

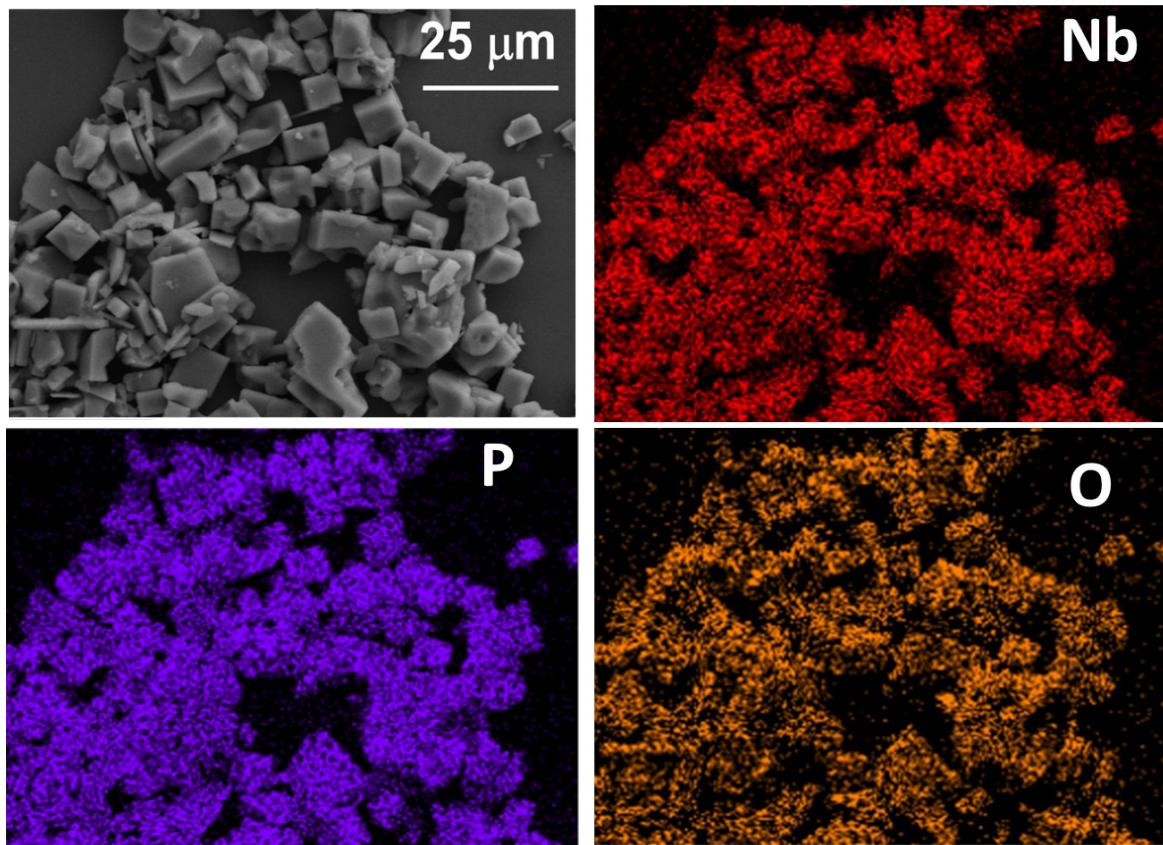
## ***Supplementary Information***

### **Unveiling a High Capacity Multi-redox ( $\text{Nb}^{5+}/\text{Nb}^{4+}/\text{Nb}^{3+}$ ) NASICON- $\text{Nb}_2(\text{PO}_4)_3$ Anode for Li- and Na-ion Batteries**

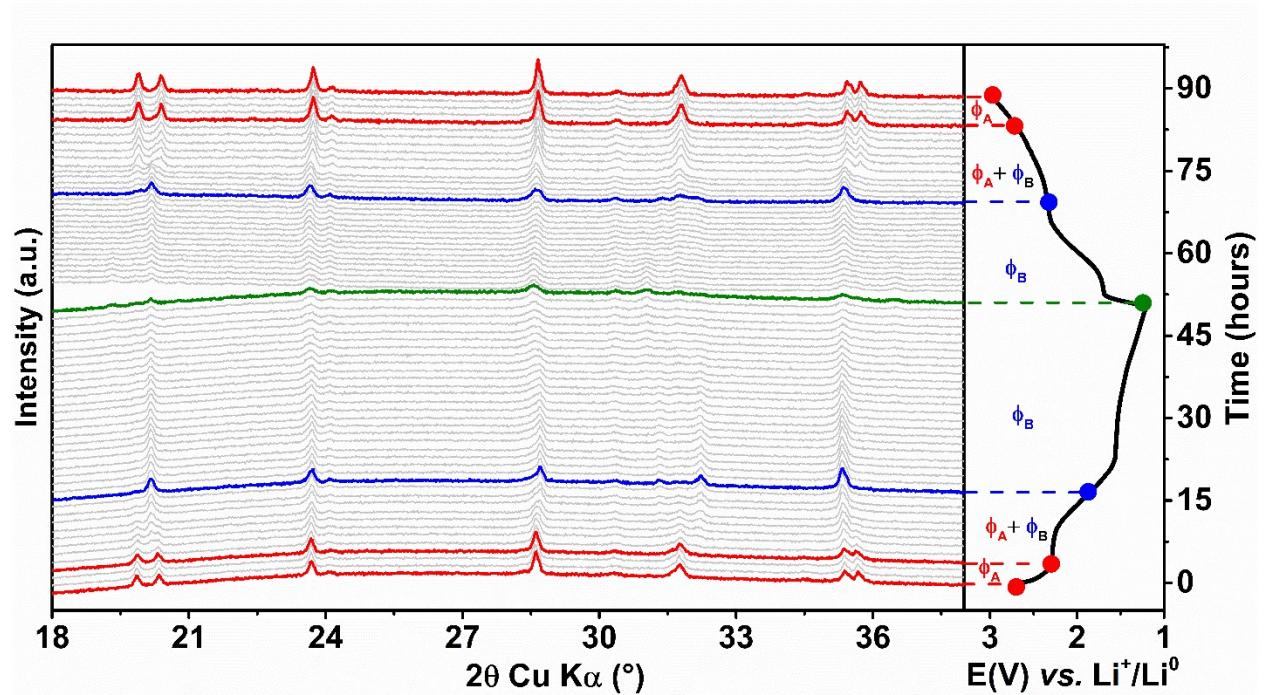
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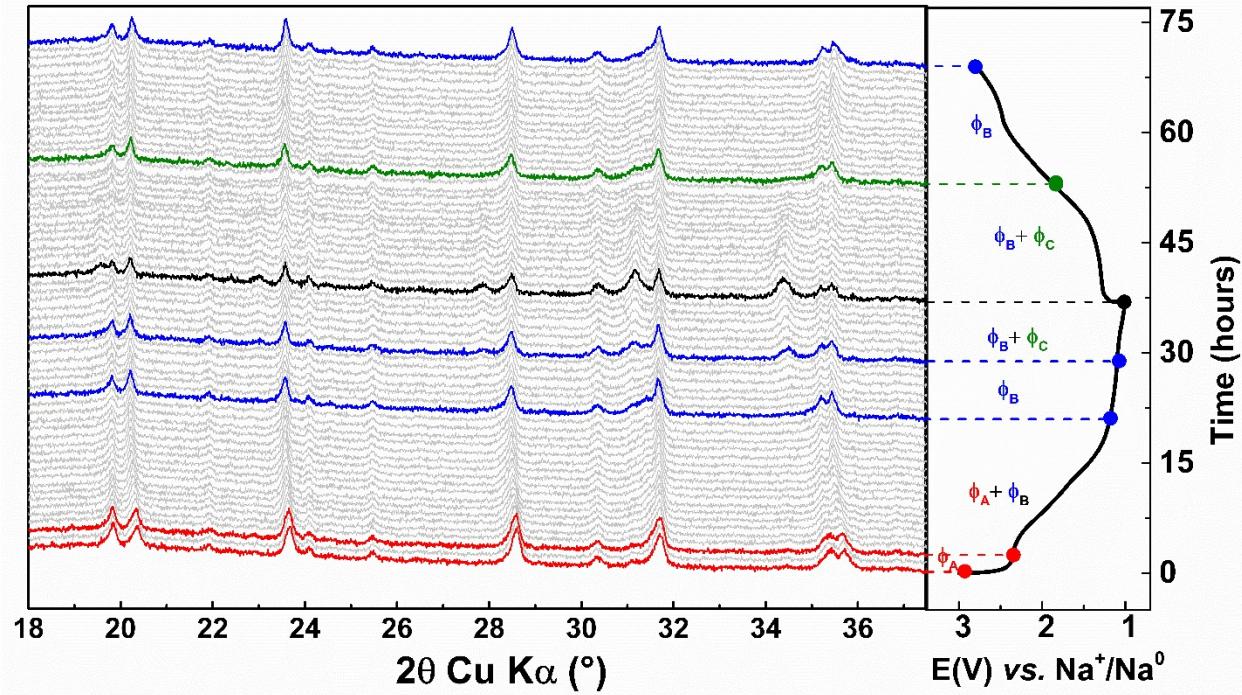
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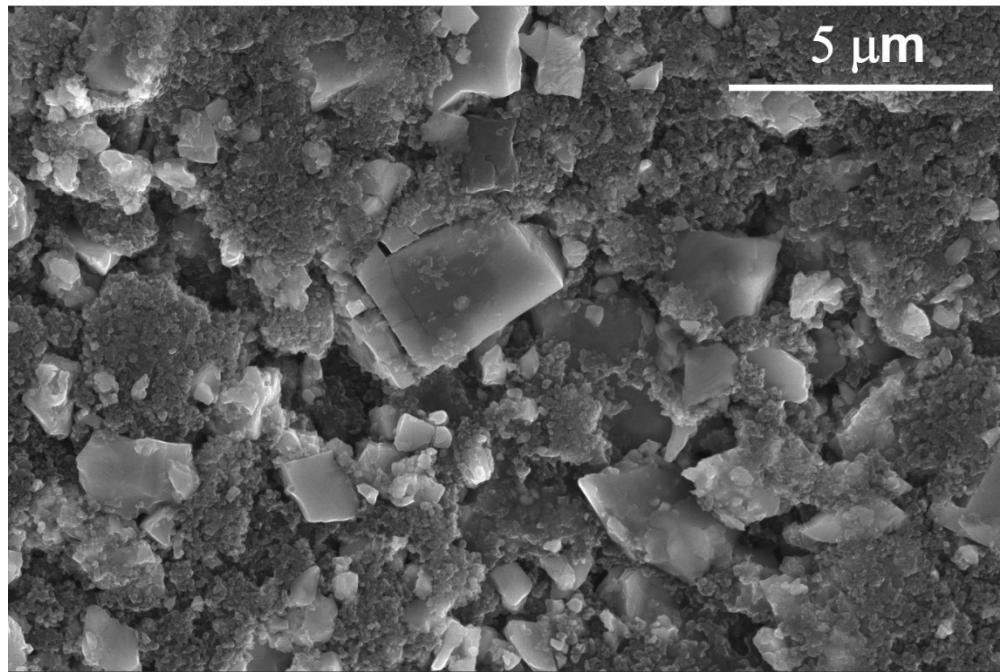
**Figure S1.** FESEM images and EDS mapping of  $\text{Nb}_2(\text{PO}_4)_3$  anode.



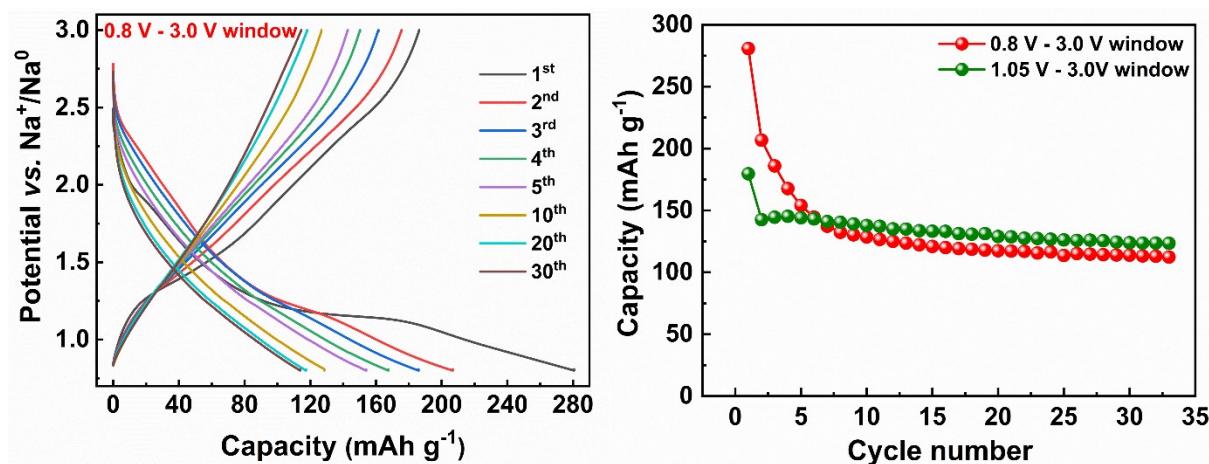
**Figure S2.** *In-situ* XRD patterns of  $\text{Nb}_2(\text{PO}_4)_3/\text{Li}$  cell collected during 2<sup>nd</sup> cycle.



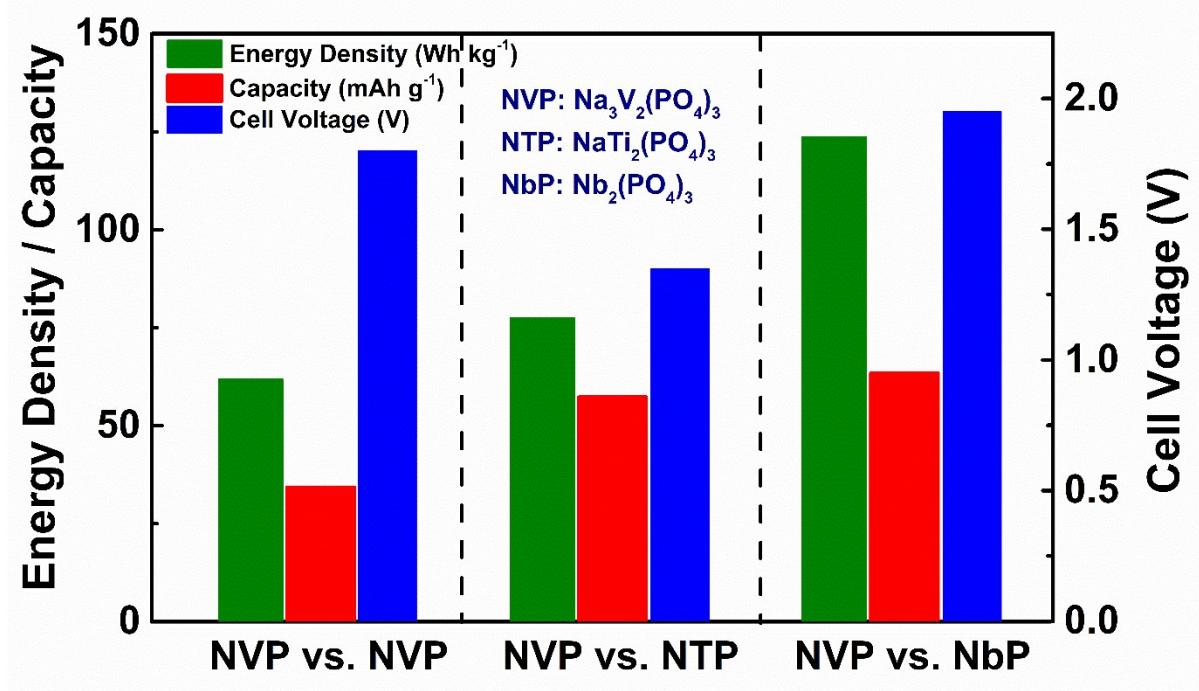
**Figure S3.** *In-situ* XRD patterns of  $\text{Nb}_2(\text{PO}_4)_3/\text{Na}$  cell collected during 2<sup>nd</sup> cycle.



**Figure S4.** SEM images of as-prepared  $\text{Nb}_2(\text{PO}_4)_3$  electrode.



**Figure S5.** Comparison of cycling performances of  $\text{Nb}_2(\text{PO}_4)_3/\text{Na}$  cells with different cut-off voltages.



**Figure S6.** A comparison of energy densities of Na-ion cells comprising  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  as the cathode and  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  or  $\text{NaTi}_2(\text{PO}_4)_3$  or  $\text{Nb}_2(\text{PO}_4)_3$  as the anode.

The energy densities of Na-ion cells comprising  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  as the cathode and  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  or  $\text{NaTi}_2(\text{PO}_4)_3$  or  $\text{Nb}_2(\text{PO}_4)_3$  as the anode are calculated using the following formula:

$$\text{Energy density } (\text{Wh kg}^{-1}_{(\text{anode+cathode})}) = \text{average cell voltage} \times \text{capacity } (\text{mAh g}^{-1}_{(\text{anode+cathode})})$$

Capacity ( $\text{mAh g}^{-1}_{(\text{anode+cathode})}$ ) is calculated using the following formula:

$$\frac{1}{C_{\text{anode} + \text{cathode}}} = \frac{1}{C_{\text{anode}}} + \frac{1}{C_{\text{cathode}}}$$

The insertion potentials and capacities of NASICON compounds (at C/10 rate) are taken from the references.<sup>1,2</sup>

**Table S1.** Crystallographic parameters and atomic coordinates of NASICON-Nb<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>

Nb <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub>						
S.G.: $\bar{R}\bar{3}c$ ; Z = 6						Chi <sup>2</sup> =4.56;
$R_{\text{bragg}} = 2.33\%$ ; $R_p = 8.23\%$ ; $R_{\text{wp}} = 11.3\%$						
$a = 8.6629(1)$ Å; $c = 22.0627(6)$ Å.						
$V = 1433.92(5)$ Å <sup>3</sup> ;						
Average Nb-O Bond distance: (Nb-O1) × 3 = 1.975(3) Å; (Nb-O2) × 3 = 1.9368(2) Å						
Atom	Wyckoff site	X	Y	Z	B (Å <sup>2</sup> )	Occ.
Nb1	12c	0	0	0.14122(2)	0.0049(2)	0.5
Nb2	12c	0	0	0.14122(2)	0.0049(2)	0.5
P	18e	0.283(2)	0	0.25	0.0145(1)	1.0
O1	36f	0.1684(2)	-0.0278(3)	0.1941(9)	0.0127(7)	1.0
O2	36f	0.1978(2)	0.1674(2)	0.0916(8)	0.0063(2)	1.0

**Table S2.** Refined parameters for the first shell of Nb K-edge EXAFS spectra collected on the pristine, discharged and charge Nb<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> anodes.

Sample	Coordination number	d(Nb-O) Å	E <sub>0</sub> (eV)	$\sigma^2$ Å <sup>2</sup>	R-factor
Pristine	3 + 3	2.095(2) x 3 1.961(2) x 3	3.00	0.0001	0.0146
Discharge 1.0V	3.98 + 2.02	2.044(5) x 3.98 2.169(8) x 2.02	5.65	0.001(9)	0.0085
Charge 3.0V	3.3 + 2.7	2.104(5) x 2.7	3.57	0.0008	0.0193

		1.962(8) x 3.3			
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## References

- 1 M. K. Sadan, A. K. Haridas, H. Kim, C. Kim, G. B. Cho, K. K. Cho, J. H. Ahn and H. J. Ahn, *Nanoscale Adv.*, 2020, **2**, 5166–5170.
- 2 M. Wu, W. Ni, J. Hu and J. Ma, *Nano-Micro Lett.*, 2019, **11**, 1–36.