

Supplementary Information for

Fabrication of porous imidazole polymerized ionic liquids with fast ion diffusing kinetics for super lithiation anode material in lithium-ion batteries

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Results and Discussion

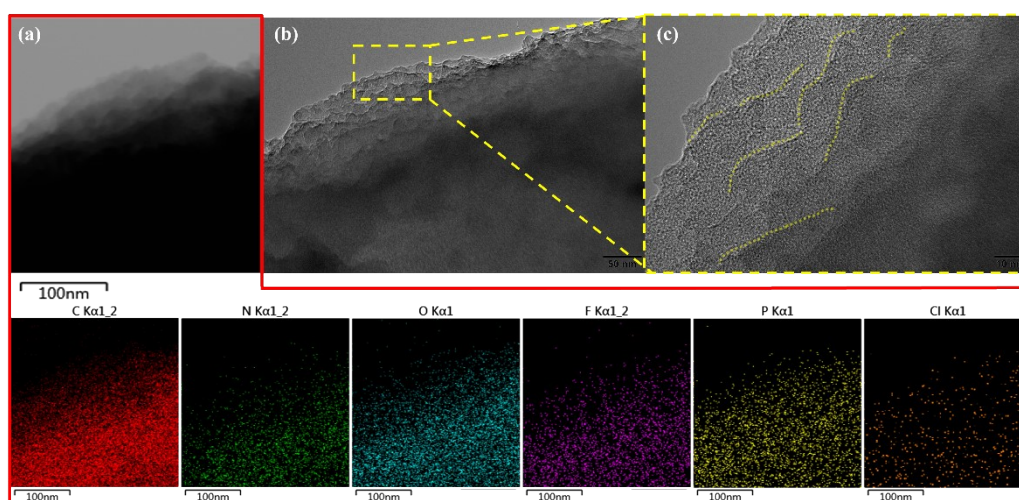


Figure S1. The EDS images (a) and HRTEM results (b, c) of PILs-Im electrode after long cycling test.

Table S1. Rate capability between PILs-Im in this work and previous organic polymer anode materials in LIBs.

Material	Charge capacity at low current density [mAh g ⁻¹]	Charge capacity at high current density [mAh g ⁻¹]	Ref.
PILs-Im	727.7 at 100 mA g ⁻¹	165.0 at 10 A g ⁻¹	This work
Tp-Azo-COF	623 at 100 mA g ⁻¹	90.76 at 2.4 A g ⁻¹	1
Covalent organic nanosheet	~ 720 at 100 mA g ⁻¹	~ 150 at 1.0 A g ⁻¹	2
Layered functionalized covalent triazine frameworks nanosheet	816 at 100 mA g ⁻¹	186 at 10 A g ⁻¹	3
Layered covalent triazine frameworks	~ 350 at 100 mA g ⁻¹	~ 50 at 10 A g ⁻¹	3
Exfoliated Schiff base network-1	542 at 100 mA g ⁻¹	212 at 5.0 A g ⁻¹	4
Covalent triazine frameworks	1418.6 at 100 mA g ⁻¹	181.8 at 10 A g ⁻¹	5
2D COF polyporphyrin	666 at 200 mA g ⁻¹	195 at 4.0 A g ⁻¹	6
Aromatic imide benzophenone-3,3',4,4'-tetracarboxylimide oligomer	1074 at 42 mA g ⁻¹	58 at 2.1 A g ⁻¹	7
Poly(chalcogenoviologen)s	799 at 50 mA g ⁻¹	252 at 2.0 A g ⁻¹	8

Table S2. Binding energy of N and C elements in different bonding states.

Potential (V vs Li ⁺ /Li)	Binding Energy (eV)					
	Imidazole N ⁺	Pyrrole N	Pyridinic N	imidazole C	pyridine C	methylene C
Pristine PILs-Im	400.4	399.7	398.5	286.3	284.9	284.4
PILs-Im at 1.0 V	400.3	399.3	398.4	286.2	284.9	284.3
PILs-Im at 0.1 V	399.7	399.2	398.2	286.0	284.9	284.2

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