

Supporting Information for

Electronic Modulation of NiO by Constructing Amorphous/Crystalline

Heterophase to Improve Photocatalytic Hydrogen Evolution

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Table S1. Mass fraction of Ni in NiO_x-CN samples with different NiO-LAL content.

Samples	NiO ₂ -CN	NiO ₄ -CN	NiO ₆ -CN	NiO ₈ -CN
Mass fraction of Ni	8.75%	17.13%	19.03%	24.13%

Table S2. BET surface areas of g-C₃N₄, NiO-LAL and NiO_x-CN samples with different NiO-LAL content.

Samples	g-C ₃ N ₄	NiO ₂ -CN	NiO ₄ -CN	NiO ₆ -CN	NiO ₈ -CN	NiO-LAL
BET Surface Area ($m^2 \cdot g^{-1}$)	199.87	90.10	100.31	92.72	84.86	4.48

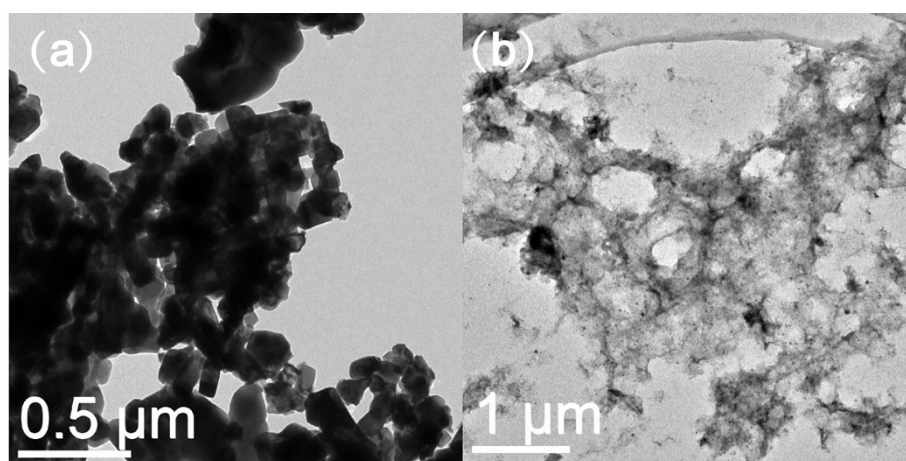


Figure S1. TEM images of NiO before laser treatment (a) and NiO-LAL at low magnification (b).

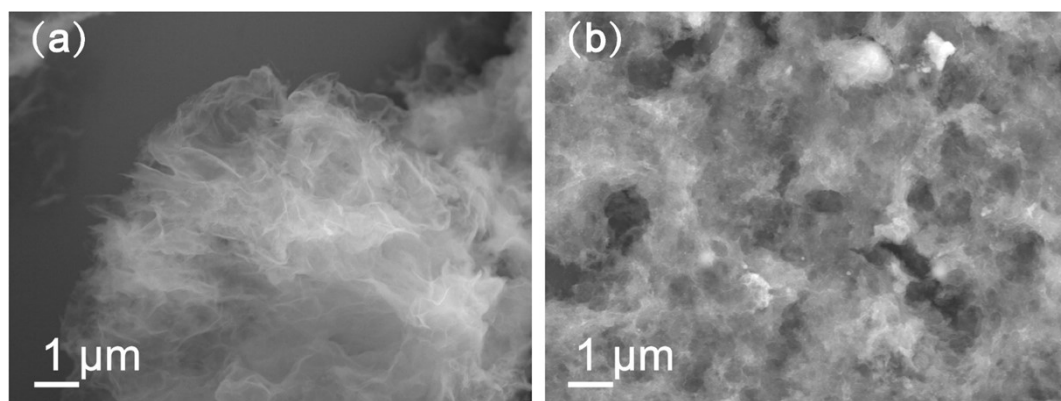


Figure S2. SEM images of g-C₃N₄ (a) and NiO-LAL (b).

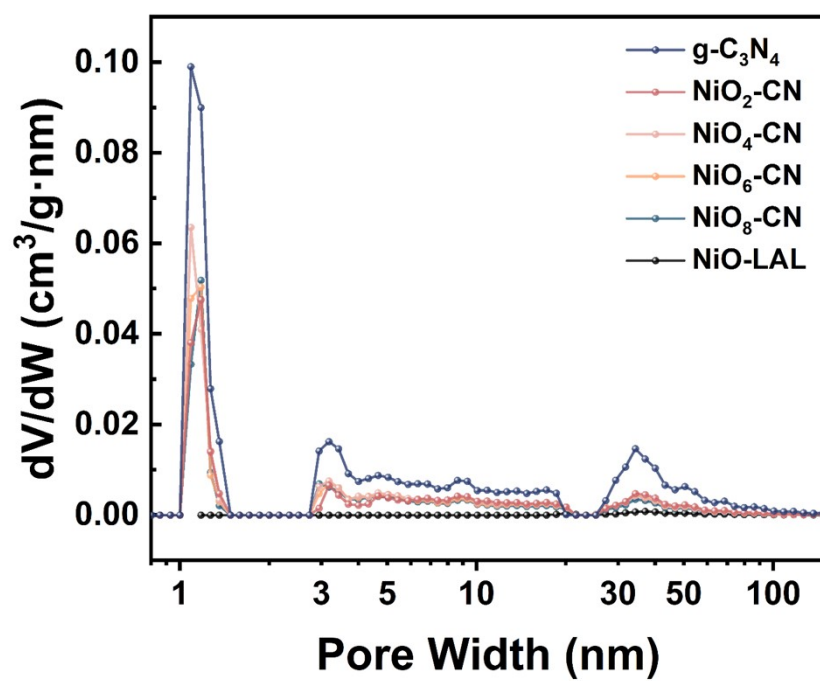


Figure S3. DFT pore volume distributions of as-prepared samples.

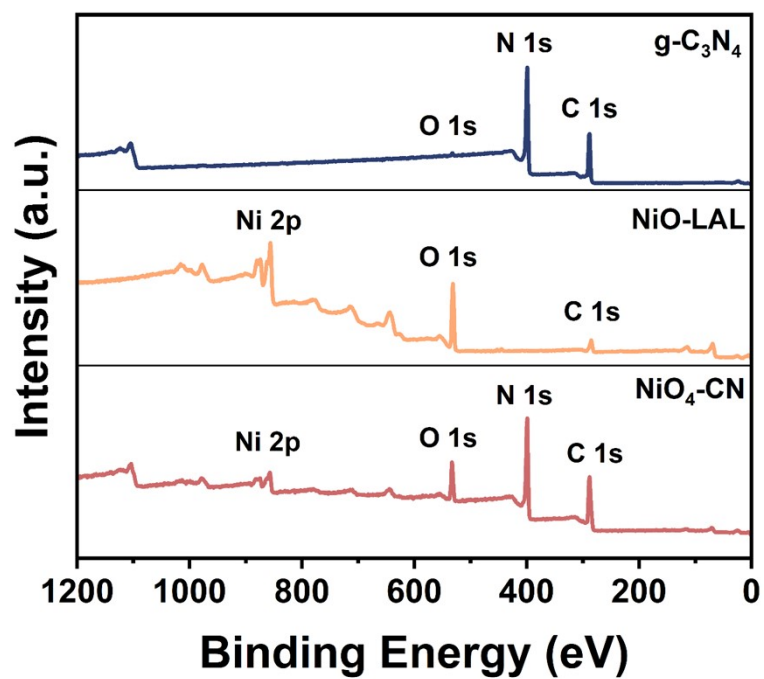


Figure S4. XPS spectra of the survey for g-C₃N₄, NiO-LAL and NiO₄-CN.

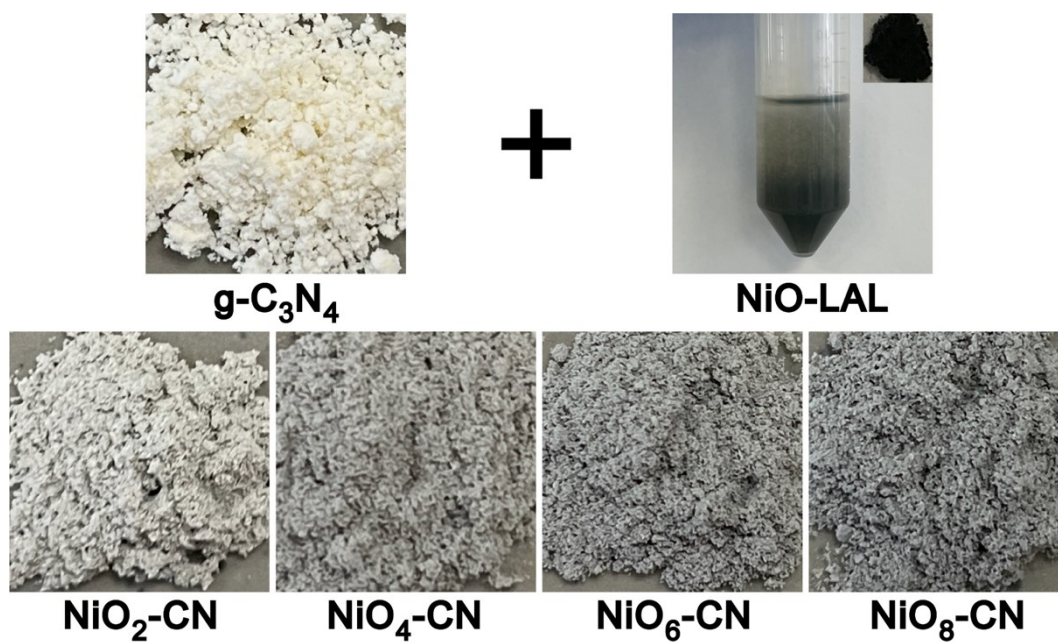


Figure S5. Photographs of $\text{g-C}_3\text{N}_4$, NiO-LAL and $\text{NiO}_x\text{-CN}$ samples with different NiO-LAL content.

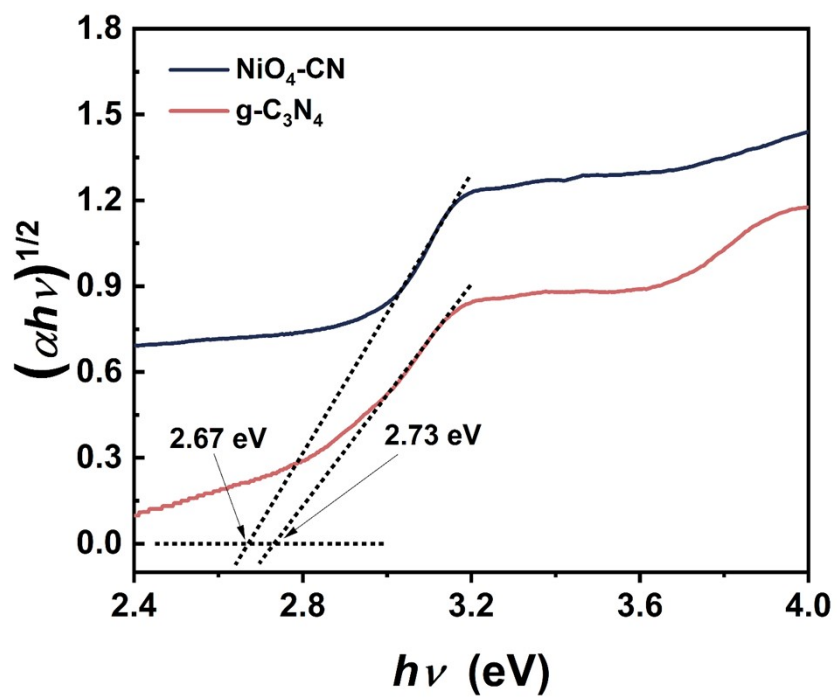


Figure S6. The plots of $(\alpha h\nu)^{1/2}$ versus $h\nu$ for the band energies of $\text{g-C}_3\text{N}_4$ and $\text{NiO}_4\text{-CN}$.

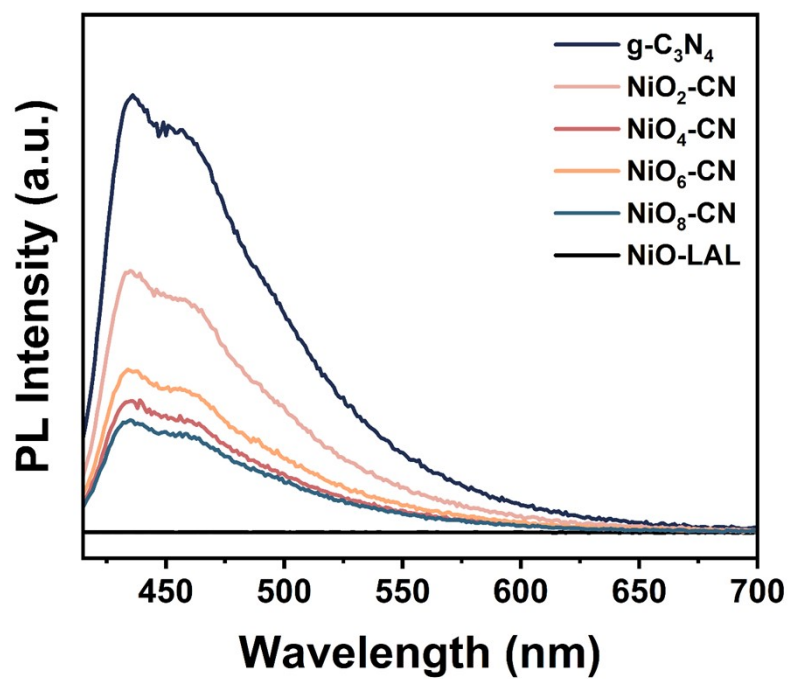


Figure S7. PL spectroscopy of g-C₃N₄, NiO-LAL and NiO_x-CN samples with different NiO-LAL content.

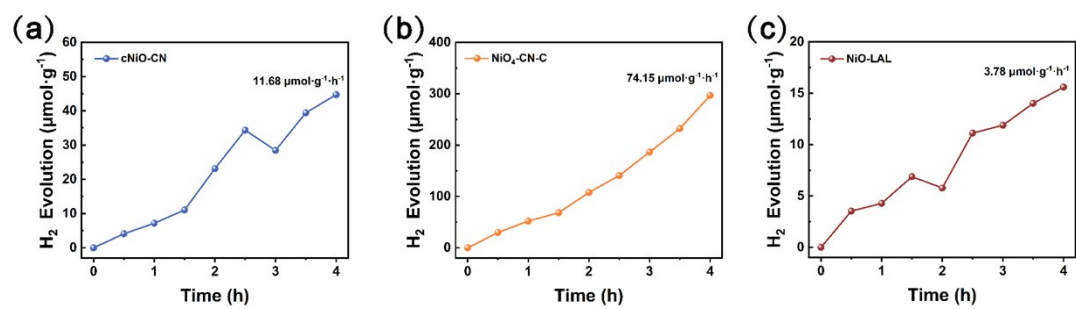


Figure S8. Photocatalytic H₂ generation performances of cNiO-CN (a), NiO₄-CN-C (b) and NiO-LAL (c).

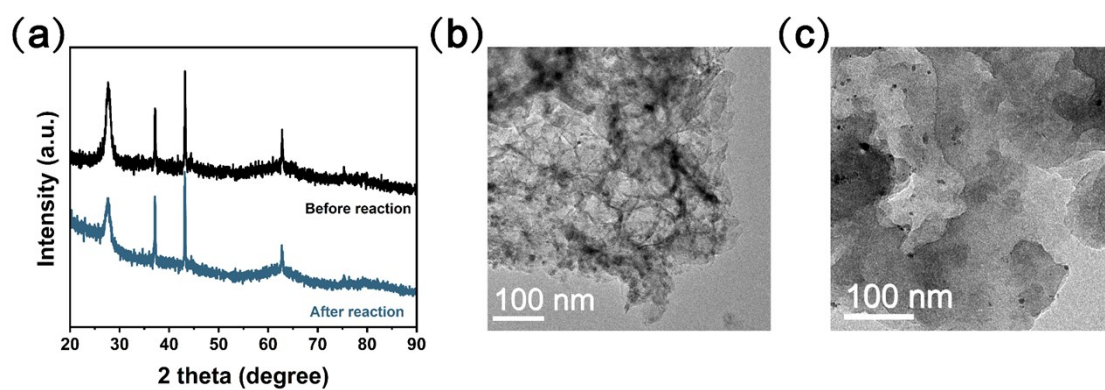


Figure S9. XRD patterns of NiO₄-CN before and after photocatalytic reaction (a), and TEM images of NiO₄-CN before (b) and after (c) photocatalytic reaction.

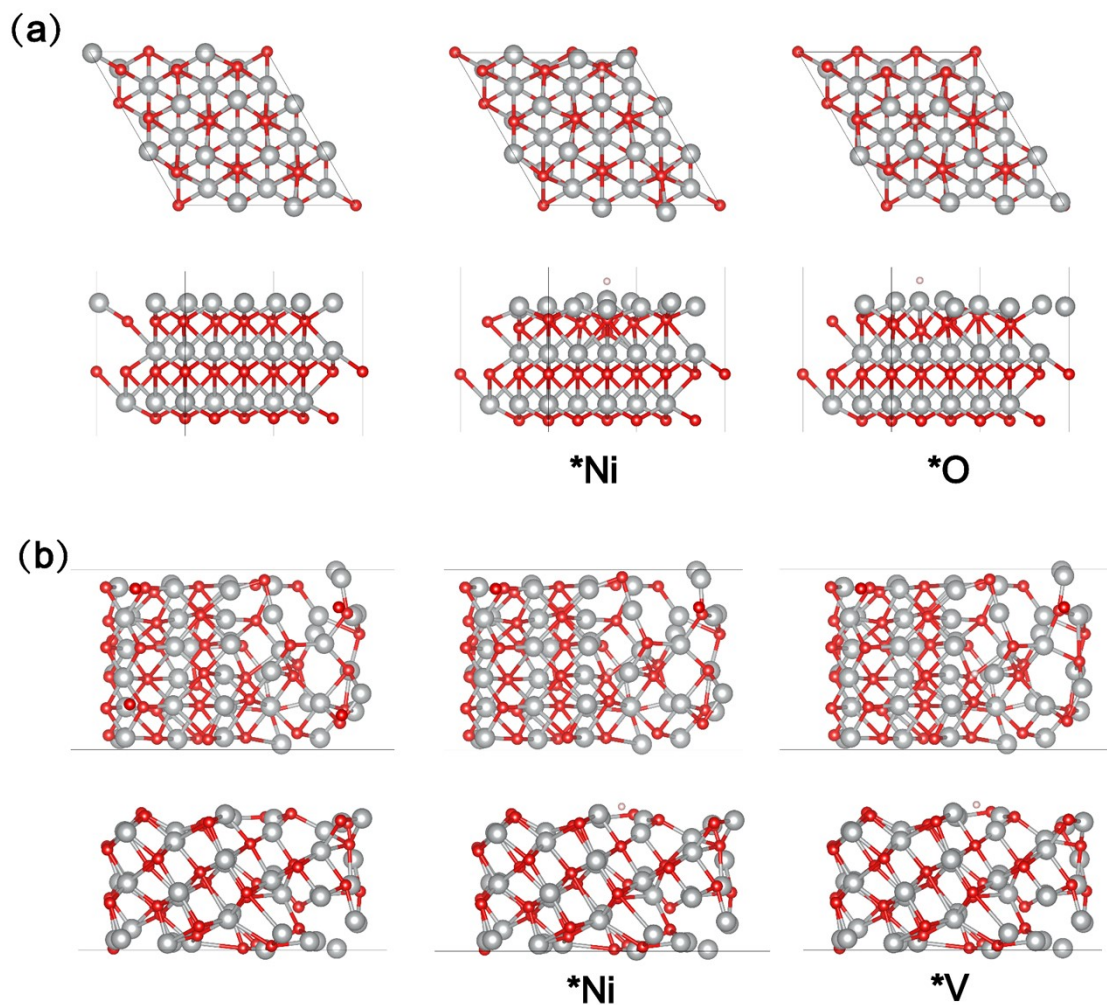


Figure S10. The structural models of pristine NiO (a) and amorphous/crystalline NiO homojunction (b).

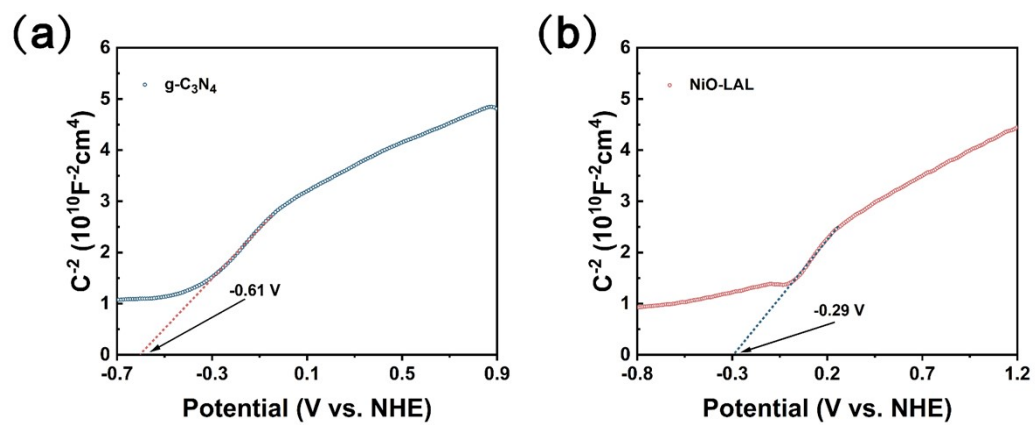


Figure S11. The Mott-Schottky plots of $g-C_3N_4$ (a) and NiO-LAL (b).