

Supporting Information

A New Ferrocene Covalent-based Organosulfur Cathode Material for High Stability Lithium-Sulfur Battery

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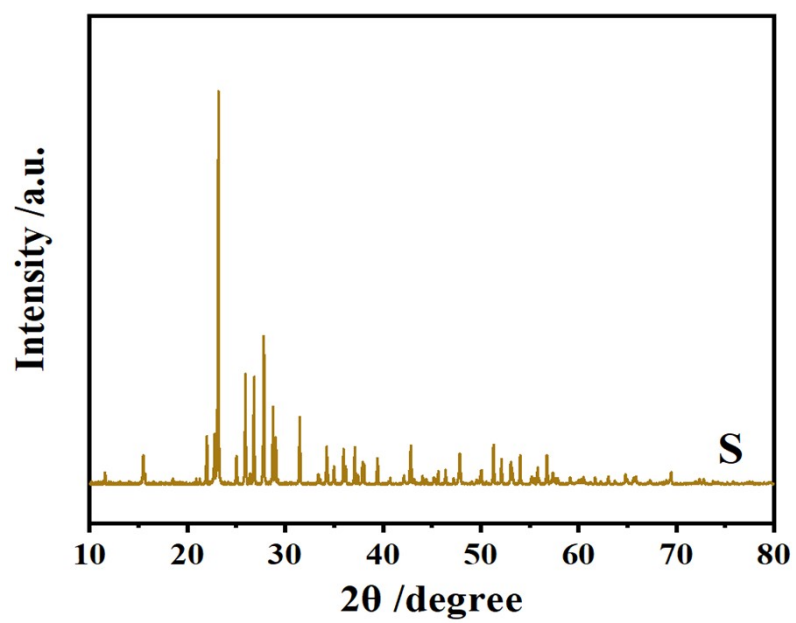


Fig. S1 The XRD analysis for sulfur.

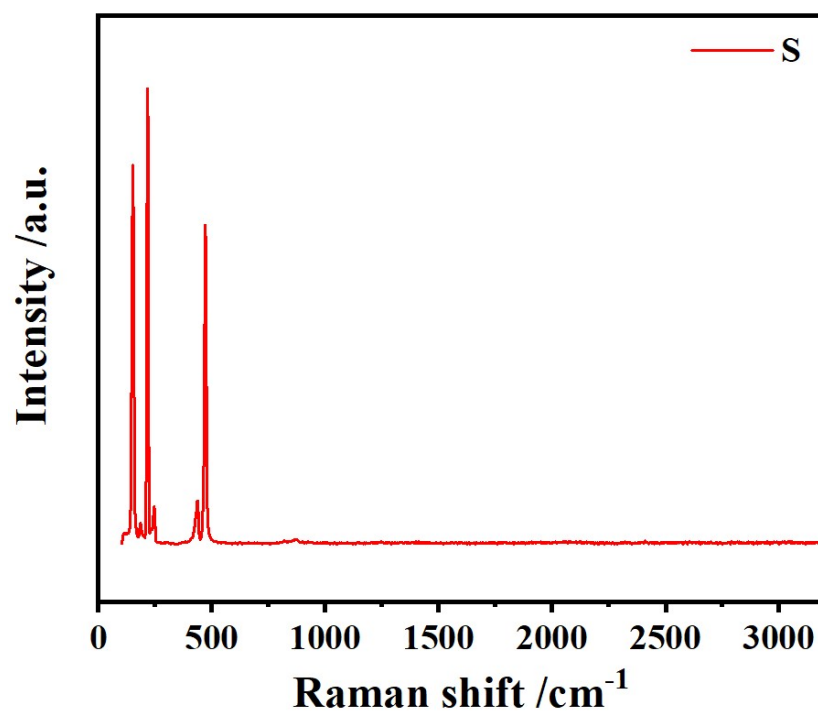


Fig. S2 The Raman spectrum of pure sulfur.

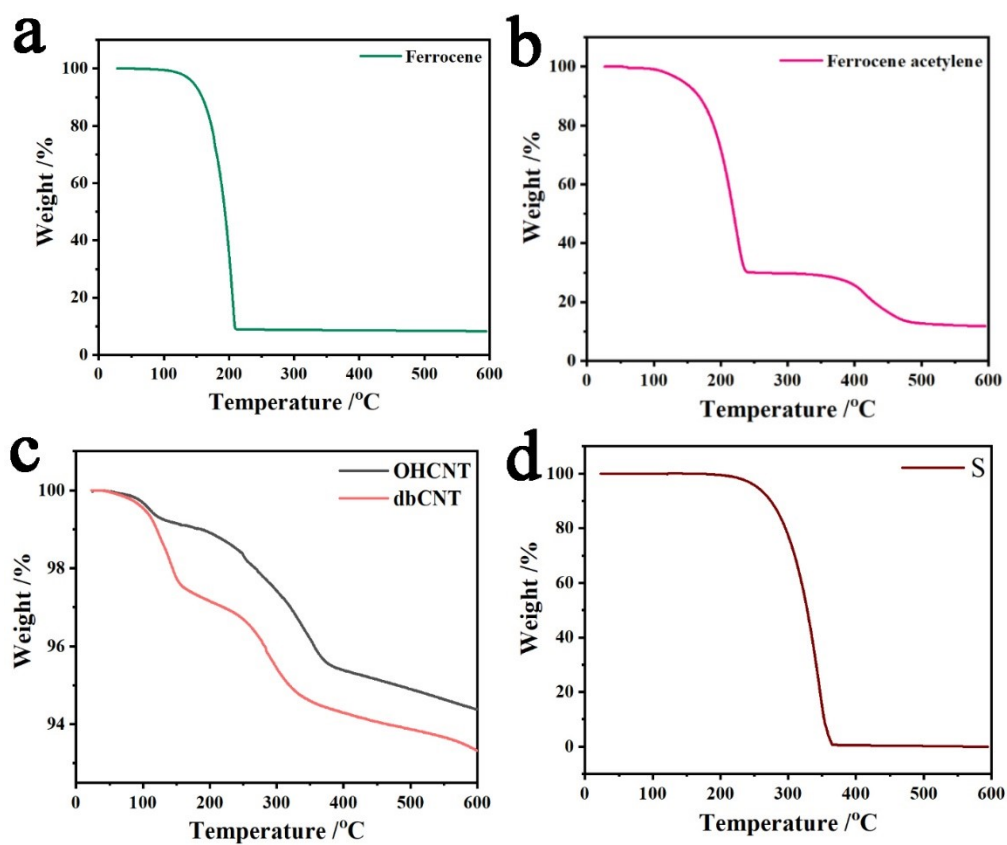


Fig. S3 The TGA analysis of ferrocene (a), ferrocene acetylene (b), OHCNT/dbCNT (c) and pure sulfur.

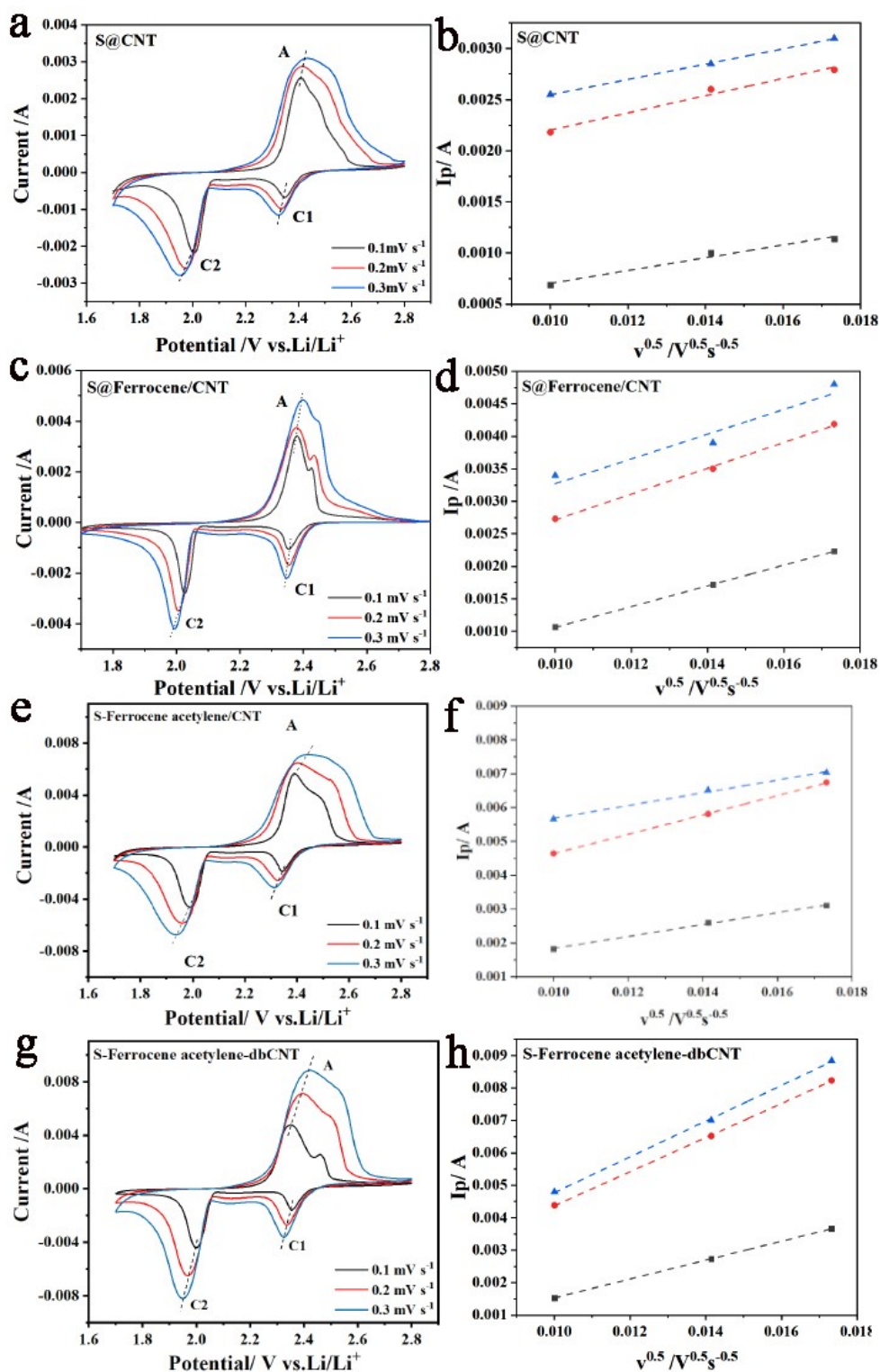


Fig. S4 The CV curves with different scan rates from 0.1 mV s⁻¹ to 0.3 mV s⁻¹ and linear fitting of the peak currents of CV curves for the cell with S@CNT (a-b), S@Ferrocene/CNT (c-d), S-Ferrocene/CNT (e-f) and S-Ferrocene acetylene-dbCNT (g-h) cathode.

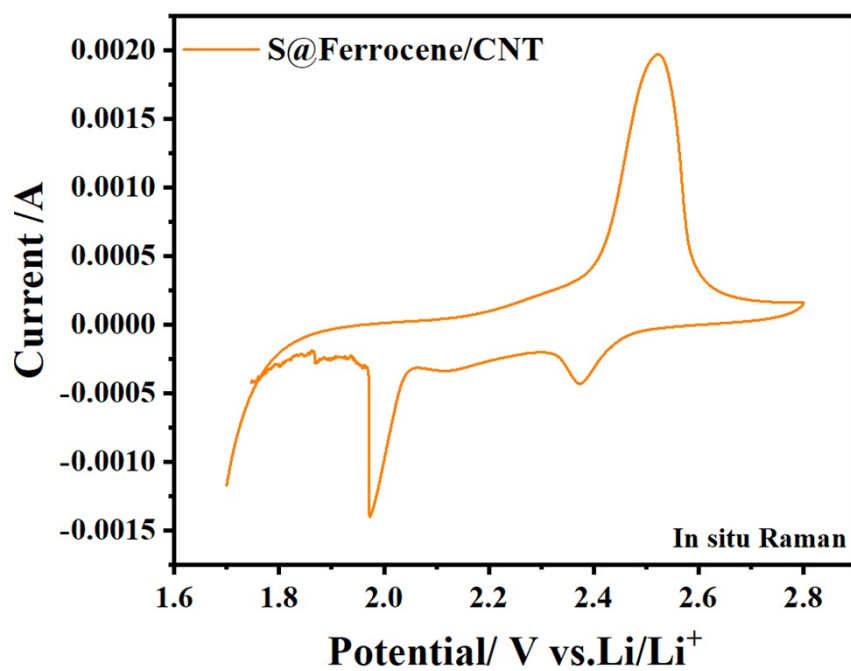


Fig. S5 (a) The CV curve of the mould with S@Ferrocene/CNT cathode at 0.3 mV s⁻¹ in in-situ Raman test.

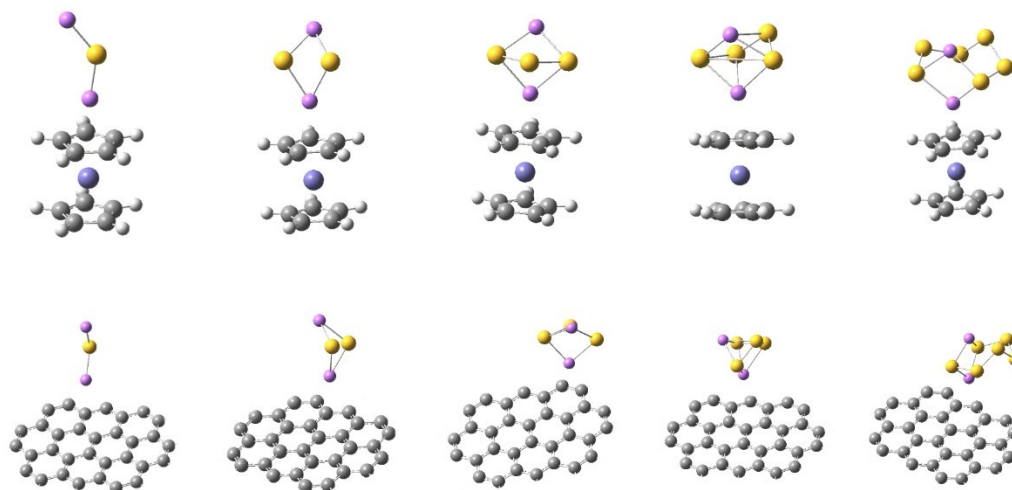


Fig. S6 The optimized structures of Li_2S , Li_2S_2 , Li_2S_3 , Li_2S_4 , Li_2S_6 on the surface of ferrocene and CNT.

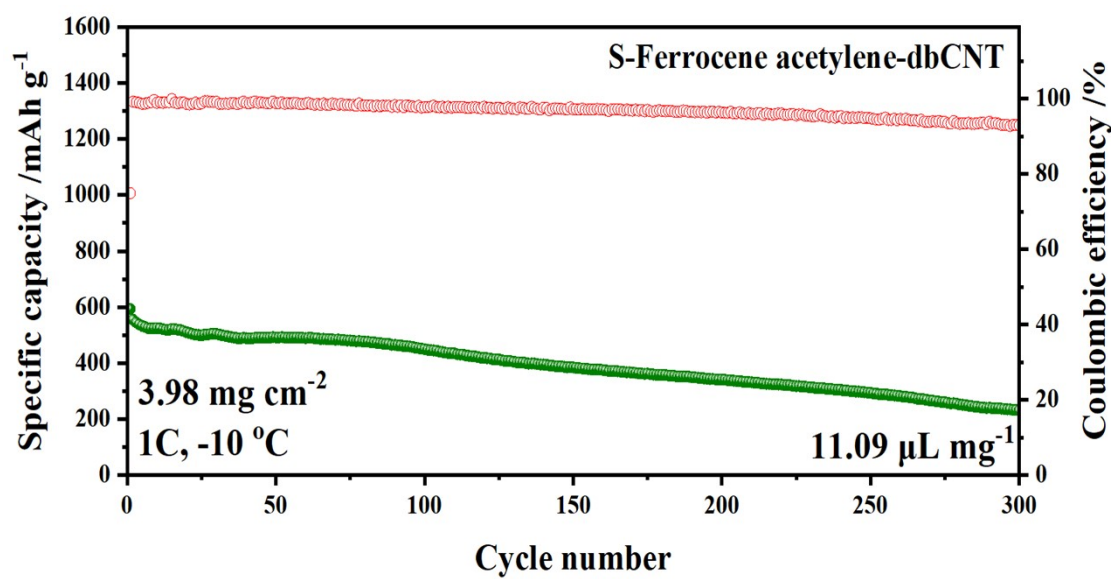


Fig. S7 Cycling performance of the cells with S-Ferrocene acetylene-dbCNT cathode at -10 °C.

Table S1. The sulfur content of S-Ferrocene acetylene-dbCNT, S-Ferrocene acetylene/CNT, S@Ferrocene/CNT and S@CNT via elemental analysis.

Name	Weight /mg	Methode	N /wt%	C /wt%	H /wt%	S /wt%	Average S /wt%
S-Ferrocene acetylene-dbCNT-1	3.136	5mg standard	---	32.270	0.880	60.756	61.560
S-Ferrocene acetylene-dbCNT-2	2.941	5mg standard	---	32.870	0.904	62.363	
S-Ferrocene acetylene/CNT-1	3.130	5mg standard	---	33.830	0.856	60.344	61.244
S-Ferrocene acetylene/CNT-2	3.020	5mg standard	---	32.590	0.842	62.143	
S@Ferrocene/CNT-1	3.182	5mg standard	---	32.440	1.059	63.647	63.561
S@Ferrocene/CNT-2	3.201	5mg standard	---	32.560	1.050	63.474	
S@CNT-1	3.096	5mg standard	---	39.910	---	62.698	62.723
S@CNT-2	3.035	5mg standard	---	40.200	---	62.748	

Table S2. The fitted impedance parameters of batteries based on S-Ferrocene acetylene-dbCNT, S-Ferrocene acetylene/CNT, S@Ferrocene/CNT, S@CNT.

Samples	R_s (Ω)	R_{ct} (Ω)
S-Ferrocene acetylene-dbCNT	3.36	24.91
S-Ferrocene acetylene/CNT	3.08	48.79
S@Ferrocene/CNT	2.45	87.21
S@CNT	5.59	144.26

Table S3. The lithium-ion diffusion coefficient for the cell with S-Ferrocene acetylene-dbCNT, S-Ferrocene acetylene/CNT, S@Ferrocene/CNT, S@CNT cathodes.

Samples	<i>C1</i>	<i>C2</i>	<i>A</i>
	($\times 10^9 \text{ cm}^2 \text{ s}^{-1}$)	($\times 10^9 \text{ cm}^2 \text{ s}^{-1}$)	($\times 10^9 \text{ cm}^2 \text{ s}^{-1}$)
S-Ferrocene acetylene-dbCNT	114	373	410
S-Ferrocene acetylene/CNT	41.5	110.7	48.3
S@Ferrocene/CNT	34.4	53.2	48.9
S@CNT	5.31	9.58	7.6

Table S4. Comparison of the electrochemical performance of the reported covalently interlinked cathode for lithium-sulfur batteries. (1C=1675 mA g⁻¹)

Cathode materials	Discharge current	Maximum Capacity (mAh g ⁻¹)	Sulfur loading (mg cm ⁻²)	Capacity retention (Cycle Numbers)	Reference
S-Ferrocene acetylene-dbCNT	0.5C	925	2.0	93.79% (400)	This work
	1 C (14.96 mA cm ⁻²)	5.38 mAh cm ⁻²	8.93	81.97% (400)	
	1 C, 0 °C	685	2.0	75.54% (600)	
SC@A	0.5C	~1000	1.0	87% (200)	(1)
SVE(1:9)	0.5C	783	1-1.5	88.8% (400)	(2)
TSP/CNT	1C	728	0.6-0.8	54.4% (600)	(3)
CNT@UiO-66-V-S	0.2C	4.25 mAh cm ⁻²	5.6	91.76% (100)	(4)
BTTPs	0.1C	901	1.5-2	68.4% (120)	(5)
GPS	0.1C	920	1.15	76.08% (100)	(6)
PTSeS/CG-CNT	0.5C	591	1.0	62% (300)	(7)

If not otherwise specified, the above batteries are tested at 30 °C.

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