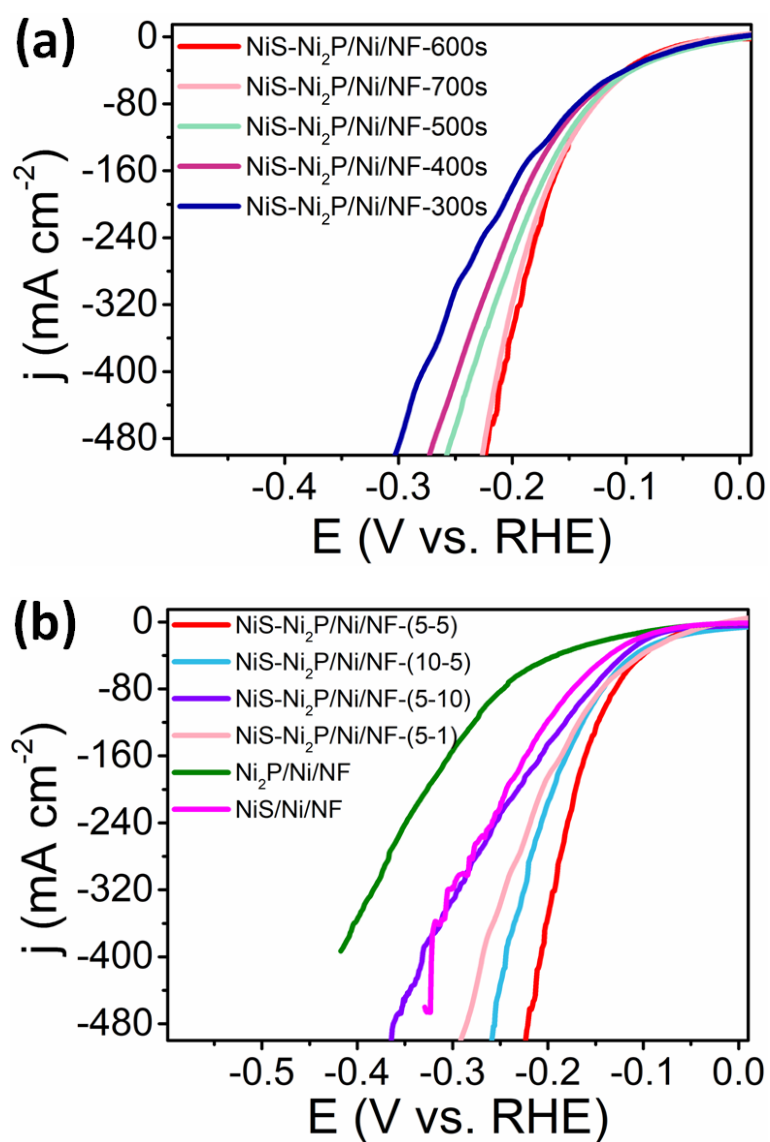


Figure S9. The Polarization curves of (a) different samples of $\text{NiS-Ni}_2\text{P/Ni/NF-a}$ (a=300, 400, 500, 600, and 700s) obtained by different electrodeposition time, (b) different samples of $\text{NiS-Ni}_2\text{P/Ni/NF-(x-y)}$ (x is the molar of S and y is the molar of NaH_2PO_2) obtained by different amount of S and P precursor.

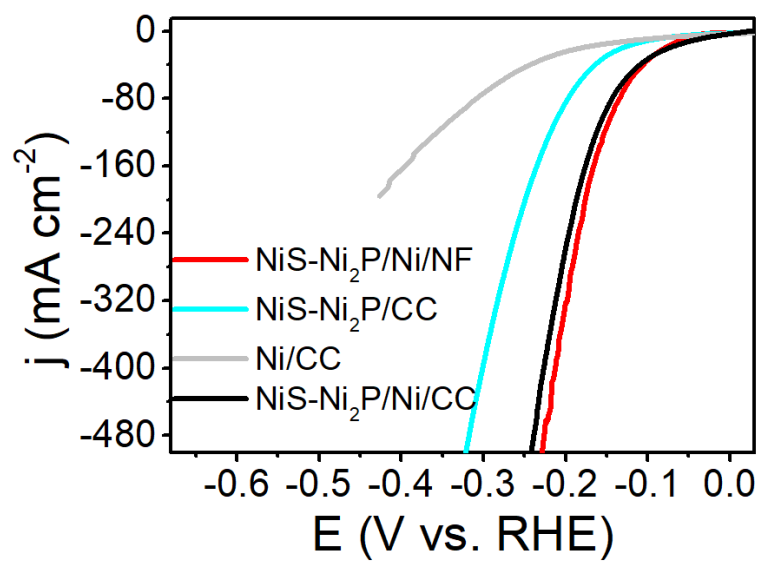


Figure S10. HER polarization curves of NiS-Ni₂P/Ni/NF, NiS-Ni₂P/CC, Ni/CC, and NiS-Ni₂P/Ni/CC in 1 M KOH solution.

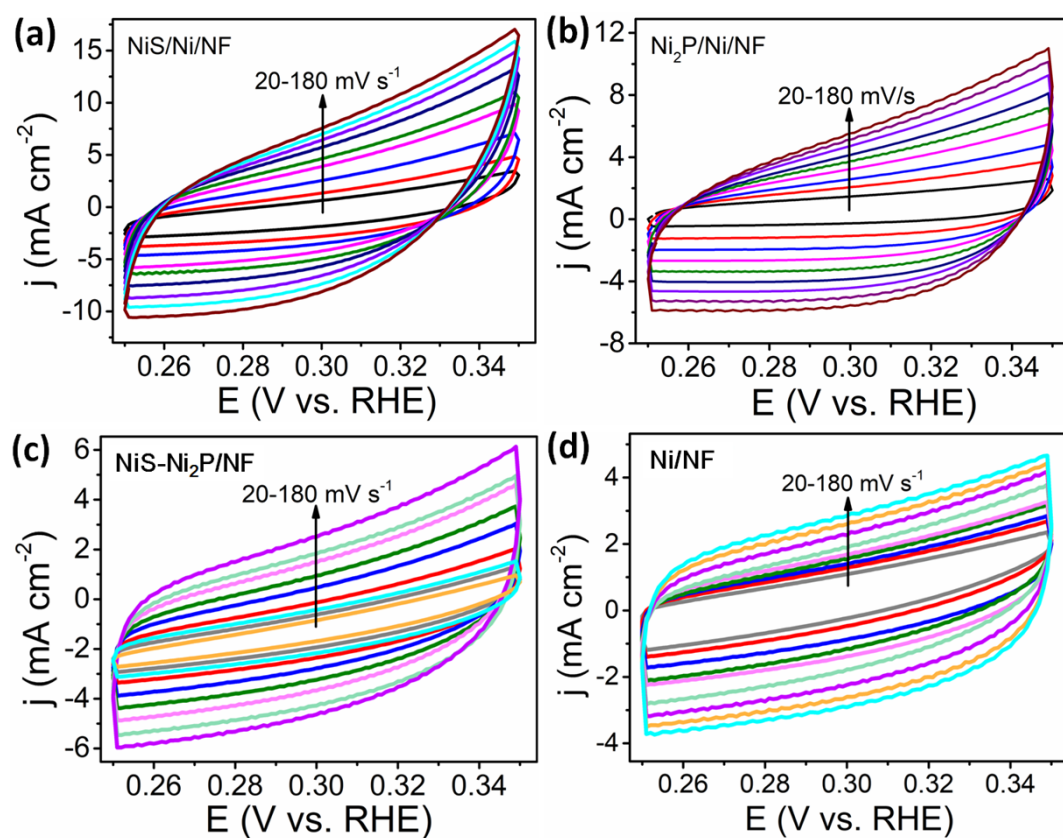


Figure S11. Cyclic voltammograms at various scan rates of 20, 40, 60, 80, 100, 120, 140, 160 and 180 mV s⁻¹ at $\eta = 0.25$ -0.35 V vs RHE for (a) NiS/Ni/NF, (b) Ni₂P /Ni/NF, (c) NiS-Ni₂P/NF, and (d) Ni/NF.

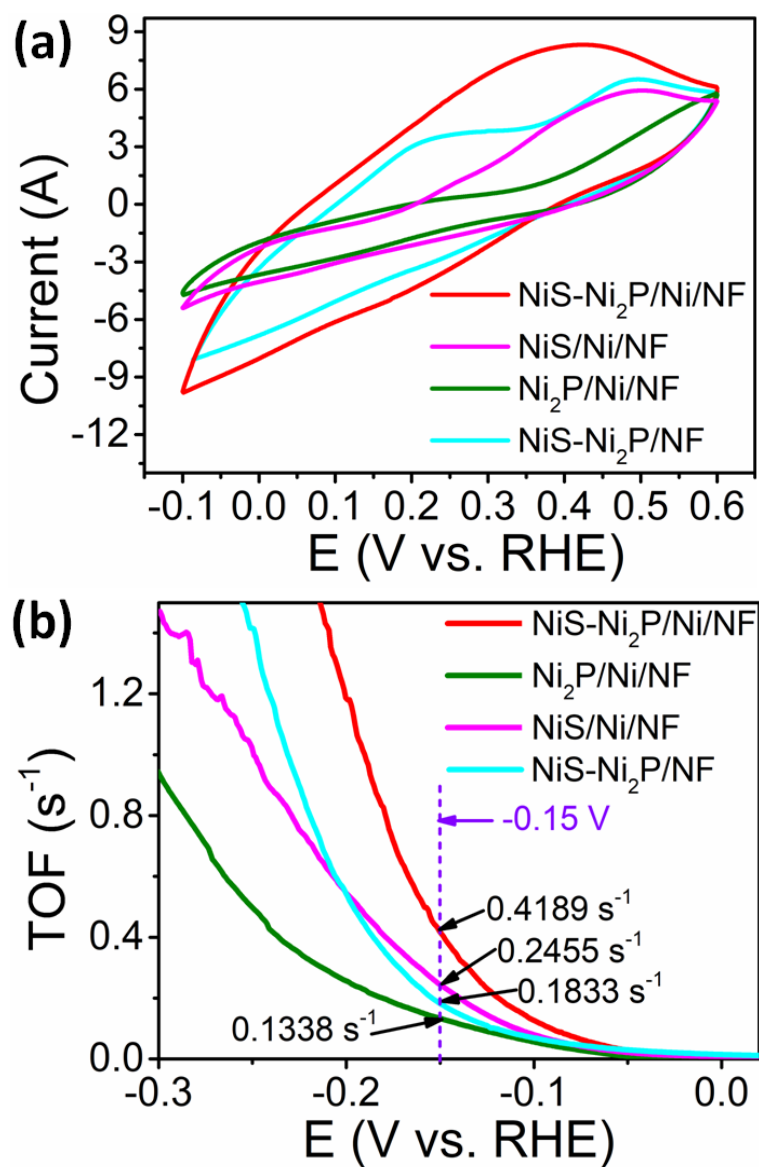


Figure S12. (a) Cyclic voltammograms (-0.1–0.6 V) of NiS-Ni₂P/Ni/NF, NiS/Ni/NF, Ni₂P/Ni/NF, and NiS-Ni₂P/NF recorded at pH = 7 phosphate buffer with a scan rate of 50 mV s⁻¹. (b) TOF curves of the NiS-Ni₂P/Ni/NF, NiS/Ni/NF, Ni₂P/Ni/NF, and NiS-Ni₂P/NF catalysts.

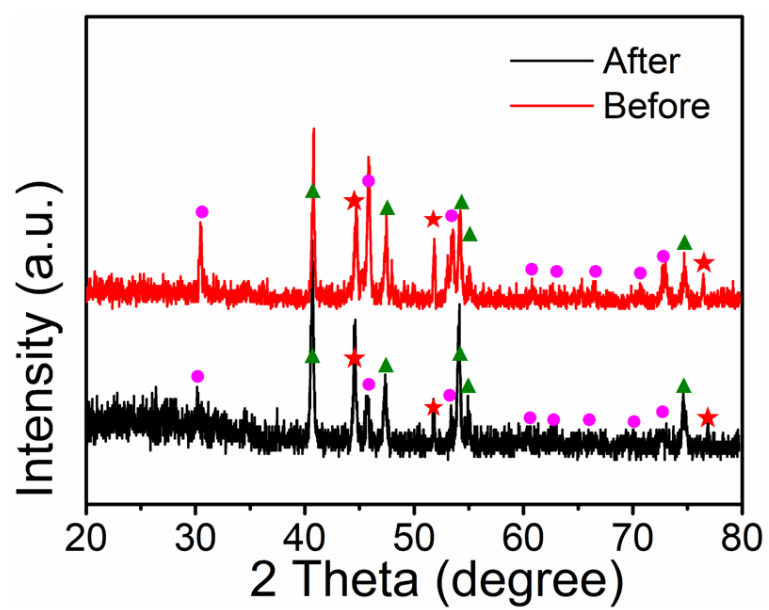


Figure S13. XRD patterns of NiS-Ni₂P/Ni/NF after HER stability test in alkaline solution.

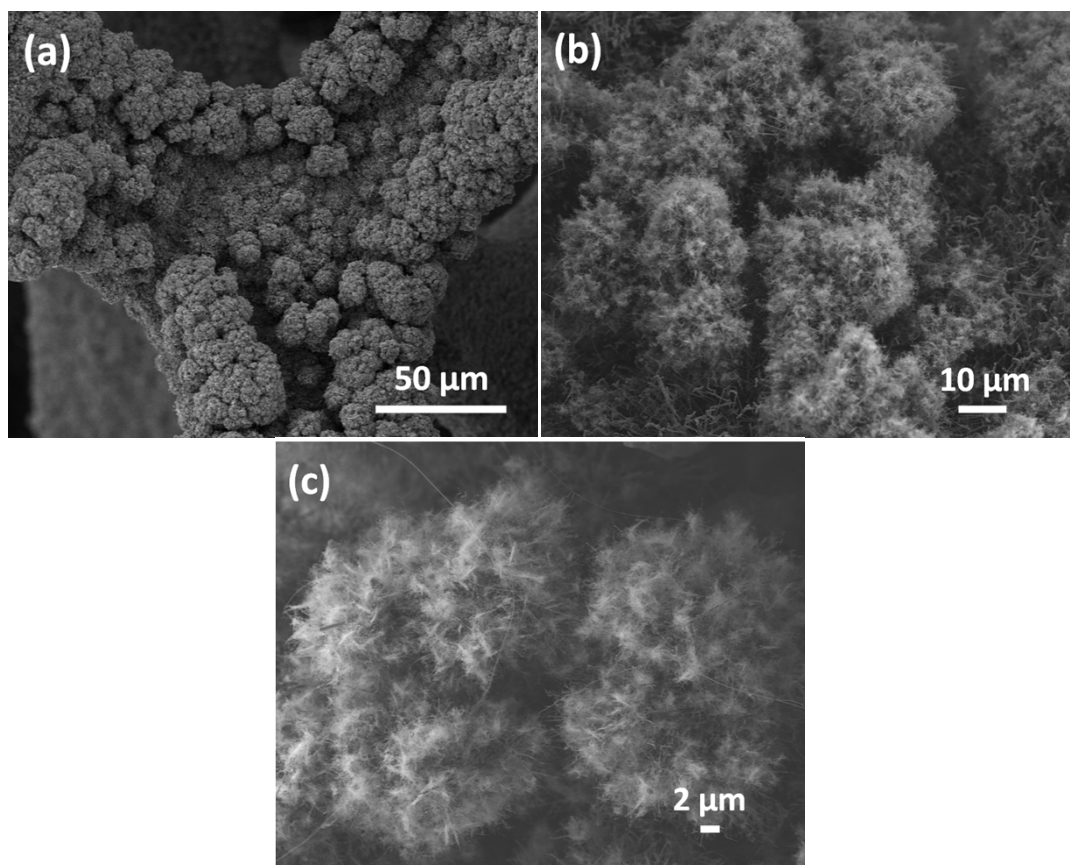


Figure S14. SEM images of NiS-Ni₂P/Ni/NF after HER stability test in alkaline solution.

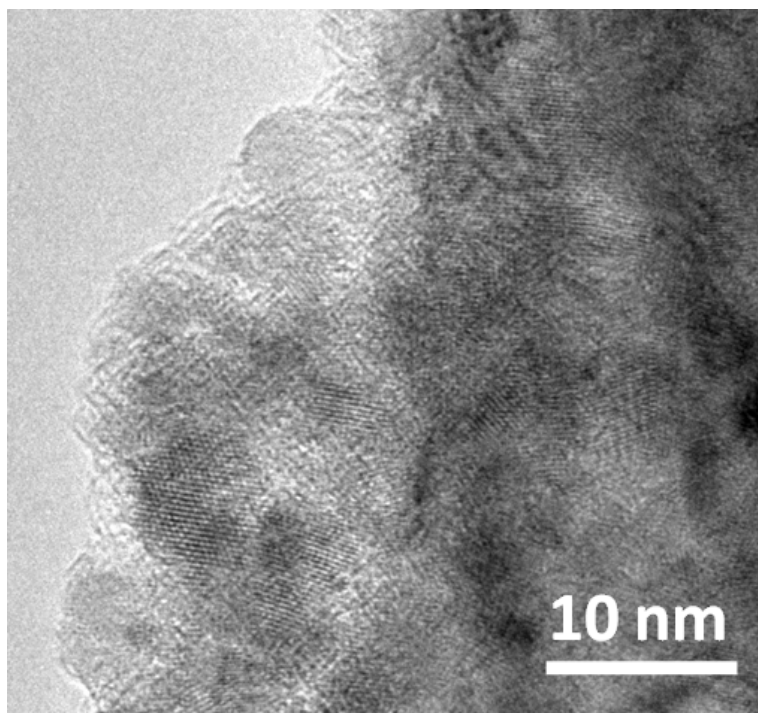


Figure S15. TEM images of NiS-Ni₂P/Ni/NF after stability test in alkaline solution.

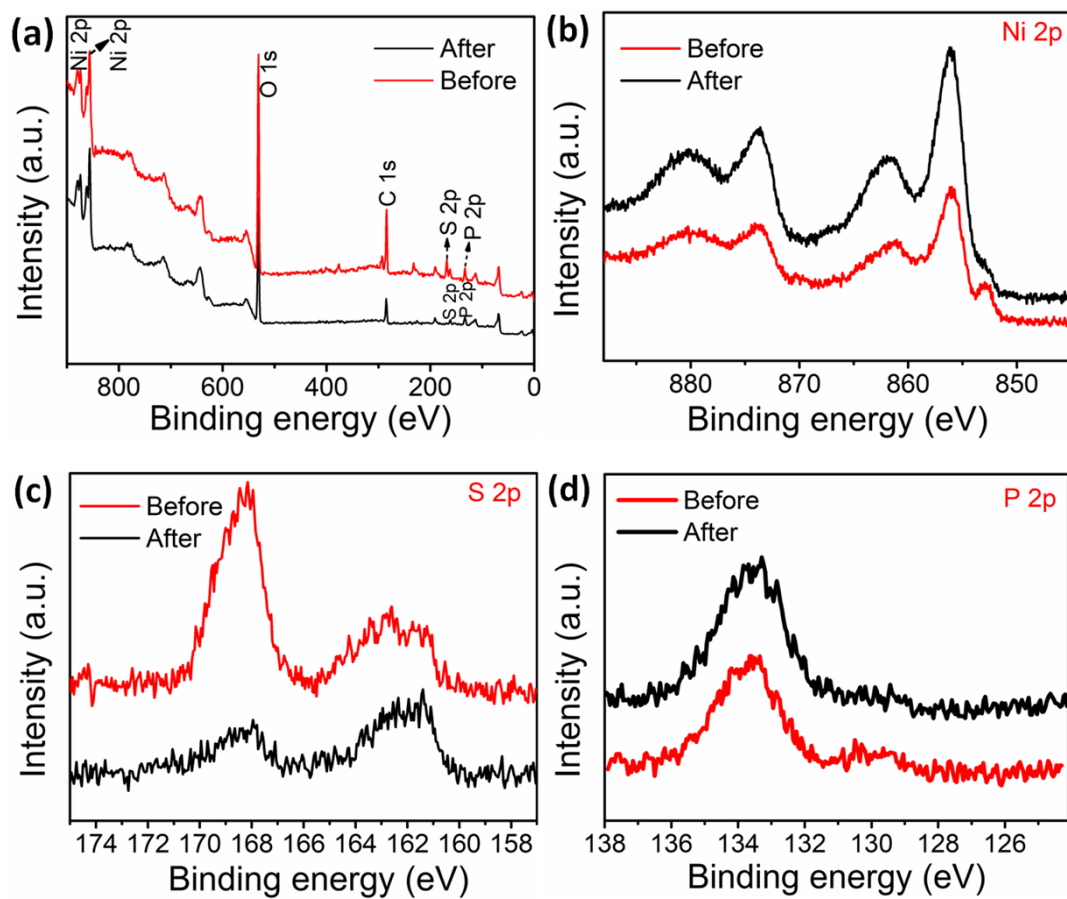


Figure S16. (a) XPS survey and (b-d) High-resolution XPS spectra: (b) Ni 2p, (c) S 2p, and (d) P 2p in NiS-Ni₂P/Ni/NF before and after stability measurement.

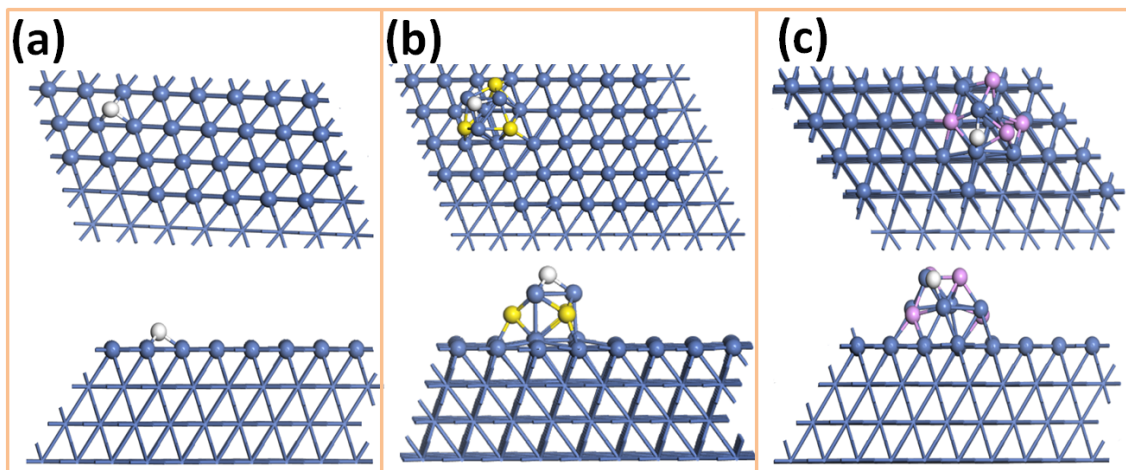


Figure S17. Optimized models for $*H$ intermediate of HER process on site of (a) Ni, (b) NiS/Ni, and (c) Ni₂P/Ni, showing both the top view (top) and side view (bottom) of each structure. Color codes: blue (Ni), pink (P), yellow (S), and white (H).

2. Supplementary Tables

Table S1. Average mass loading on Ni foam (NF) current collector.

Samples	Mass loading ^[a]
	(mg cm ⁻²)
NiS-Ni ₂ P/Ni/NF	17.1
NiS/Ni/NF	16.8
Ni ₂ P/Ni/NF	17.0
NiS-Ni ₂ P/NF	15.6
Ni/NF	15.1
20wt% Pt/C/NF	1.62

Table S2.The Ni, P, and S content of NiS-Ni₂P/Ni/NF electrocatalysts.

Samples	Composition (wt%)		
	Ni ^[a]	p ^[a]	S ^[b]
NiS-Ni ₂ P/Ni/NF	61.35	8.16	14.74

^[a]Ni and P content by ICP-AES. ^[b]S content by elemental analyzer.

Table S3. The composition of NiS-Ni₂P/Ni/NF, Ni₂P/Ni/NF, and NiS/Ni/NF from XPS.

Catalysts	C (wt.%)	Ni (wt.%)	S (wt.%)	P (wt.%)	O (wt.%)
NiS-	36.19	19.22	13.5	10.23	20.86
Ni ₂ P/Ni/NF					
Ni ₂ P/Ni/NF	40.88	14.79	-	18.51	25.82
NiS/Ni/NF	37.26	15.75	16.84	-	30.15

Table S4. HER performance of the as-prepared samples in 1.0 M KOH solution.

Catalysts	Overpotential at 10	Overpotential at 100
	mA cm ⁻² (mV)	mA cm ⁻² (mV)
NiS-Ni ₂ P/Ni/NF	53	139
NiS/Ni/NF	79	188
Ni ₂ P/Ni/NF	90	265
NiS-Ni ₂ P/NF	107	218
Ni/NF	121	356
20wt% Pt/C/NF	45	177

Table S5. Comparison of HER performance of the developed NiS-Ni₂P/Ni/NF electrocatalyst in this work with other high-performance catalysts in 1.0 M KOH.

Catalysts	Overpotential	Ref.
NiS-Ni₂P/Ni/NF	$\eta_{10}=53$ mV; $\eta_{100}=139$ mV	This work
Ni ₂ P-Ni ₃ S ₂ HNAs on NF	$\eta_{10}=80$ mV	1
Co _{0.9} S _{0.58} P _{0.42}	$\eta_{10}=141$ mV	2
Ni ₃ S ₄ @MoS ₂ /CC-3	$\eta_{10}=97$ mV	3
NiSP/NF	$\eta_{10}=93$ mV	4
Ni ₂ P/Ni ₃ Se ₄ -5.0	$\eta_{10}=57$ mV	5
V-Ni ₃ S ₂ /Ni _x P _y /NF nanosheet	$\eta_{10}=90$ mV	6
NiNS	$\eta_{100}=197$ mV	7
NiS ₂ -MoS ₂	$\eta_{10}=76$ mV	8
O-NiCoP/Ni ₂ P	$\eta_{10}=58$ mV	9
Ni-S-P/NF	$\eta_{10}=120$ mV	10
NiS/Ni ₂ P/CC	$\eta_{20}=111$ mV	11
CP/Ni ₂ P/NiS	$\eta_{10}=103$ mV	12
Ni-P-S nanosheets array	$\eta_{10}=101.9$ mV	13
N-NiMoO ₄ /NiS ₂	$\eta_{10}=99$ mV	14
Ni ₂ P@FePOx heterostructure	$\eta_{10}=75$ mV	15
Ni ₂ P/NF	$\eta_{10}=116$ mV	16
v-NiS ₂ /CeO ₂ HSs	$\eta_{10}=123$ mV	17
NiS-Gr-CC	$\eta_{10}=71$ mV	18
NiS/NiS ₂ /Ni ₃ S ₄	$\eta_{10}=68$ mV	19
Mo-NiS/Ni(OH) ₂	$\eta_{10}=74$ mV	20
Ni ₂ P/Ni@C	$\eta_{10}=149$ mV	21
Ni ₂ P-Cu ₃ P@NiCuC	$\eta_{10}=78$ mV	22
NiCoP/rGO	$\eta_{10}=209$ mV	23

Table S6. The C_{dl} and ECSA of the as-prepared samples.

Catalysts	C_{dl} (mF cm ⁻²)	ECSA(cm ⁻²)
NiS-Ni ₂ P/Ni/NF	60.99	1524
NiS/Ni/NF	41.90	1047
Ni ₂ P/Ni/NF	29.45	731
NiS-Ni ₂ P/NF	19.26	482
Ni/NF	13.10	328
Pt/C/NF	33.31	833

Table S7. HER performance of the as-prepared samples in 1 M PBS.

Catalysts	Overpotential at 10	Overpotential at 50
	mA cm ⁻² (mV)	mA cm ⁻² (mV)
NiS-Ni ₂ P/Ni/NF	115	241
NiS/Ni/NF	129	316
Ni ₂ P/Ni/NF	175	524
NiS-Ni ₂ P/NF	148	427
Ni/NF	221	-
20wt% Pt/C/NF	101	247

Table S8. Comparison of HER performance of the as-synthesized NiS-Ni₂P/Ni/NF electrocatalyst in this work with other high-performance catalysts in neutral.

Catalysts	Media	Overpotential	Ref.
NiS-Ni₂P/Ni/NF	1M PBS	η_{10}=115 mV; η_{50}=241 mV	This work
NiCo ₂ S ₄ /N,S-rGO	1M PBS	η_{10} =216 mV	24
Co ₉ S ₈ @C	1M PBS	η_{10} =280 mV	25
Ti@Ni(OH) ₂ -NiMoS	1M PBS	η_{10} =198 mV	26
CoSe ₂ /a-CoP,	1M PBS	η_{10} =185 mV	27
CoNi ₂ S ₄ /WS ₂ /Co ₉ S ₈	1M PBS	η_{10} =146 mV	28
Ni ₁₂ P ₅ -Ni ₂ P/Ni/NF	0.1M PBS	η_{10} =112 mV	29
W- Ni/Ni(OH) ₂ /Cu foam	1M PBS	η_{100} =310 mV	30
Cu-Ni ₃ S ₂ /NF	1M PBS	η_{10} =228 mV	31
MoP-Ru ₂ P/NPC	1M PBS	η_{10} =126 mV	32
Fe@Fe _x NiO/Ni@Ni _y CoP	1M PBS	η_{100} =386 mV	33
CN/CNL/MoS ₂ /CP	1M PBS	η_{10} =145 mV	34
PSS-PPy/Ni-Co-P	1M PBS	η_{10} =106 mV	35
Mo-CoP/NC/TF	1M PBS	η_{10} =130 mV	36
FeP NPs@NPC	1M PBS	η_{10} =386 mV	37
Ni _{0.1} Co _{0.9} P	1M PBS	η_{10} =125 mV	38
Cu _{0.075} Co _{0.925} P/CP	1M PBS	η_{10} =120 mV	39
Ni(S _{0.5} Se _{0.5}) ₂	1M PBS	η_{10} =124 mV	40

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