

*Electronic Supplementary Information*

**MXene/ZnO flexible freestanding film as dendrite-free support  
in lithium metal batteries**

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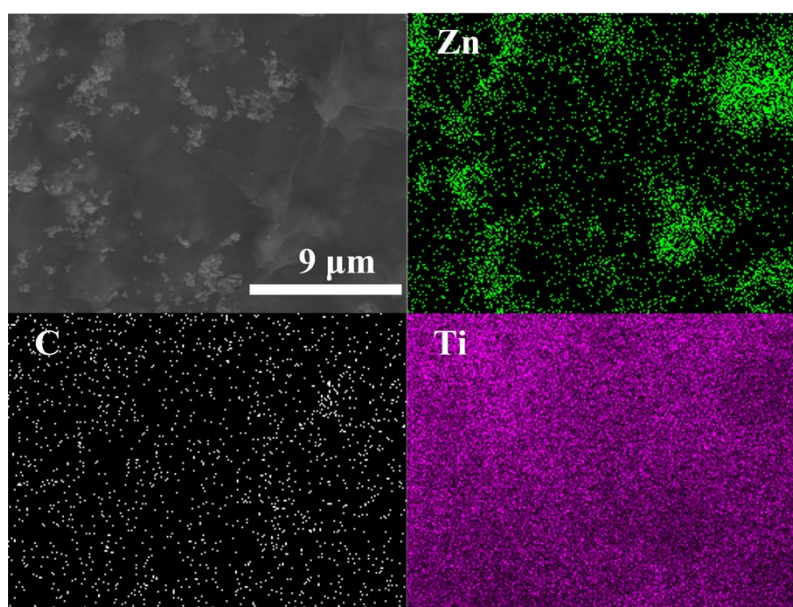


Figure S1 EDS mapping of top surface of MXene/ZnO substrate.

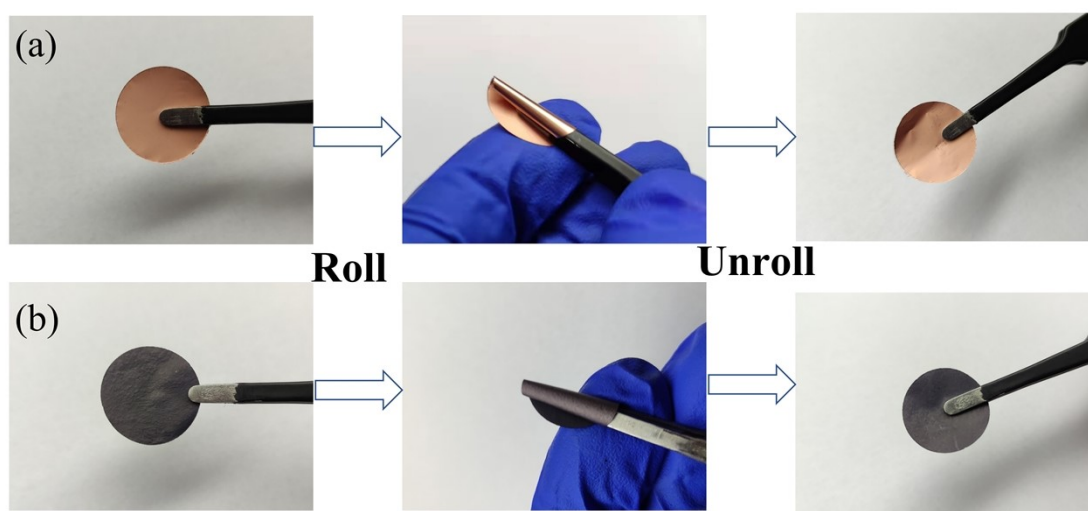


Figure S2 Optical images of roll and unroll of Cu and MXene/ZnO substrates.

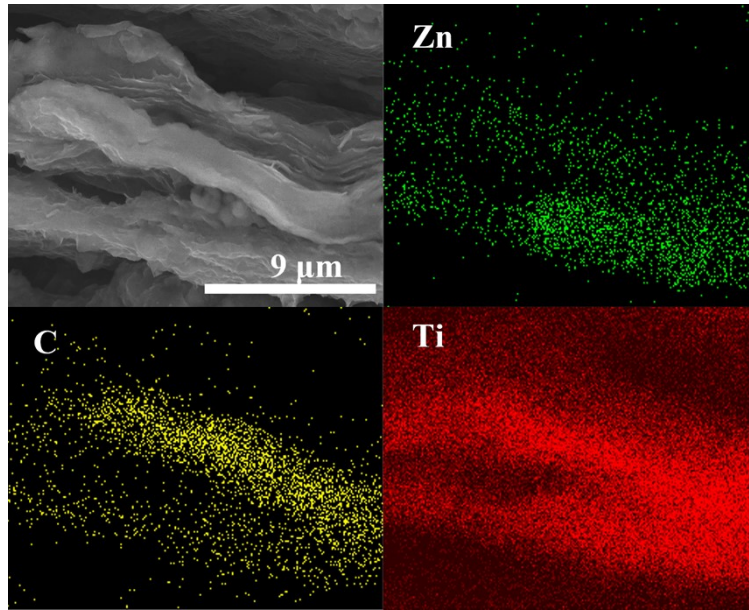


Figure S3 EDS mapping of cross-sectional surface of MXene/ZnO substrate.

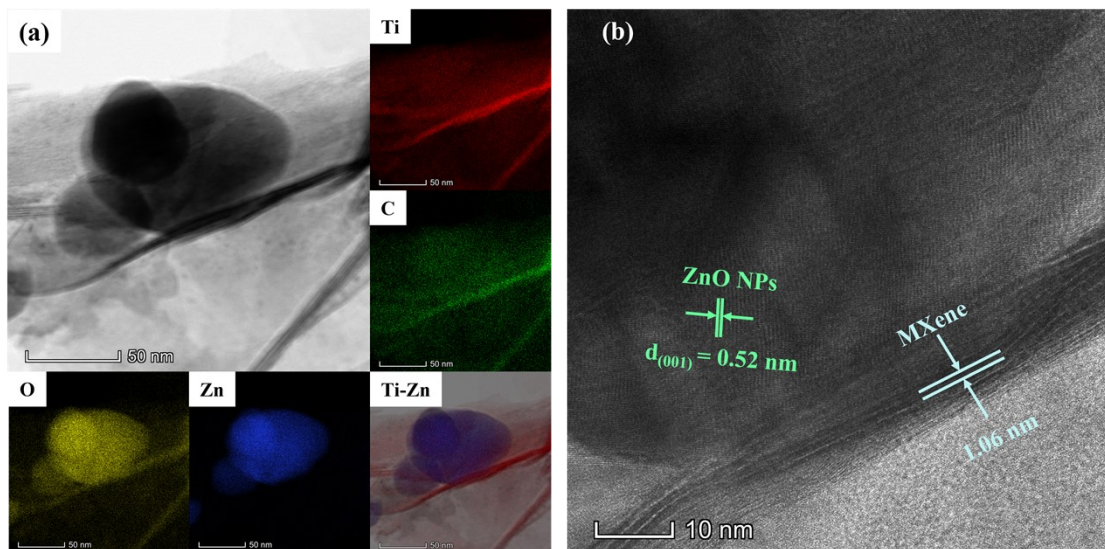


Figure S4 (a) TEM image of MXene/ZnO and corresponding elemental mapping, (b) HRTEM image of MXene/ZnO.

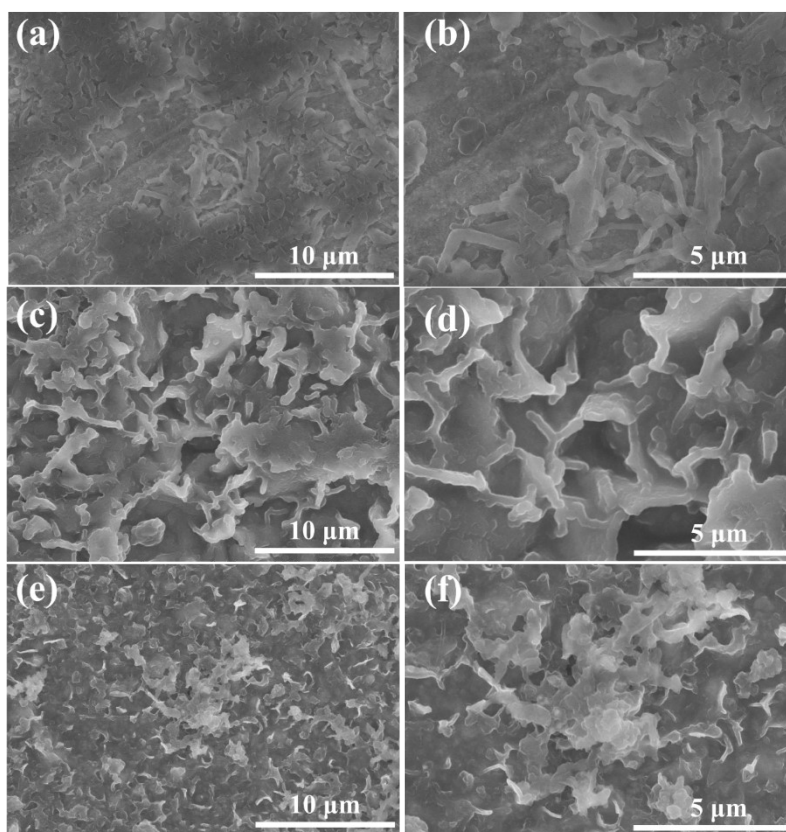


Figure S5 Morphologies of Li plating 0.2 mAh in Cu foil (a-b), MXene substrate (c-d), and MXene/ZnO (e-f).

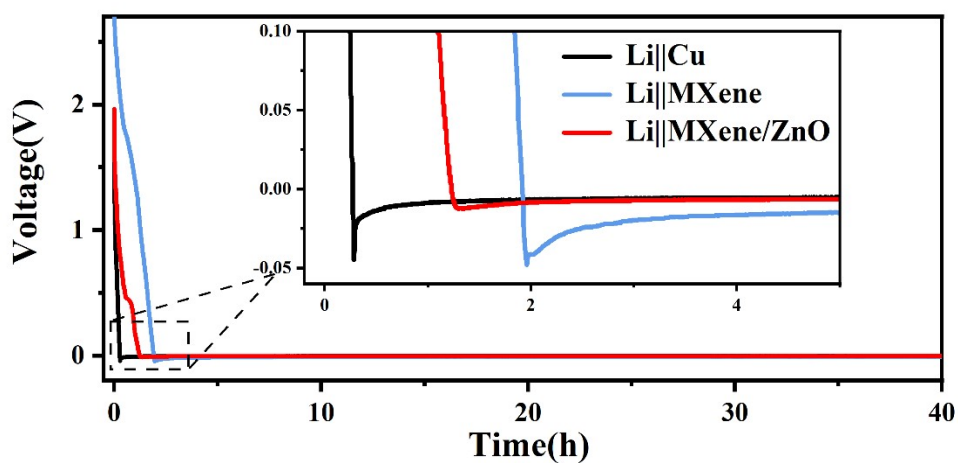


Figure S6 Voltage profiles of Li deposition on the Cu, MXene, and MXene/ZnO substrates at the current density of  $0.05 \text{ mA cm}^{-2}$ .



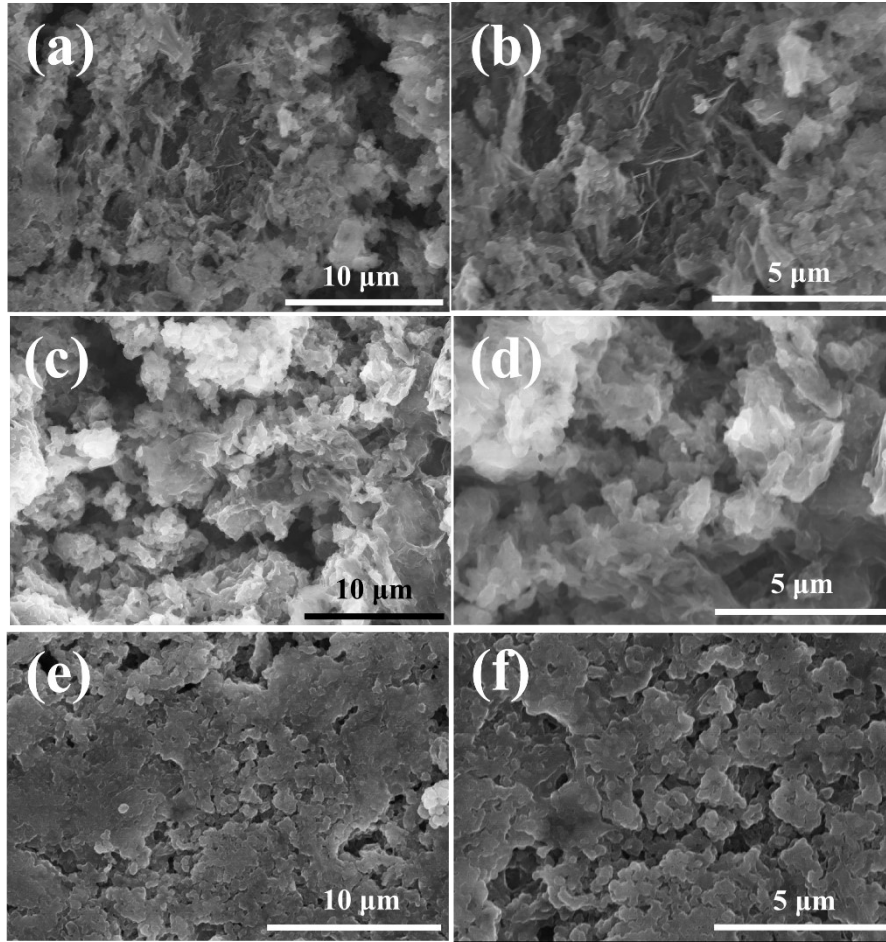


Figure S7 SEM images of Li stripping after 30 cycles (a-b) Cu, (c-d) MXene, and (e-f) MXene/ZnO substrate.

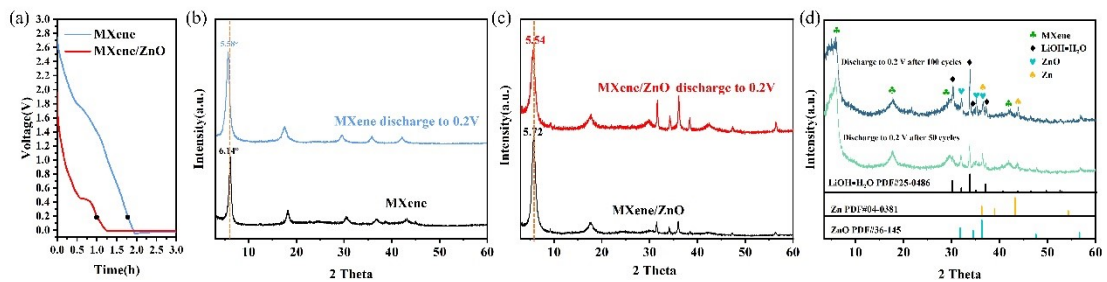


Figure S8 Discharge curves of Li||MXene and Li|| MXene/ZnO cell (a); XRD patterns of MXene (b); and MXene/ZnO (c) substrates when the half-cell was discharged to 0.2 V (vs Li/Li<sup>+</sup>). (d) XRD patterns of MXene/ZnO substrates when the half-cell was discharged to 0.2 V (vs Li/Li<sup>+</sup>) after 50 and 100 cycles at the current density of 3 mA cm<sup>-2</sup> and the capacity of 1 mAh cm<sup>-2</sup>.

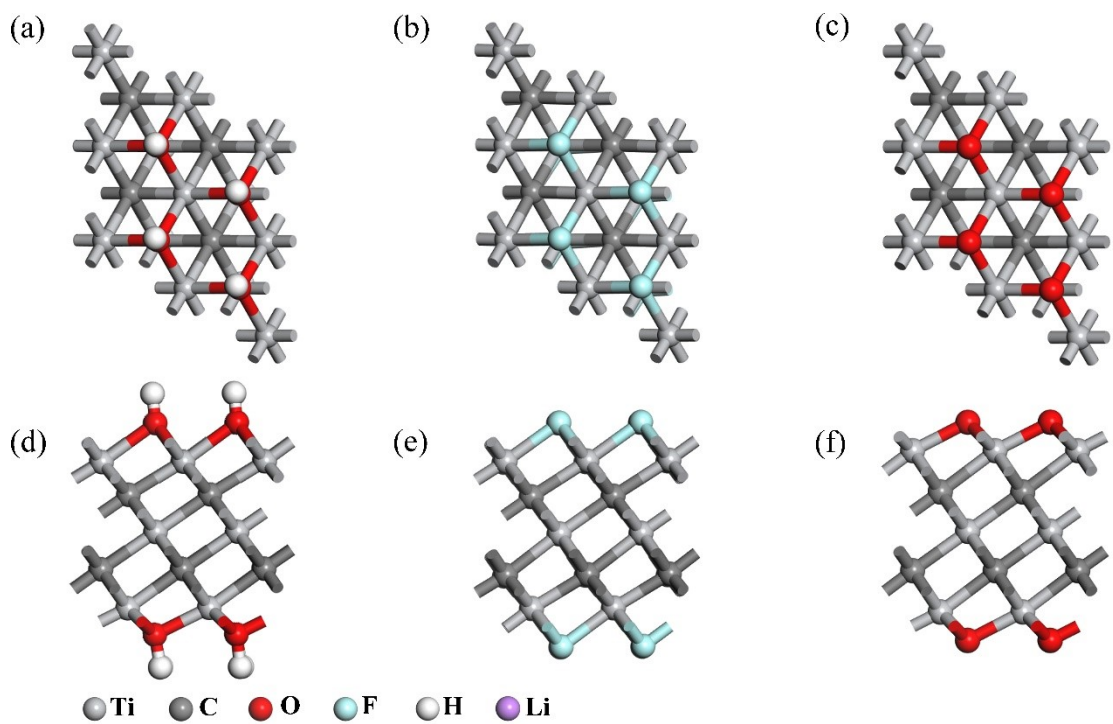


Figure S9 The top and side view of  $\text{Ti}_3\text{C}_2\text{T}_x$  with different terminating group.

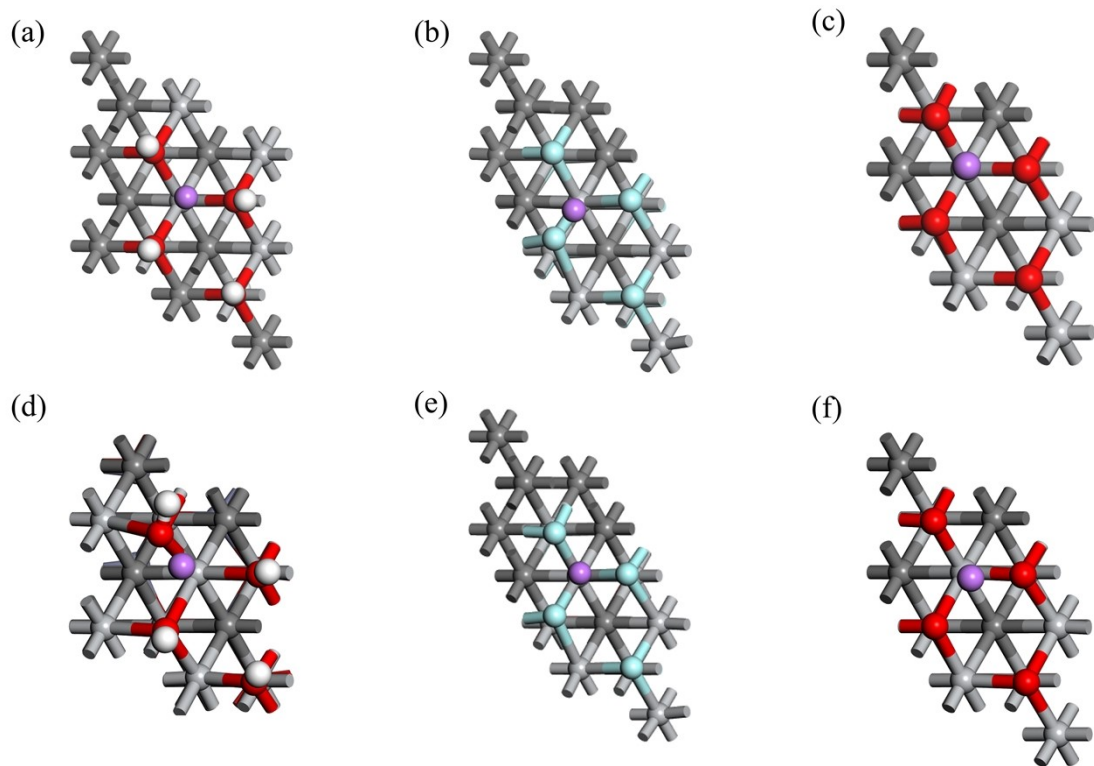


Figure S10 The top view of the optimized site of adsorbed Li on  $\text{Ti}_3\text{C}_2(\text{OH})_2$  (a),  $\text{Ti}_3\text{C}_2\text{F}_2$  (b),  $\text{Ti}_3\text{C}_2\text{O}_2$  (c),  $\text{Ti}_3\text{C}_2(\text{OH})_2@\text{ZnO}$  (d),  $\text{Ti}_3\text{C}_2\text{F}_2@\text{ZnO}$  (e), and  $\text{Ti}_3\text{C}_2\text{O}_2@\text{ZnO}$  (f).

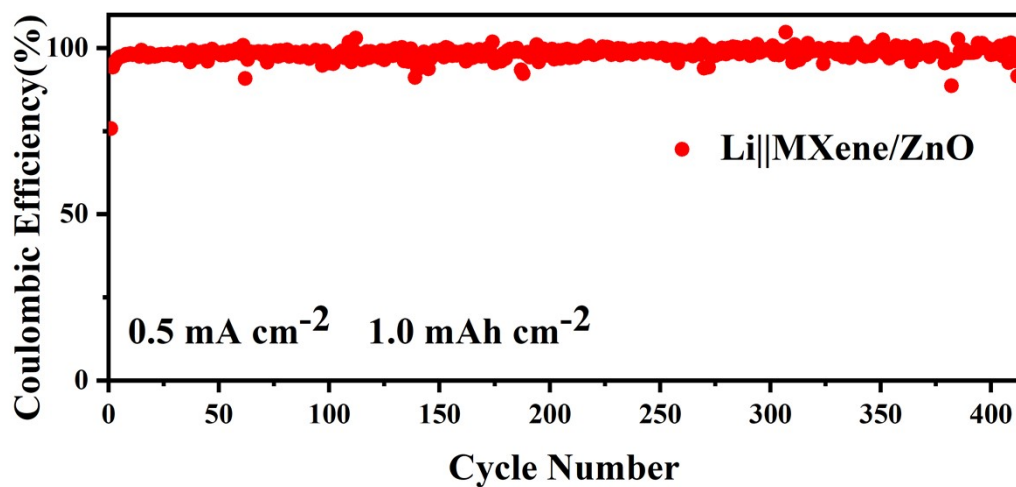


Figure S11 Cycling performance of repeated Li plating/stripping on MXene/ZnO at the current density of  $0.5 \text{ mA cm}^{-2}$

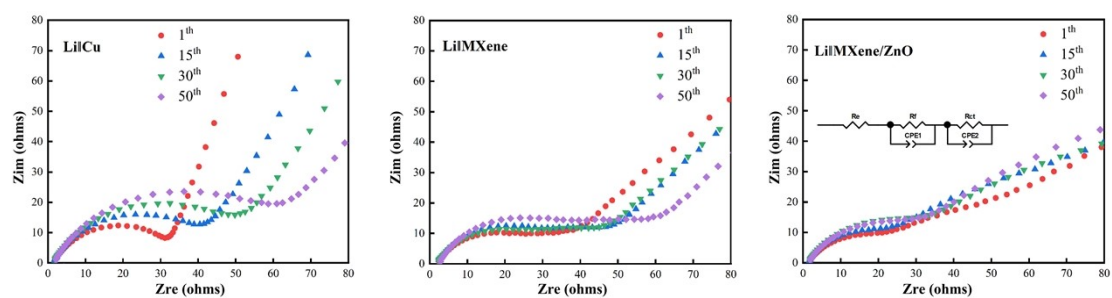


Figure S12 Nyquist plots and equivalent circuit diagram in Li || Cu (a), Li || MXene (b), and Li || MXene/ZnO (c) half-cells at different cycles.

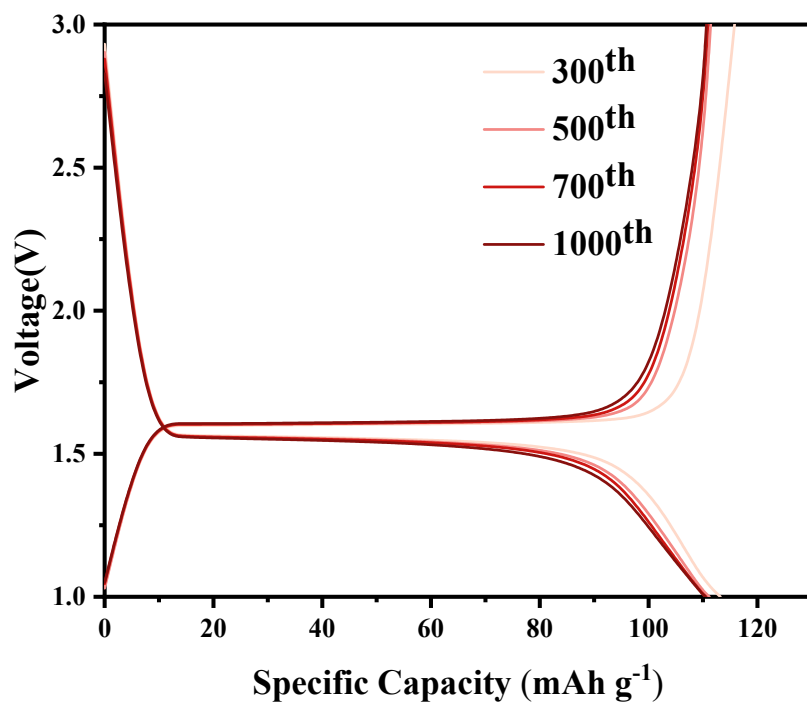


Figure S13 Galvanostatic charge-discharge profiles at 300<sup>th</sup>,500<sup>th</sup>,700<sup>th</sup> and 1000<sup>th</sup> of LTO || MXene/ZnO@Li full cell.

**Table S1** Comparison of the prepared current collector in Li metal half-cell

Materials	Current density & Capacity	Coulombic efficiency	Cycling times	Ref
<b>MXene/ZnO film</b>	0.5mA cm <sup>-2</sup> 1 mA cm <sup>-2</sup> 3 mA cm <sup>-2</sup>	98.3% 97.5% 96.0%	410 330 150	This work
<b>ZnO-CuZn mesh</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	97.48%	100	1
<b>ZnO in 3D conductive carbon nanofibers</b>	1 mAcm <sup>-2</sup> , 0.5 mAh cm <sup>-2</sup>	-	200	2
<b>ZnO/Carbon nanofibers</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	89.8% to 98%	150	3
<b>MXene@COFs</b>	1 mAcm <sup>-2</sup> , 2 mAh cm <sup>-2</sup>	~ 99%	300	4
<b>MXene/Graphene farmework</b>	1 mAcm <sup>-2</sup> , 2 mAh cm <sup>-2</sup>	-	90	5
<b>TiO<sub>2</sub> nanoparticles on Ti<sub>3</sub>C<sub>2</sub>Tx</b>	0.2 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	95.4%	100	6
<b>Au@MXene</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	99%	200	7
<b>Au NPs in rGO</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	99.1%	300	8
<b>Supersized graphite carbon tubes (S-GCTs)</b>	0.5 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	-	350	9
<b>N, O co-doped carbon nanospheres</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	97.65 %	275	10
<b>Nitrogen-doped hollow porous carbon spheres</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98.5 %	270	11
<b>TiN@Cu foam</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	~ 98 %	300	12
<b>Li<sub>2</sub>O on Cu foil</b>	0.25 mA cm <sup>-2</sup> , 0.25 mAh cm <sup>-2</sup>	97.8 %	100	13
<b>Li<sub>2</sub>S@Ni-Nickel foam</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98.1%	100	14
<b>Carbon paper/Sn/SnO composite</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98.8 %	300	15
<b>3D hierarchical porous Cu@Sn</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98 %	240	16
<b>Ultrathin TiO<sub>2</sub> on a Cu current collector</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	97.4%	100	17
<b>Cu<sub>0.7</sub>Zn<sub>0.3</sub>/Cu foam</b>	1.5 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98%	300	18
<b>Oxygen vacancy-rich Co<sub>3</sub>O<sub>4</sub>/nickel foam</b>	1 mAcm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	98%	200	19
<b>3D porous poly-melamine formaldehyde (PMF)/Li</b>	1 mA cm <sup>-2</sup> , 1 mAh cm <sup>-2</sup>	95.8%	300	20



**Table S2** Comparison of the prepared current collector in lithium metal full cell

Materials	Cathode	Current density	Cycling times	Capacity retention	Ref
MXene/ZnO film	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	5 C	1200	86.7 %	This work
ZnO-CuZn mesh	LiNi <sub>1/3</sub> Co <sub>1/3</sub> Mn <sub>1/3</sub> O	1 C	150	94.3 %	1
ZnO in 3D conductive carbon nanofibers	LiFePO <sub>4</sub>	1 C	300	95.7 %	2
MXene/Graphene framework	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	1 C	450	~ 85 %	5
Au@MXene	LiFePO <sub>4</sub>	1 C	200	97.7 %	7
TiN@Cu foam	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	2 C	700	93.4 %	12
3D porous poly-melamine-formaldehyde (PMF)/Li	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	2 C	1000	~ 92%	20
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> /g-C <sub>3</sub> N <sub>4</sub>	LiFePO <sub>4</sub>	0.5 C	320	85.5 %	21
In@Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> paper	LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub>	200 mA g <sup>-1</sup>	130	80 %	22
Vertically Aligned MXene Nanosheet	LiFePO <sub>4</sub>	0.5 C	350	85.4 %	23

(1 C = 0.175 mAg<sup>-1</sup> for Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub>, 1 C = 0.17 mAg<sup>-1</sup> for LiFePO<sub>4</sub>, 1 C = 0.278 mAg<sup>-1</sup> for LiNi<sub>1/3</sub>Co<sub>1/3</sub>Mn<sub>1/3</sub>O).

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