

Supporting Information

Revealing the Chemical Structure Dependent Carrier Trapping in One-Dimensional Antimony Selenide Photovoltaic Materials

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Experimental Section

Fabrication of Sb₂Se₃ films

Sb₂Se₃ films were fabricated on insulating substrate by thermal evaporation deposition method under a pressure of 5×10^{-4} Pa. Sb₂Se₃ (99.99%, zhongnuoxincai) powder was utilized as thermal evaporation source, and suitable Sb (99.999%, Sinopharm) powder was co-evaporated to adjust the stoichiometry of Sb₂Se₃ films. We selected a suitable evaporated rate of around 5 nm s^{-1} and obtained Sb₂Se₃ films around 300 nm. The substrate was preheated at 315 °C. Eventually, Sb₂Se₃ films were annealed at 380 °C for 8 min in glove box filled with N₂.

Fabrication of Sb₂Se₃ solar cells

The FTO glass (TEC-A7) was cleaned in the order of DI water, isopropanol, acetone, ethanol, then it was cleaned by UV ozone for 15 min. Afterwards, CdS electron transporting layer was deposited on as-cleaned FTO glass by CBD method. Then, Sb₂Se₃ film was deposited on CdS by thermal evaporation method. Afterwards, Spiro-OMeTAD was spin-coated on Sb₂Se₃ film as hole transport layer. Eventually, Au electrode was evaporated on the Spiro-OMeTAD layer under a pressure of 5×10^{-4} Pa. The active area was defined as 0.04 cm^2 by a metallic mask.

Characterizations of films and devices

The crystallinity of Sb₂Se₃ films was characterized by XRD (Bruker Advance D8 diffractometer) with Cu K α radiation ($\lambda = 1.5406 \text{ \AA}$). The SEM images of Sb₂Se₃ thin films were measured by FE-SEM (Hitachi SU8220) equipped with an EDS (Bruker) module. Raman spectroscopy (Horiba JobinYvon, LabRAM HR800) was measured to analyze chemical bonds of the films with 532-nm laser excitation. XPS results of the films were carried out with a monochromatic Al K α X-ray source to characterize the elemental composition of the surface by a Thermo Scientific K-Alpha+ instrument. UV-vis-NIR spectrophotometer (SOLID 3700) was used to measure the steady-state optical absorption of Sb₂Se₃ films. The *J-V* performance of Sb₂Se₃ films was measured by a Keithley 2400 equipment under solar-simulated AM 1.5G sunlight (a typical solar simulator based on xenon-lamp (Oriel Sol 3A)). Halogen lamp connected with monochromator (SPIEQ200) was as the single illuminated source to characterize the EQE of Sb₂Se₃ solar cells.

Table S1. The statistical chemical composition of Sb_2Se_3 films with 10 samples each for Sb_2Se_3 , Se-rich and Sb-rich films probed by energy dispersive X-ray spectroscopy.

Sb_2Se_3 films	Se (at%)	Sb (at%)	Se/Sb
Sb-rich	58.40±0.92	41.60±0.92	1.41±0.05
Se-rich	60.95±0.35	39.05±0.35	1.56±0.02
Sb_2Se_3	59.96±0.36	40.04±0.36	1.50±0.02

Table S2. Results of biexponential fitting of GSB decay.

Sb_2Se_3 films	A_1	τ_1 (ps)	A_2	τ_2 (ps)	τ_{ave} (ps)
Sb-rich	-0.135	0.08	-0.865	0.31	0.30
Se-rich	-0.031	0.07	-0.969	0.20	0.20

Table S3. Results of biexponential fitting that rely on Levenberg-Marquardt algorithm.

Sample	A_1	τ_1 (ps)	A_2	τ_2 (ps)	τ_{ave} (ps)
Sb-rich Sb_2Se_3	0.115	14.6	0.649	3891.1	3888.5
Se-rich Sb_2Se_3	0.128	20.2	0.703	4256.3	4252.7
Sb-rich Sb_2Se_3 on CdS	0.024	8.3	0.447	1972.7	1972.2
Se-rich Sb_2Se_3 on CdS	0.075	4.7	0.553	1410.7	1410.1

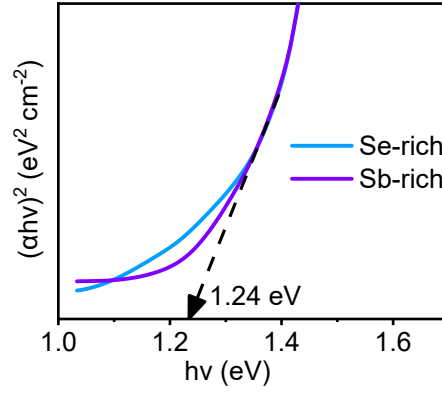


Figure S1. UV-vis spectroscopy of Sb-rich and Se-rich Sb_2Se_3 films.

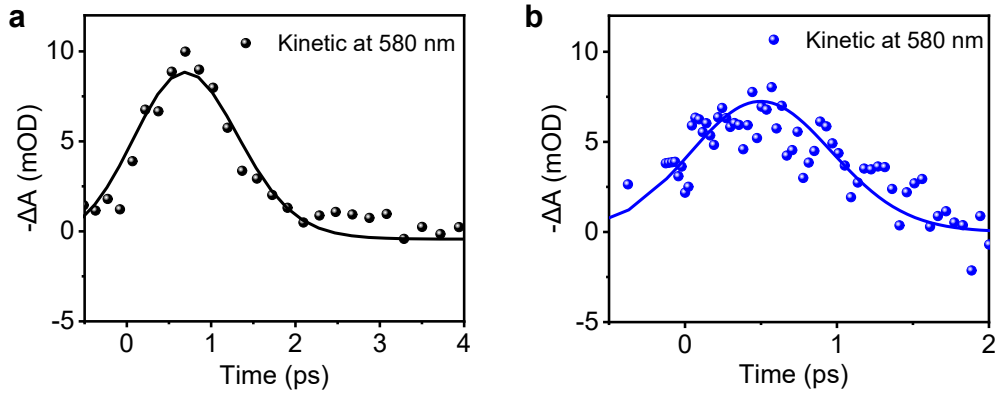


Figure S2. Transient kinetic traces of (a) Sb-rich Sb_2Se_3 films and (b) Se-rich Sb_2Se_3 films and their fits (solid lines).

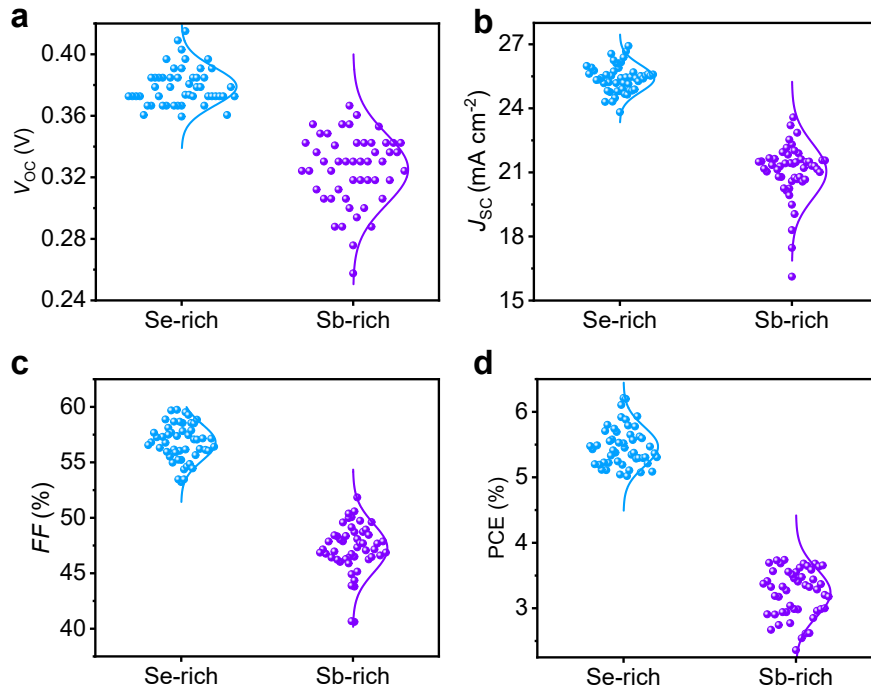


Figure S3. (a-d) Statistical V_{OC} , J_{SC} , FF and PCE of Se-rich and Sb-rich Sb_2Se_3 devices.

Note S1. TA characterization.

The carrier dynamics was obtained from femtosecond transient absorption spectroscopy (fs-TAS). The fs-TAS instrument we used comprises ultrafast laser system, light conversion, sample transport, delay generation and probe detection. Broad UV/vis probe band is achieved by generating continuous white light in different media (e.g., CaF₂, sapphire); the choice of media and seed energy determine the range of spectrum and shape of the generated continuous light.¹ Transient absorption (TA) measurements are performed on a pump-probe system (Helios, Ultrafast System) with the maximum delay time of ~8000 ps using a motorized optical delay line under ambient conditions. The pump pulses at 400 nm wavelength (~800 μW average power at the sample) are delivered by an ultrafast optical parametric amplifier (OPera Solo) excited by a regenerative amplifier (Coherent Astrella, 800 nm, 1 kHz), seeded with a mode-locked Ti:sapphire oscillator (Coherent Vitera, 800 nm, 80 MHz) and pumped with a LBO laser (Coherent Evolution-50C, 1 kHz system). A small amount of 800 nm femtosecond pulses from the regenerative amplifier are used to pump a sapphire crystal to create a 320-780 nm white light continuum as probe pulses. Interpulse fluctuations of continuous white light probe pulses are corrected using a self-splitting reference beam (Reference). The mechanical chopper operates at a frequency of 500 Hz and is used for modulating the pump pulses. The TA decay is fitted with biexponential function:

$$\Delta A(t) = \sum_{i=1}^N A_i \exp(-t/\tau_i) \quad (1)$$

where t is the probe delay time, A_i and τ_i are amplitudes of TA and decay lifetimes, respectively. Here, the number of components (N) to satisfactorily fit the experimental data is 2. The average lifetime (τ_{ave}) was estimated from the fitting parameters according to the following equation.

$$\tau_{ave} = \sum_{i=1}^2 A_i \tau_i^2 / A_i \tau_i \quad (2)$$

Note S2. Calculation of trap-assisted recombination rate coefficient.

Equation S3 shows the evolution of carrier density after pump excitation.

$$-\frac{dN(t)}{dt} = k_1N + k_2N^2 \quad (3)$$

where $N(t)$ is the carrier density; k_1 is the monomolecular recombination rate coefficient at low carrier density; k_2 is the bimolecular recombination rate coefficient at high carrier density. Herein, we only focus on k_1 because trap-assisted recombination is one of monomolecular recombination. Thus, we can simplify the Equation S3 as follows:

$$-\frac{dN(t)}{dt} = k_1N \quad (4)$$

For further adjustment,

$$\frac{dN(t)}{N} = -k_1dt \quad (5)$$

Integrate both sides of the Equation S5 at the same time.

$$\int_0^N \frac{dN}{N} = -k_1 \int_0^t dt \quad (6)$$

Then,

$$\ln N = -k_1t + C \quad (7)$$

Where C is a constant. When $t = 0$, $N = N_0$ (N_0 is the initial carrier density). Thus, $C = \ln N_0$.

Half-lifetime ($t_{1/2}$) is generally used for first-order decay process. Suppose that when N_0 decays to half, $t = t_{1/2}$,

$$\ln \frac{1}{2}N_0 = -k_1t_{1/2} + \ln N_0 \quad (8)$$

Therefore,

$$t_{1/2} = \frac{\ln 2}{k_1} \quad (9)$$

Changing the excitation intensity will vary the initial carrier density. Since k_1 is the intrinsic decay rate constant, the time for the initial carrier density decaying to half ($t_{1/2}$) is fixed, thus the decay curve of free electron-hole recombination remains unchanged and independent of excitation intensity.

If trap-assisted recombination of electrons and holes occurs, we assume that holes are trapped firstly and then recombine with electrons. In this case, we denote N_{TH} and N_{FH} as density of trapped and free holes, respectively; N_{TE} and N_{FE} as density of electrons bound to trapped holes and free electrons, respectively; k_{TH} as intrinsic trapping rate constant of holes and k_{TE} as trapping rate constant of electrons captured by pre-trapped holes; N_0 as the initial carrier density; t_{TH} as hole capture time and t_{TE} as electron capture time. Since the holes are trapped at first, we have following equations.

$$\frac{dN_{TH}}{dt} = N_{FH}k_{TH} = (N_0 - N_{TH})k_{TH} \quad (10)$$

Integrate both sides of Equation S10 at the same time.

$$\int_0^N \frac{dN_{TH}}{N_0 - N_{TH}} = k_{TH} \int_0^{t_{TH}} dt \quad (11)$$

According to the integration result, when $N = N_{TH}$,

$$\ln\left(1 - \frac{N_{TH}}{N_0}\right) = -k_{TH}t_{TH} \quad (12)$$

The trapped holes then continue to attract free electrons,

$$\frac{dN_{TE}}{dt} = N_{FE}(N_{TH} - N_{TE})k_{TE} \approx (N_{TH} - N_{TE})k_{TE} \quad (13)$$

Integrate both sides of Equation S13 at the same time.

$$\int_0^N \frac{dN_{TE}}{N_{TH} - N_{TE}} = k_{TE} \int_0^{t_{TE}} dt \quad (14)$$

According to the integration result, when $N = N_{TE}$,

$$\ln\left(1 - \frac{N_{TE}}{N_{TH}}\right) = -k_{TE}t_{TE} \quad (15)$$

It can be seen that the t_{TE} is related to N_{TH} . When the N_{TH} increases, the electron capture rate increases, and the recombination rate of electrons and holes increases, thus the dynamic decay accelerates. The variability of excitation intensity will change the N_0 , which in turn influences the N_{TH} and carrier capture time t_{TE} , leading to different decay curves.

Reference

- 1 L. A. Baker, S. E. Greenough and V. G. Stavros, *J. Phys. Chem. Lett.*, 2016, **7**, 4655-4665.