Electronic supplementary information

Stable Zn electrodes enabled by an ultra-thin Zn phosphate protective layer

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Optimization of protective layer thickness

Zn foils were treated in PPA solutions at different concentrations (1, 5, and 20 wt.%). The resulting Zn foils were denoted as 1wt% PPA-Zn, 5wt% PPA-Zn, and 20wt% PPA-Zn, respectively. Next, symmetric cells were assembled to compare their internal resistance. As shown in Figure S1, the charge transfer resistance increases with the PPA concentration, suggesting that excessive Zn phosphate protective layers are detrimental to long-term cycling stability. Furthermore, at the current density of 5 mA cm⁻² and the areal capacity of 1 mAh cm⁻², the symmetric cell with 5wt% PPA-Zn anode delivered the highest cycle stability (Figure S2). Therefore, we set the concentration of the PPA solution to 5 wt.% for the rest of our studies.



Figure S1. EIS profiles of symmetric cells assembled using Zn foils with Zn phosphate protective layers in different thicknesses: bare Zn, 1wt% PPA-Zn, 5wt% PPA-Zn, and 20wt% PPA-Zn.



Figure S2. (a) Cycle stability and (b) voltage profiles of symmetric cells assembled using Zn foils with Zn phosphate protective layers in different thicknesses: bare Zn, 1wt% PPA-Zn, 5wt% PPA-Zn, and 20wt% PPA-Zn.



Figure S3. Element mapping results obtained by SEM-EDX for (a) PPA-Zn and (b) Zn



Figure S4. FTIR spectra of PPA-Zn and Zn foil.



Figure S5. The XPS depth profile of P 2p for (a) 1 wt% PPA-Zn and (b) 20 wt% PPA-Zn.



Figure S6. XRD patterns of PPA-Zn and Zn foil.



Figure S7. SEM images of PPA-Zn foils after plating Zn at different current densities, areal capacities, and charge/discharge cycles: (a) 10 charge/discharge cycles and (b) 100 charge/discharge cycles at the current density of 10 mA cm⁻² and the areal capacity of 1 mAh cm⁻², (c) 10 charge/discharge cycles at the current density of 10 mA cm⁻² and the areal capacity of 2 mAh cm⁻², (d) 10 charge/discharge cycles at the current density of 10 mA cm⁻² and the areal capacity of 5 mAh cm⁻². SEM images of pristine Zn foils after plating Zn at different current densities, areal capacities, and cycles: (e) 10 charge/discharge cycles at the current density of 10 mA cm⁻² and the areal capacities, areal capacities, and cycles: (e) 10 charge/discharge cycles and (f) 100 charge/discharge cycles at the current density of 10 mA cm⁻² and the areal capacity of 5 mAh cm⁻².



Figure S8. XPS spectrum of PPA-Zn after Zn plating at the current density of 10 mA cm⁻² and areal capacity of 1 mAh cm⁻².



Figure S9. The discharge-charge curve of a symmetric cell assembled using PPA-Zn. Because there is Zn at the initial stage, the cut-off voltage cannot reach 0.5 V.



Figure S10. The cycle stability of PPA- Zn/V_2O_5 and the Zn/V_2O_5 full cells.

Anode	Coating thickness	Volume increment after coating	Current density (mA cm ⁻²)	Areal Capacity (mAh cm ⁻²)	DOD	Cycle Life (h)	Cumulative plating capacity (Ah cm ⁻²)	Reference
PPA-Zn	45 nm	0.09%	2	1	3.4%	6500	6.5	This work
PPA-Zn	45 nm	0.09%	10	5	17.0%	300	1.5	This work
MOF@Zn	8-10 μm	/	3	0.5	/	500	0.75	ACS Appl. Mater. Interfaces, 2019 , 11, 32046 ¹
rGO@Zn	/	/	10	2	C 3.1%	200	1.00	ACS Appl. Mater. Interfaces, 2018 , 10, 25446 ²
TiN@Zn	1.3 µm	0.65%	2	2	C 1.7%	1000	1	ACS Energy Lett., 2022, 7, 197 ³
ZnS@Zn	0.5 µm	5%	2	2	C 34.2%	1100	1.10	<i>Adv. Mater.</i> , 2020 , 32, 2003021 ⁴
Zn3(PO4)2@Zn	140 nm	1.4%/ 0.2%	1	5	C 14.2%	800	0.40	<i>Adv. Mater.</i> , 2021 , 33, 2007416 ⁵
Zn@CNT	/	/	5	2.5	C 37.5%	100	0.25	Adv. Mater., 2019, 31, 1903675 ⁶
Zn@3D- CNT/EVA	/	/	10	1	/	100	0.50	Adv. Mater., 2022 , 34, 2200860 ⁷
CaCO ₃ @Zn	/	/	0.25	0.05	/	400	0.05	Adv. Energy Mater., 2018 , 8, 1801090 ⁸
Zn@3D-Ni	/	/	5	2	/	200	0.50	Adv. Energy Mater., 2021 , 11, 2003927 ⁹
Kaolin@Zn	21 µm	1%	4.4	1.1	C 1.9%	800	1.76	Adv. Funct. Mater., 2020, 30, 2000599 ¹⁰
PVB@Zn	1 µm	10%	0.5	0.5	С	2200	0.55	Adv. Funct. Mater., 2020, 30, 2001263 ¹¹

Table S1. Comparison of the electrochemical performance of Zn metal-based symmetric cells reported in recently published studies. For the studies in which the DODs were not mentioned, we calculated the DODs according to their supplied information and marked as C.

					8.5%			
NaTi2(PO4)3@Zn	10-25 μm	/	1	1	/	250	0.13	Adv. Funct. Mater., 2020,30, 2004885 ¹²
TiO ₂ +PVDF@Zn	5 µm	6.25%	8.85	8.85	60%	250	1.11	Adv. Funct. Mater., 2020 , 30, 2001867 ¹³
Zn@ZnO	//	/	5	2.5	4.3%	100	0.25	Adv. Funct. Mater., 2020 , 30, 2004210 ¹⁴
ZrO2@Zn	4 µm	/	5	1	/	2100	5.25	Adv. Funct. Mater., 2020 , 13, 1908528 ¹⁵
Mxene@Zn	200 nm	0.4%	0.2	0.2	C 0.68%	850	0.09	Angew. Chem. Int. Ed., 2021, 60, 2861 ¹⁶
CNT@Zn	/	/	1	1	/	400	0.20	Chem. Eng. J., 2020 , 384, 123355 ¹⁷
In@Zn	8 µm	8%	4	1	C 1.7%	400	0.80	Chem. Eng. J., 2020 , 396, 125363 ¹⁸
Zn@Graphite fiber	/	/	1	1	C 16.7%	700	0.35	Electrochim. Acta, 2017 , 244, 172 ¹⁹
Zn@ZnO-3D	/	/	5	1.25	1.3%	500	1.25	Energy Environ. Sci., 2020 , 13, 503 ²⁰
PA@ Zn	40 µm	200%/ 28.6%	10	10	85%	75	0.38	Energy Environ. Sci., 2019, 12, 1938 ²¹
3D-gradient alloy	/	/	3	3	/	700	1.05	Energy Environ. Sci., 2022, 15, 1086 ²²
Cu–Zn/Zn	132.2 μm	/	1	0.5	/	1500	0.75	<i>Energy Stor. Mater.</i> , 2020 , 27, 205 ²³
3D porous Zn	/	/	5	10	71%	200	0.50	Energy Stor: Mater., 2020 , 30, 104 ²⁴
C sphere@Zn	3.9 µm	3.9%	1	1	C 1.7%	1000	0.50	Energy Stor. Mater., 2020 , 25, 858 ²⁵
Zn@MOF	/	/	1	1	C 20%	50	0.03	Joule, 2019 , 3, 1289 ²⁶
Al ₂ O ₃ @Zn	10 nm	/	1	1	C 0.85%	500	0.25	J. Mater. Chem. A, 2020 , 8,7836 ²⁷

Liquid metal@Zn	1 mm	/	2.5	2.5	/	200	0.25	J. Mater. Chem. A, 2021, 9, 5597 ²⁸
AgZn ₃ @Zn	14 µm	/	0.25	0.5	/	1750	0.218	J. Mater. Chem. A, 2021 , 9,8452 ²⁹
Montmorillonite @Zn	/	/	2	1	C 5.7%	700	0.7	J. Mater. Chem. A, 2021, 9,16814 ³⁰
TiO ₂ @Zn	20 µm	66.7%	2	2	C 11.4%	280	0.28	<i>Nature Commun.</i> , 2020 , 11, 3961 ³¹
Zn@3D In	/	/	1	1	/	1000	0.50	Nano Energy, 2022 , 99, 107331 ³²

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