

Hydrolysis of ionic clusters to induce interconnective sieving pores in ion-conductive membranes for vanadium flow batteries

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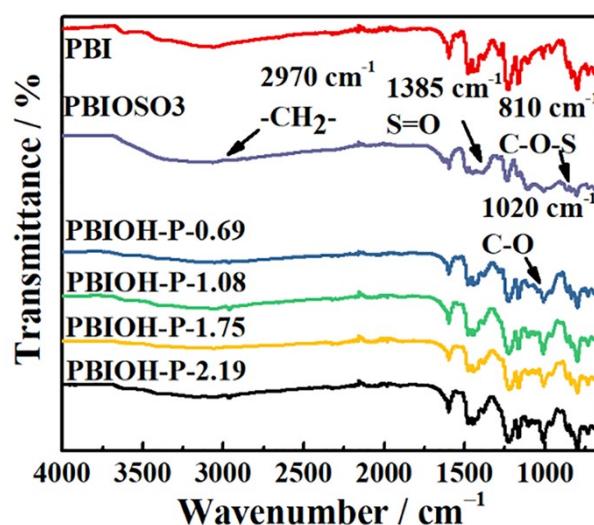


Fig. S1. The FTIR patterns of PBI, PBIOSO3 and PBIOH-P membranes.

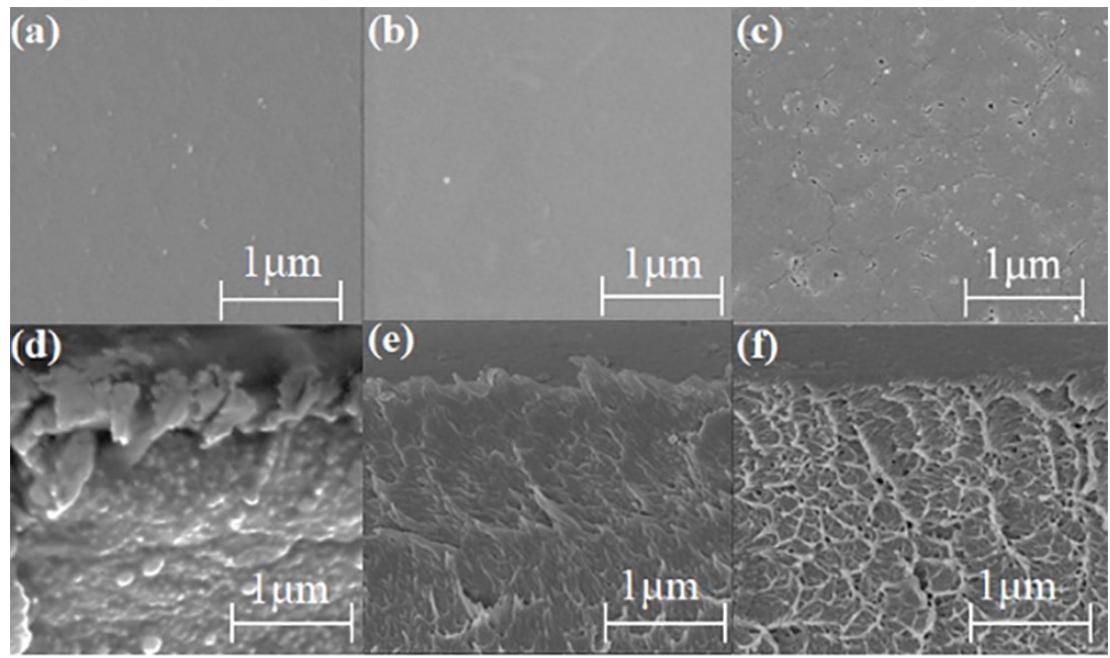


Fig. S2. The SEM surface images of (a) PBI, (b) PBIOSO₃ and (c) PBIOH-P-1.75 membranes; cross-section images of (d) PBI, (e) PBIOSO₃ and (f) PBIOH-P-1.75 membranes.

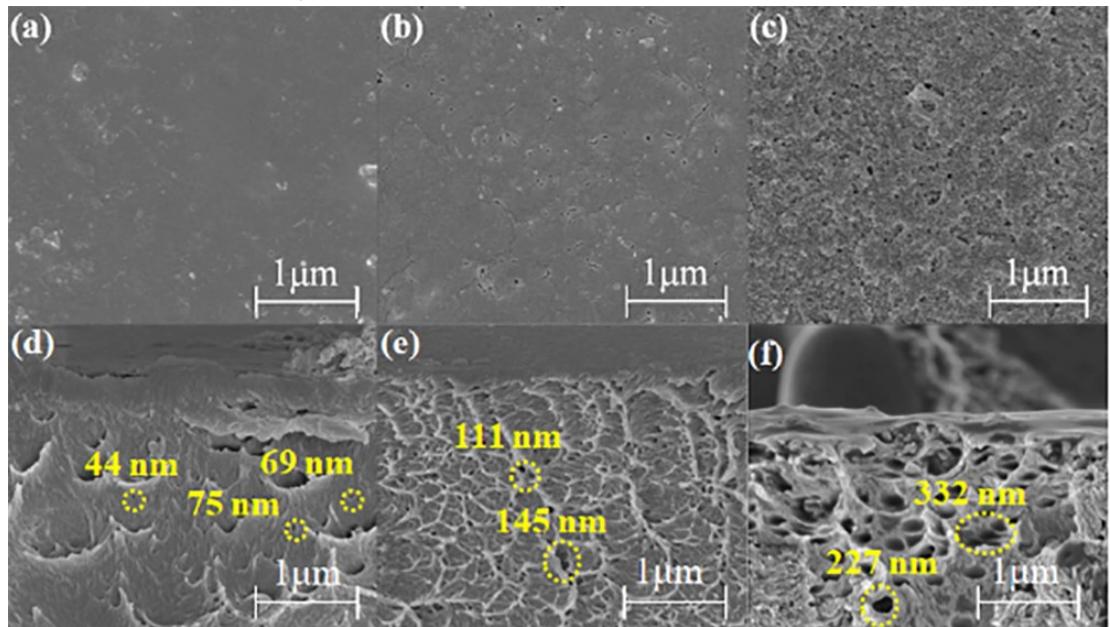


Fig. S3. The SEM surface images of (a) PBIOH-P-0.69, (b) PBIOH-P-1.75 and (c) PBIOH-P-2.19 membranes; cross-section images of (d) PBIOH-P-0.69, (e) PBIOH-P-1.75 and (f) PBIOH-P-2.19 membranes

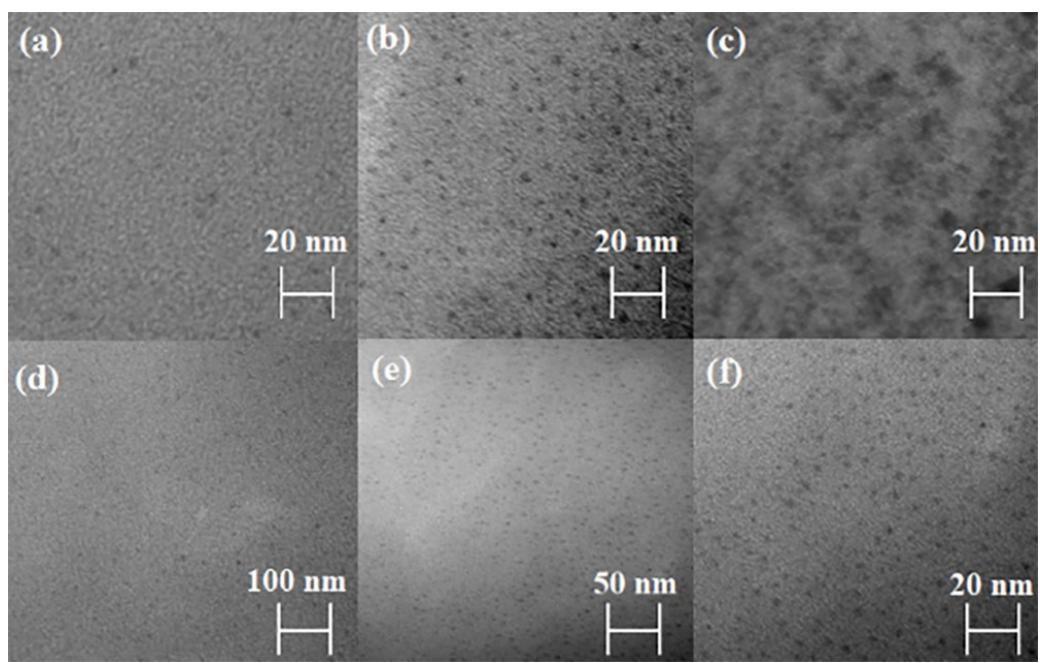


Fig. S4. The TEM ions clusters images of (a) OPBIOSO3-0.69, (b) OPBIOSO3-1.75 and (c) OPBIOSO3-2.19 membranes; TEM ions clusters images of OPBIOSO3-1.75 membrane with 100 nm, 50 nm and 20 nm.

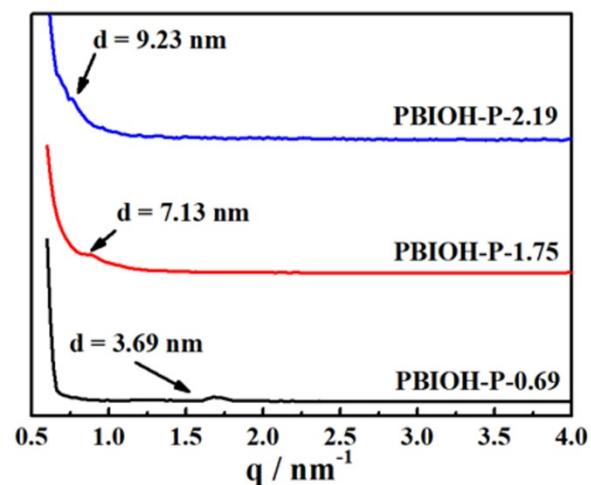


Fig. S5. SAXS curves of the PBIOH-P membranes.

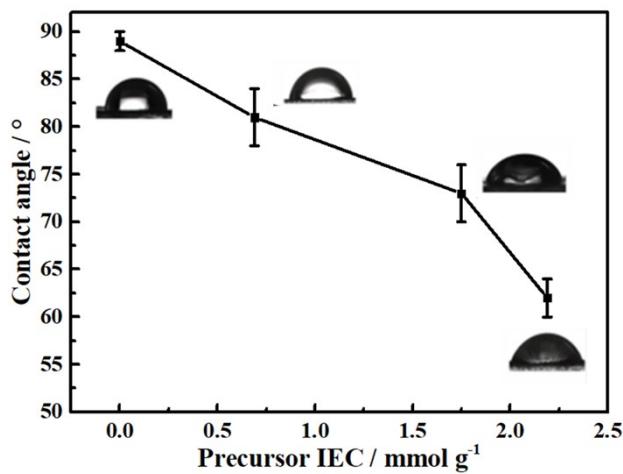


Fig. S6. The contact angle of the PBI and PBIOH-P membranes.

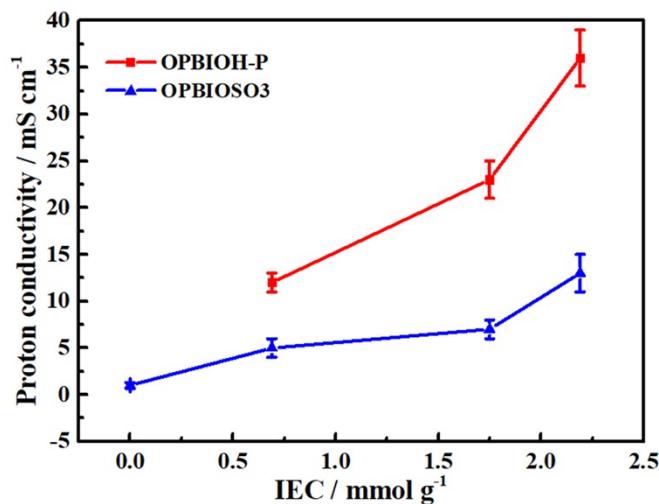


Fig. S7. The proton conductivity of the PBIOH-P and PBIOSO₃ membranes.

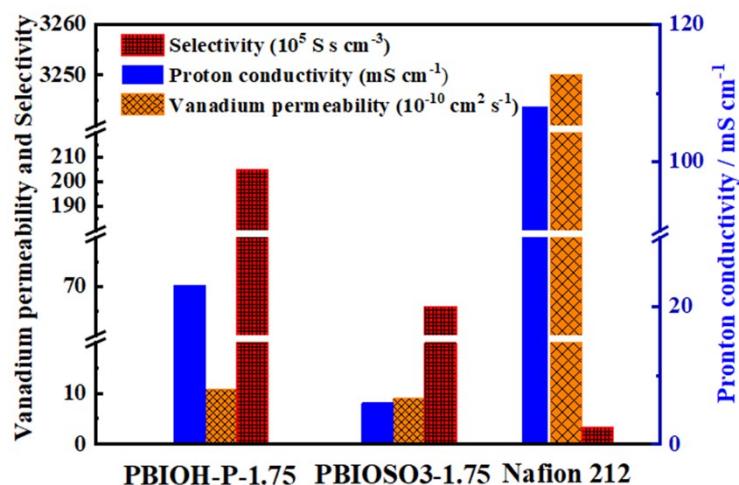


Fig. S8. The Selectivity, proton conductivity and vanadium permeability of the PBIOH-P, PBIOSO₃-1.75 and Nafion 212 membranes. (Selectivity is denoted as proton conductivity divided by vanadium permeability)

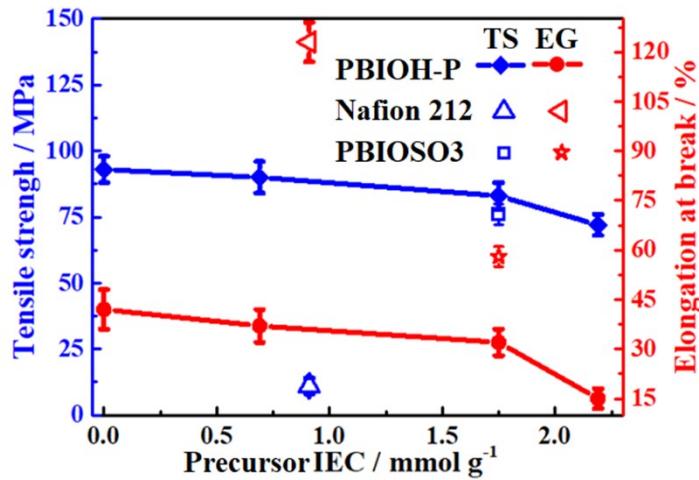


Fig. S9. Tensile strength and elongation at break of the PBIOH-P and PBIOSO3 membranes with precursor IEC of 1.75, and Nafion 212 membrane.

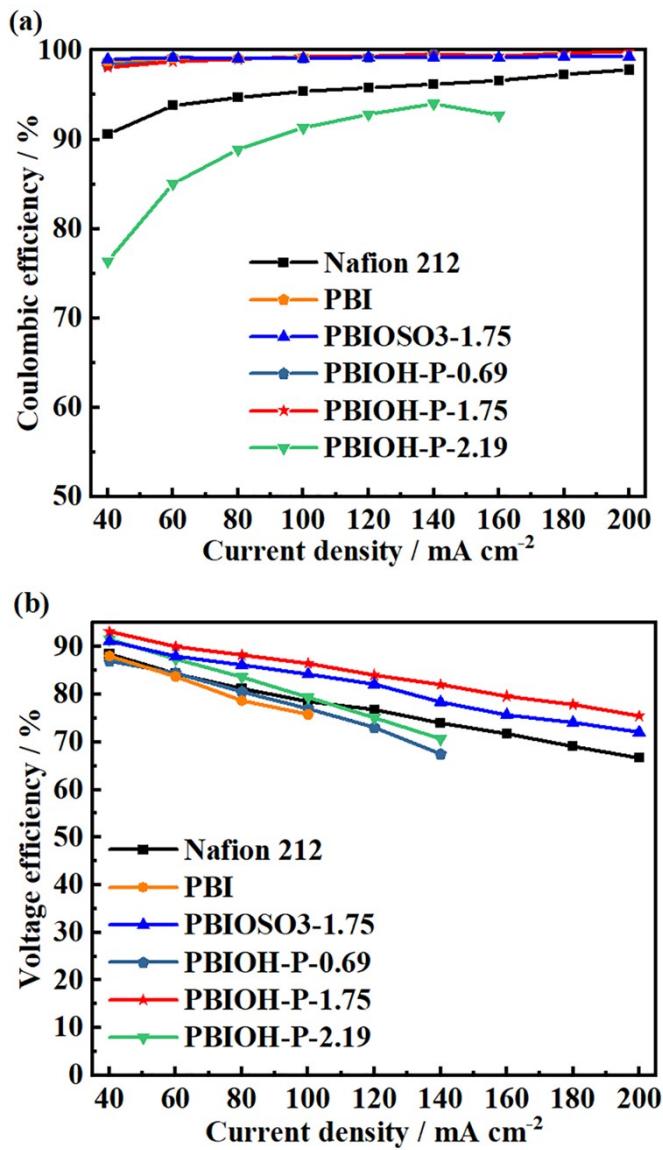


Fig. S10. Columbic efficiency and voltage efficiency of the PBIOH-P, Nafion 212, PBI and PBIOSO3-1.75 based cells.

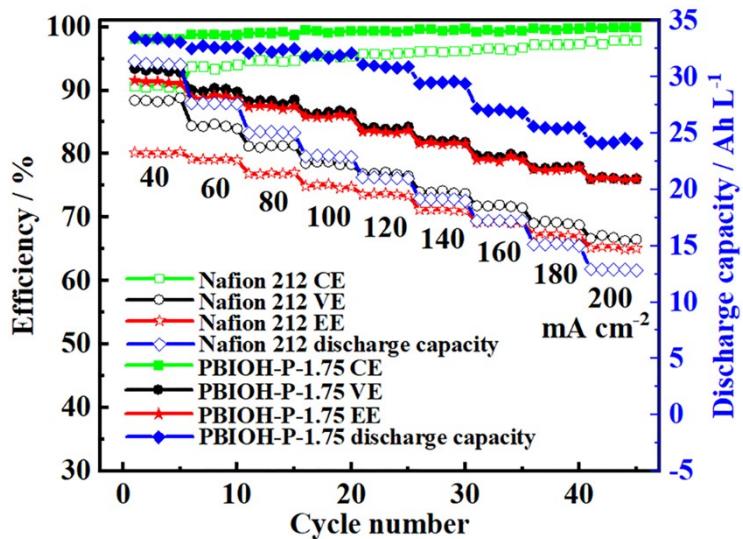


Fig. S11. The rate performance of the PBIOH-P-1.75 and Nafion 212 based VFBs.

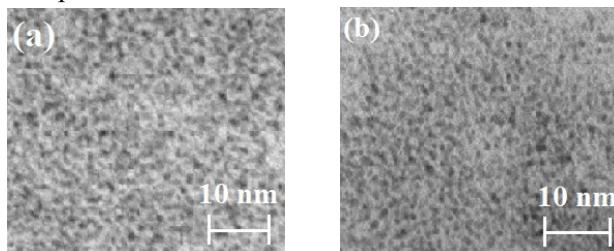


Fig. S12. The TEM images of pore morphology (a) before and (b) after VFB cycling test

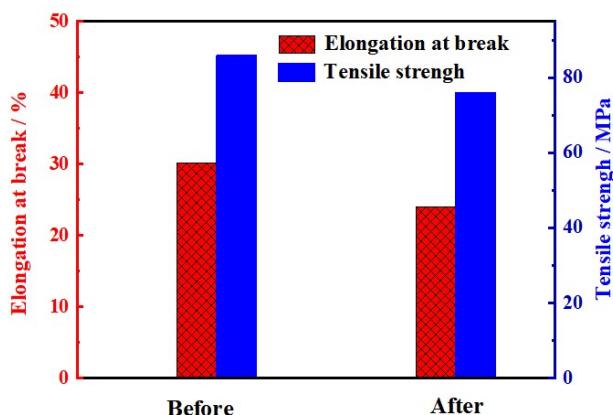


Fig. S13. The mechanical properties before and after VFB cycling test

Table S1 The comparison of H^+/V^{n+} ion selectivity with recently reported porous membranes, composite membranes, and anion exchange membranes

Membranes	Category	H^+/V^{n+} Selectivity	Conductivity	Reference

		$\times 10^5 S \text{ min cm}^{-3}$	$mS \text{ cm}^{-1}$	
<i>This work</i>	<i>Porous membranes</i>	3.42	70	
<i>SPEEK/SPAES-15</i>	<i>Composite membrane</i>	0.29	29.3	1
<i>SPI/2% PMPP</i>	<i>Composite membrane</i>	1.42	116	2
<i>SPEEK/Hi m-pS-3</i>	<i>Composite membrane</i>	0.40	31.54	3
<i>S/801&808 -4</i>	<i>Composite membrane</i>	2.2	78	4
<i>SPEEK/SP PTA-25</i>	<i>Composite membrane</i>	0.23	29.08	5
	<i>Anion exchange membrane</i>			
<i>QCPPAE</i>	<i>Anion exchange membrane</i>	0.96	21	6
<i>CAPSU</i>	<i>Anion exchange membrane</i>	2.08	15	7
<i>PyPPSU</i>	<i>Anion exchange membrane</i>	2.44	8.8	8
<i>PVDF/SiO₂-SO₃H-42</i>	<i>Porous membranes</i>	1.25	14	9
<i>2# ABPBI</i>	<i>Porous</i>	0.79	30	10

Reference:

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