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Supporting Information

Boosting the Oxygen Reduction Reaction Behaviour of Atomic Fe-N₄ Active Sites

in the Porous Honeycomb-like Carbon via P Heteroatom Doping

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Fig. S1. The XRD pattern of Fe-N₄/NP-PHC.



Fig. S2. SEM image of Fe-N₄/NP-PHC.



Fig. S3. TEM image of Fe-N₄/NP-PHC.



Fig. S4. N_2 adsorption-desorption isotherms and the corresponding pore distribution of Fe-N₄/NP-PHC.



Fig. S5. SEM images of Fe-M-CA.



Fig. S6. TEM images of Fe-M-CA.



Fig. S7. XRD pattern of Fe-M-CA.



Fig. S8. The survey XPS spectra of Fe-N₄/NP-PHC.



Fig. S9. EXAFS fitting results of Fe-N₄/NP-PHC at k space.



Fig. S10. (a) SEM and (b) TEM (c) HAADF-STEM and (d) elemental mapping images of NP-PHC.



Fig. S11. CV curves of Fe-N₄/NP-PHC, Pt/C and NP-PHC.



Fig. S12. Kinetic current density (j_k) of Fe-N₄/NP-PHC, Pt/C and NP-PHC at 0.85 V.



Fig. S13. Rotating ring disk electrode polarization curves of Fe-N₄/NP-PHC and Pt/C.



Fig. S14. (a) ORR polarization curves and (b) corresponding Tafel plots of Fe-N₄/NP-PHC and Fe-N₄/N-PHC.



Fig. S15. ORR polarization curves before and after 3000 potential cycles for Fe- $N_4/NP\text{-}PHC$ catalyst.



Fig. S16. SEM of Fe-N₄/NP-PHC after ORR durability measurement.



Fig. S17. TEM of Fe-N₄/NP-PHC after ORR durability measurement.



Fig. S18. (a) HAADF-STEM and (b) mapping images of Fe-N₄/NP-PHC after ORR durability measurement.



Fig. S19. XPS of Fe-N₄/NP-PHC after ORR durability measurement.



Fig. S20. XRD spectra of Fe-N₄/NP-PHC after stability test.



Fig. S21. Raman spectra of Fe-N₄/NP-PHC after stability test.

Samples	Shell	CN	R(Å)	σ ² (Å ²)	ΔE_0 (eV)	R-factor
Fe-N ₄ /NP-PHC	Fe-N	4.1	2.01	0.006	-4.58	0.012

Table S1. Structural parameters extracted from EXAFS data fitting of $(S_0^2 = 1)$

CN is the coordination number; R is interatomic distance; σ^2 is Debye-Waller factor (a measure of thermal and static disorder); ΔE_0 : edge-energy shift (the difference between the zero kinetic energy value of the sample and that of the theoretical model); R factor is used to evaluate the goodness of the fitting.

Catalyst	Eonset(V vs.RHE)	E _{1/2} (V vs.RHE)	References
Fe-N ₄ /NP-PHC	1.0 V	0.89V	This work
Ni-N ₄	0.97 V	0.86 V	[1]
Fe/OES	1.0 V	0.85 V	[2]
FePc&rGO	0.98 V	0.89 V	[3]
Co-SAC/NC	1.019 V	0.884 V	[4]
Co-N-C SA/HCF	0.928 V	0.801 V	[5]
Fe ₃ C	0.987 V	0.855 V	[6]
Co-N ₃ C ₁	0.904 V	0.824 V	[7]
FeSAs/PTF-600	1.01 V	0.87 V	[8]
CNT@SAC-Co/NCP		0.87 V	[9]
SACe-N/PC	1.0 V	0.88 V	[10]
NBCNT-10	0.958 V	0.82 V	[11]
CoFe/S-N-C		0.855 V	[12]
WN-Ni@N,P-CNT	1.02 V	0.84 V	[13]
Co,Nb-MoS ₂ /TiO ₂	0.96 V	0.86 V	[14]
3DOM P-Co ₃ O _{4-δ}	0.99 V	0.82 V	[15]
Co ₉ S ₈ @N, S–C	1.03 V	0.887 V	[16]
PdNi/Ni@N-C	1.01 V	0.89 V	[17]
Fe-doped MOF CuCoSe@HCNF		0.756 V	[18]
Co/MnO@NC	0.96 V	0.83 V	[19]

 Table S2. Performance of the reported state-of-the-art ORR SACs in 0.1 M KOH.

Table S3. Calculated ΔG for 4e⁻ ORR pathway on Fe-N₄ and Fe-N₄-P at U = 0 V vs. RHE.

$\mathbf{U} = 0 \mathbf{V}$	O ₂ (eV)	OOH* (eV)	O* (eV)	OH* (eV)	OH ⁻ (eV)
Fe-N ₄ /NP-CM	4.92	3.39978	1.17276	0.1461	0.00
Fe-N ₄ /N-CM	4.92	3.2197	1.18296	0.05421	0.00

Table S4. Calculated ΔG for 4e⁻ ORR pathway on Fe-N₄ and Fe-N₄-P at U = 1.23 V (vs. RHE).

U = 1.23 V	O ₂ (eV)	OOH* (eV)	O* (eV)	OH* (eV)	OH ⁻ (eV)
Fe-N ₄ /NP-CM	0	-0.29022	-1.28724	-1.0839	0.00
Fe-N ₄ /N-CM	0	-0.4703	-1.27704	-1.17579	0.00

Catalyst	Peak power density (mW cm ⁻²)	Charge/discha rge voltage gap (V) at 10mA cm ⁻²	Battery stability	References
Fe-N ₄ /NP-PHC	200	1.0	20 min/cycle for 100 h	This work
BCN/rGO-Co	157	1.235	for 200 h	[20]
Fe/N-G-SAC	120	0.78	1 h/cycle for 240 h	[21]
Ni ₃ Fe/Co-N-C	68	0.78	for 65 h	[22]
Pt@CoS2-NrGO	114	0.88	for 55 h	[23]
CoNP-PTCOF	53	1.2	10 min/cycle for 120 h	[24]
3DOM Co3O4		0.96	2 h/cycle for 400 h	[25]
Cop@CoNC	188.8	0.80	for 360 h 10 min	[26]
CoNP-s-IMCOF	48	1.1	cycle for 157h 20	[27]
P-CoO@PWC-2	73	0.83	min/cycle for 233.3 h	[28]
Fe-Co ₂ P@Fe-N-C	81	0.73	min/cycle for 283.3 h	[29]
NiFe@C@Co CNFs	130	0.73	20 min/cycle for 400 h	[30]

 Table S5. Performance of rechargeable Zn-air batteries assembled using SACs.

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