

Electronic Supplementary Information

LiV_3O_8 as an Intercalation-type Cathode for Aqueous Aluminum-Ion Batteries†

Vaiyapuri Soundharrajan,^a Subramanian Nithiananth^b Jun Lee,^a Jung Ho Kim,^c Jang-Yeon Hwang^{a,*} and Jaekook Kim^{a,*}

^a Department of Materials Science and Engineering, Chonnam National University, Gwangju 61186, Republic of Korea.

^b Graduate School of Science and Technology, Shizuoka University, 3-5-1 Johoku, Naka-Ku, Hamamatsu, Shizuoka 432-8011, Japan.

^c Institute for Superconducting and Electronic Materials (ISEM), Australian Institute of Innovative Materials (AIIM), University of Wollongong, North Wollongong, New South Wales 2500, Australia.

E-mail: hjy@jnu.ac.kr; jaekook@chonnam.ac.kr; Fax: +82-62-530-1699; Tel: +82-62-530-17.

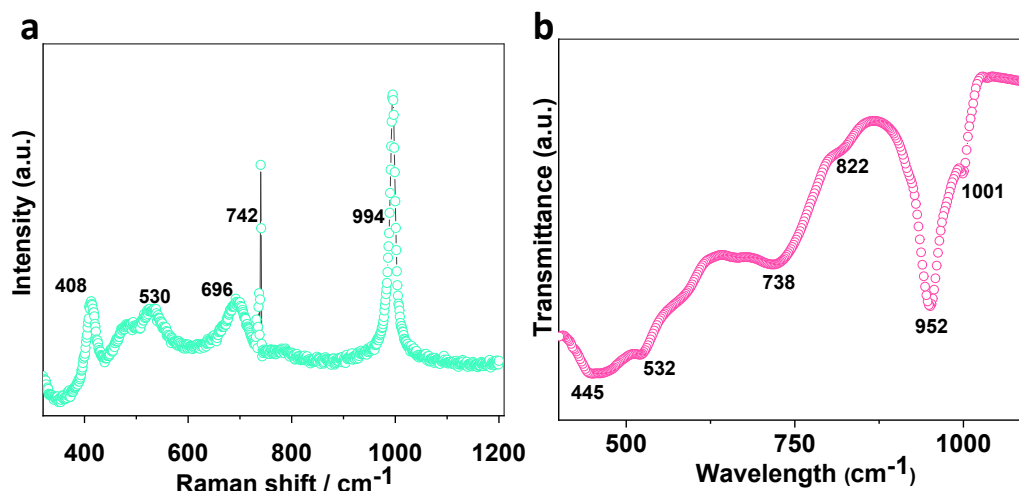


Fig. S1 (a) Raman profile of LVO cathode and (b) FTIR output of LVO cathode.

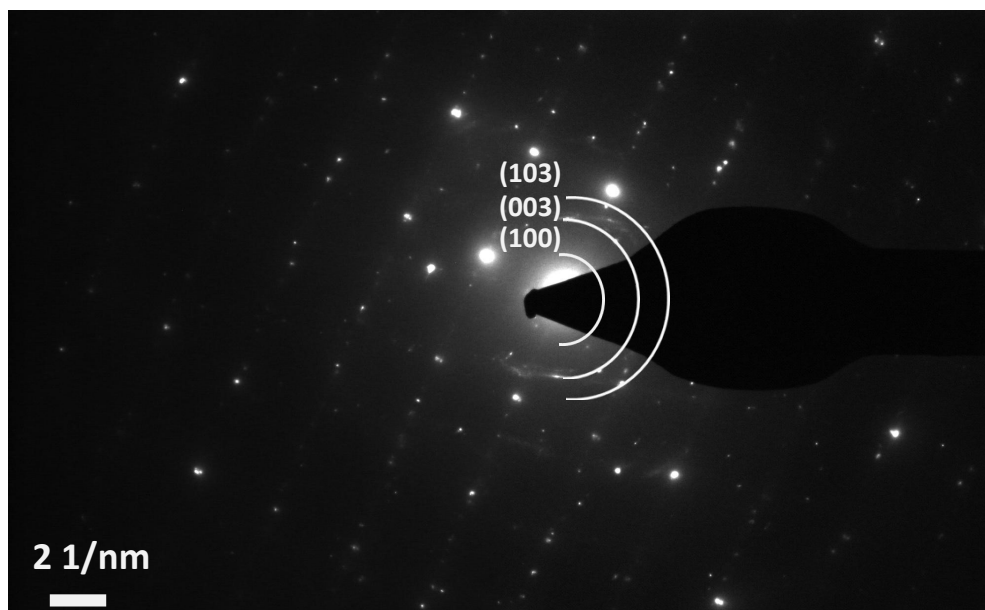


Fig. S2 SAED pattern of LVO sample.

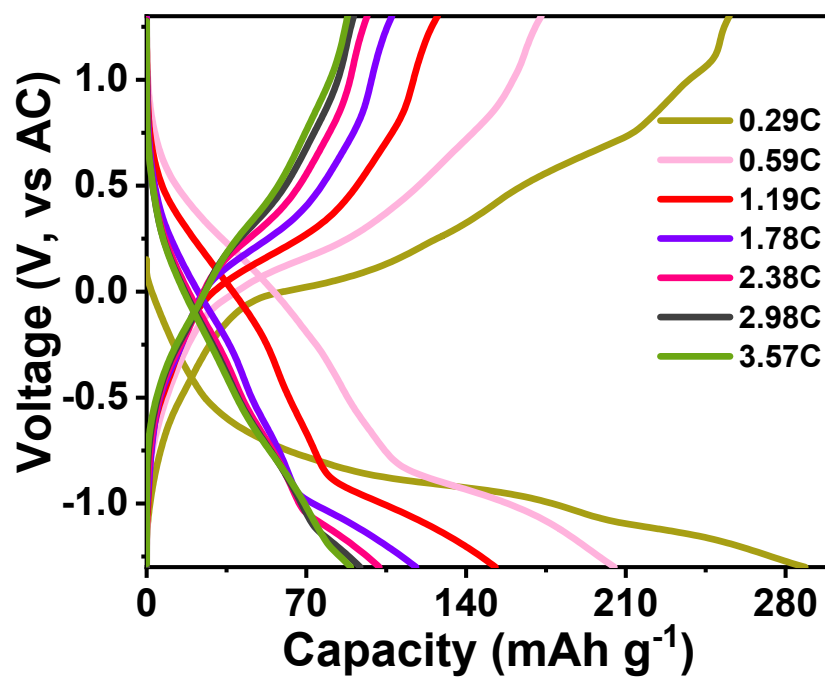


Fig. S3 Selected charge/discharge profile of LVO cathode at different C rates.

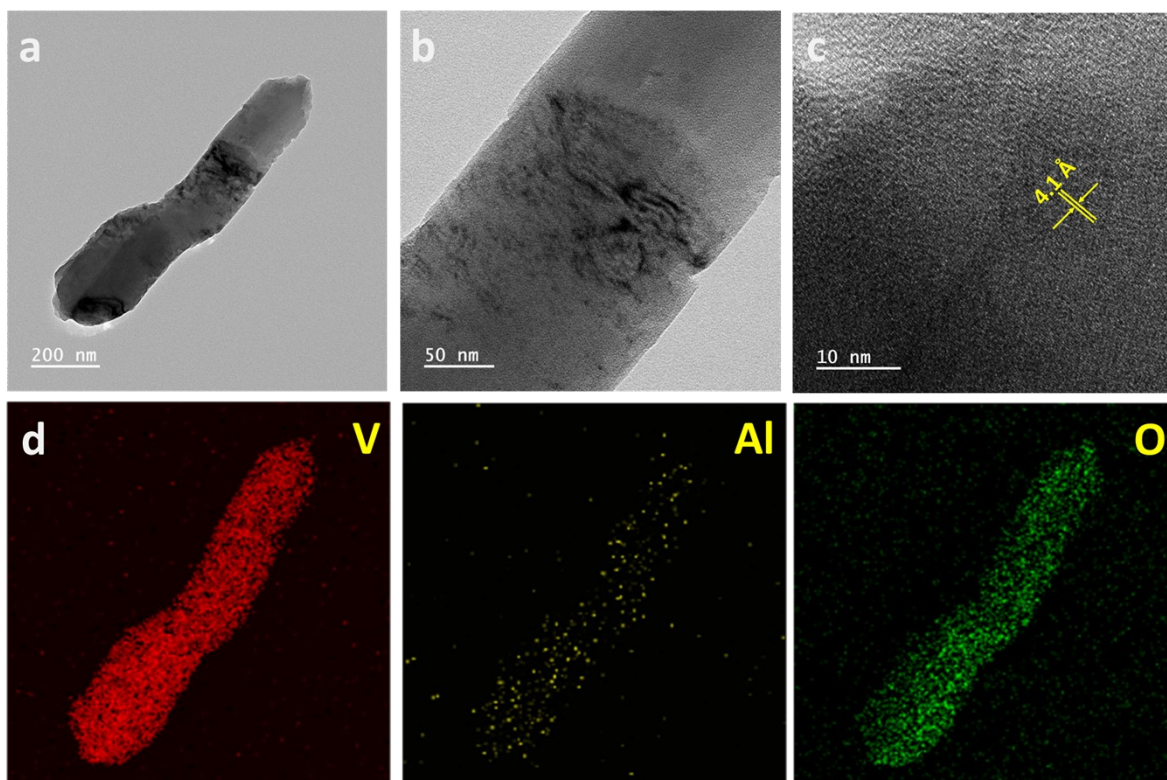


Fig. S4 Ex situ TEM analysis of LVO cathode after the 1st discharge process.

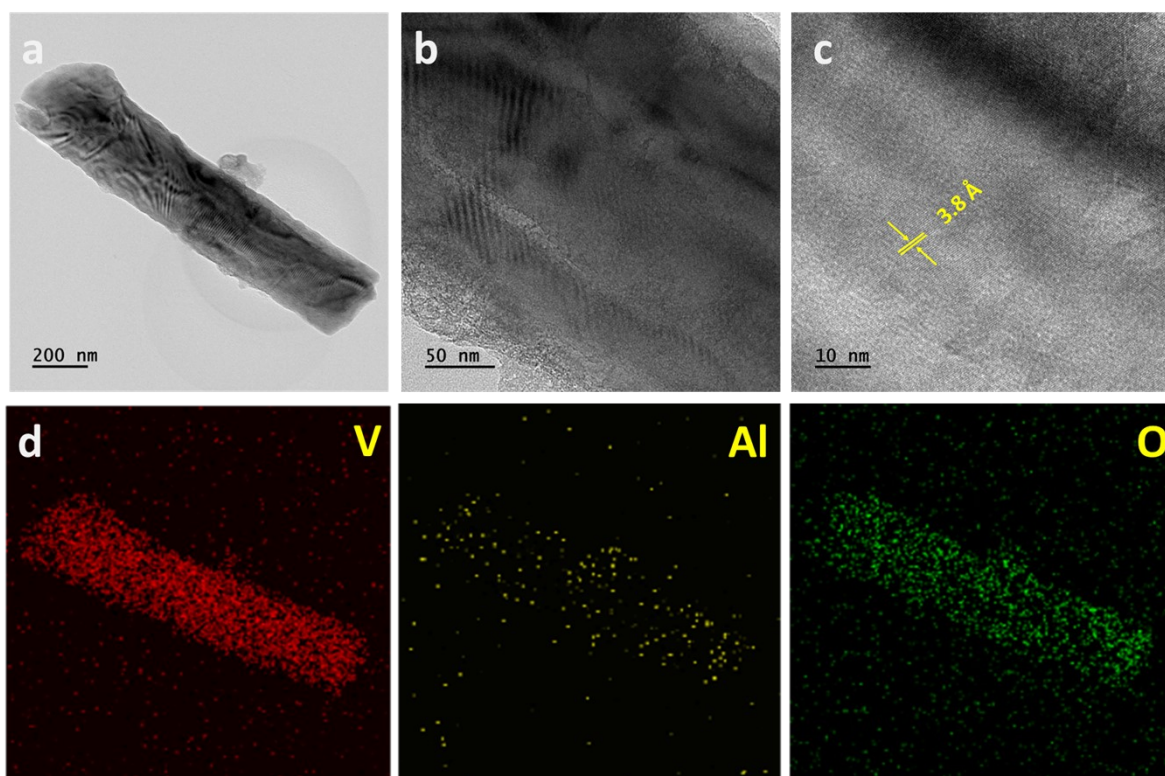


Fig. S5 Ex situ TEM analysis of LVO cathode after the 1st charge process.

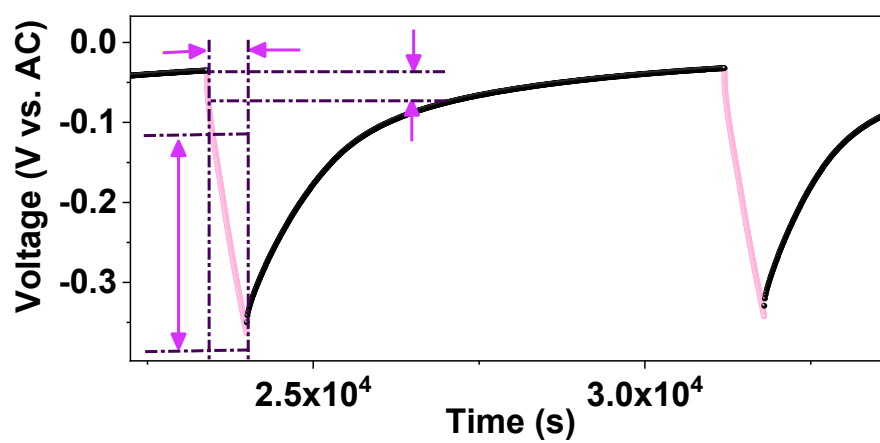


Fig. S6 The plot of single-stage GITT output for the 1st discharge curve.

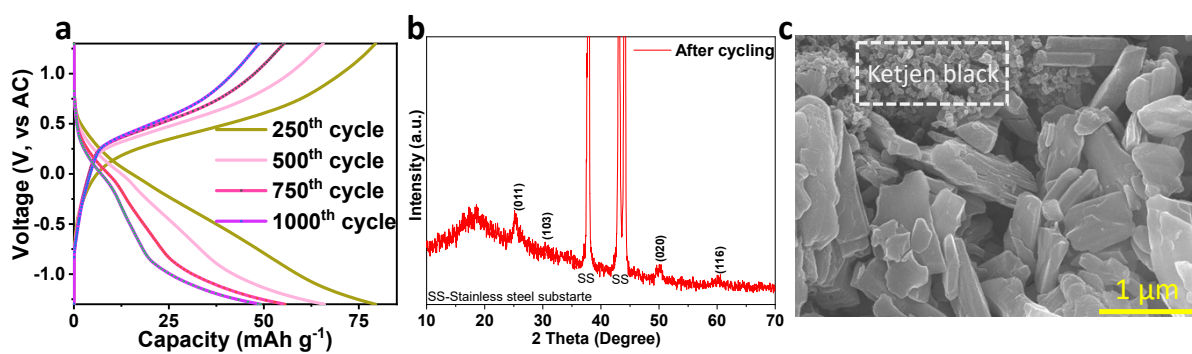


Fig. S7 (a) Galvanostatic charge/discharge profile at 3.57C, Post-mortem outputs of the LVO cathode at 3.57C after 1000 cycles (b) ex-situ XRD profile, and (c) ex-situ SEM images.

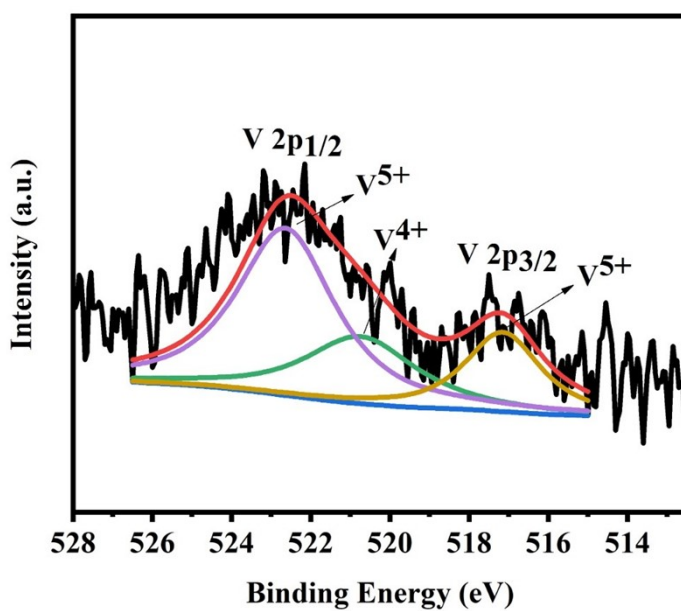


Fig. S8 Ex-situ XPS analysis of LVO cathode after the 1000th cycle.

Table. S1 Comparison of vanadium-based cathodes established for AAlBs.

S.No	Cathode	Counter/ Reference electrode	Electrolyte	Reversible capacity/ Current density	Cycle number/ capacity retention
1	$\text{VOPO}_4 \cdot 2\text{H}_2\text{O}^1$	Aluminium foil	$\text{Al}(\text{CF}_3\text{SO}_3)_3$	$125.4 \text{ mAh g}^{-1}/20 \text{ mA g}^{-1}$	40/ 60%
2	VOPO_4^2	Pt sheet / Ag/AgCl	$\text{Al}(\text{NO}_3)_3$	$80.1 \text{ mAh g}^{-1}/1 \text{ A g}^{-1}$	2800/86.2%
3	FeVO_4^3	Carbon paper/ Ag/AgCl	AlCl_3	$350 \text{ mAh g}^{-1}/60 \text{ mA g}^{-1}$	1
4	$\text{VO}_2\text{-B}^4$	Graphite paper/ Ag/AgCl	$\text{Al}(\text{TOF})_3$	$234 \text{ mAh g}^{-1}/150 \text{ mA g}^{-1}$	1000/77.2%
5	VO_2^5	Carbon rod / saturated calomel electrode (SCE)	$\text{Al}(\text{TOF})_3$	$235 \text{ mAh g}^{-1}/200 \text{ mA g}^{-1}$	200/32%
6	V_2O_5^6	Aluminum foil	$\text{Al}(\text{TOF})_3$	$200 \text{ mAh g}^{-1}/40 \text{ mA g}^{-1}$	50/60%
7	LiV_3O_8 (This work)	Activated Carbon (AC)	$\text{Al}(\text{CF}_3\text{SO}_3)_3$	$147 \text{ mAh g}^{-1}/0.57\text{C}$ or 0.5 A g^{-1}	500/77.3%

Reference

1. Q. Pang, S. Yang, X. Yu, W. He, S. Zhang, Y. Tian, M. Xing, Y. Fu and X. Luo, *J. Alloys Compd.*, 2021, **885**, 161008.
2. P. Wang, Z. Chen, H. Wang, Z. Ji, Y. Feng, J. Wang, J. Liu, M. Hu, J. Fei, W. Gan and Y. Huang, *Energy Storage Mater.*, 2020, **25**, 426–435.
3. S. Kumar, R. Satish, V. Verma, H. Ren, P. Kidkhunthod, W. Manalastas and M. Srinivasan, *J. Power Sources*, 2019, **426**, 151–161.
4. Y. Cai, S. Kumar, R. Chua, V. Verma, D. Yuan, Z. Kou, H. Ren, H. Arora and M. Srinivasan, *J. Mater. Chem. A*, 2020, **8**, 12716–12722.
5. Y. Wang, X. Shi, J. Wang, X. Liu and X. Lu, *Mater. Today Energy*, 2021, **19**, 100578.
6. Q. Zhao, L. Liu, J. Yin, J. Zheng, D. Zhang, J. Chen and L. A. Archer, *Angew. Chemie Int. Ed.*, 2020, **59**, 3048–3052.