

Supporting Information

Suppressing Water Clusters by “Hydrotropic” Ionic Liquid for Highly Stable Aqueous Lithium-ion Batteries

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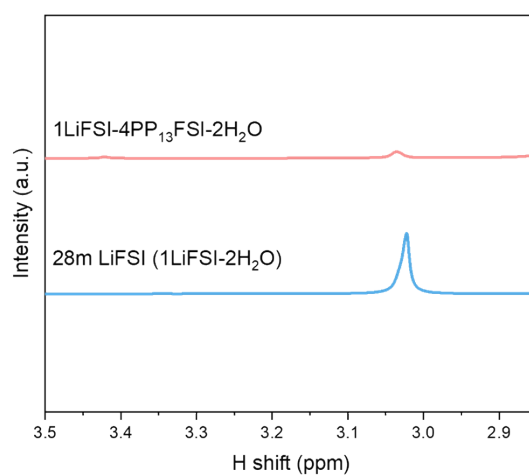


Figure S1. ^1H measurement for different aqueous electrolytes with and without PP₁₃FSI.

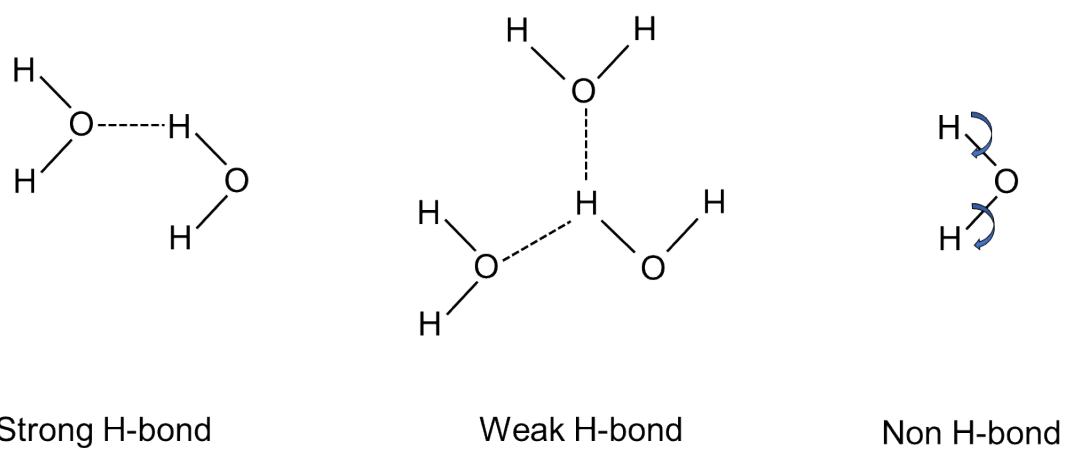


Figure S2. Corresponding vibrational structure of three types of O-H bonds.

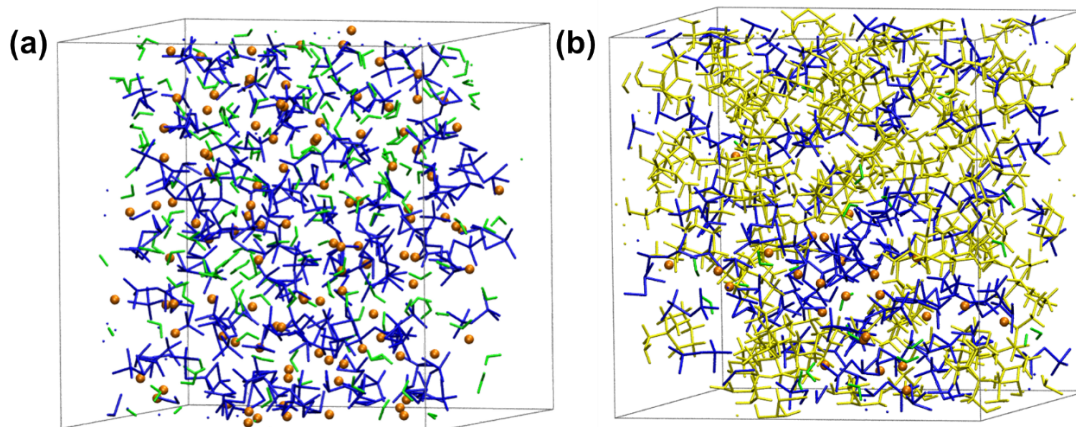


Figure S3. MD simulations results for **a)** 35 m LiFSI and **b)** 1LiFSI-4PP₁₃FSI-1H₂O.

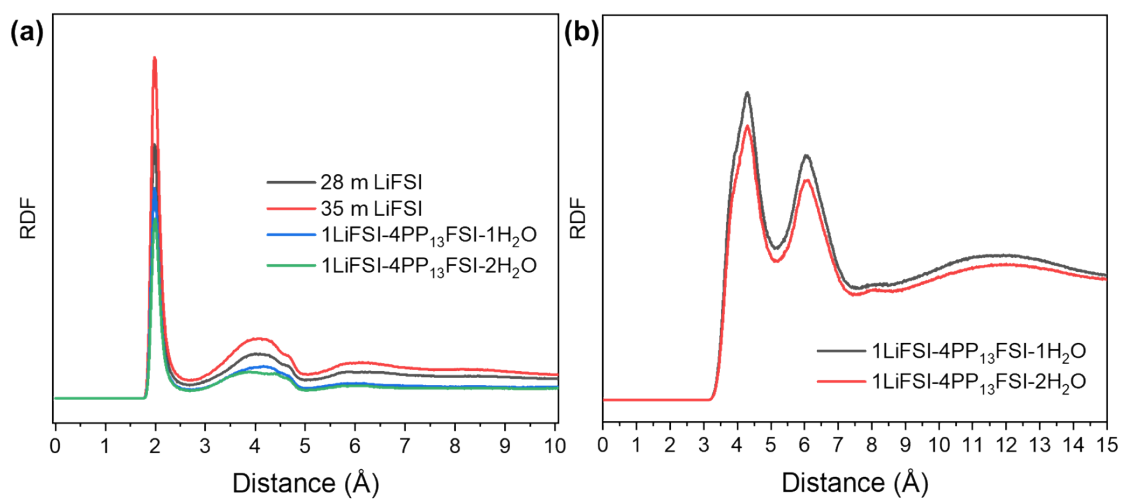


Figure S4. RDF plots for **a)** Li-O (FSI⁻) and **b)** Li-N (PP₁₃⁺) in different electrolytes.

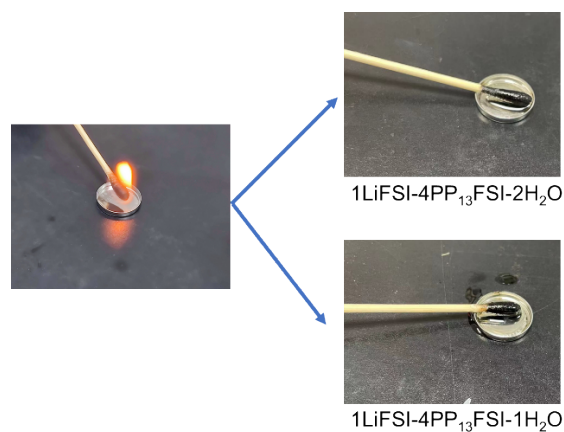


Figure S5. Flammability testing. An ignited cotton swab was immersed in the 1LiFSI-4PP₁₃FSI-2H₂O a) and 1LiFSI-4PP₁₃FSI-1H₂O b) electrolyte. The fire of cotton swab was extinguished in 1LiFSI-4PP₁₃FSI-xH₂O-based electrolytes.

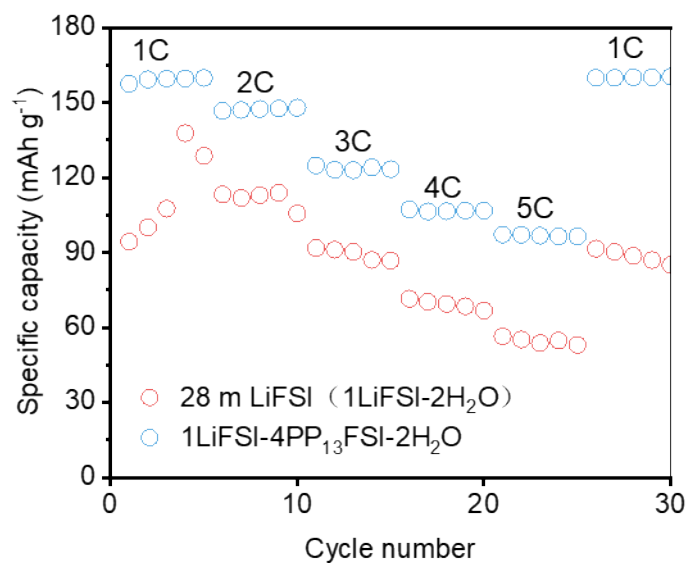


Figure S6. Rate performance of aqueous L-Li₄Ti₅O₁₂||LiMn₂O₄ full cells with the 28 m LiFSI electrolyte (red circles) and the 1LiFSI-4PP₁₃FSI-2H₂O electrolyte (blue circles).

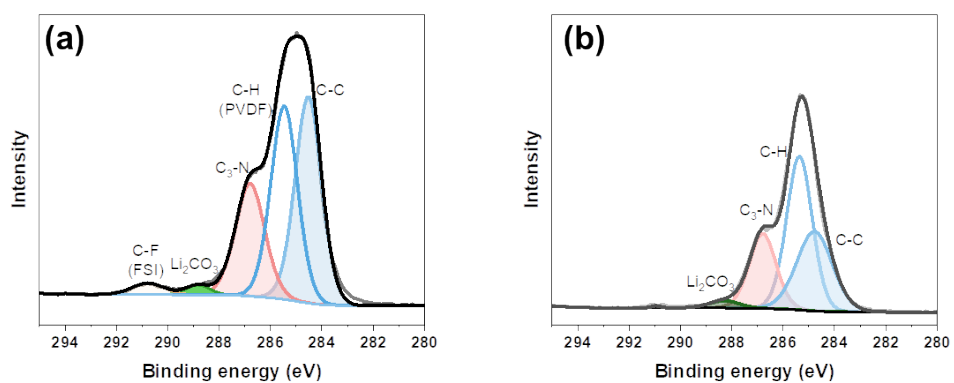


Figure S7. C 1s XPS spectra of cycled $\text{Li}_4\text{Ti}_5\text{O}_{12}$ electrodes in: **a)** 1LiFSI-4PP₁₃FSI-2H₂O electrolyte, **b)** 1LiFSI-4PP₁₃FSI-1H₂O electrolyte. (30th cycle)

Table S1. Ratios of FSI⁻ anions coordinated with Li⁺ in the first coordination shell based on MD simulations for different electrolytes.

Electrolytes	Ratios of Li-FSI coordination in the first coordination shell of Li ⁺
28 m LiFSI	80.11%
35 m LiFSI	82.38%
1LiFSI-4PP ₁₃ FSI-2H ₂ O	87.79%
1LiFSI-4PP ₁₃ FSI-1H ₂ O	95.99%

Table S2. The number of molecules employed for MD simulations of different aqueous electrolytes.

Electrolytes	Number of molecules		
	LiFSI	H ₂ O	PP ₁₃ FSI
28 m LiFSI	200	400	0
35 m LiFSI	200	320	0
1LiFSI-4PP ₁₃ FSI-2H ₂ O	100	200	400
1LiFSI-4PP ₁₃ FSI-1H ₂ O	100	100	400