

Supporting Information for

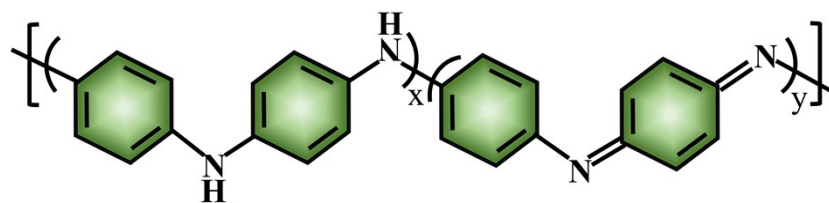
Interface functionalization of mesoporous ruthenium film with polyaniline for enhanced hydrogen evolution electrocatalysis at all-pH value

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Scheme. S1 Molecular structure of PANI.

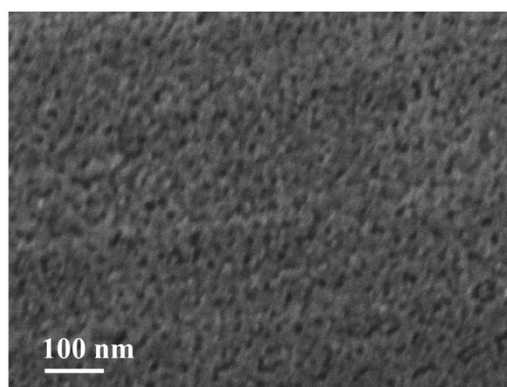


Fig. S1 SEM image of the mRu film.

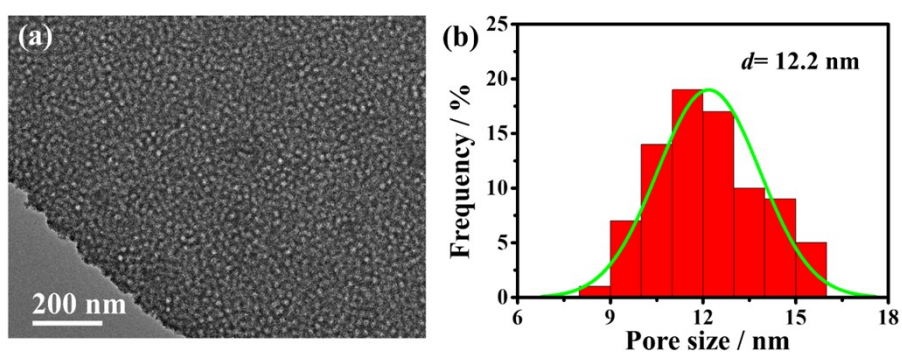


Fig. S2 (a) TEM image and (b) corresponding pore size distribution of the mRu@PANI/CP. The inserted values indicate the average values.

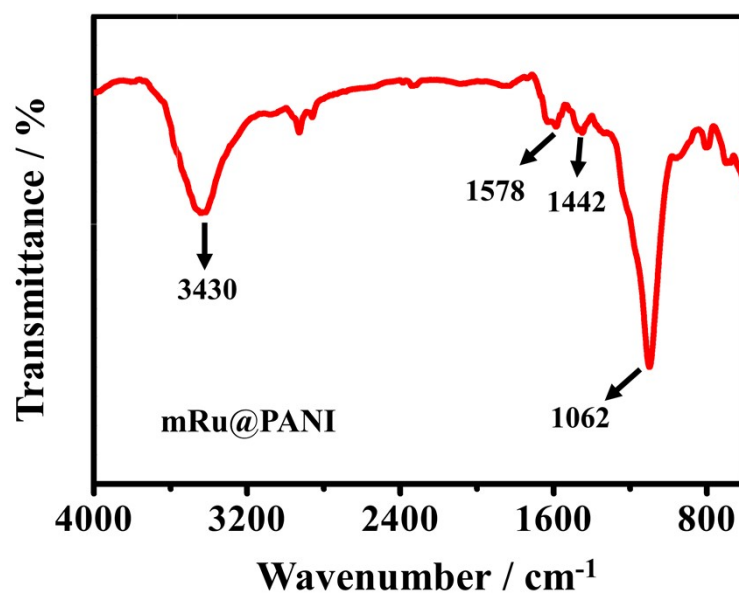


Fig. S3 FTIR spectra of mRu@PANI.

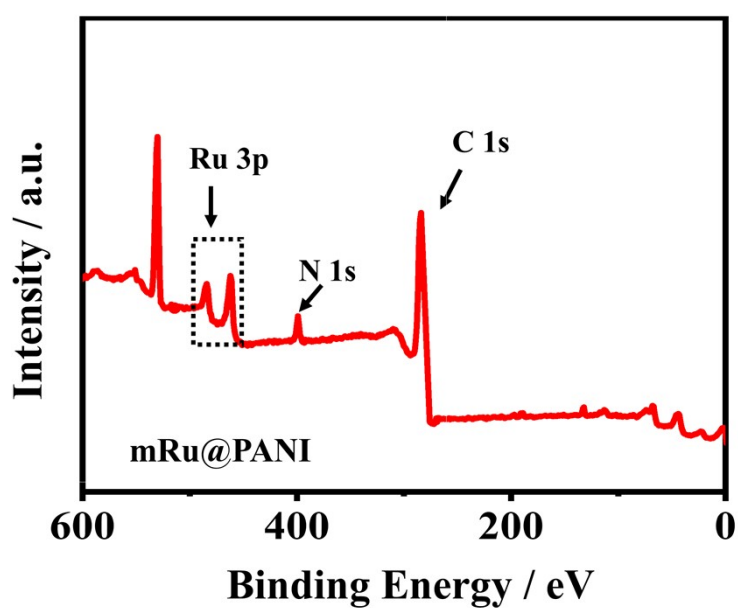


Fig. S4 XPS survey spectrum of mRu@PANI.

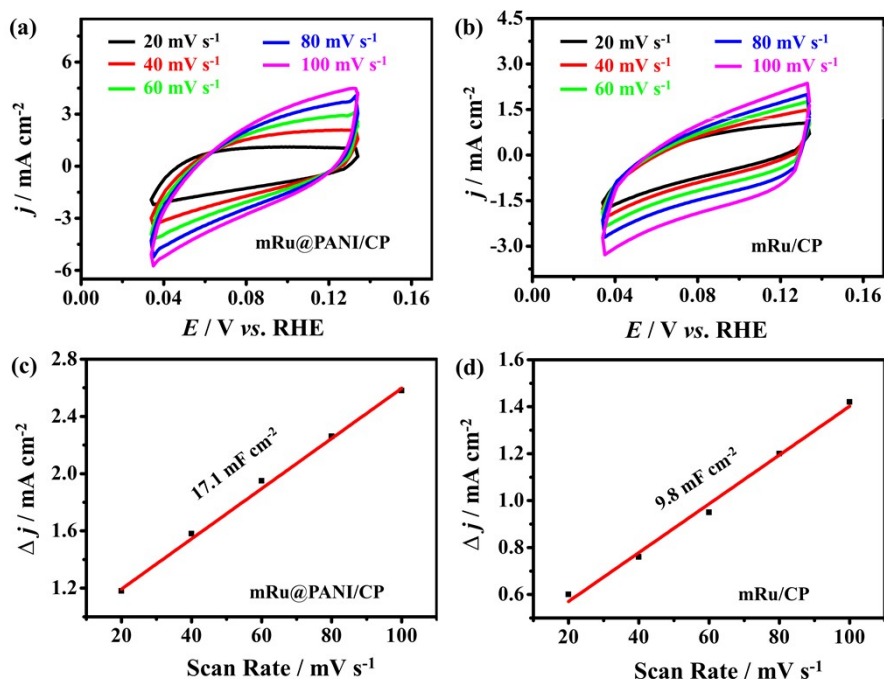


Fig. S5 Cyclic voltammogram (CV) curves of (a) mRu@PANI/CP and (b) mRu with different scan rates in 0.5 M H₂SO₄. The corresponding capacitive current densities of (c) mRu@PANI/CP and (d) mRu/CP.

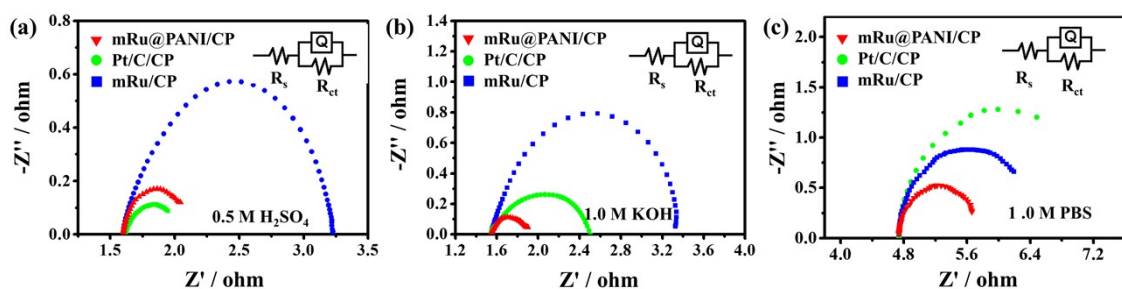


Fig. S6 EIS spectra of various catalysts in (a) 0.5 M H₂SO₄, (b) 1.0 M KOH and (c) 1.0 M PBS. The inserts show corresponding equivalent circuits.

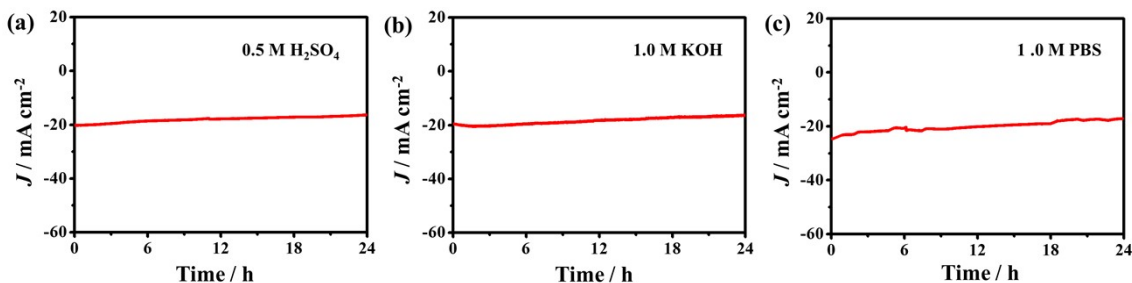


Fig. S7 Chronoamperometry (i-t) curves of mRu@PANI/CP in (a) 0.5 M H₂SO₄, (b) 1.0 M KOH and (c) 1.0 M PBS.

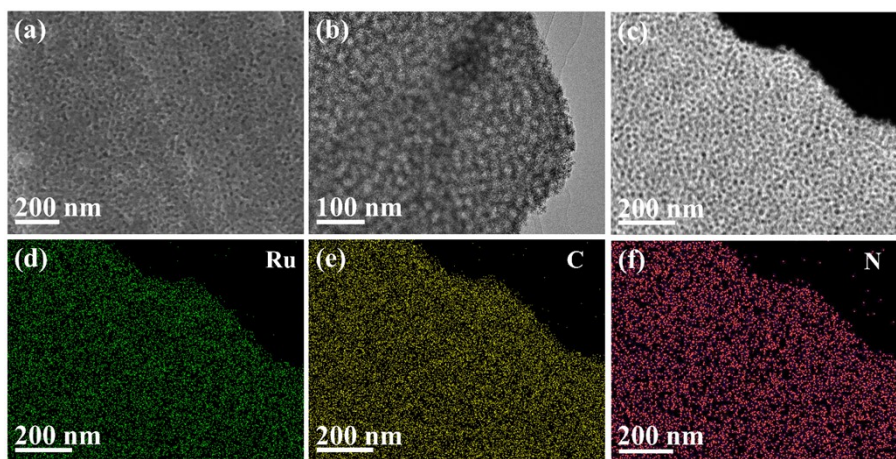


Fig. S8 The (a) SEM, (b) TEM, (c) HAAD-STEM and (d) corresponding EDS mapping of mRu@PANI/CP after stability test in 0.5 M H₂SO₄.

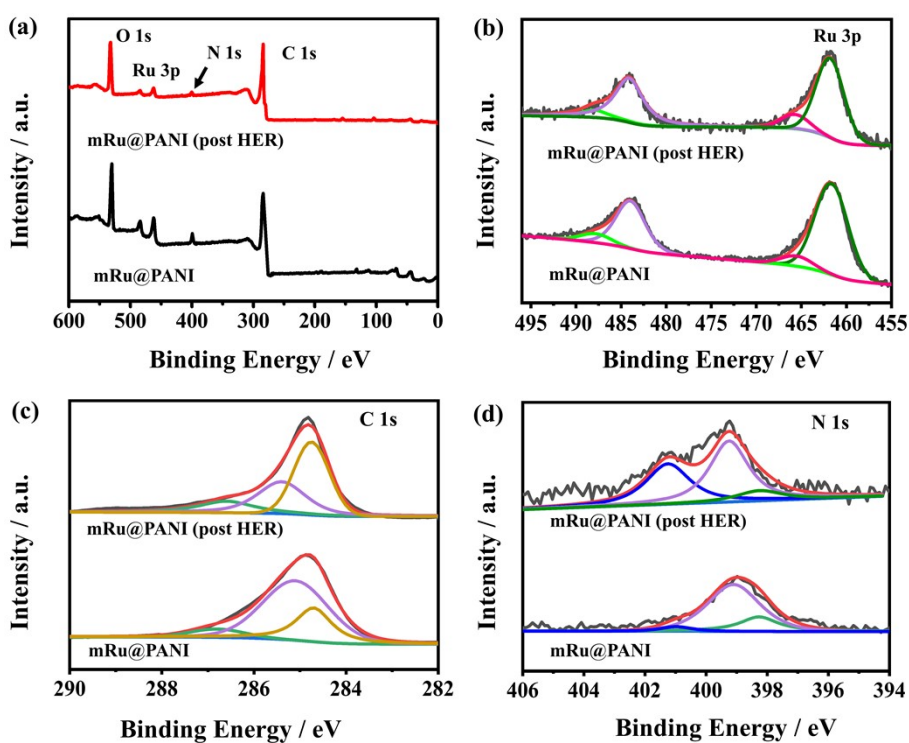


Fig. S9 (a) XPS survey spectra, (b) Ru 3p, (c) C 1s, and (d) N 1s spectra of mRu@PANI before and after stability test in 0.5 M H₂SO₄.

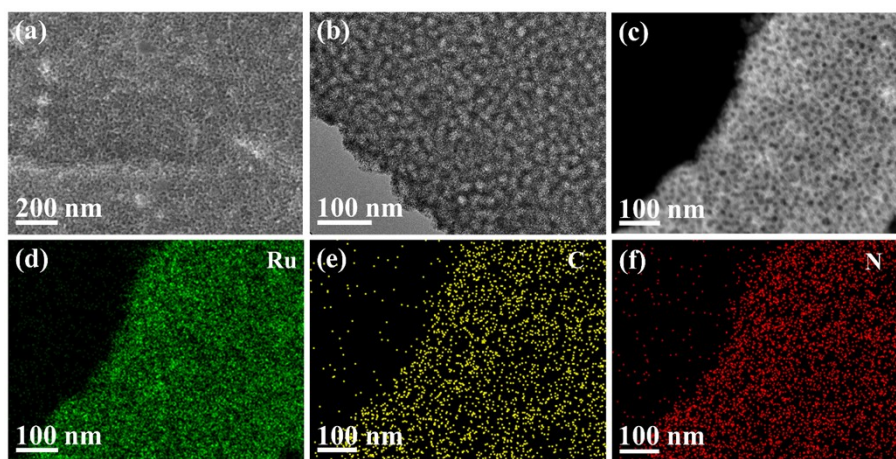


Fig. S10 The (a) SEM, (b) TEM, (c) HAAD-STEM and (d) corresponding EDS mapping of mRu@PANI/CP after stability test in 1.0 M KOH.

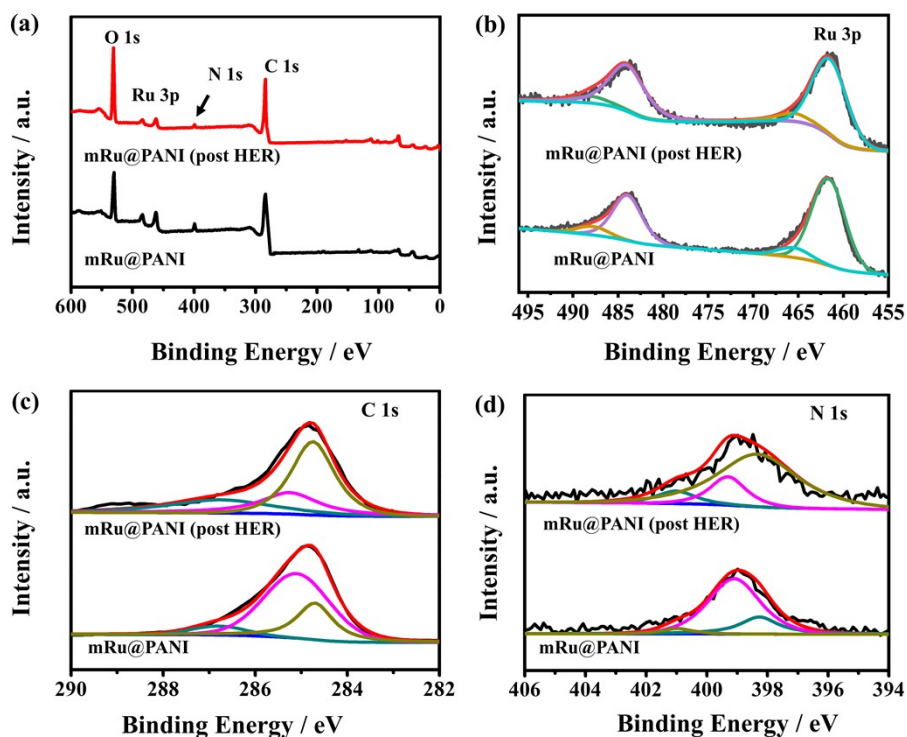


Fig. S11 (a) XPS survey spectra, (b) Ru 3P, (c) C 1s, and (d) N 1s spectra of mRu@PANI before and after stability test in 1.0 M KOH.

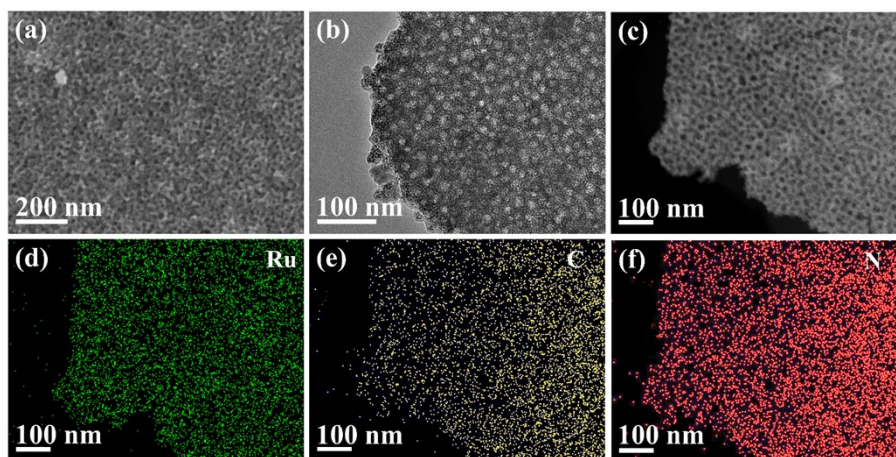


Fig. S12 The (a) SEM, (b) TEM, (c) HAAD-STEM and (d) corresponding EDS mapping of mRu@PANI/CP after stability test in 1.0 M PBS.

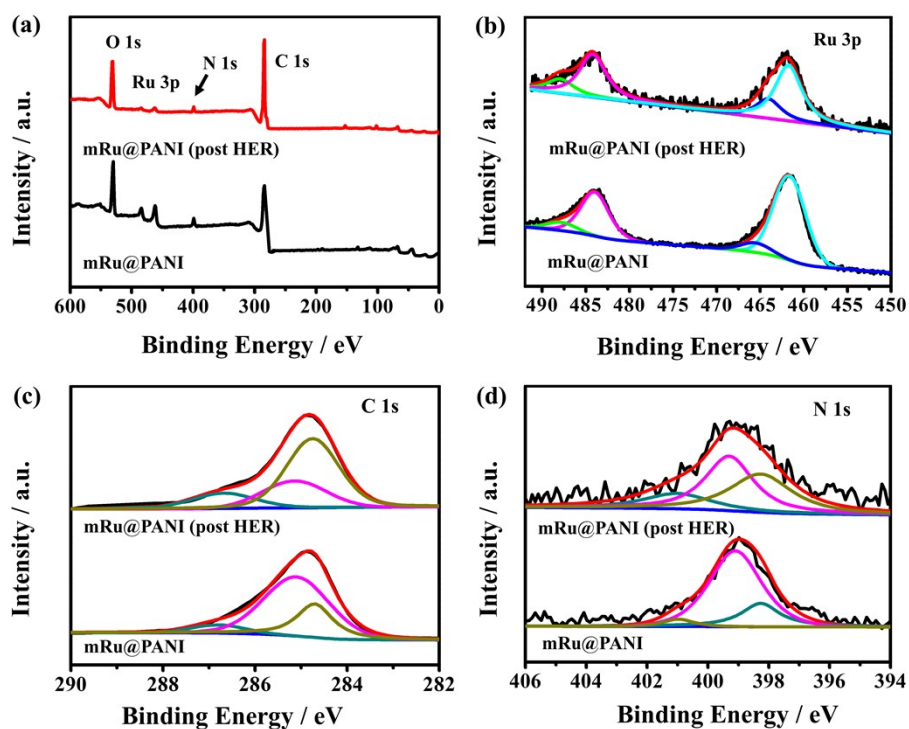


Fig. S13 (a) XPS survey spectra, (b) Ru 3P, (c) C 1s, and (d) N 1s spectra of mRu@PANI before and after stability test in 1.0 M PBS.

Table S1. HER performance comparison between the mRu@PANI/CP and some other reported Ru-based electrocatalysts in 0.5 M H₂SO₄.

Catalyst	Electrolytes	Overpotential at 10 mA cm ⁻² (mV)	Ref.
mRu@PANI/CP	0.5 M H ₂ SO ₄	30	This work
Ru _x Fe _y P-NCS/CNF	0.5 M H ₂ SO ₄	66	1
RuNi NSs@PANI	0.5 M H ₂ SO ₄	42	2
Ru _x Se@MoS ₂	0.5 M H ₂ SO ₄	120	3
Rh ₂ P	0.5 M H ₂ SO ₄	55	4
Ru-MC	0.5 M H ₂ SO ₄	40	5
Ru@N/S/TiO ₂ /rGO	0.5 M H ₂ SO ₄	60	6
Ni-Ru-P	0.5 M H ₂ SO ₄	35	7
RuC	0.5 M H ₂ SO ₄	35	8

Table S2. HER performance comparison between the mRu@PANI/CP and some other reported electrocatalysts in 1.0 M KOH.

Catalyst	Electrolytes	Overpotential at 10 mA cm ⁻² (mV)	Ref.
mRh@PANI/CP	1.0 M KOH	27	This work
Ru-MoS ₂ /CC	1.0 M KOH	41	9
Ru-MoO ₂	1.0 M KOH	29	10
Ru@CN-0.16	1.0 M KOH	32	11
Ru/Cu ₂₊₁ O NT/CuF	1.0 M KOH	44	12
Ru-Ru ₂ P/PC	1.0 M KOH	43.4	13
RuCu NWs	1.0 M KOH	32	14
S-RuP@NPSC-900	1.0 M KOH	92	15
CuRu	1.0 M KOH	85	16

Table S3. HER performance comparison between the mRu@PANI/CP and some other reported electrocatalysts in 1.0 M PBS.

Catalyst	Electrolytes	Overpotential at 10 mA cm ⁻² (mV)	Ref.
mRh@PANI/CP	1.0 M PBS	24	This work
Ru@CN-0.16	1.0 M PBS	100	11
RuCu	0.1 M PBS	73	14
CuRu	1.0 M PBS	91	16
Ni-Ru	1.0 M PBS	80	17
Ni ₂ P-Ru ₂ P/CCG	1.0 M PBS	113	18
Ru-rGo	1.0 M PBS	72	19
RuSA–N-Ti ₃ C ₂ Tx	1.0 M PBS	81	20
Rh _x Ru _{100-x} @UiO-66-NH ₂	1.0 M PBS	114	21

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