

Supporting Information

Solvent Exchange-Induced Facile Recrystallisation and Particle Size Control of Sulphide Solid Electrolytes for All-Solid-State Li-Ion Batteries

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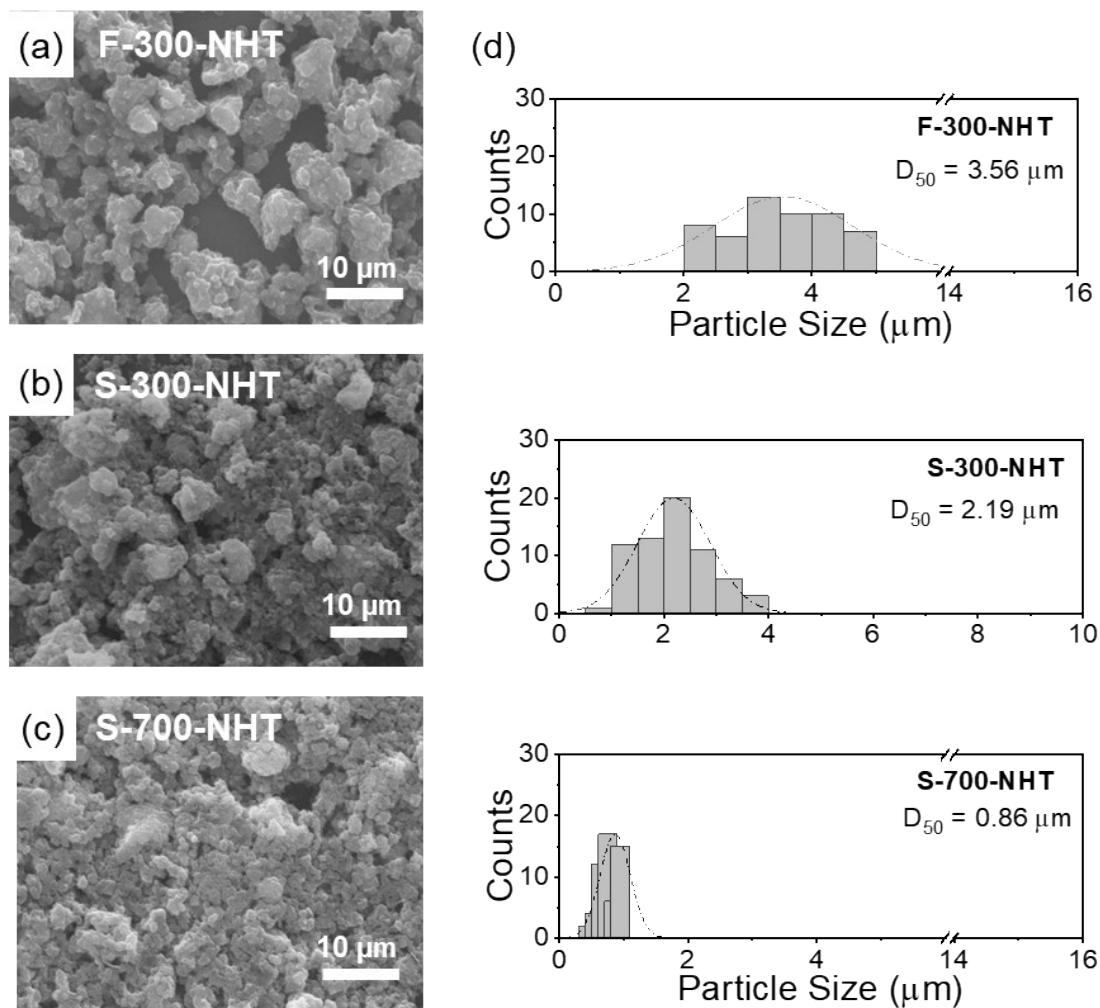
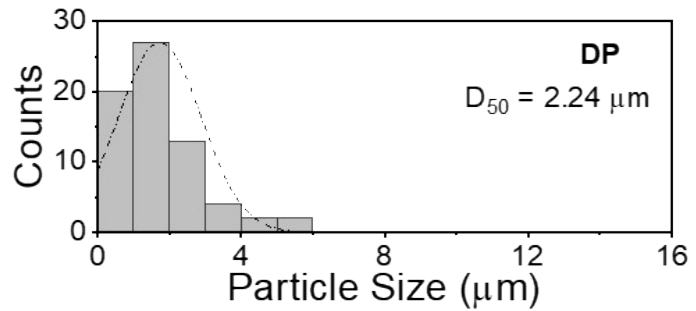
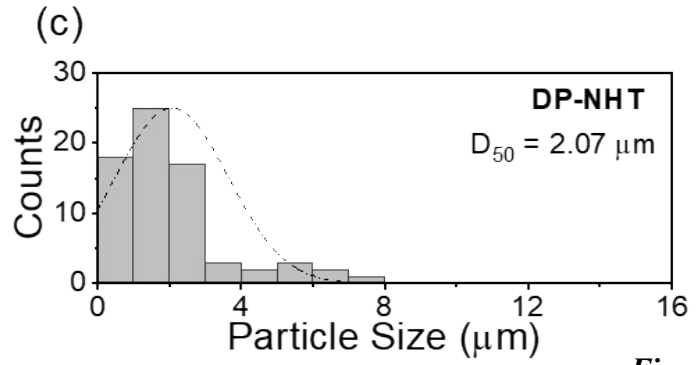
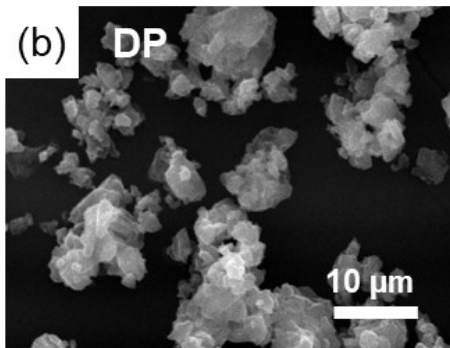
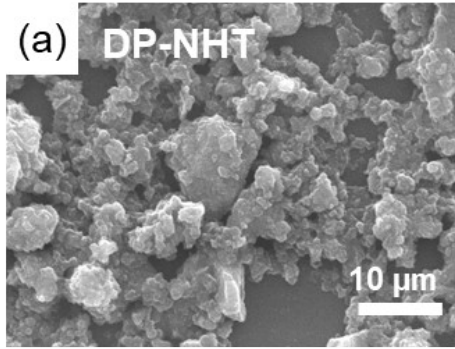


Figure S1. (a-c) SEM images of SSE powders prepared by the solvent exchange technique (a: F-300-NHT b: S-300-NHT, and c: S-700-NHT). (d) Particle size distribution of F-300-NHT, S-300-NHT, and S-700-NHT SSE powders for the estimation of the D_{50} values.



Figure

S2. (a-b) SEM images of the DP SSE powders (a:

DP-NHT and b: DP). (c) Particle size distribution of DP-NHT and DP SSE powders for the estimation of the D_{50} values.

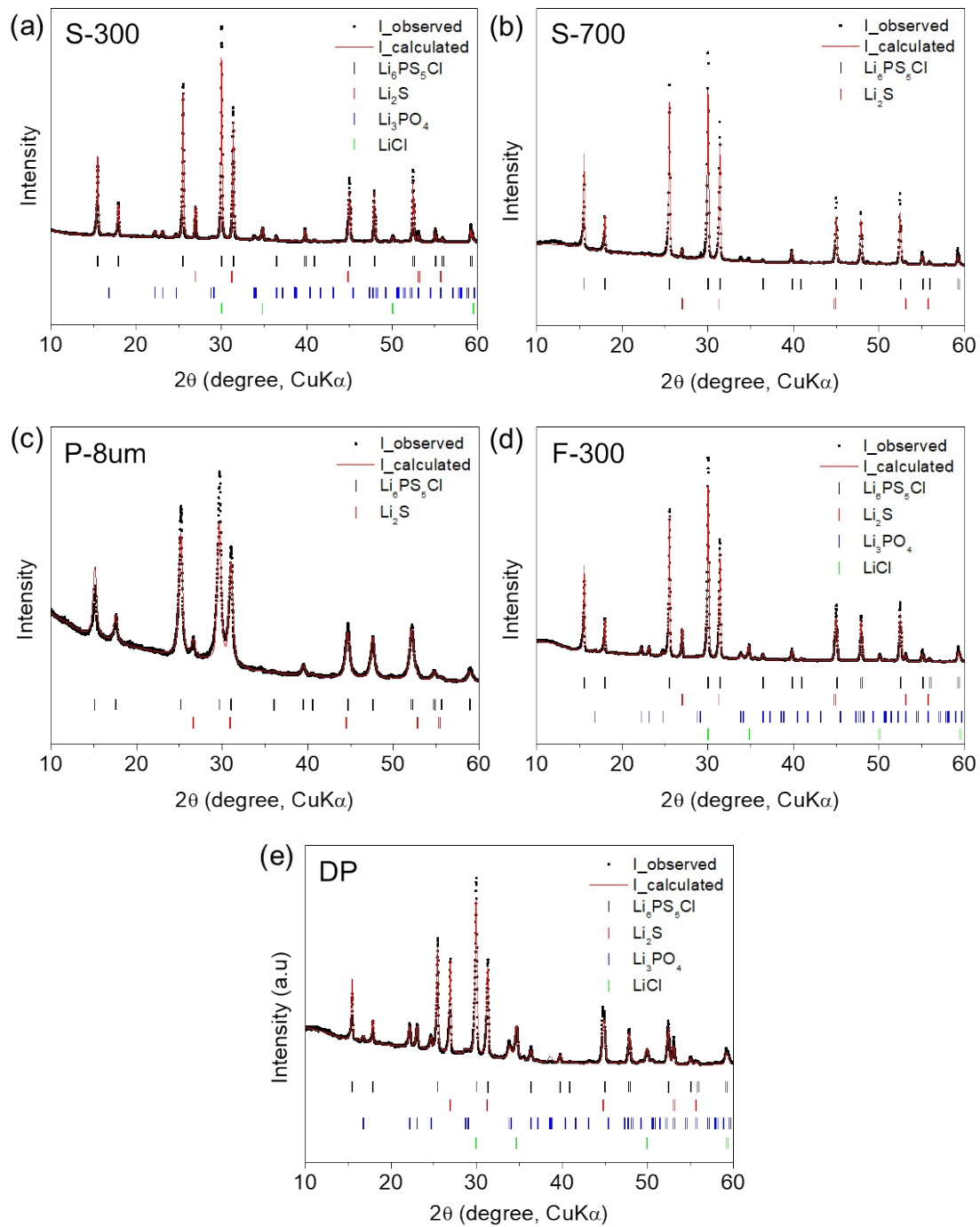


Figure S3. Rietveld refinement results for a) S-300, b) S-700, c) P-8 μm , d) F-300, and e) DP SSE powders.

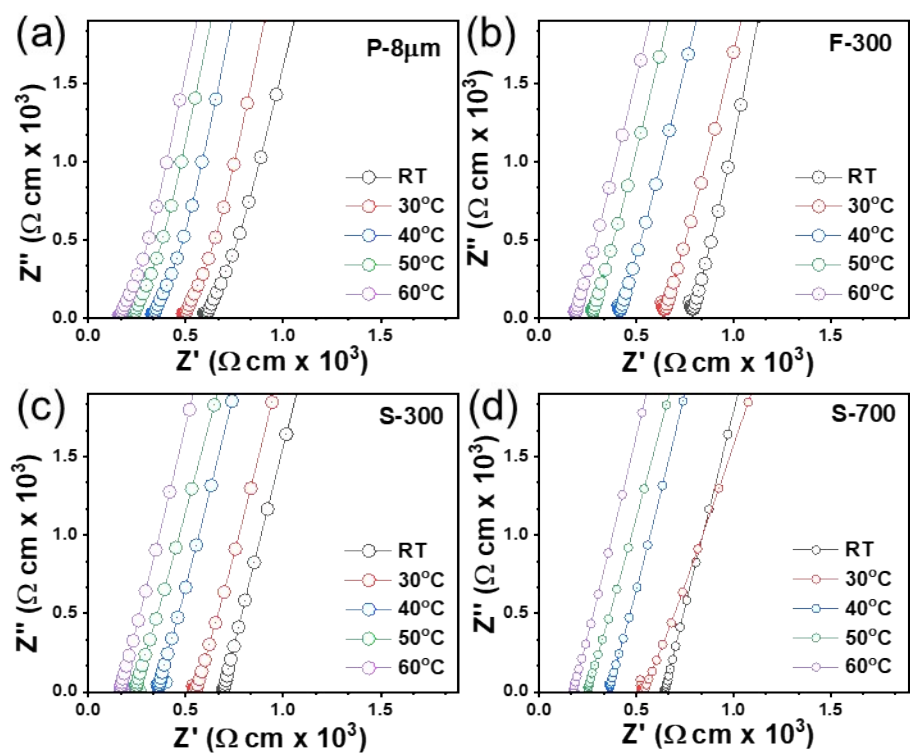


Figure S4. (a-d) EIS Spectra of P-8 μm , F-300, S-300, and S-700 SSEs under varying temperatures between 25 $^\circ\text{C}$ to 60 $^\circ\text{C}$.

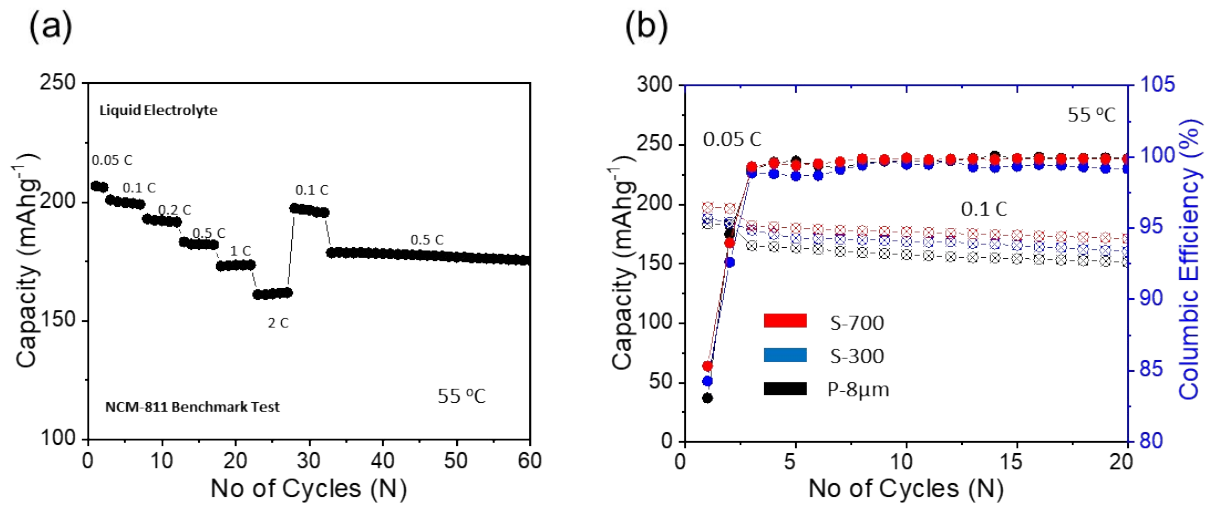


Figure S5. (a) Benchmark test for LNO-coated NCM-811 with liquid electrolyte. (b) Capacity retention and Coulombic efficiency of the ASSLBs with P-8μm, S-300, and S-700, cycled at 0.1 C for 20 cycles.

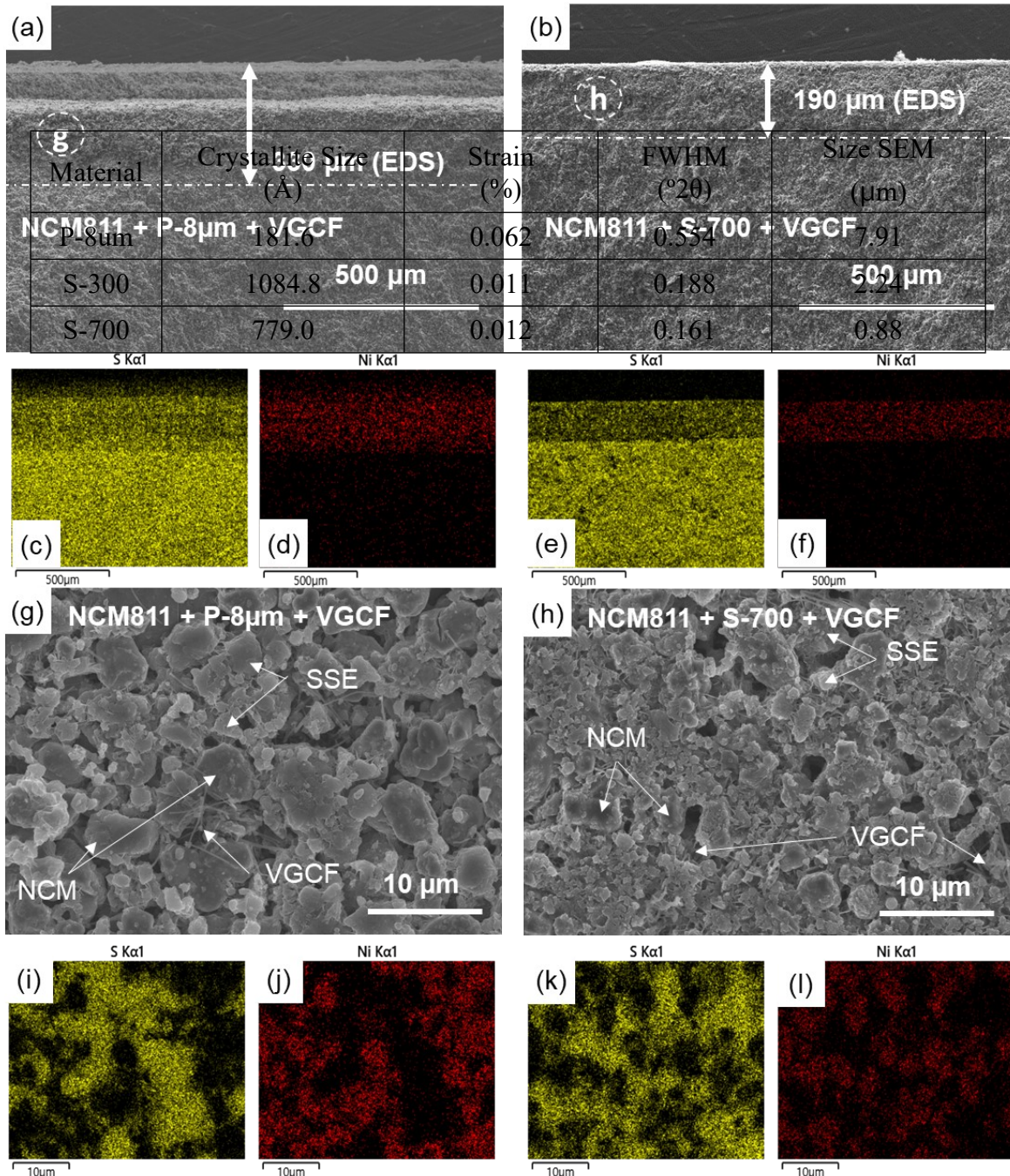


Figure S6. (a and b) The cross-sectional SEM images of cathode composite | SSE pellet containing P-8 μ m and S-700 SSE respectively. (c-f) EDS mapping for sulfur (S) and nickel (Ni) to differentiate cathode composite and SSE part as well as calculating the thickness. (g-f) cross-sectional SEM images of cathode composite for NCM811 + P-8 μ m + VGCF and NCM811 + S-700 + VGCF samples, respectively. The arrows indicate dispersed NCM811, SSE and VGCF in cathode composite. (i-l) EDS mapping for S and Ni to indicate the dispersion of SSE and NCM in cathode composite for NCM811 + P-8 μ m + VGCF, and NCM811 + S-700 + VGCF samples, respectively.

Table S1. Summary of the strain and FWHM analysis of P-8 μ m, S-300, and S-700 SSE powders.