

Electronic Supporting Information

Solid Polymer Electrolyte of Ionic Liquids via Bicontinuous Ion Transport Channel for Lithium Metal Batteries

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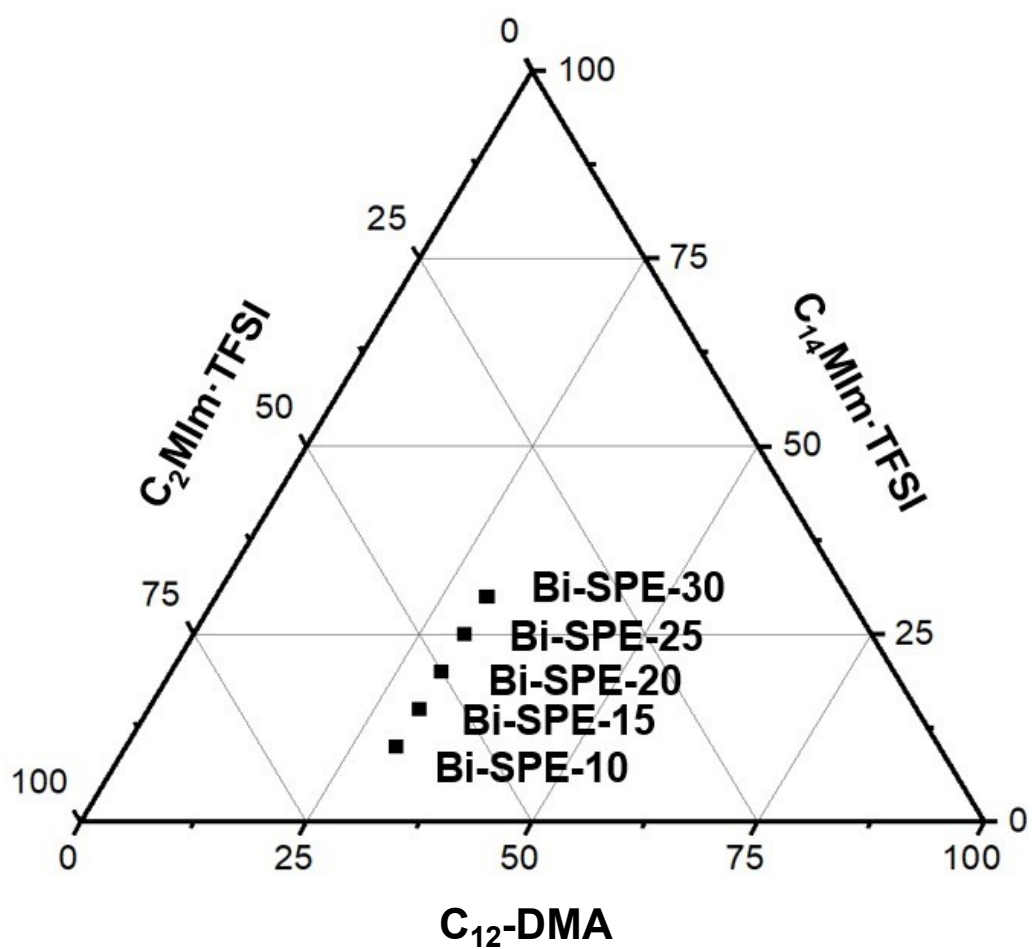


Figure S1. Ternary diagram of C₂Mim·TFSI, C₁₄Mim·TFSI, and C₁₂-DMA for Bi-SPE series.



Figure S2. Digital image of Bi-SPE-5, which is opaque, indicating that thermodynamically unstable macroemulsions were formed.

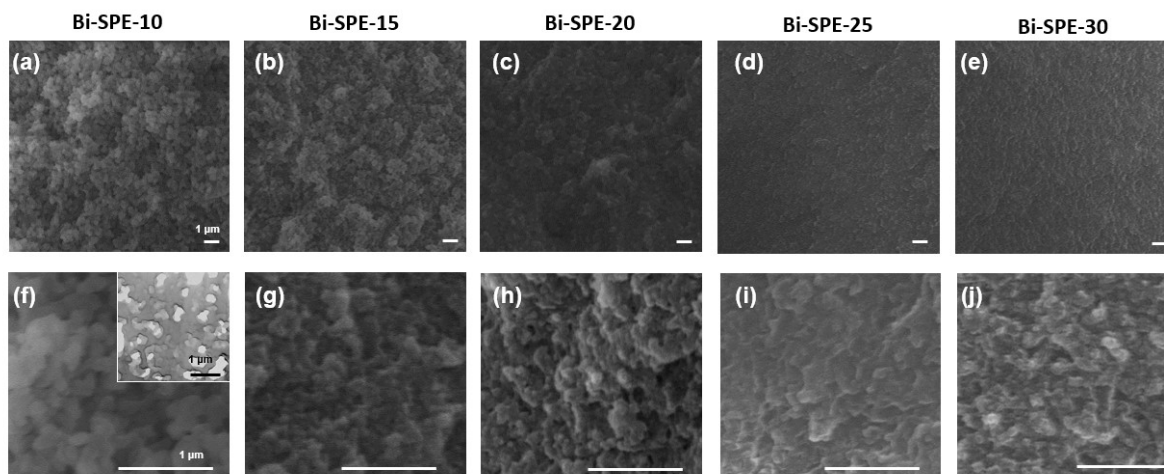


Figure S3. (a-e) Low- and (f-j) high-magnification scanning electron microscopy (SEM) images of (a,f) Bi-SPE-10, (b,g) Bi-SPE-15, (c,h) Bi-SPE-20, (d,i) Bi-SPE-25, and (e,j) Bi-SPE-30. Inset of (f) shows the cross-sectional TEM image of Bi-SPE-10. Scale bars are 1 μm.

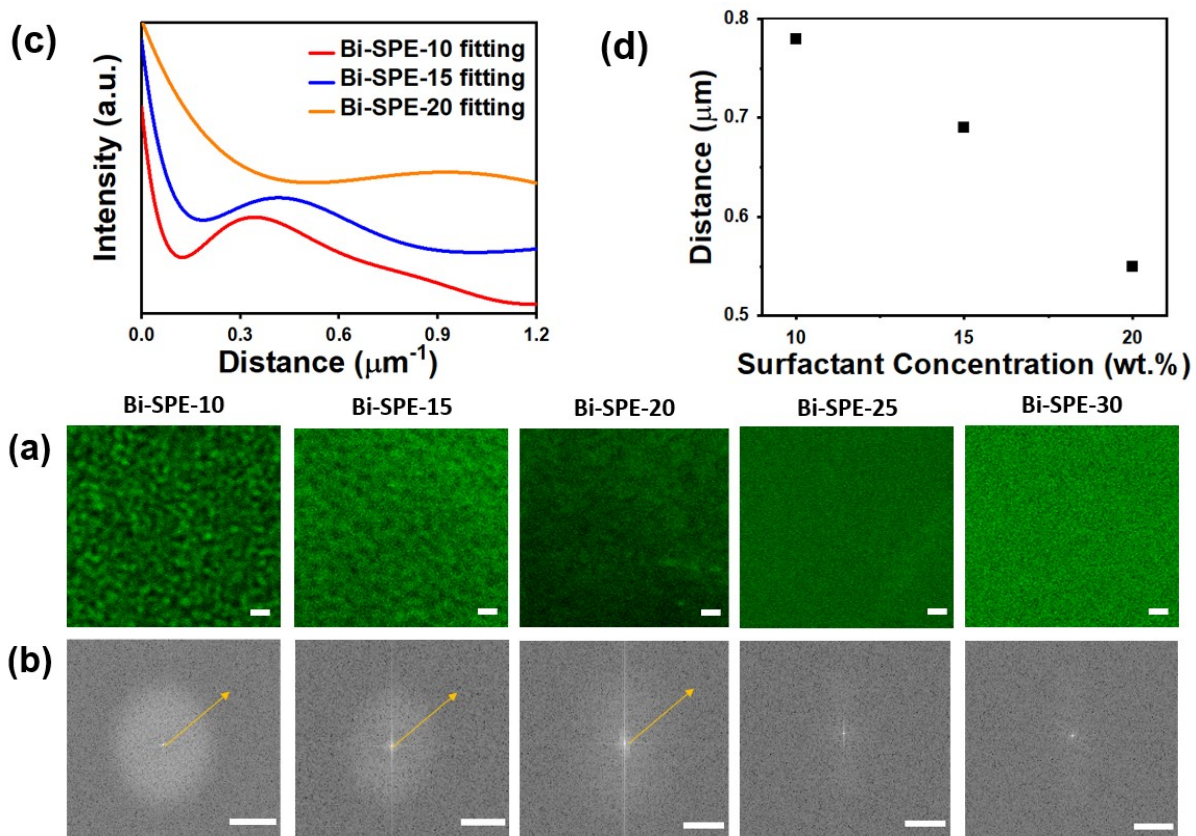


Figure S4. (a) Confocal microscopy images of the Bi-SPE series with green-colored solidified C12-DMA (scale bar = 1 μm). (b) FFT images from confocal microscopy (scale bar = 0.5 μm⁻¹). (c) Line intensities along the yellow arrows in FFT images of Bi-SPE-10, -15, -20. (d) Interdistances domain of Bi-SPE-10, -15, -20.

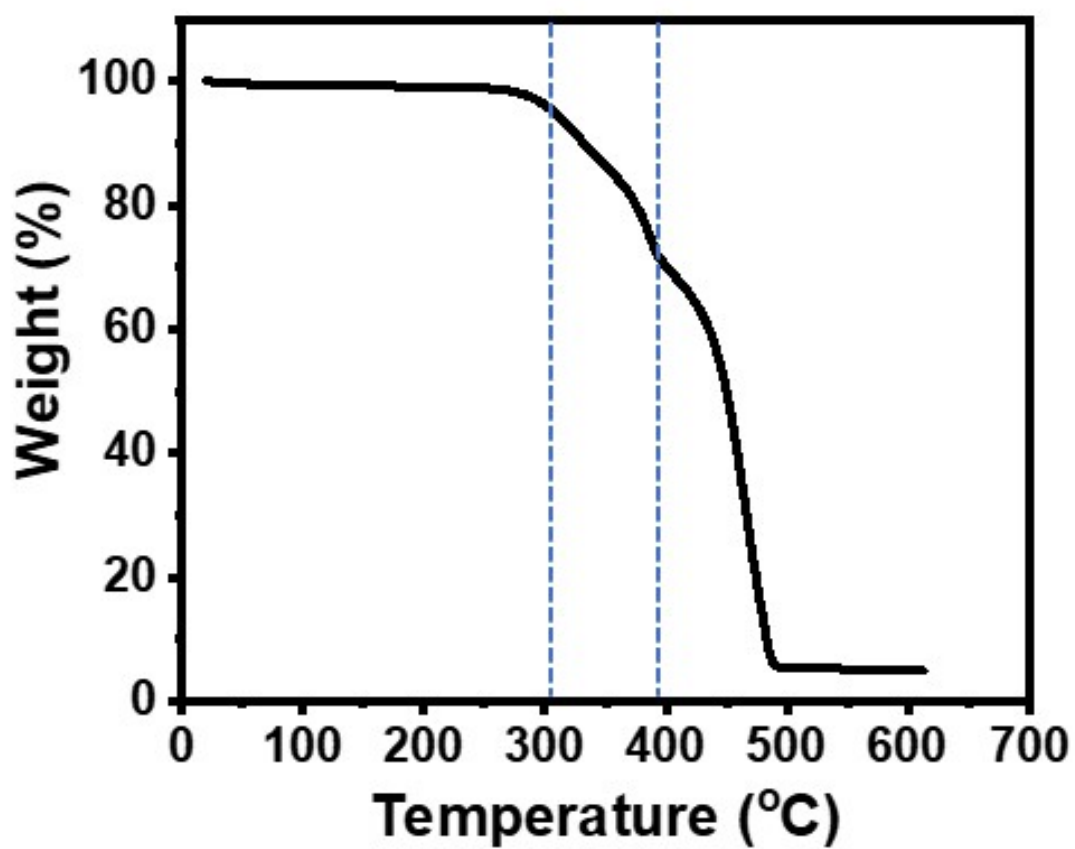


Figure S5. TGA curve of Bi-SPE-10.

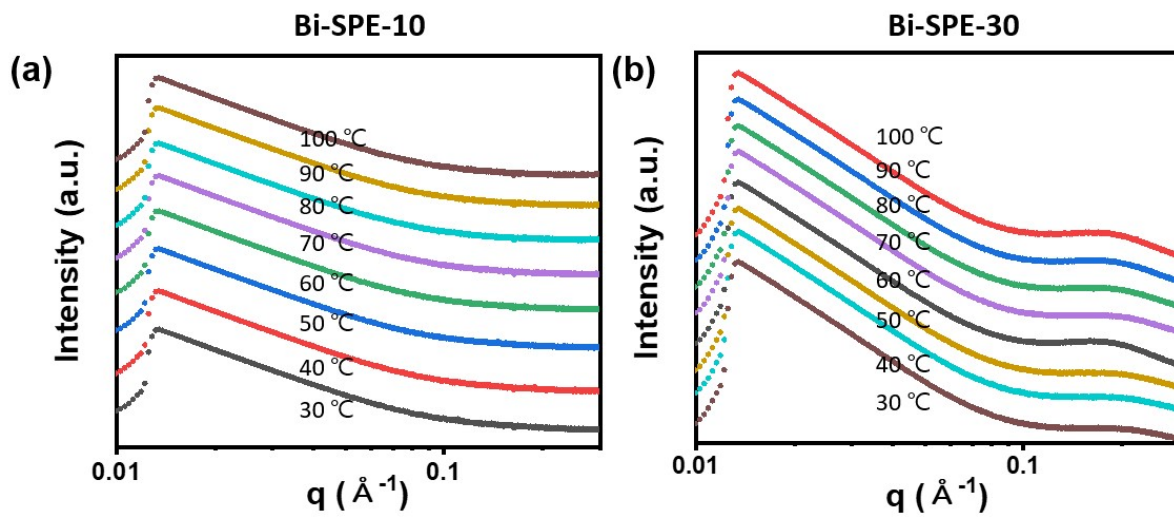


Figure S6. SAXS fitting curves with the function of temperature from 30 to 100 °C.

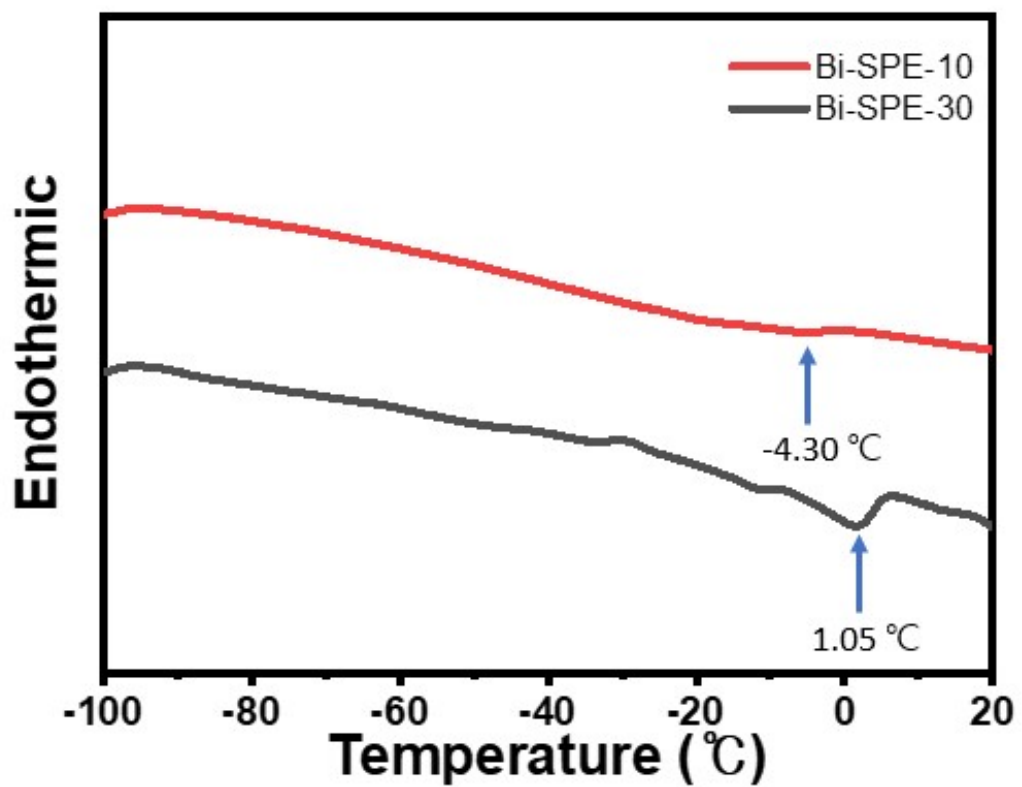


Figure S7. DSC curves of Bi-SPE-10 and Bi-SPE-30.

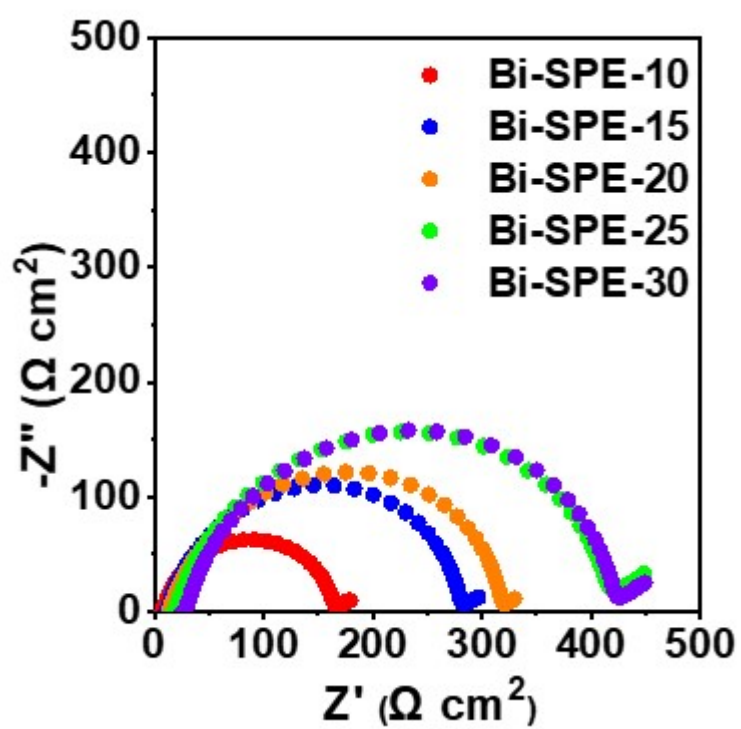


Figure S8. Impedance of symmetric lithium cell with five Bi-SPEs after steady-state.

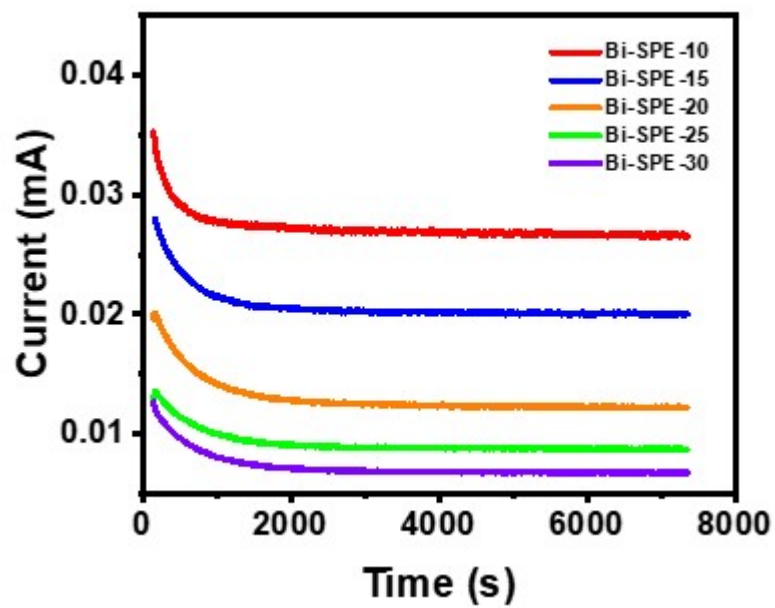


Figure S9. Time-dependent current profiles at 10 mV of symmetric lithium cells with five Bi-SPEs.

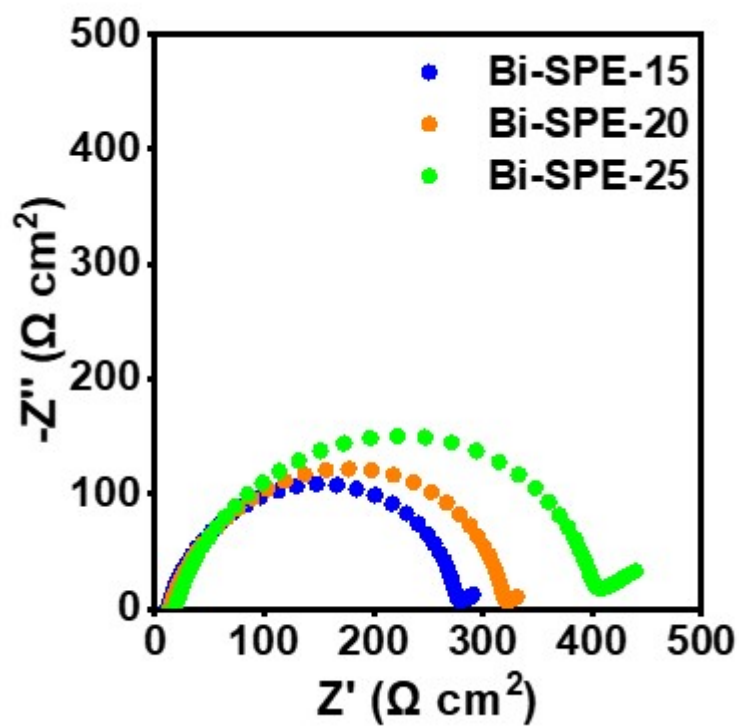


Figure S10. Impedance spectra of symmetric lithium cells with Bi-SPE-15, -20 and -25 before steady-state.

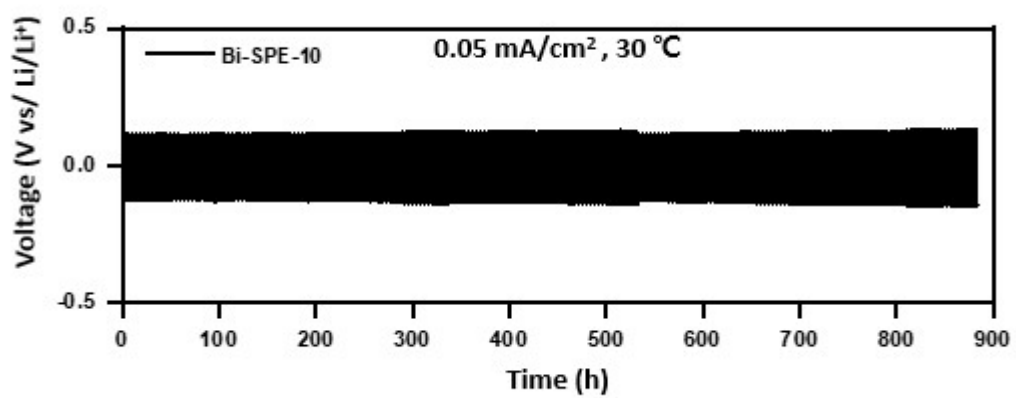


Figure S11. Time-dependent voltage profile of Li|Bi-SPE-10|Li symmetric cell at 0.05 mA/cm²

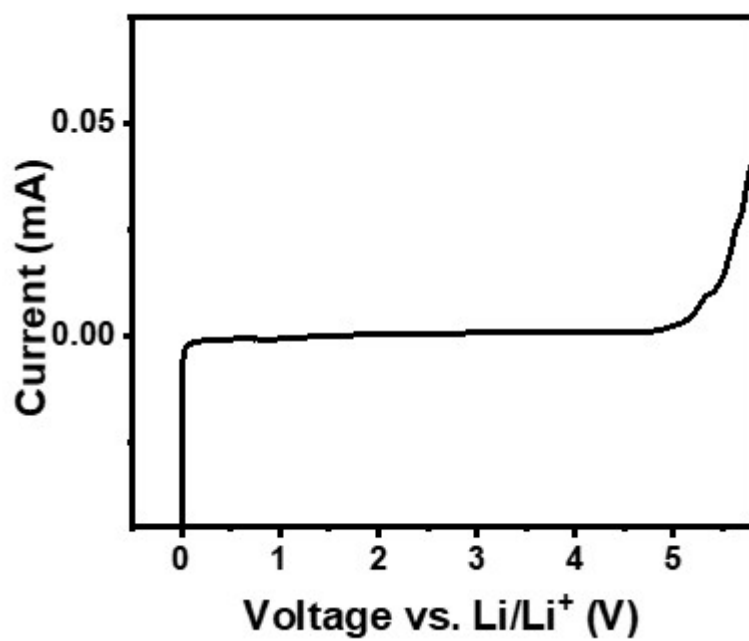


Figure S12. Linear sweep voltammogram of Bi-SPE-10.

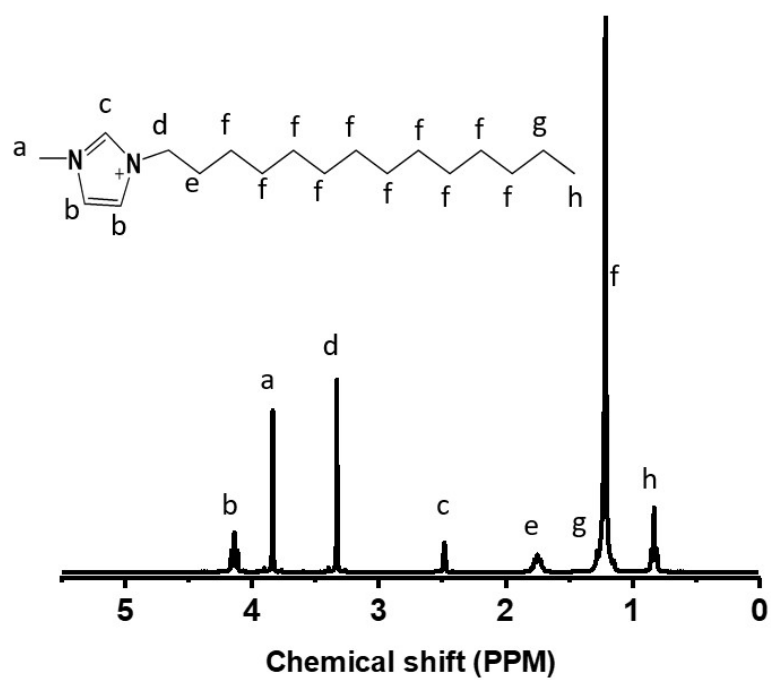


Figure S13. 1H NMR spectrum of C_{14} MI⁺·Cl with corresponding assignment of the signals of imidazolium unit and aliphatic chain.

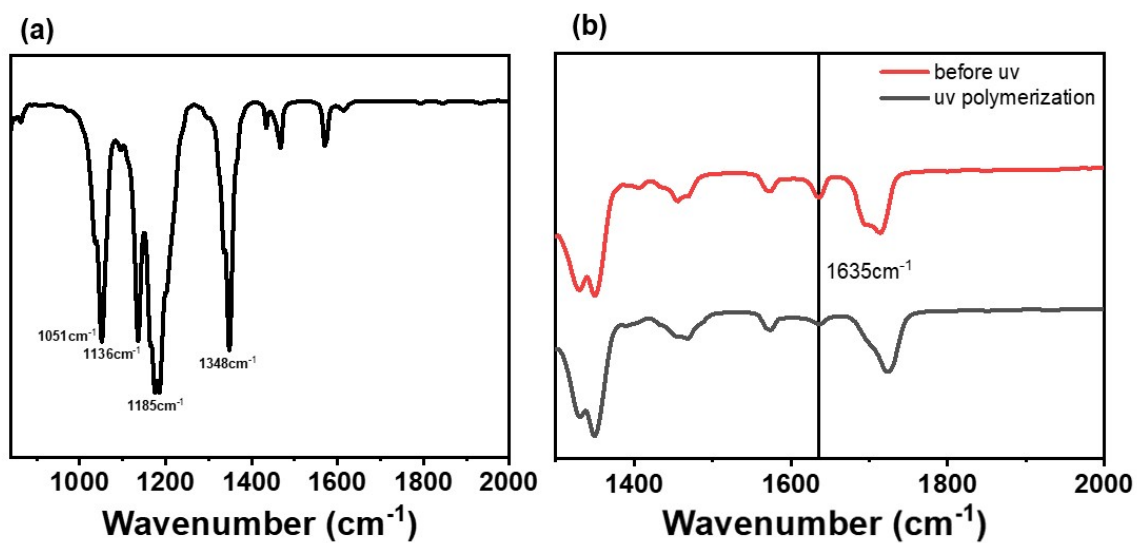


Figure S14. FT-IR peaks of (a) C₁₄MIm·TFSI and (b) Bi-SPE-10 before and after UV polymerization, in which the peak of C=C in acrylate monomers decreased at 1635 cm⁻¹.

Table S1. Initial and equilibrium impedances (R_o , R_{ss}), initial and equilibrium currents (I_o , I_{ss}), and Lithium transference number (t_+) of five Bi-SPEs.

Sample name	R_o (ohm)	R_{ss} (ohm)	I_o (mA)	I_{ss} (mA)	t_+
Bi-SPE-10	165.0	165.8	0.035	0.0265	0.58
Bi-SPE-15	275.2	279.2	0.027	0.021	0.48
Bi-SPE-20	311	311.7	0.019	0.012	0.42
Bi-SPE-25	388.3	410.3	0.016	0.01	0.39
Bi-SPE-30	407.4	430.2	0.018	0.0067	0.35