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## Supporting Information

## The Structural Evolution of Poly(ethylene terephthalate) Oligomers Produced via Glycolysis Depolymerization

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Figure S1- NMR spectra for the starting PET material.



Figure S2-NMR spectra for the depolymerized PET from a 10-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)<sub>2</sub> mole ratio of 100:1.



Figure S3- NMR spectra for the depolymerized PET from a 20-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)<sub>2</sub> mole ratio of 100:1.



Figure S4-NMR spectra for the depolymerized PET from a 30-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S6-NMR spectra for the depolymerized PET from a 50-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S8-NMR spectra for the depolymerized PET from a 70-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S9-NMR spectra for the depolymerized PET from an 80-minute glycolysis reaction run at 165 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S10-NMR spectra for the depolymerized PET from a 10-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S11-NMR spectra for the depolymerized PET from a 20-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S12-NMR spectra for the depolymerized PET from a 30-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S13-NMR spectra for the depolymerized PET from a 40-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S14-NMR spectra for the depolymerized PET from a 50-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S15-NMR spectra for the depolymerized PET from a 60-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S16-NMR spectra for the depolymerized PET from a 70-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S17-NMR spectra for the depolymerized PET from an 80-minute glycolysis reaction run at 175 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S18-NMR spectra for the depolymerized PET from a 10-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S19-NMR spectra for the depolymerized PET from a 20-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S20-NMR spectra for the depolymerized PET from a 30-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S21-NMR spectra for the depolymerized PET from a 40-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S22-NMR spectra for the depolymerized PET from a 50-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S23-NMR spectra for the depolymerized PET from a 60-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S24-NMR spectra for the depolymerized PET from a 70-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S25-NMR spectra for the depolymerized PET from an 80-minute glycolysis reaction run at 185 °C, PET:Zn(Ac)2 mole ratio of 100:1.



Figure S26- NMR spectra for the depolymerized PET from a 2-minute homogenous glycolysis reaction in N-methyl-2-pyrrolidone.



Figure S27-NMR spectra for the depolymerized PET from a 4-minute homogenous glycolysis reaction in N-methyl-2-pyrrolidone.



Figure S28-NMR spectra for the depolymerized PET from a 6-minute homogenous glycolysis reaction in N-methyl-2-pyrrolidone.

## 2.3.3 Calibration Standard Chromatograms



Figure S29- Data for PET starting materials from soft drink bottle analyzed via size exclusion chromatography analysis by PolyAnalytik.