

Supporting Information

Nano-engineering induced Bi dots *in-situ* anchored into modified porous carbon with superior sodium ion storage

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Detailed steps for in situ XRD testing:

The Bi@MC slurry was pasted onto an aluminum net, followed by drying at 120°C under vacuum for 12 h. The dried material was used as the electrode. Sodium foil and glass fiber was severed as the counter electrode and separator, respectively, and 1.0 M NaPF₆ in DME as electrolyte. The XRD chamber was equipped with a Be window to permit X-ray passage. *In situ* XRD patterns of the Bi@MC electrodes were tested on Bruker D8 Advanced (Cu Kα) and studied by a specially-made chamber (Beijing scistar technology co. ltd) in a scanning range of 10-90°(2θ) with step increment of 2.5°. The Be foil is the X-ray penetrator. The corresponding galvanostatic charge-discharge cycling was carried out at a current density of 100 mA g⁻¹ between 0.1 and 1.6 V versus Na⁺/Na.

The Bi content was calculated based on the following equation:

$$Bi(wt\%) = \frac{\text{final weight of } Bi_2O_3}{\text{initial weight of Bi@MC}} \times \frac{2 * \text{molecular weight of Bi}}{\text{molecular weight of } Bi_2O_3} \times 100\%$$

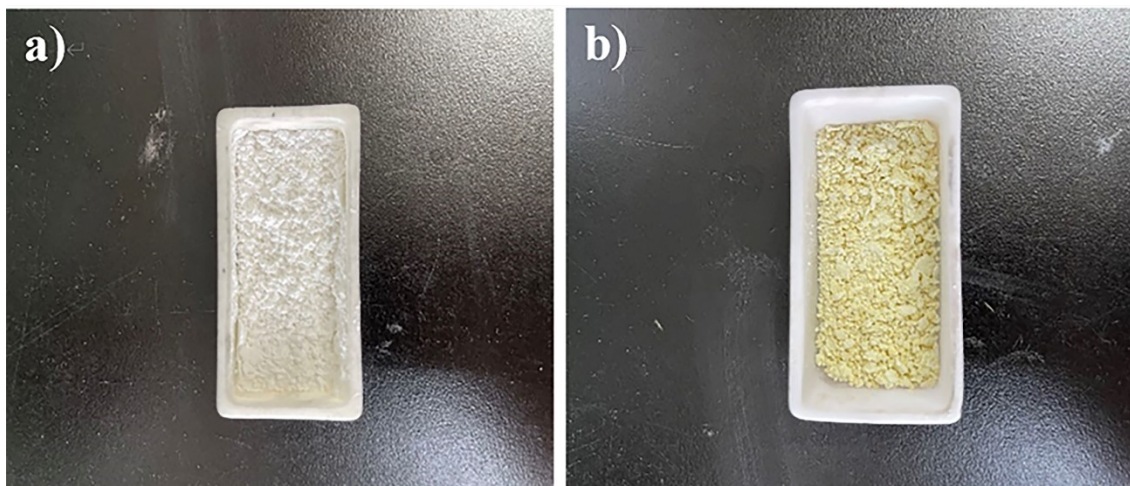


Fig. S1 Digital photographs of a) bismuth subsalicylate (BS) and b) a mixture of BS and KOH.

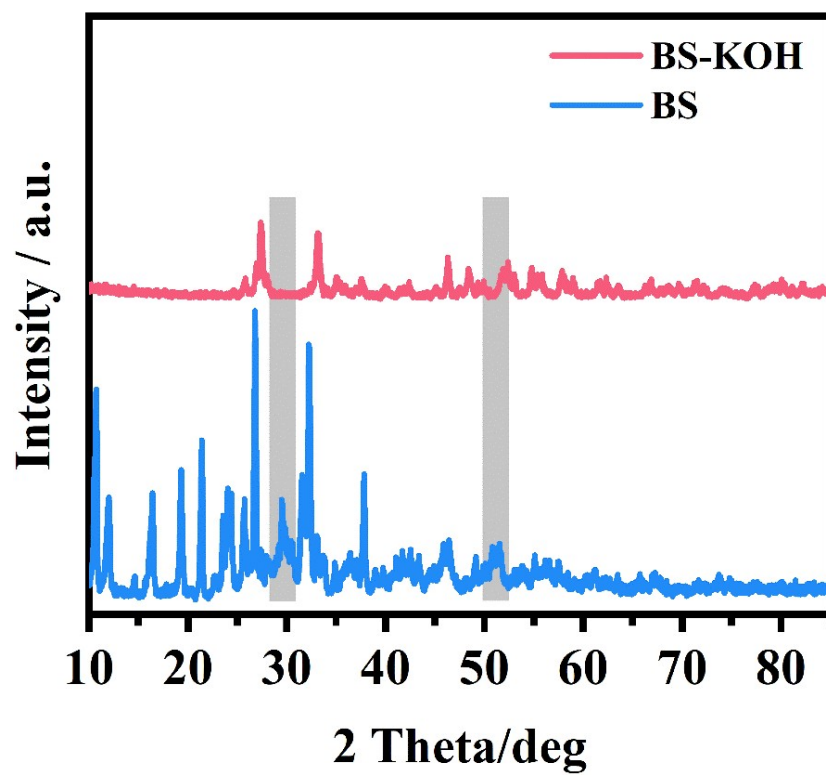


Fig. S2 XRD patterns of BS and the mixture of BS and KOH.

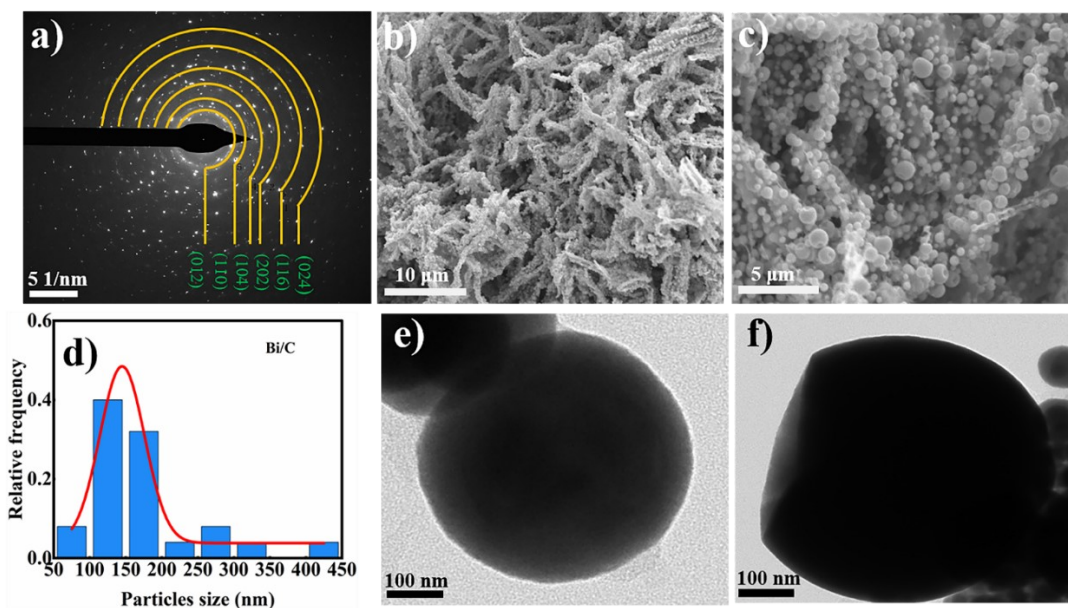


Fig. S3 a) SAED image of Bi@MC, b-c) SEM images of Bi/C, d) the size distribution of Bi/C, e-f) TEM images of Bi/C.

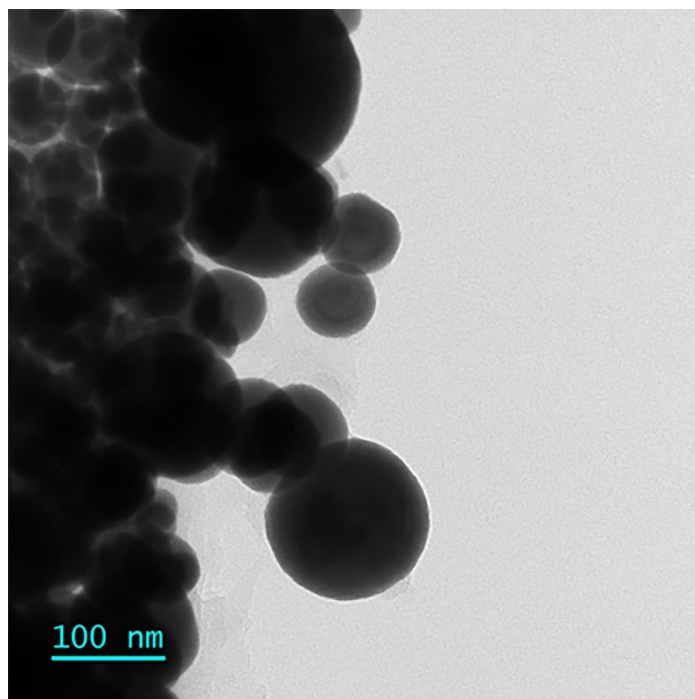


Fig. S4 TEM image was used to calculate the particle size distribution of Bi/C.

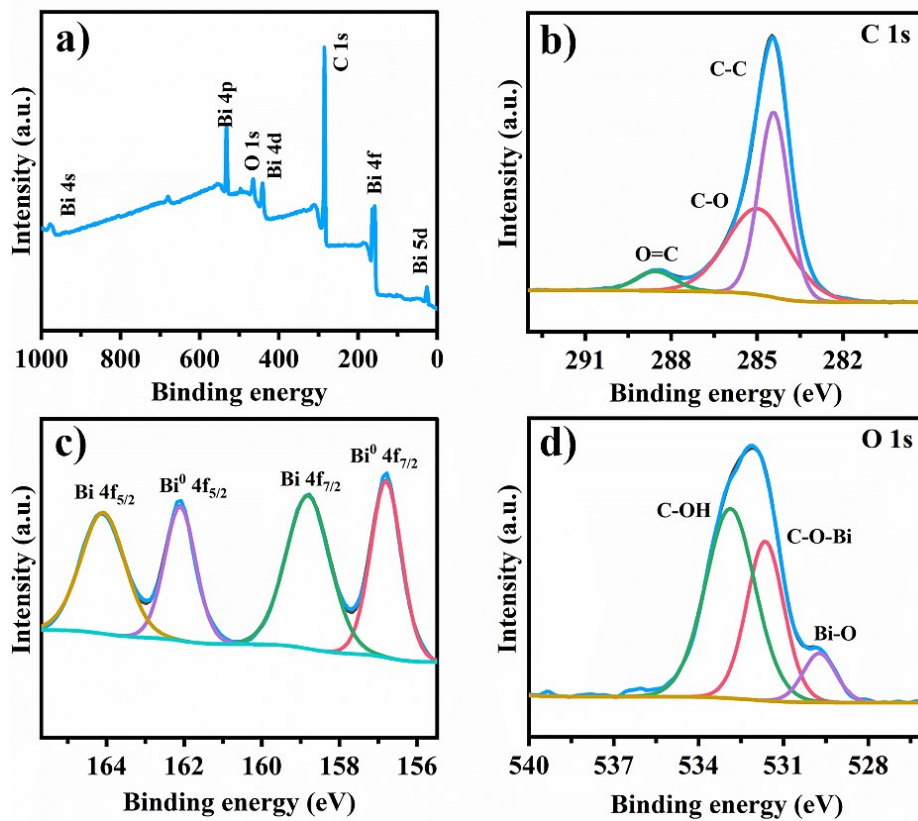


Fig. S5 XPS spectra of (a) full survey, (b) high-resolution C 1s, (c) high-resolution Bi 4f, and (d) high-resolution O1s for Bi/C.

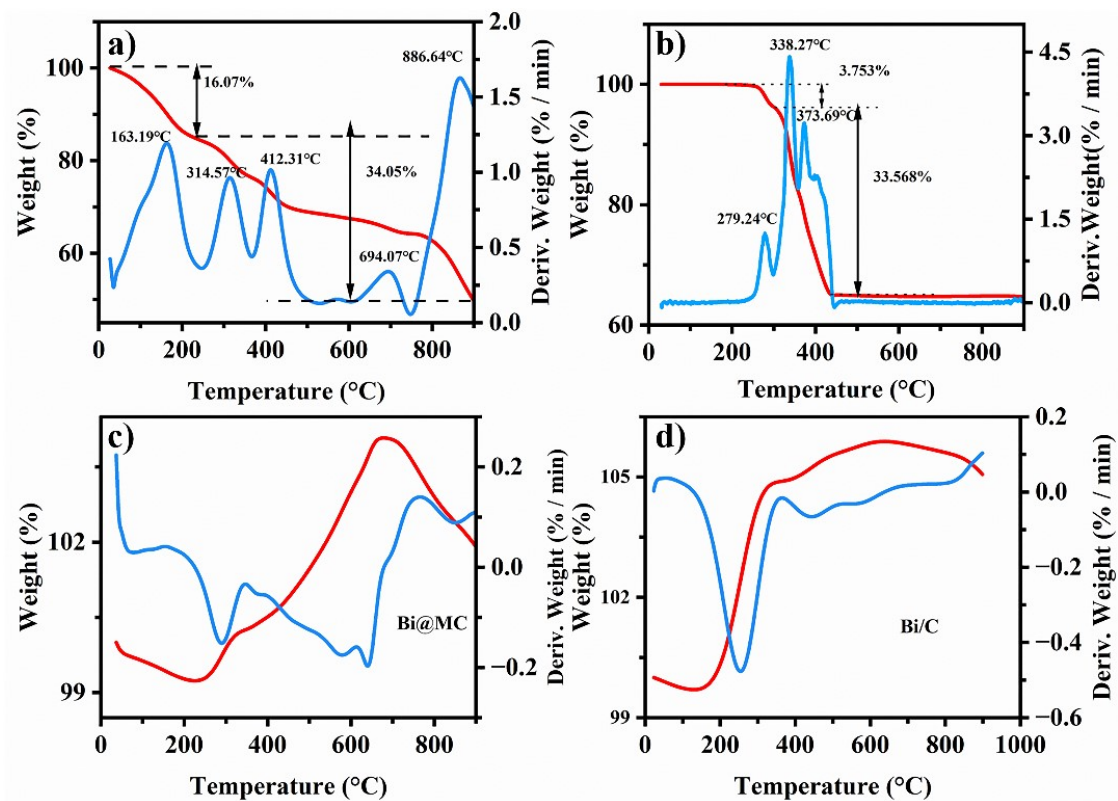


Fig. S6 The TGA curves of a) mixture of BS and KOH, b) BS in Ar, c) Bi@MC and d) Bi/C in air.

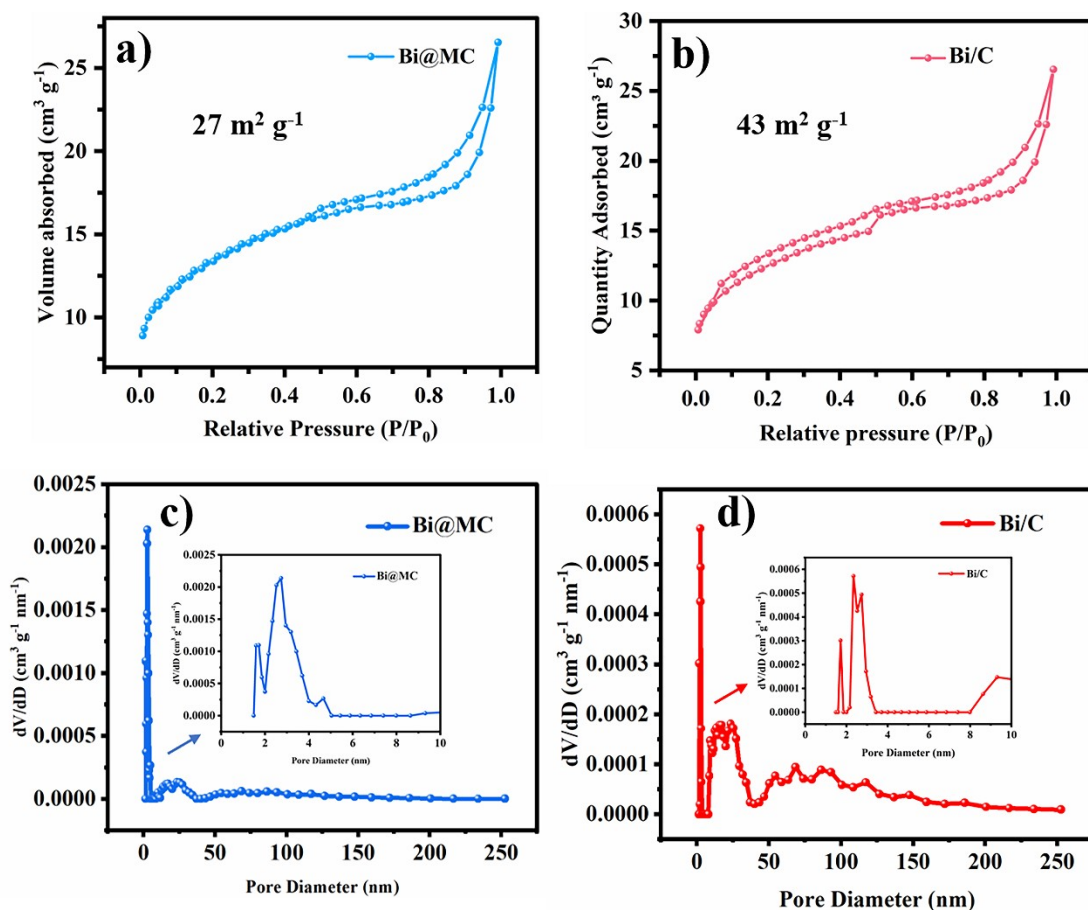


Fig. S7 N_2 adsorption-desorption isotherms spectra of a) Bi@MC and b) Bi/C. The pore size distribution of c) Bi@MC and d) Bi/C.

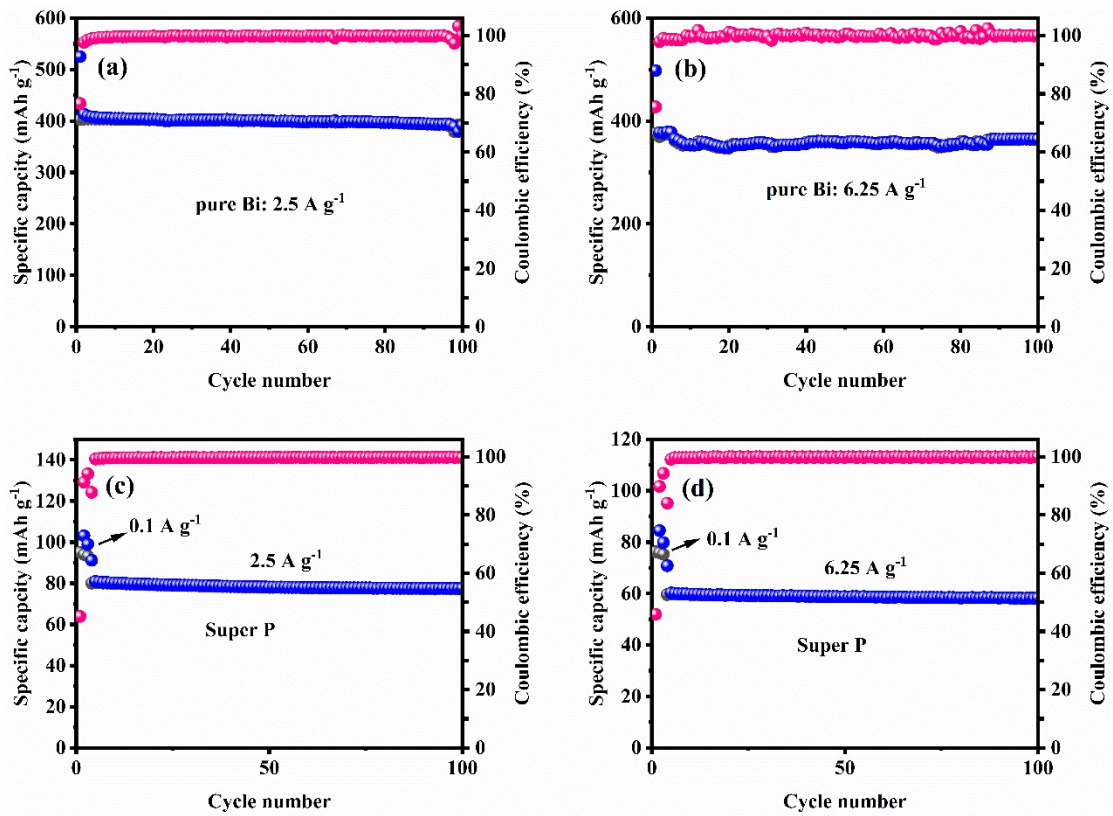


Fig. S8 Cycle performance of pure Bi and super P at a, c) 2.5 A g⁻¹, b, d) 6.25 A g⁻¹.

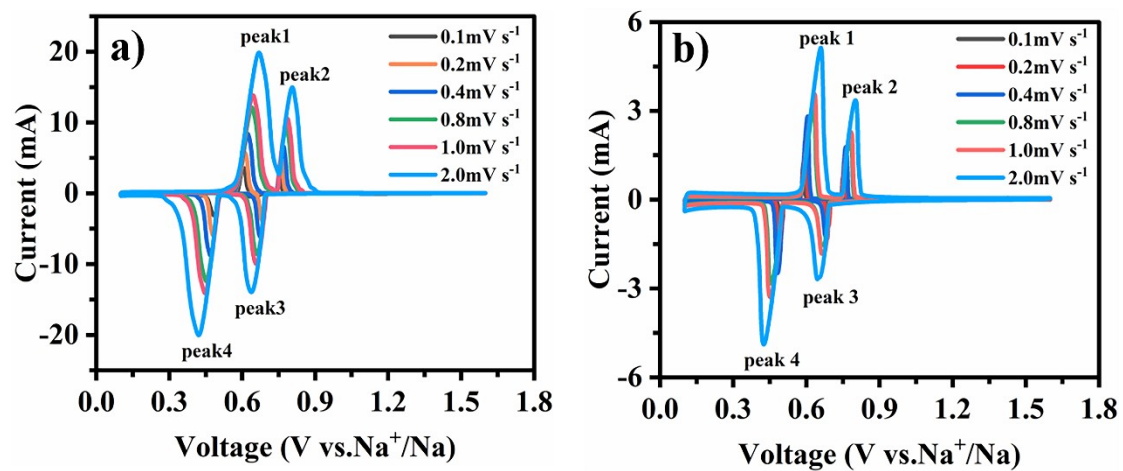


Fig. S9 CV curves of a) Bi@MC and b) Bi/C at different scan rates (0.1, 0.2, 0.4, 0.8, 1, and 2 mV s⁻¹, respectively).

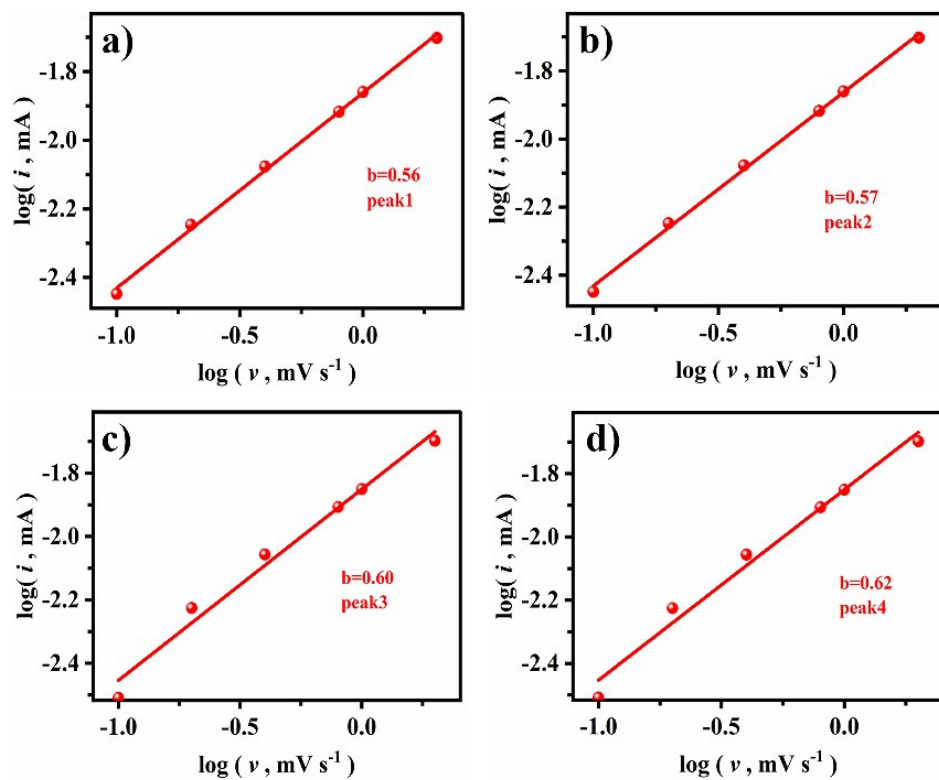


Fig. S10 The fitted linear relationship of $\log(v)$ versus $\log(i)$ for Bi/C based on the CV curves at different rates shown in Fig S8 b): a) peak1, b) peak2, c) peak3, d) peak4.

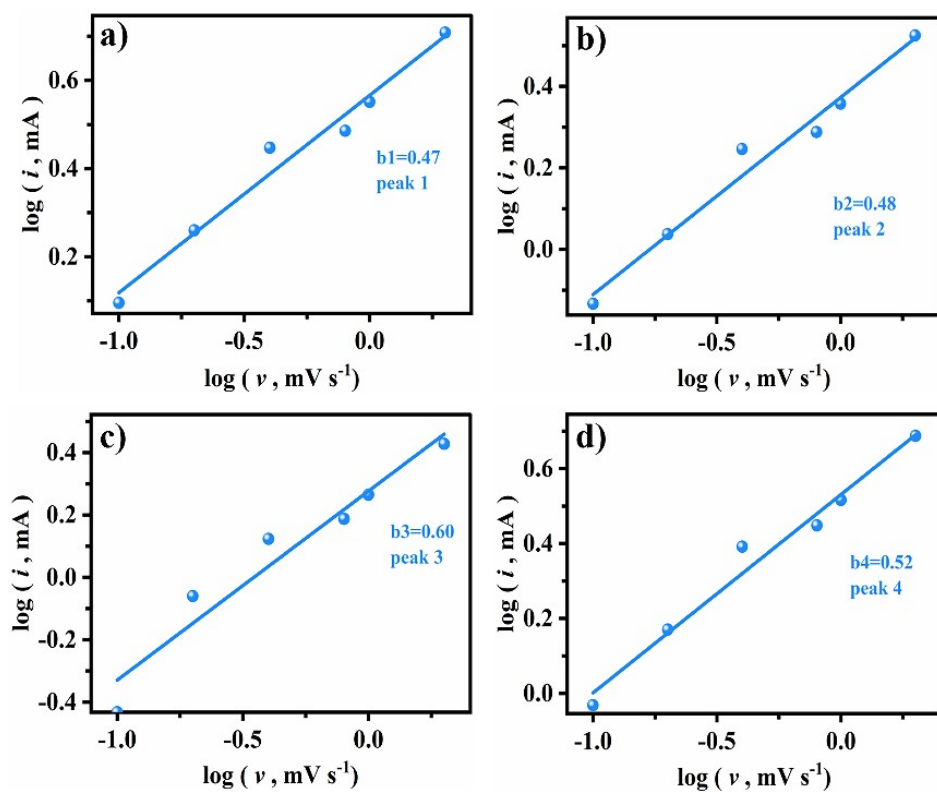


Fig. S11 The fitted linear relationship of $\log(v)$ versus $\log(i)$ for Bi@MC based on the CV curves at different rates shown in Fig S8 a): a) peak1, b) peak2, c) peak3, d) peak4.

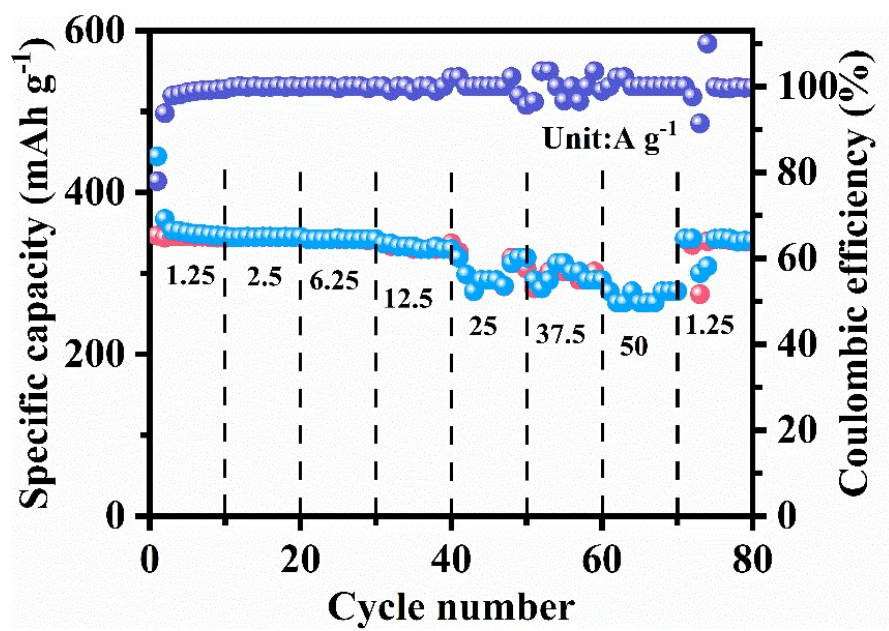


Fig. S12 Rate performance of the Bi/C electrode.

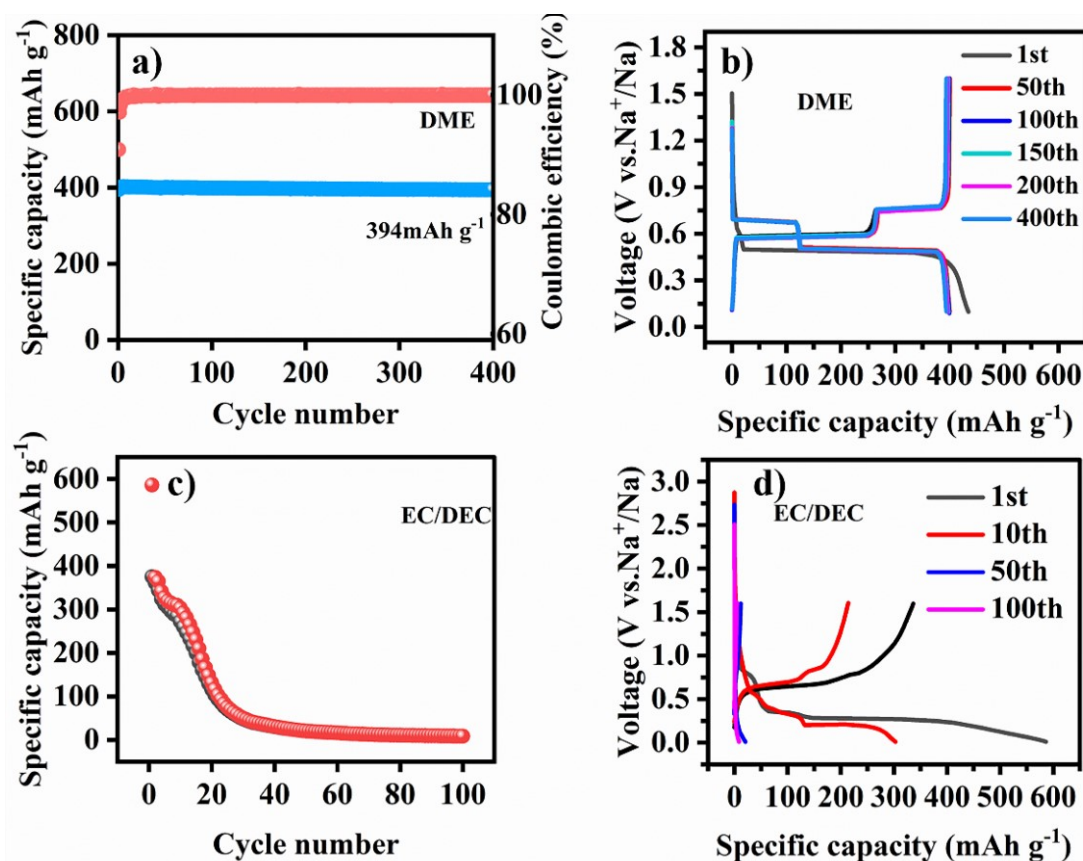


Fig. S13 a) Electrochemical performance and b) selected charge/discharge curves of Bi@MC in DME-based electrolyte at a current density of 1.25 A g⁻¹, c) electrochemical performance and d) selected charge/discharge curves of Bi@MC in EC/DEC-based electrolyte at a current density of 1.25 A g⁻¹.

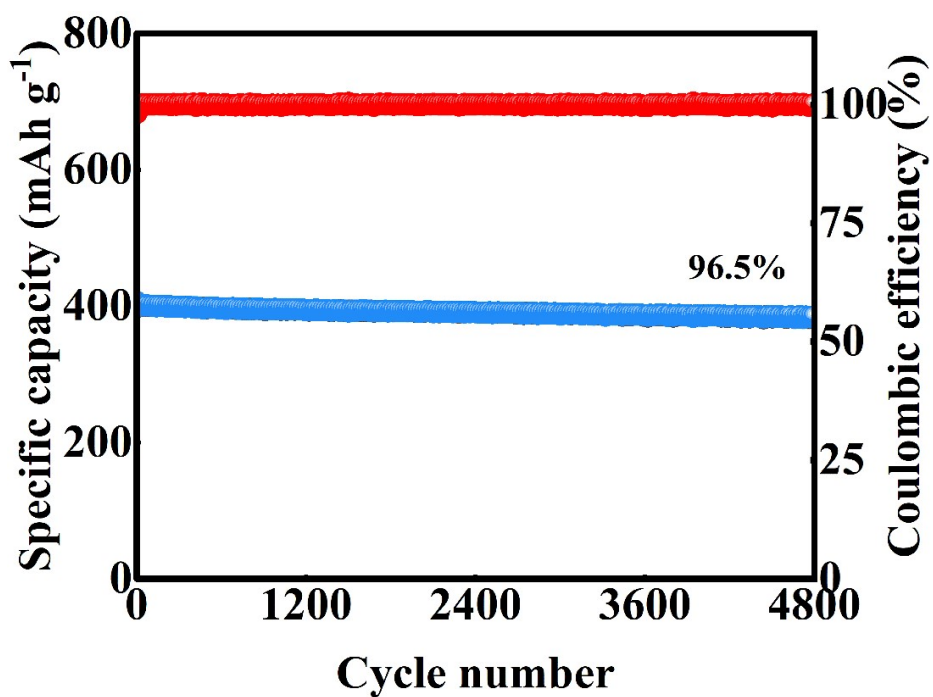


Fig. S14 Long cycle test of Bi@MC electrode at 5 A g⁻¹.

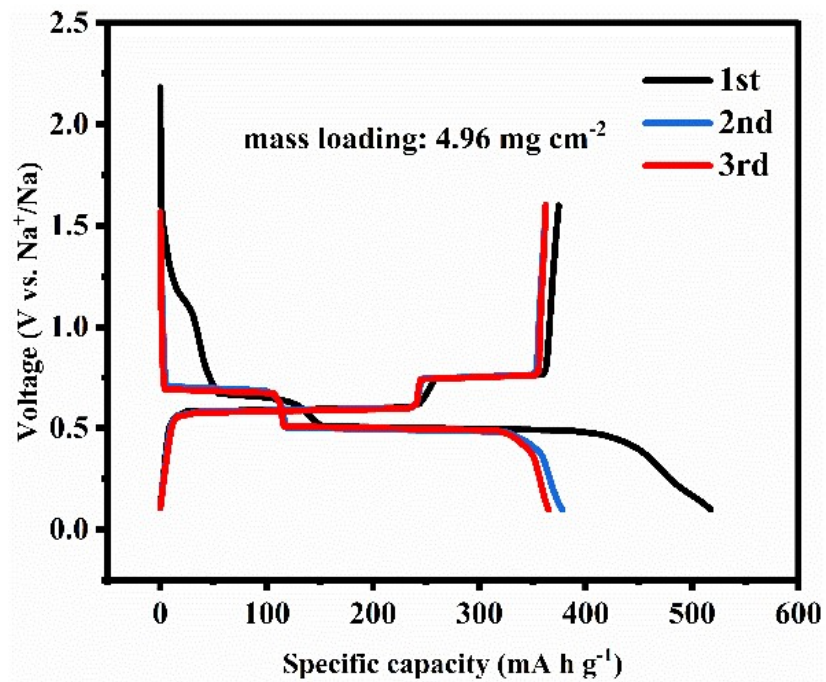


Fig. S15 The electrochemical performance of Bi@MC with mass loading of 4.96 mg cm⁻².

Table S1. Electrochemical comparison of this work with other reported Bi-based anodes for SIBs.

Electrode materials	Capacity @ cycle number at current density	Rate performance (mAh g⁻¹)	Reference
Bi@MC	378 mAh g⁻¹@ 10000 at 2.5 A g⁻¹ 379 mAh g⁻¹@ 5000 at 6.25 A g⁻¹	345 mAh g⁻¹at 50000mA g⁻¹	This Work
Bi@C=CFs	305 @ 5000 at 5 A g ⁻¹	308.8 at 80000 mA g ⁻¹	1
Bulk Bi	360 @ 3500 at 7.7 A g ⁻¹	325 at 57750 mA g ⁻¹	2
H-BiS/EGF	261@17300 at 5 A g ⁻¹	228 at 25000 mA g ⁻¹	3
HBiC	264@15000 at 5 A g ⁻¹	297 at 10000 mA g ⁻¹	4
Bi@graphene	200@50 at 40 mA g ⁻¹	250 @1280 mA g ⁻¹	5
Bi@C	300@100 at 150 mA g ⁻¹	—	6
Bi@C microsphere	123.5@100 at 100 mA g ⁻¹	83.4 @2000 mA g ⁻¹	7
Bi nanorod bundle	301.9@150 at 50 mA g ⁻¹	102.3@2000 mA g ⁻¹	8
Bulk Bi	389@2000 at 400 mA g ⁻¹	355@2000 mA g ⁻¹	9
Bi-NS@C	106@1000 at 200 mA g ⁻¹	110@2000 mA g ⁻¹	10
Bi/CFC	350@300 at 50 mA g ⁻¹	120@2000 mA g ⁻¹	11
Bi/C nanofibers	273.2@500 at 100 mA g ⁻¹	69@3200 mA g ⁻¹	12
Bi nanoflakes	302.4@100 at 200 mA g ⁻¹	206.4@2000 mA g ⁻¹	13
Bi@graphite	~140@10000 at 3.2A g ⁻¹	113@48000 mA g ⁻¹	14
Bi@C nanoplates	200@200 at 150 mA g ⁻¹	74@2000 mA g ⁻¹	15
Bi@N-C	235@2000 at 10 A g ⁻¹	178@100000 mA g ⁻¹	16
Bi@3DGF	185.2@2000 at 10 A g ⁻¹	194@50000 mA g ⁻¹	17
Bi@N-C	302@1000 at 1 A g ⁻¹	368@2000 mA g ⁻¹	18

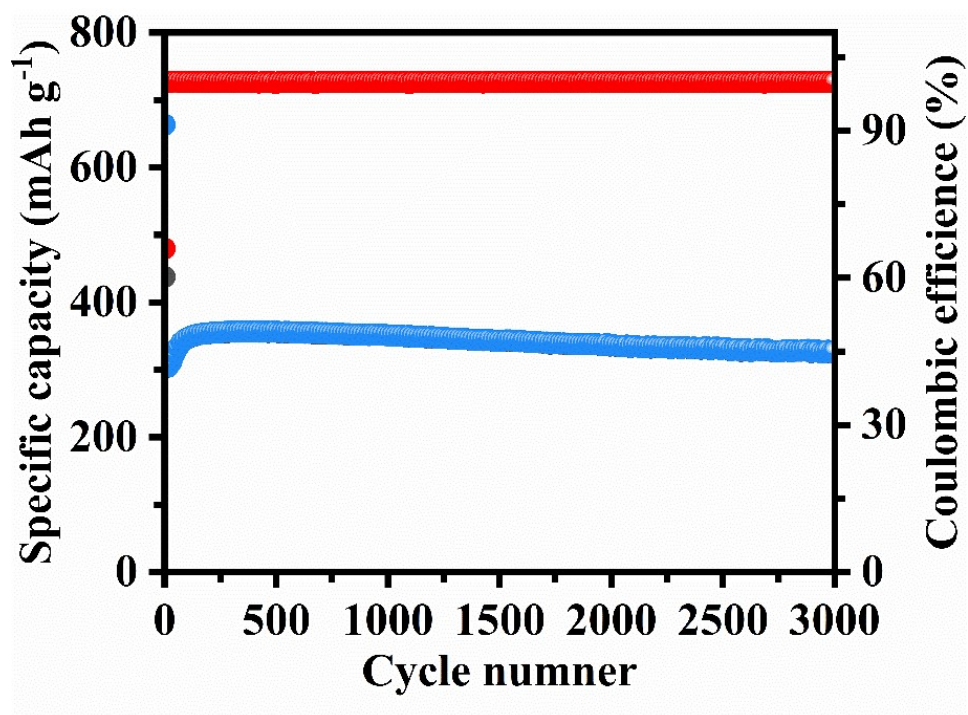


Fig. S16 Cycling performance of Bi/C electrode at a current density of 1.25 A g⁻¹ after 3000 cycles.

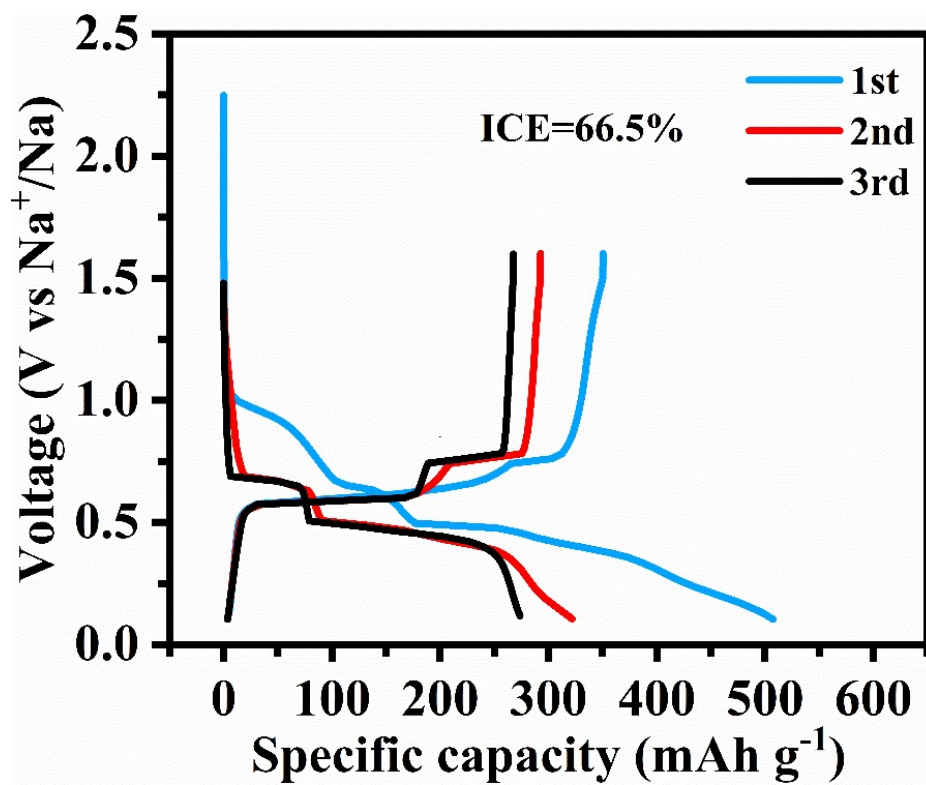


Fig. S17 First three charge/discharge curves of Bi/C electrode.

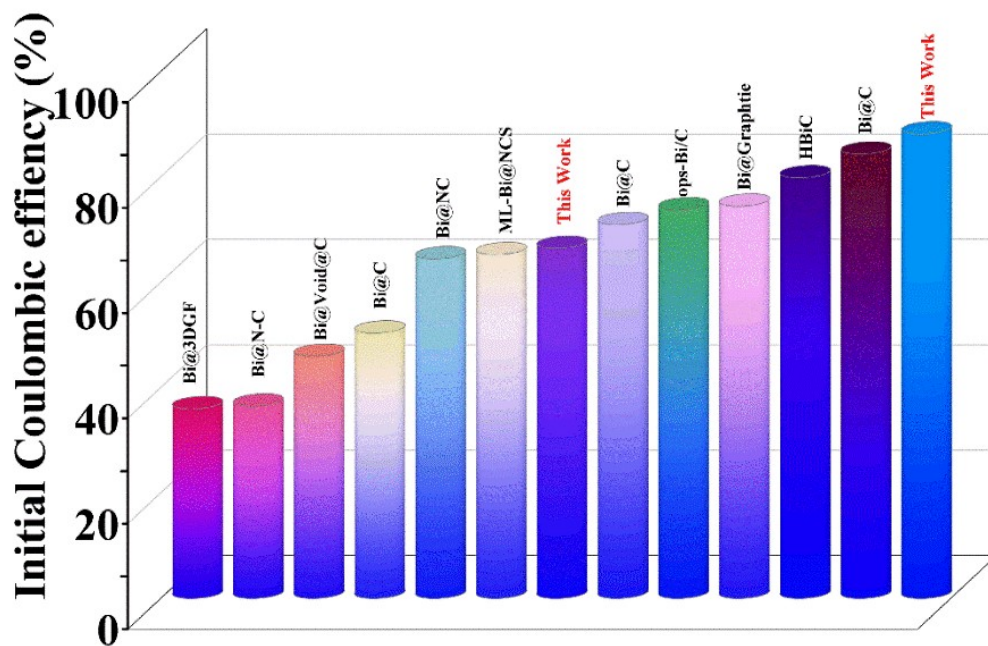


Fig.S18 Initial Coulombic efficiency comparison of this work with other reported Bi-based anodes for SIBs.

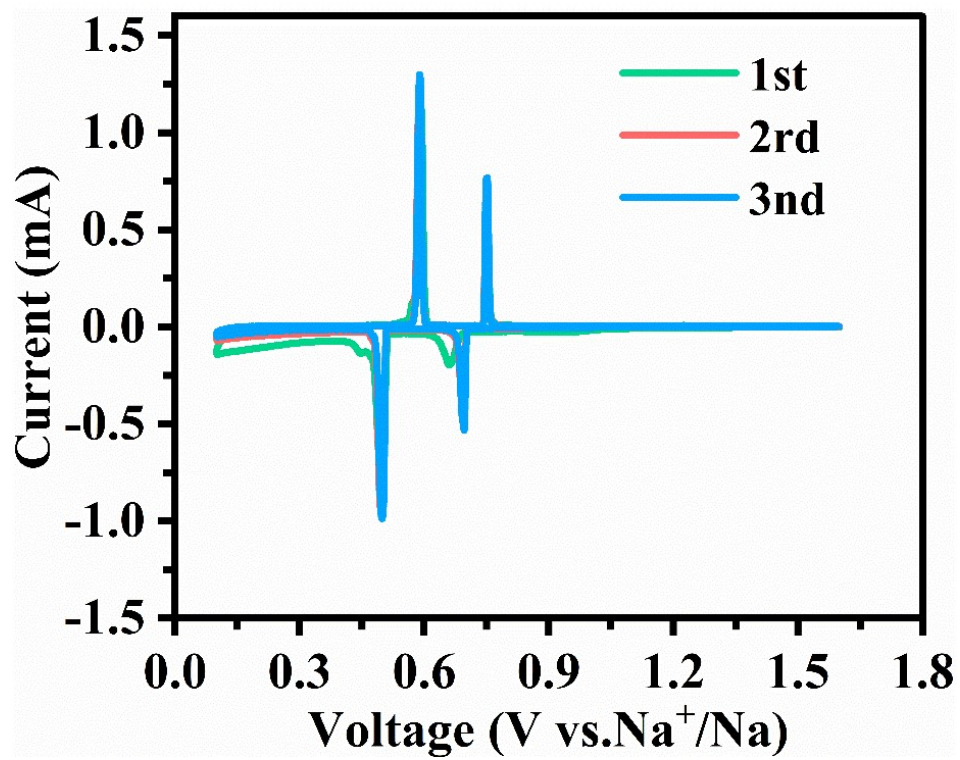


Fig.S19 First three CV curves of Bi/C electrode.

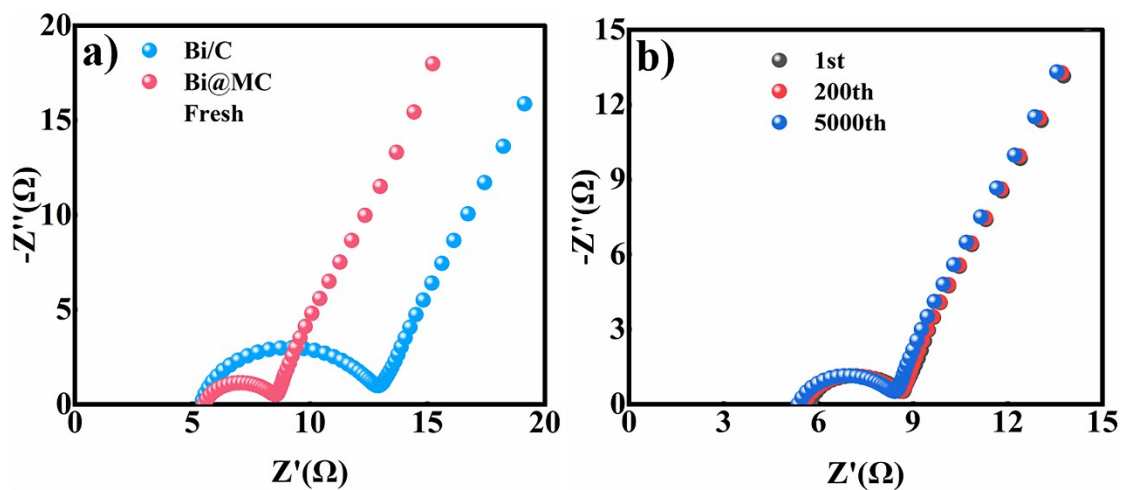


Fig. S20 a) Nyquist plots of fresh Bi@MC and Bi/C composite electrodes, b) Nyquist plots of Bi@MC electrode at 1st, 200th and 5000th cycle, respectively.

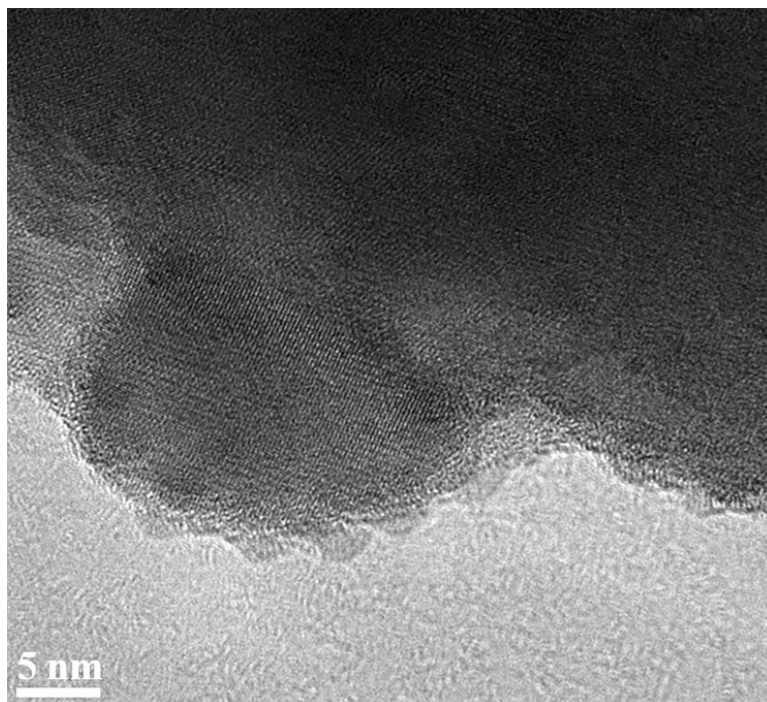


Fig. S21 the overview image of Fig. 4f.

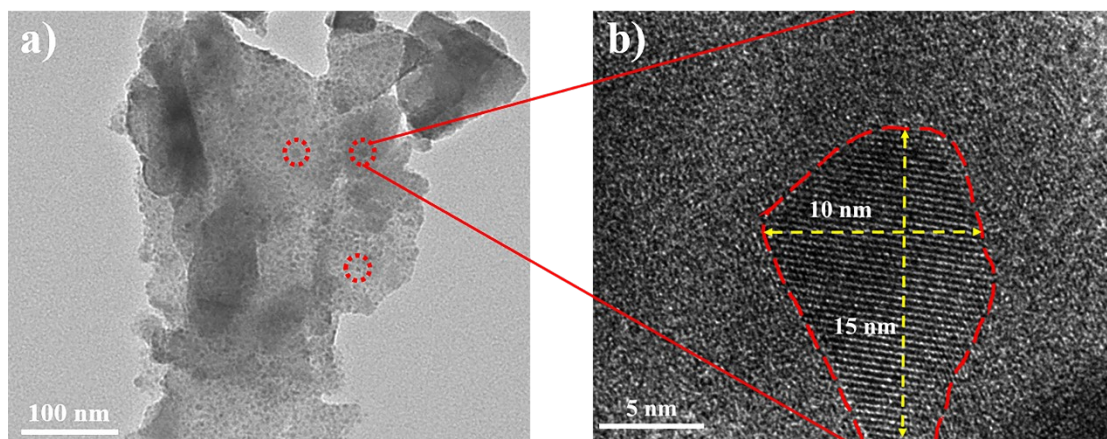


Fig. S22 HRTEM image of the Bi@MC electrode after 200 cycles.

Table S2. The distribution of two types oxygen-functional groups on Bi@MC and Bi/C composites.

Samples	O/C	C-OH (%)
Bi@MC	0.187	18.02
Bi/C	0.135	13.1

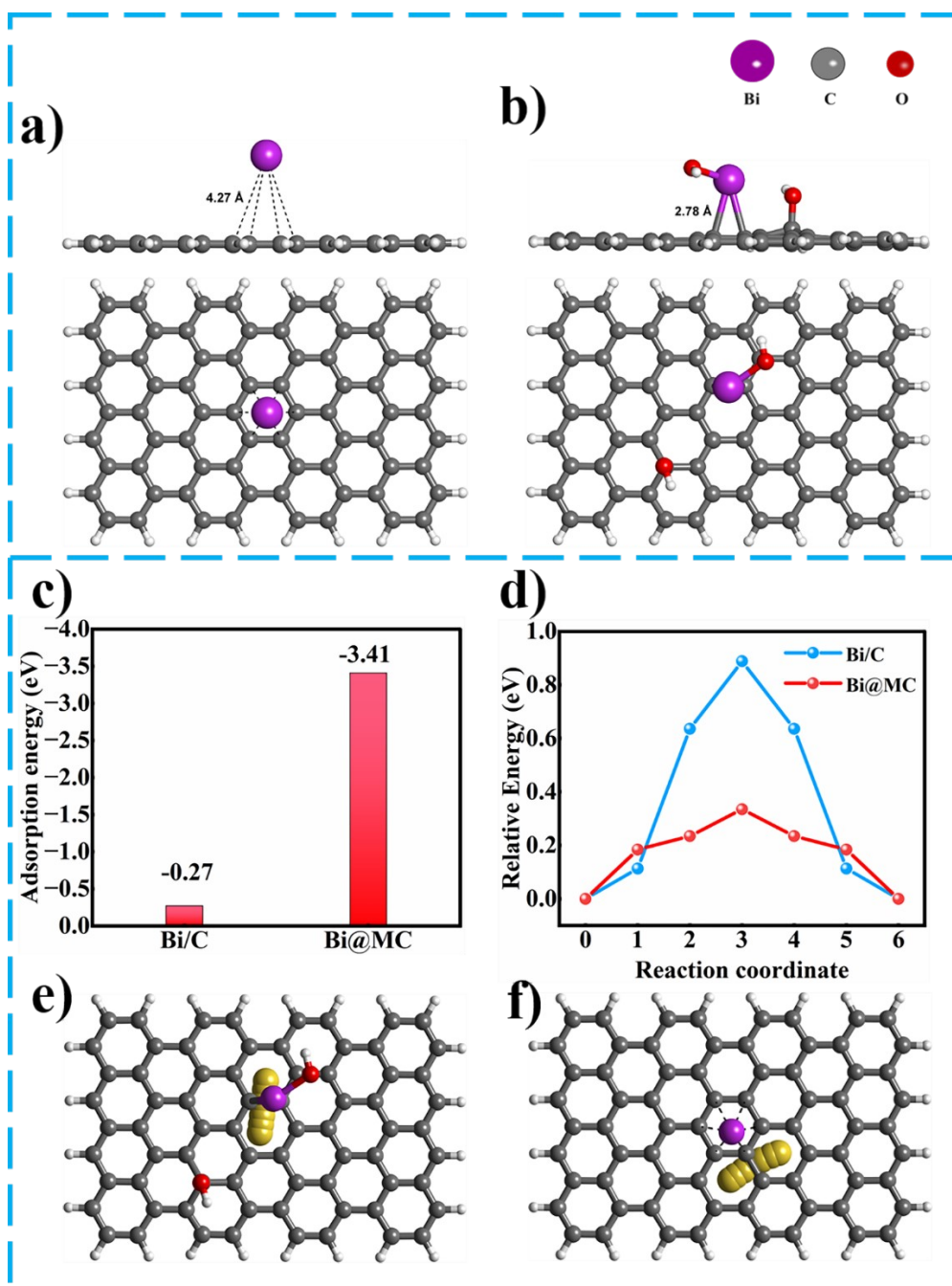


Fig. S23 Optimized structures of Bi adsorption on (a) graphene and (b) KOH-modified graphene. c) The calculated adsorption energy of Bi and d) Na diffusion energy barrier on graphene and KOH-modified graphene, respectively. Schematic representation of Na⁺ diffusion path on (e) Bi-graphene and (f) Bi-graphene-KOH.

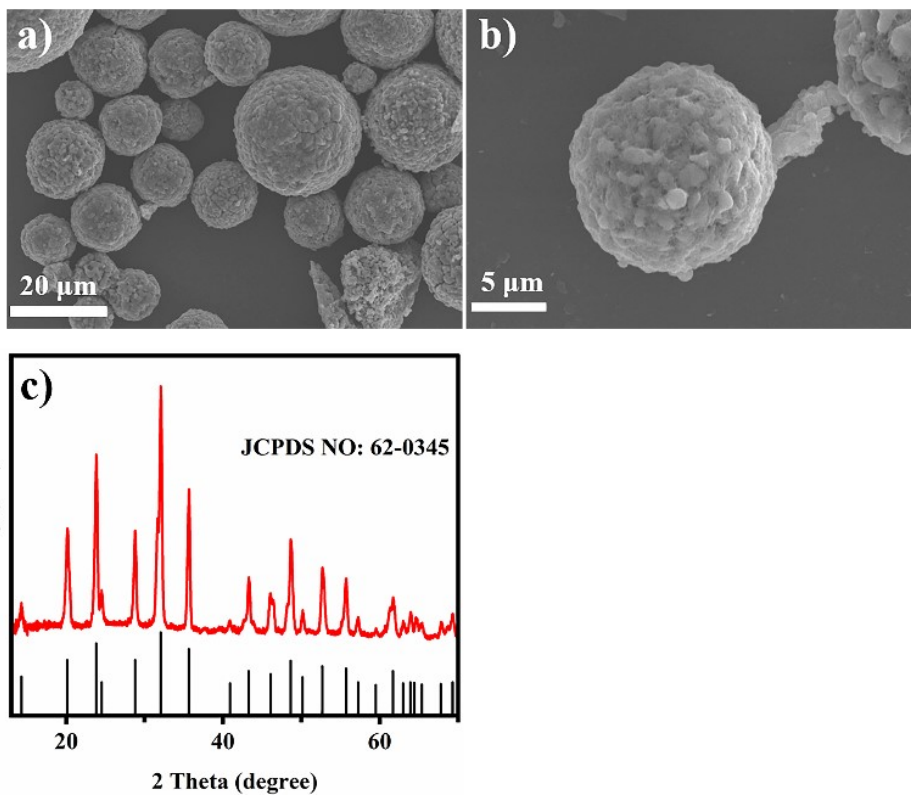


Fig. S24 The a, b) SEM image and c) XRD pattern of NVP.

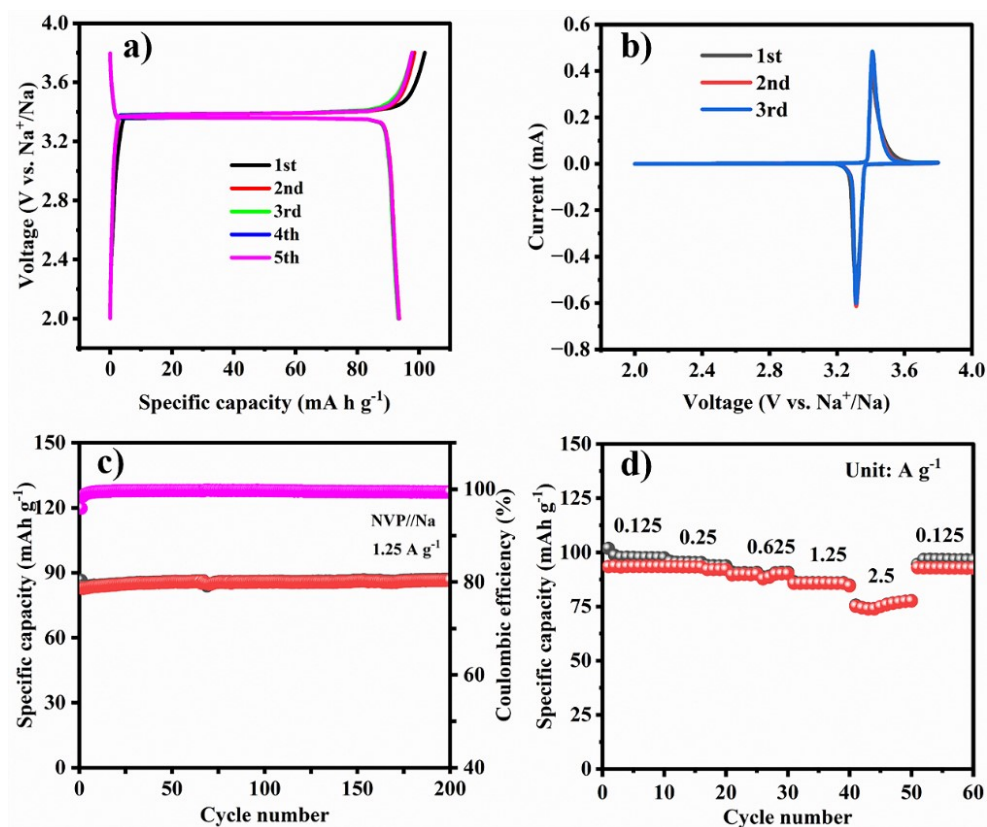


Fig. S25 a) Discharge/charge curve of NVP//Na at 1.25 A g^{-1} , b) first three CV (0.1 mV s^{-1}) curves of NVP//Na. c) electrochemical cycling performance of NVP//Na at 1.25 A g^{-1} , b) rate performance of NVP//Na at different current densities.

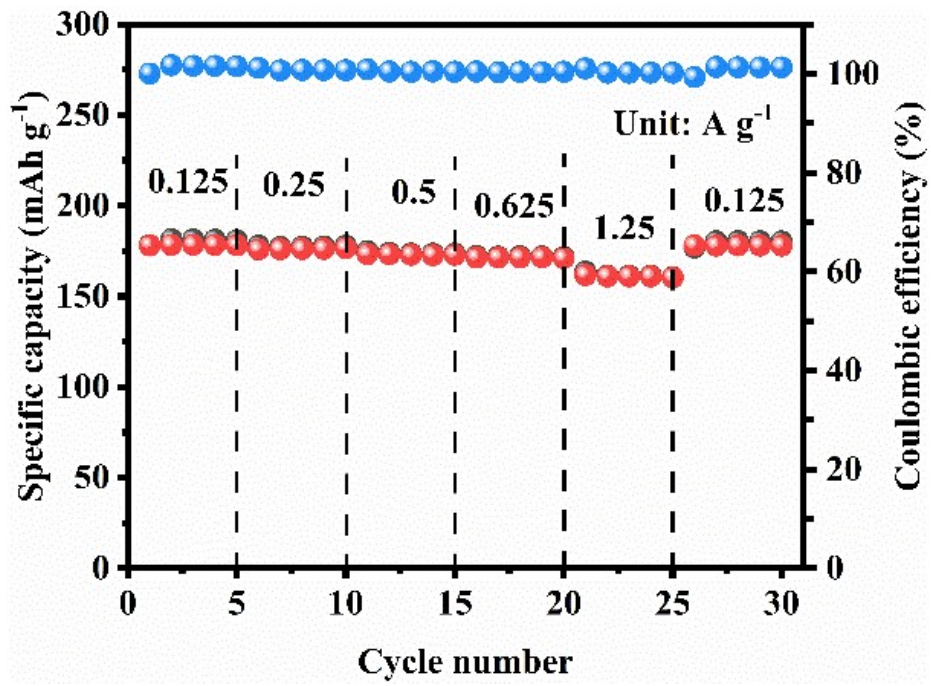


Fig. S26 The rate performance of full cell (NVP | Bi@MC).

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