## Electronic Supporting Information (ESI)

## In-depth structural characterization of the influence of Li<sup>+</sup> excess on spherical, Co-free layered LiMn<sub>0.5</sub>Ni<sub>0.5</sub>O<sub>2</sub> cathode material using correlative Raman-SEM microscopy

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**Figure S1.** SEM images of the  $(Mn_{0.5}Ni_{0.5})(OH)_2$  precursor: a) top view of spherical particle morphology and b) particle cross-section of dense spheres.



**Figure S2.** XRD pattern of Li<sub>1.00</sub> (blue), Li<sub>1.07</sub> (orange), Li<sub>1.10</sub> (red) and Li<sub>1.16</sub> (green): a) overview and b) details of selected reflections.

**Table S1.** Calculated Rietveld refinement parameters based on the structural model for  $LiMn_{0.5}Ni_{0.5}O_2$  of Bréger *et al.*<sup>1</sup>: space group, lattice parameters, Cell Volume *V*, cation occupancies, atom coordinate *z* of the oxide ions, the estimated crystallite size (integral breadth) and the *R* factor.

	Li <sub>1.00</sub>	Li <sub>1.07</sub>	Li <sub>1.10</sub>	Li <sub>1.16</sub>
Space group	R3m (166)	R3m (166)	R3m (166)	R3m (166)
a / Å	2.890	2.880	2.874	2.867
c / Å	14.306	14.289	14.272	14.255
V / ų	103.5	102.6	102.1	101.5
Li <sub>Li</sub> (3a: 0, 0, 0)	0.91	0.92	0.93	0.94
Ni <sub>Li</sub> (3a: 0, 0, 0)	0.09	0.08	0.07	0.06
Li <sub>тм</sub> (3b: 0, 0, 1/2)	0.09	0.08	0.07	0.06
Мп <sub>™</sub> (3b: 0, 0, 1/2)	0.50	0.50	0.50	0.50
Ni <sub>TM</sub> (3b: 0, 0, 1/2)	0.41	0.42	0.43	0.44
z [O (6c: 0, 0, z)]	0.242	0.243	0.243	0.243
Crystallite Size / nm	149	385	284	293
R <sub>wp</sub> / %	5.8	5.9	4.6	5.3



**Figure S3.** XRD pattern (black) of a)  $Li_{1.00}$ , b)  $Li_{1.07}$ , c)  $Li_{1.10}$  and d)  $Li_{1.16}$  in comparison with the calculated pattern from Rietveld refinement (red) and the difference between both (blue).



Figure S4. Li/Ni disorder as a function of the amount of excess Li<sup>+</sup> in the structure determined via Rietveld refinement.



**Figure S5.** Overview SEM images of  $Li_{1.16}$  particle cross section showing an irregular morphology and the local phase separation.



**Figure S6.** EBSD results of Li<sub>1.16</sub>: EDX mapping of a) Ni and b) Mn, orientation distribution maps with inverse pole figure color key in direction c) Z, d) X and e) Y with respect to the acquisition system. The specimen is located in the X-Y plane and is oriented perpendicularly to the Z-axis. Furthermore, g) the inverse pole figures for the three directions X, Y and Z and h) pol figures for the families {0001}, {10-10} and {11-20} indicating both no significant preferred crystallographic orientation of the crystallites. This is confirmed by f) the correlation of the Mackenzie plot (grey) with the random pair distribution (orange) of 99 % and an M-index of 0.1, where a M-index of 1 represents a single crystal and a value of 0 represents a purely random orientation distribution.<sup>2</sup>



**Figure S7.** Histograms of the fitted position of the Raman vibration band at approximately 600 cm<sup>-1</sup> of  $Li_{1.00}$  (blue),  $Li_{1.07}$  (orange),  $Li_{1.10}$  (red) and  $Li_{1.07}$  (green).



**Figure S8.** Long-term cycling performance of  $Li_{1.07}$  and  $Li_{1.10}$  at RT (2.5 - 4.7 V): a) Specific discharge capacities and b) discharge voltage profiles of cycles with lower discharge current (12 mA g<sup>-1</sup>). The small fluctuation of the data points in a) is related to slight variations of the room temperature.



**Figure S9.** Electrochemical analysis of the  $1^{st}$  cycle specific capacity behavior in different voltage ranges: OCV – 4.4 V (black) and 4.4 – 4.7 V (red) as well as the irreversible initial capacity loss. The given numbers are mean values of three cells.



**Figure S10.** Comparison of the discharge capacity development during cycling (12 mA  $g^{-1}$ ) of Li<sub>1.00</sub> using different upper cut-off voltages: 4.3 V (blue), 4.5 V (yellow) and 4.7 V (green).



**Figure S11.** Electrochemical characterization of  $Li_{1.00}$ ,  $Li_{1.07}$ ,  $Li_{1.10}$  and  $Li_{1.16}$  during the initial cycles (12 mA g<sup>-1</sup>): supplementary dQ/dV plots of the cycling data presented in Figure 9.

## References

- 1 J. Bréger, Y. S. Meng, Y. Hinuma, S. Kumar, K. Kang, Y. Shao-Horn, G. Ceder and C. P. Grey, *Chem. Mater.*, 2006, **18**, 4768.
- 2 P. Skemer, I. Katayama, Z. Jiang and S. Karato, *Tectonophysics*, 2005, **411**, 157.