

Supporting Information

Constructing FePSe₃-FeSe₂ heterojunctions uniformly in Ketjen Black carbon matrix for superior potassium ion batteries

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Table S1. The results of ICP-OES of FePSe₃-FeSe₂/C composite.

Elements	Mass percentage (%)	Molar percentage (%)	Molar ratio of FePSe ₃ and FeSe ₂
Fe	9.82	0.175	
P	4.77	0.154	7.5: 1.0
Se	39.42	0.499	

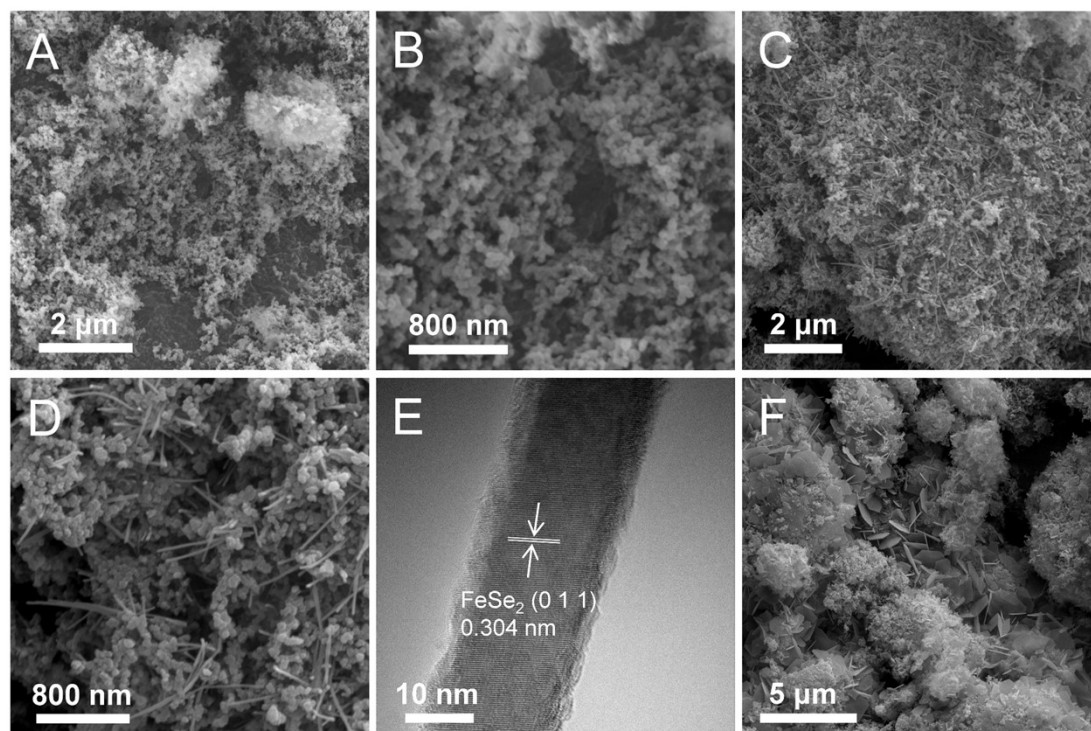


Figure S1. (A-B) SEM images of KB carbon. (C-D) SEM images and (E) HRTEM image of FeSe₂/C. (F) SEM image of FePSe₃/C.

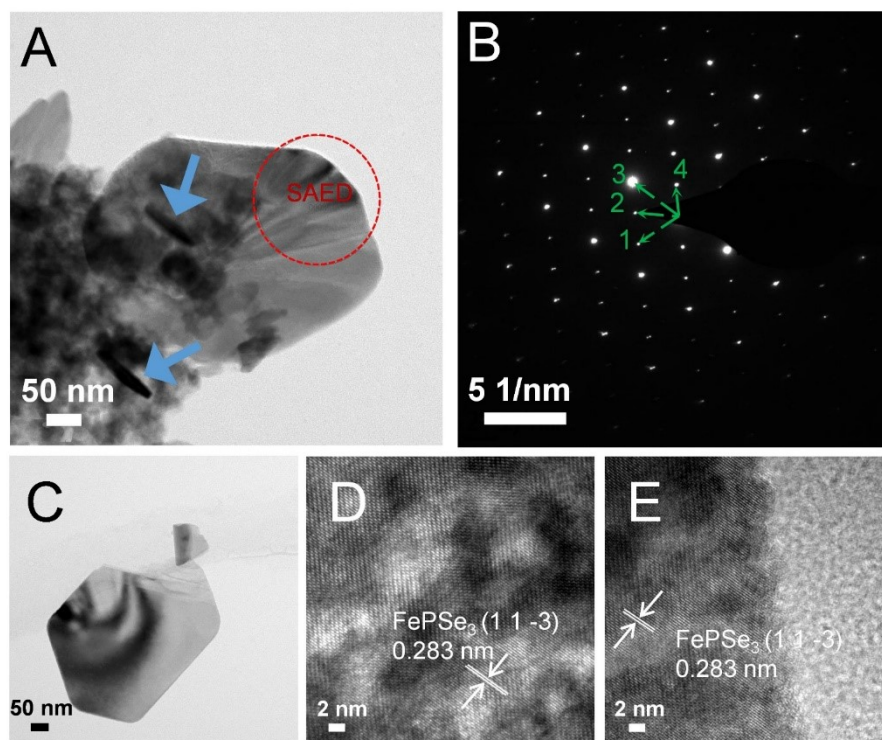


Figure S2. (A) TEM image, (B) SAED pattern, (C) TEM image and (D) & (E) HRTEM images of FePSe₃/C. In more details, points 1-4 marked in green in the SAED pattern correspond to the (-1 -1 -6), (0 -1 -5), (1 -1 -4) and (0 0 0) planes of FePSe₃, respectively.

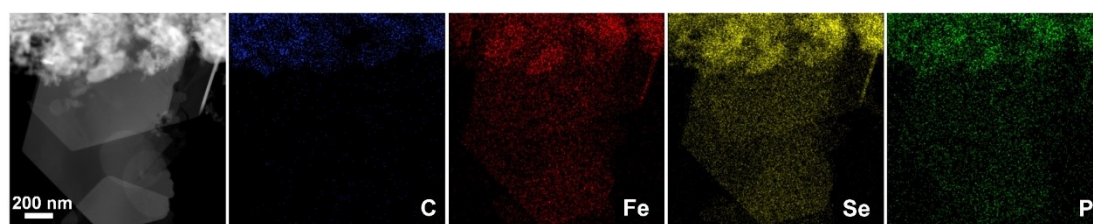


Figure S3. HAADF-STEM image with the corresponding EDS images of FePSe₃/C.

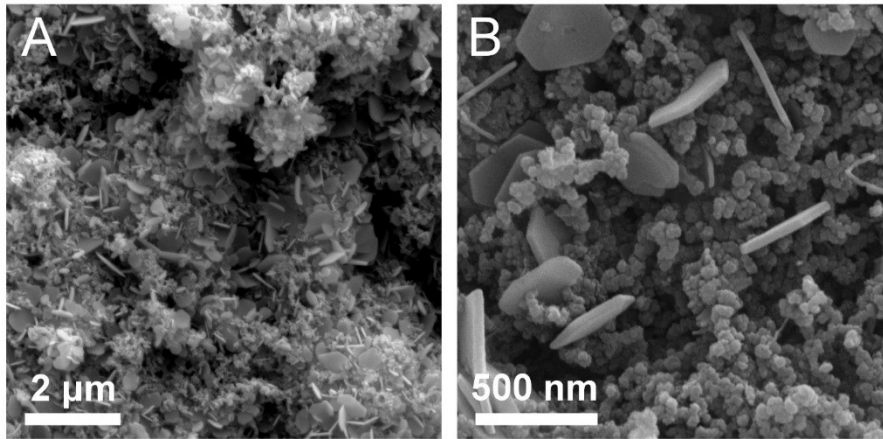


Figure S4. The high-resolution SEM images of FePSe₃-FeSe₂/C.

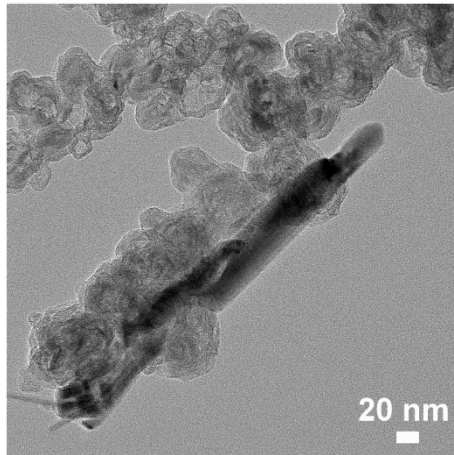


Figure S5. The TEM image of the side-view of FePSe₃-FeSe₂/C.

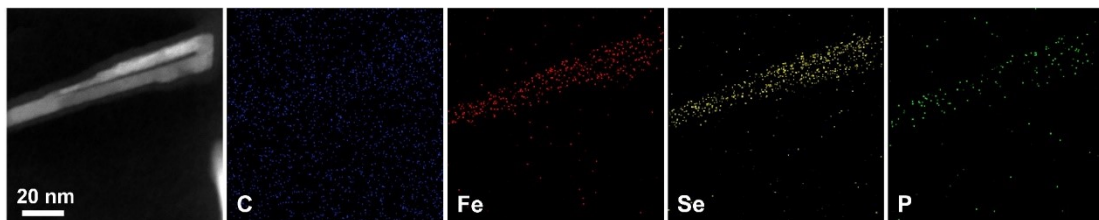


Figure S6. HAADF-STEM image with the corresponding EDS images of the side-view of FePSe₃-FeSe₂/C.

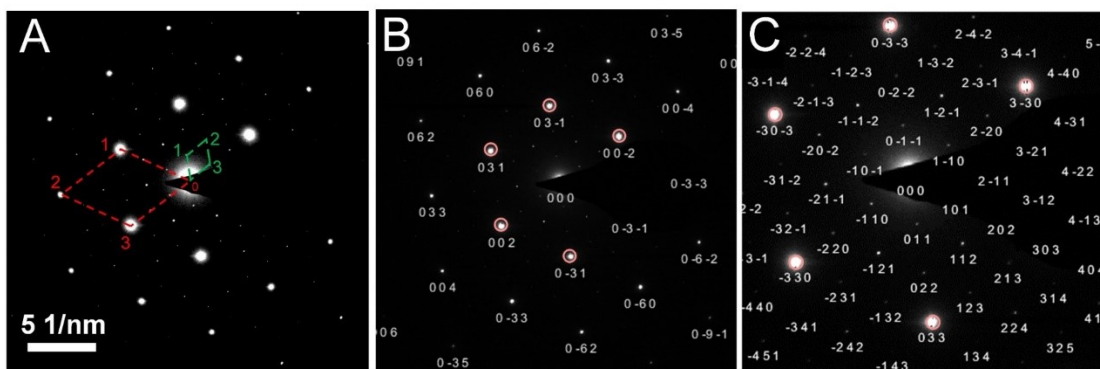


Figure S7. (A) SAED pattern of FePSe₃-FeSe₂/C. (B) The FeSe₂ crystal planes and (C) the FePSe₃ crystal planes corresponding to the diffraction spots in SAED.

This SAED pattern can be readily indexed as the 2D in-plane reflections of FePSe₃ with (-1 -1 1) zone axis direction and FeSe₂ with (1 0 0) zone axis direction. In more details, points 1-3 marked in green correspond to the (0 -1 -1), (1 -2 -1), and (1 -1 0) planes of FePSe₃, respectively, and the lattice planes of FeSe₂ represented by the red marked dots 1-3 in turn are (0 3 1), (0 3 3), and (0 0 2), respectively.

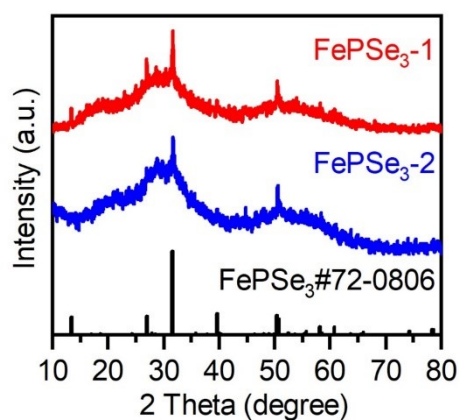


Figure S8. PXRD patterns of FePSe₃-1 and FePSe₃-2.

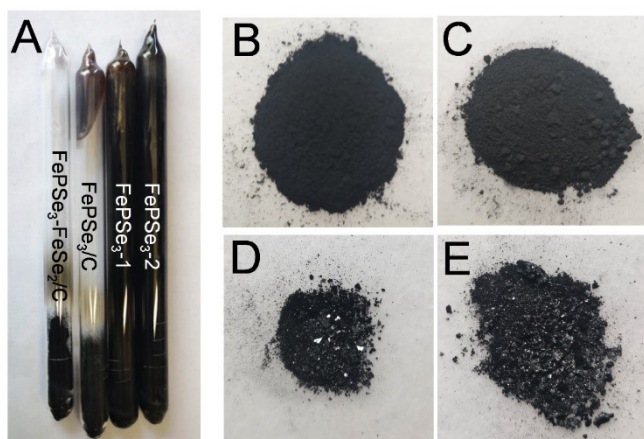


Figure S9. (A) The quartz tubes contain FePSe₃-FeSe₂/C, FePSe₃/C, FePSe₃-1 and FePSe₃-2 products in the order from left to right. (B) FePSe₃-FeSe₂/C products. (C) FePSe₃/C products. (D) FePSe₃-1 products. (E) FePSe₃-2 products.

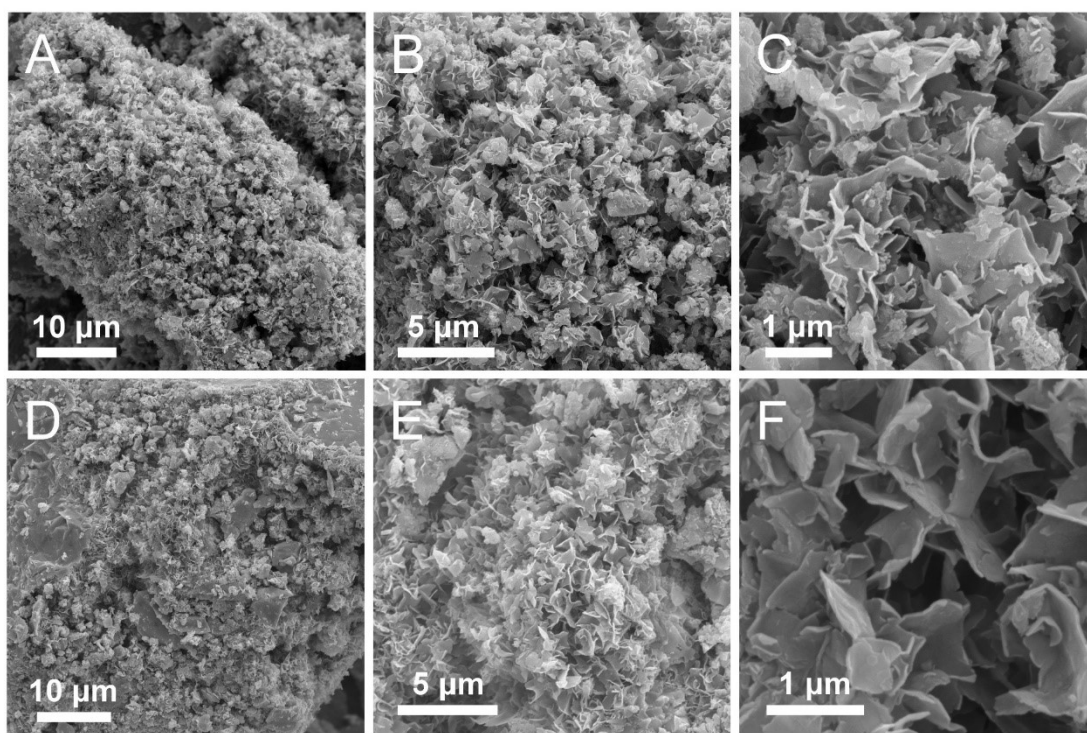


Figure S10. SEM images of (A-C) FePSe₃-1 and (D-F) FePSe₃-2.

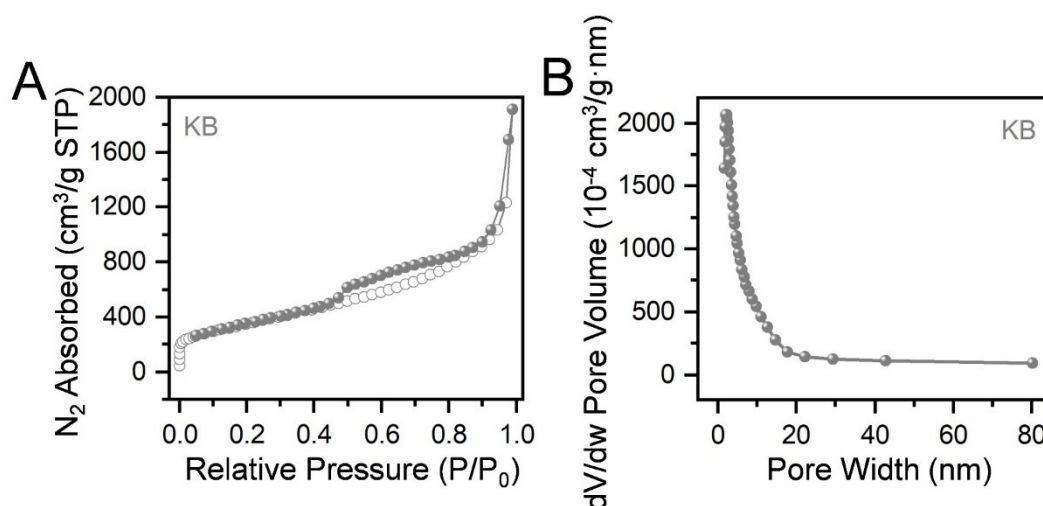


Figure S11. (A) N_2 adsorption-desorption isotherms and (B) the corresponding pore size distribution of KB carbon.

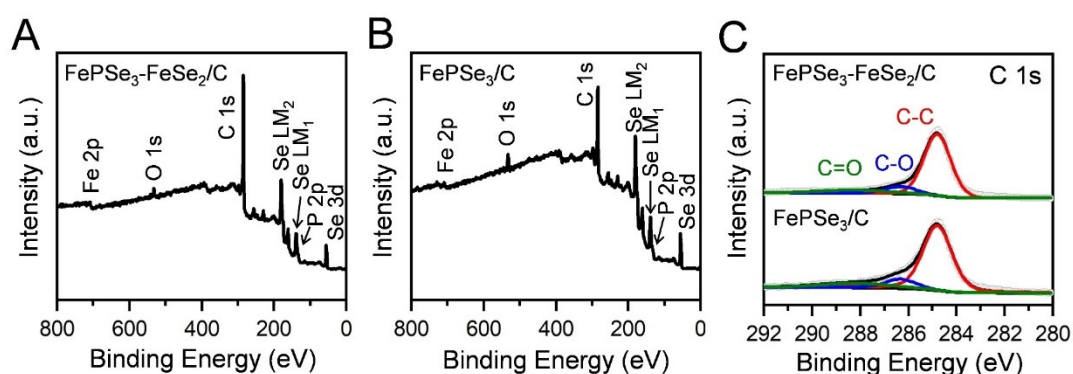


Figure S12. XPS survey spectra of (A) $FePSe_3-FeSe_2/C$ and (B) $FePSe_3/C$. (C) High-resolution XPS C 1s spectra of $FePSe_3-FeSe_2/C$ and $FePSe_3/C$.

Table S2. Brunauer Emmett Teller (BET) specific surface areas of KB, $FePSe_3-FeSe_2/C$ and $FePSe_3/C$.

Sample	KB	$FePSe_3-FeSe_2/C$	$FePSe_3/C$
BET Surface Area (m^2/g)	1263.9	107.8	18.9

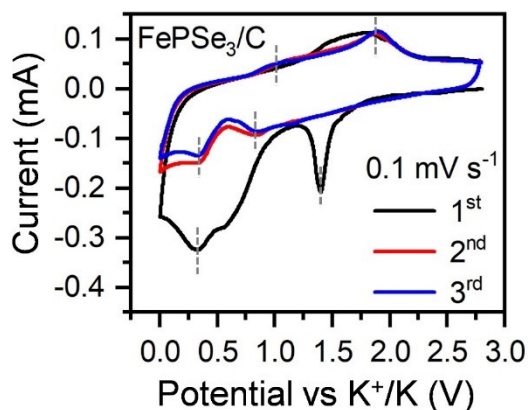


Figure S13. CV curves for initial three cycles of FePSe₃/C electrode for PIBs at 0.1 mV s⁻¹ between 0.005 and 2.8 V.

Table S3. The electrochemical reactions of FePSe₃-FeSe₂/C and FeSe₂/C with K⁺ on anode.

Sample	Potential	Reaction
FePSe ₃ - FeSe ₂ /C	Discharged to 1.40 V	$FePSe_3 + 6K^+ + 6e^- \rightarrow Fe + P + 3K_2Se$
	Discharged to 1.20 V	$FeSe_2 + 4K^+ + 4e^- \rightarrow Fe + 2K_2Se$
	Discharged to 0.48V	$3P + 4K^+ + 4e^- \rightarrow K_4P_3$
	Charged to 1.04 V	$2Fe + 2K_2Se \rightarrow FeSe_2 + 4K^+ + 4e^-$
	Charged to 1.74 V	$K_4P_3 + 9K_2Se + 3Fe \rightarrow 3FePSe_3 + 22K^+ + 22e^-$
FePSe ₃ /C	Discharged to 1.40 V	$FePSe_3 + 6K^+ + 6e^- \rightarrow Fe + P + 3K_2Se$
	Discharged to 0.32 V	$3P + 4K^+ + 4e^- \rightarrow K_4P_3$
	Charged to 1.08 V	$K_4P_3 \rightarrow 3P + 4K^+ + 4e^-$
	Charged to 1.80 V	$Fe + P + 3K_2Se \rightarrow FePSe_3 + 6K^+ + 6e^-$

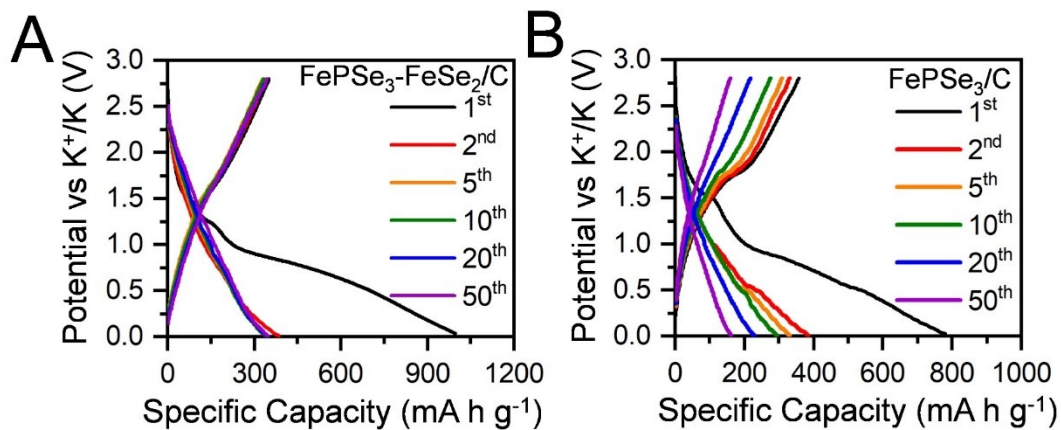


Figure S14. Charge–discharge profiles of (A) $\text{FePSe}_3\text{-FeSe}_2/\text{C}$ and (B) FePSe_3/C at the current density of 0.1 A g^{-1} .

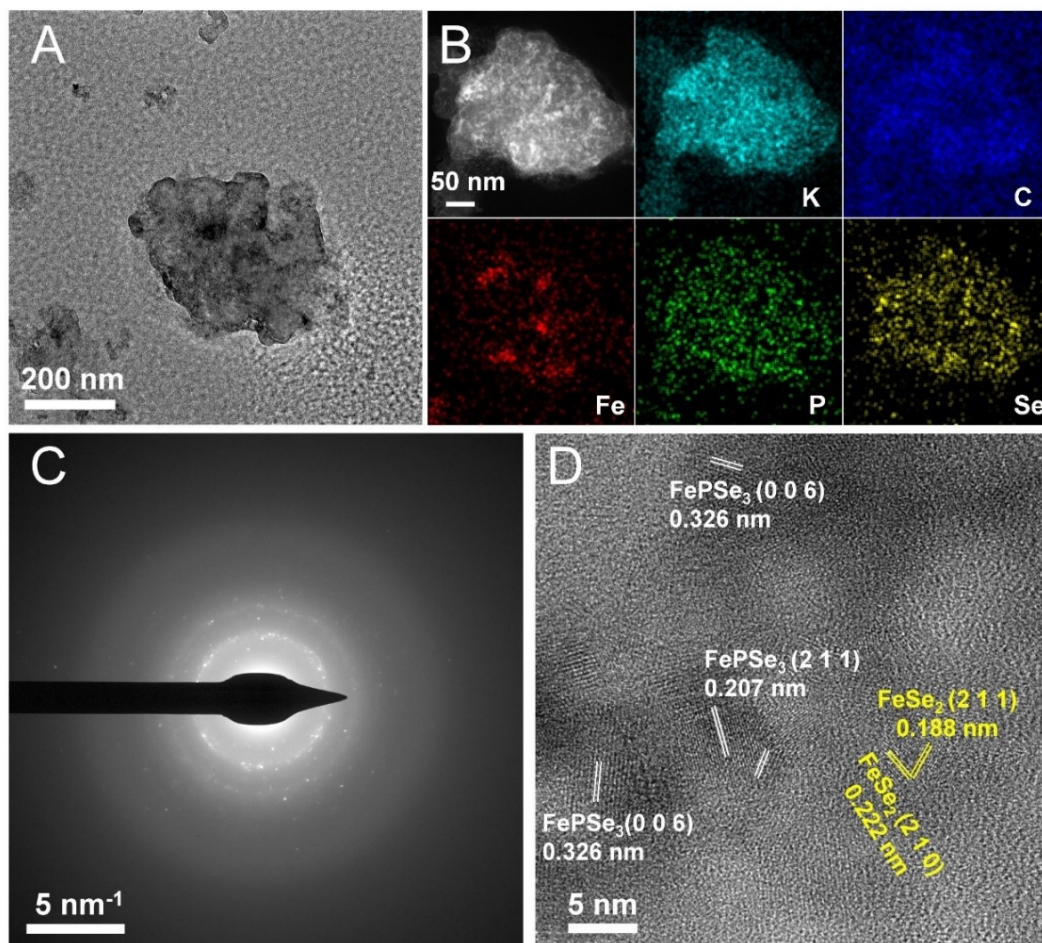


Figure S15. (A) TEM image, (B) HAADF-STEM image with the corresponding EDS images, (C) the SAED pattern and (D) HRTEM image of FePSe₃-FeSe₂/C at the current density of 0.1 A g⁻¹ after 260 cycles.

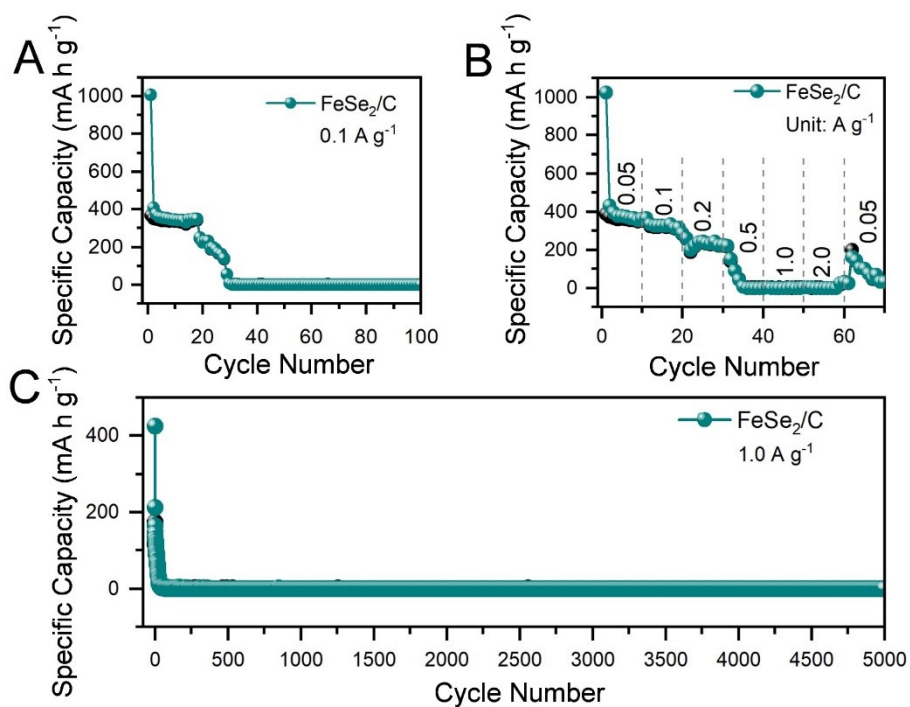


Figure S16. (A) Cycling performance, (B) rate performance and (C) long-term cyclic performance of FeSe₂/C electrodes in the voltage range of 0.005-2.8 V.

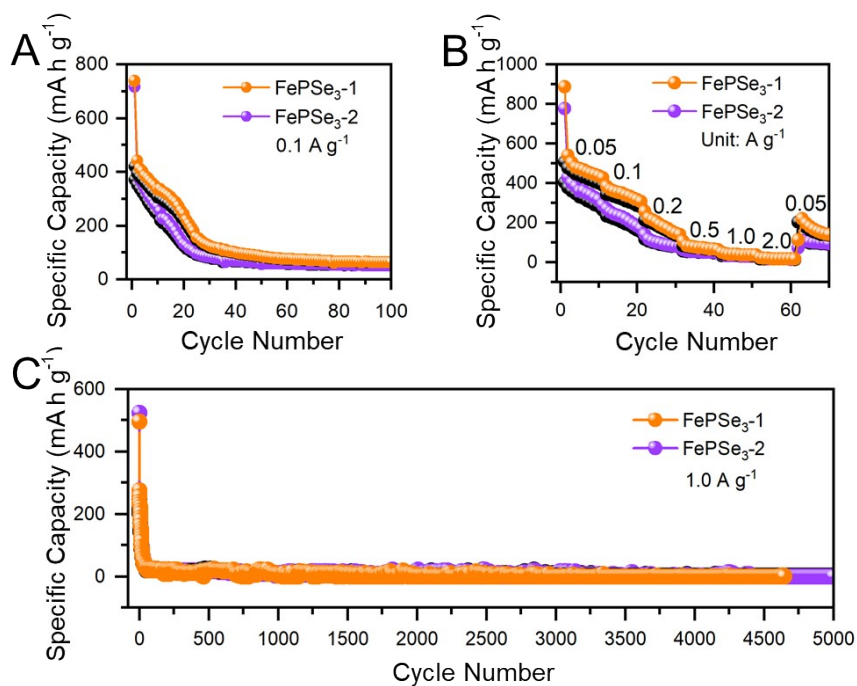


Figure S17. (A) Cycling performance of FePSe₃-1 and FePSe₃-2 electrodes at the current density of 0.1 A g⁻¹ in the range of 0.005-2.8 V. (B) Rate performance of FePSe₃-1 and FePSe₃-2 electrodes at various current densities from 0.05 to 2.0 A g⁻¹. (C) Long-term cyclic performance of FePSe₃-1 and FePSe₃-2 electrodes at 1.0 A g⁻¹.

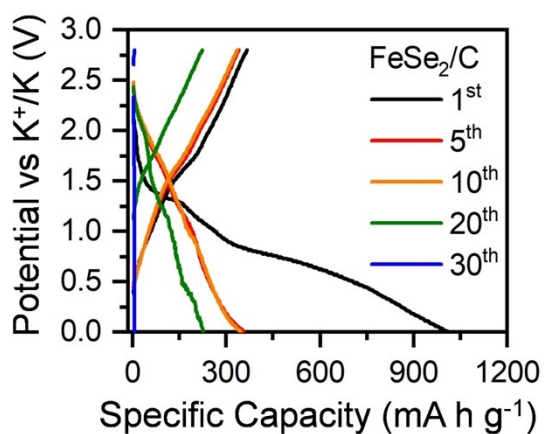


Figure S18. The galvanostatic charge/discharge profiles of FeSe₂/C at the current density of 0.1 A g⁻¹.

Table S4. Comparison of the cycling performance of FePSe₃-FeSe₂/C and other FeSe₂-based or FePSe₃-based materials for PIBs anodes.

Material	Current density (A g ⁻¹)	Cycle number	Capacity (mA h g ⁻¹)	Mass loading (mg cm ⁻²)	Calculate based on	Ref
FePSe ₃ -FeSe ₂ /C	0.1	100	352	1.0	composite	This work
FeSe ₂ /C	1.0	3700	224	0.7		
FeSe ₂ /NC	1.0	250	301	1.0	composite	1
FeSe ₂ @NC	0.5	100	280	1.0-1.5	composite	2
FeSe ₂	0.4	80	441	1.0-1.2	composite	3
ZnSe-FeSe ₂ /RGO	0.05	100	363	/	composite	4
FeSe ₂ /N-C	0.2	100	256	1.0	composite	5
FeSe ₂ @C NBs	0.1	700	221	/	composite	6
f-FePSe ₃ /CNT	0.5	1000	223	0.8	composite	7

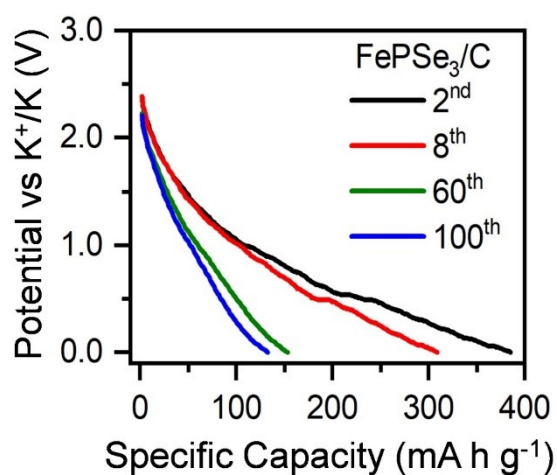


Figure S19. The galvanostatic discharge profiles of FePSe₃/C at the 2nd, 8th, 60th and 100th cycles at 0.1 A g⁻¹ in the voltage range of 0.005-2.8 V.

Table S5. The specific capacities of FePSe₃-FeSe₂/C electrode in three voltage ranges.

	2 nd cycle	8 th cycle	60 th cycle	100 th cycle
Voltage range (V)	Specific capacity (mA h g⁻¹)			
0.005-1.00	260.2	201.6	189.1	179.9
1.00-1.58	73.7	72.0	77.0	78.7
1.58-2.80	51.9	63.6	92.0	93.7
Full range	385.8	337.2	358.1	352.3

Table S6. The capacity increments of FePSe₃-FeSe₂/C electrode in three voltage ranges.

	ΔC_{8-2}	ΔC_{60-8}	ΔC_{100-60}
Voltage range (V)	Specific capacity (mA h g⁻¹)		
0.005-1.00	-58.6	-12.5	-9.2
1.00-1.58	-1.7	5.0	1.7
1.58-2.80	11.8	28.4	1.7
Full range	-48.5	20.9	-5.8

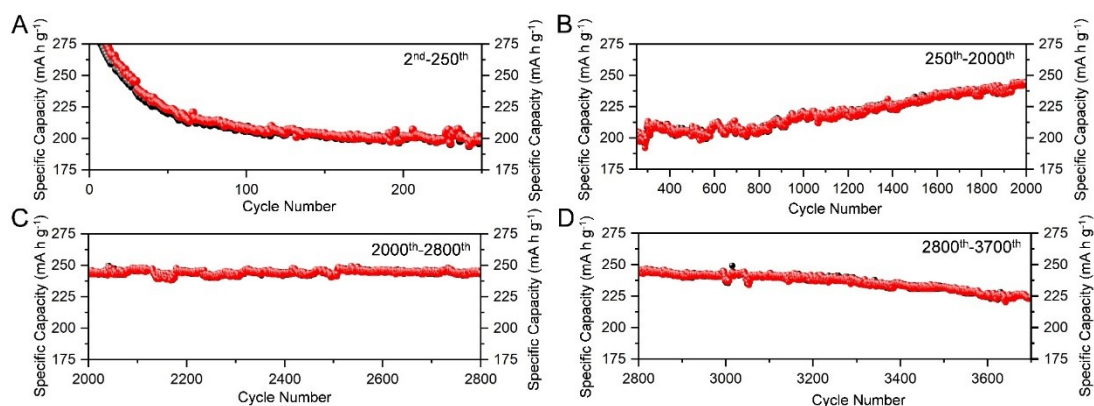


Figure S20. Cycling performance of FePSe₃-FeSe₂/C at a current density of 1.0 A g⁻¹ in different cycle intervals: (A) from the 2nd to the 250th cycle, (B) from the 250th to the 2000th cycle, (C) from the 2000th to the 2800th cycle and (D) from the 2800th to the 3700th cycle.

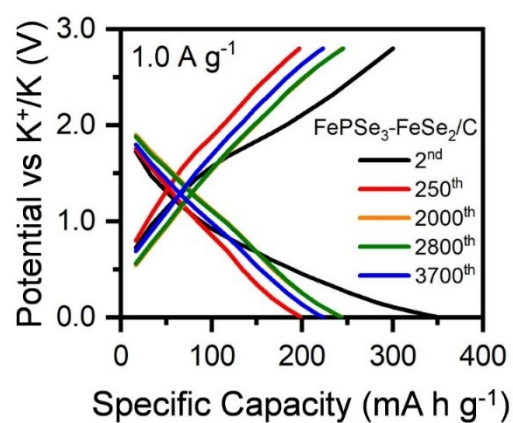


Figure S21. The galvanostatic charge/discharge profiles of FePSe₃-FeSe₂/C at the 2nd, 250th, 2000th, 2800th and 3700th cycles at 1.0 A g⁻¹ in 0.005-2.8 V.

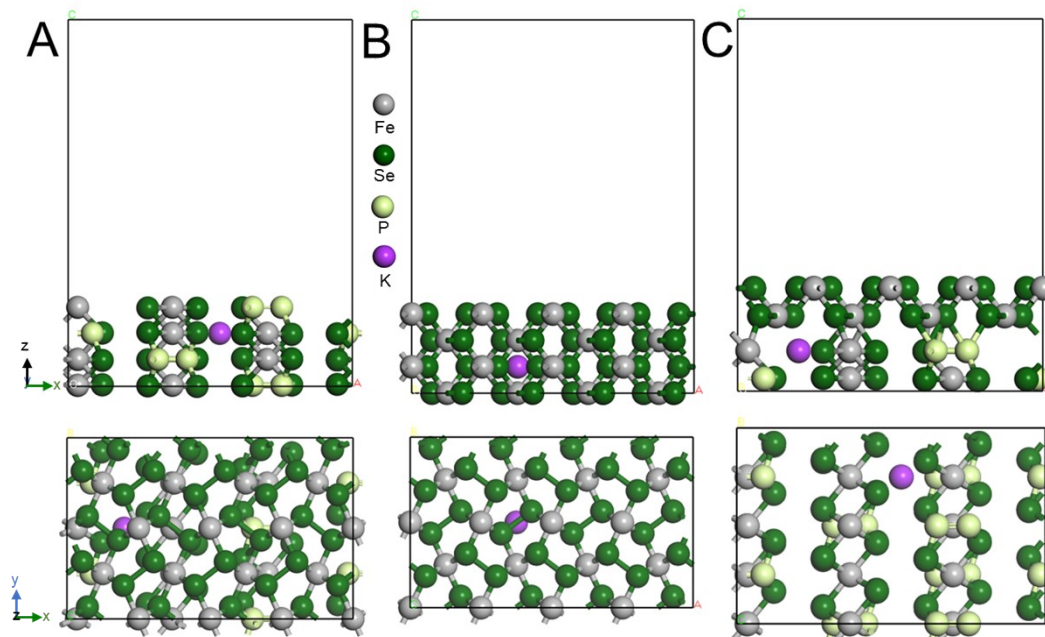


Figure S22. Side-view (up) and top-view (down) of built models of (A) FePSe_3 , (B) FeSe_2 , and (C) $\text{FePSe}_3\text{-FeSe}_2$.

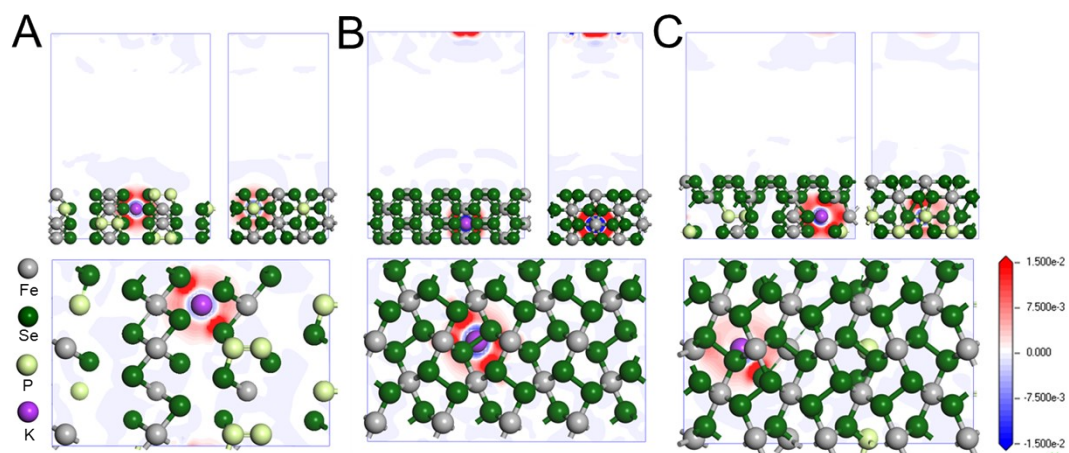


Figure S23. Three slices of electron density map of (A) FePSe_3 , (B) FeSe_2 , and (C) $\text{FePSe}_3\text{-FeSe}_2$.

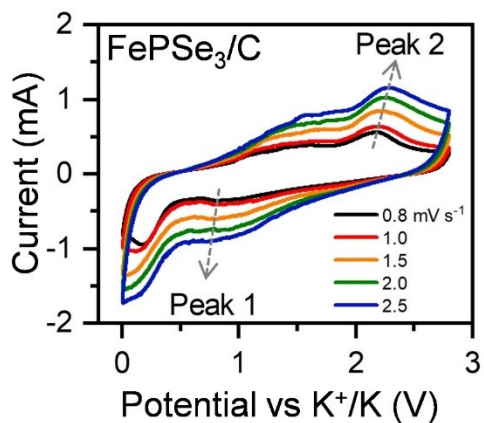


Figure S24. CV curves of FePSe₃/C at different scan rates from 0.8 to 2.5 mV s⁻¹.

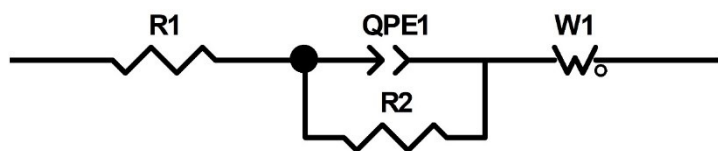


Figure S25. Equivalent circuit for measured EIS spectra.

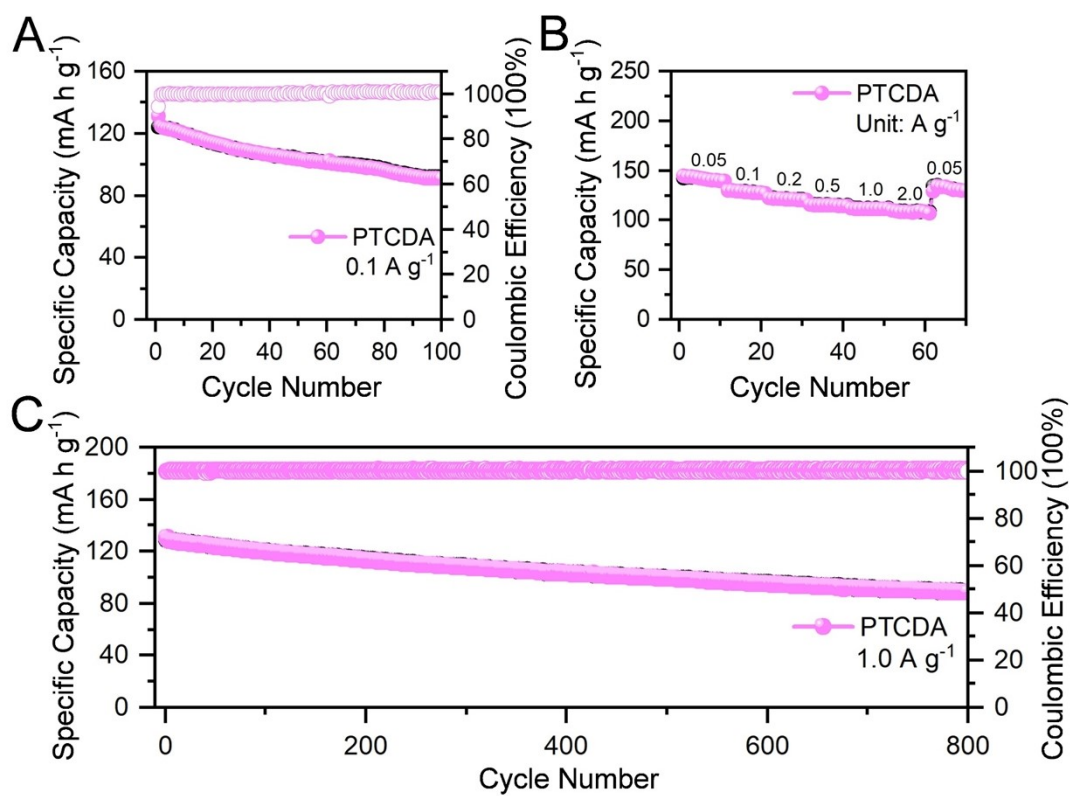


Figure S26. (A) Cycling performance, (B) rate performance and (C) long-term cyclic performance of PTCDA half cell in the voltage range of 1.5-3.5 V.

References

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7. Huang, Y.-F.; Yang, Y.-C.; Tuan, H.-Y., Construction of strongly coupled few-layer FePSe₃-CNT hybrids for high performance potassium-ion storage devices. *Chemical Engineering Journal* **2023**, *451* (4), 139013.