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Supporting Information

High Photoactive Black Phase Stability of CsPbl₃ Nanocrystals under

Damp-heat Conditions of 85°C and 85% Relative Humidity

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Figure S1. XRD pattern analysis of undoped and doped (Ni-3.65%) CsPbI₃ NCs.



Figure S2. TEM image and size distribution statistics of undoped CsPbI₃ NCs.



Figure S3. TEM images and size distribution statistics of (a) Ni-1.25%, (b) Ni-3.65%, (c) Ni-4.42%, (d) Ni-5.28% CsPbI₃ NCs.



Figure S4. HRTEM image and corresponding FFT of Ni-1.25% $CsPbI_3 NCs$.



Figure S5. EDS spectrum of Ni-3.65% $CsPbI_3 NCs$.



Figure S6. Full spectrum of the $85^{\circ}C/85\%$ RH damp-heat stability test of undoped and doped CsPbI₃ NCs films.



Figure S7. The stability of doped (Ni-3.65% CsPbI₃ NCs, left) and undoped (right) CsPbI₃ NCs solutions mixed with equal volume of water, the solutions were intensely sloshed by stirrers in the whole test process.



Figure S8. Doped (Ni-3.65% CsPbI₃ NCs, left) and undoped (right) CsPbI₃ NCs films soaking in the water.



Figure S9. The stability of doped (Ni-3.65% CsPbI₃ NCs, left) and undoped (right) CsPbI₃ NCs films soaking in water.



Figure S10. The stability of doped and undoped CsPbI₃ NCs solutions under ambient condition.



Figure S11. DFT calculation results. (a) Stability regions of Cs, Pb, and I compounds against Cs and Pb chemical potentials. The shaded region is the stability region for the synthesis of CsPbI₃, and the different patterns indicate whether the formation of Ni_{Cs} or Ni_{Pb} is preferred. The formation of Ni_i is not favored over the whole region. (b) Crystal formation energies of CsPbI₃ with different Ni/Pb doping concentrations.



Figure S12. Absorption spectrum (a) and photoluminescence (PL) spectrum (b) of the undoped $CsPbI_3$ and Ni^{2+} doped $CsPbI_3$ (4% Ni/Pb) bulk films.



Figure S13. XRD pattern of the undoped CsPbI₃ and Ni²⁺ doped CsPbI₃ (4% Ni/Pb) bulk films.



Figure S14. Stabilities of the Ni²⁺ doped CsPbI₃ (4% Ni/Pb) and undoped CsPbI₃ bulk films at 150 $^{\circ}$ C under the N₂ atmosphere.



Figure S15. Stabilities of the Ni²⁺ doped CsPbI₃ (4% Ni/Pb) and undoped CsPbI₃ bulk film at 85 $^{\circ}$ C under the N₂ atmosphere.

| Samples | Nil ₂ /Pbl ₂ | Reaction Temperatures (°C) | ICP-MS results (Ni/Pb) | Sizes (nm) |
|----------|------------------------------------|----------------------------|------------------------|------------|
| undoped | 0:1 | 170 | - | 15.23 |
| Ni-1.25% | 1:1 | 170 | 0.0125:1 | 11.08 |
| Ni-3.65% | 3:1 | 170 | 0.0365:1 | 6.95 |
| Ni-4.42% | 3:1 | 185 | 0.0442:1 | 7.76 |
| Ni-5.28% | 3:1 | 200 | 0.0528:1 | 10.03 |

Table S1. ICP-MS results of Ni doped CsPbI₃ NCs with different reaction temperature and NiI₂/PbI₂ loading ratio.

Table S2. Fitted TRPL decay results, average lifetimes (τ_{ave}), PLQYs, radiative (Γ_{rad}) and nonradiative ($\Gamma_{non-rad}$) decay rates of undoped and doped NCs. Crystal sizes were also given. The methods of calculation referenced previous study¹. The bi-exponential function was used to fit the decay curves except Ni-3.65% CsPbI₃ NCs, because it sufficient to be fitted with a single-exponential function for Ni-3.65% CsPbI₃ NCs.

| Samples | A1 | τ1 (ns) | A2 | τ2 (ns) | τave (ns) | PLQY (%) | Γrad (μs⁻¹) | Γnon-rad (μs ⁻¹) | Sizes (nm) |
|----------|------|---------|------|---------|-----------|----------|-------------|------------------------------|------------|
| undoped | 0.66 | 22.97 | 0.33 | 106.34 | 81.08 | 0.31 | 3.88 | 8.45 | 15.23 |
| Ni-1.25% | 0.82 | 26.97 | 0.21 | 95.54 | 59.27 | 0.57 | 9.56 | 7.31 | 11.08 |
| Ni-3.65% | 1.03 | 18.94 | 0.00 | 0.00 | 18.94 | 0.88 | 46.31 | 6.50 | 6.95 |
| Ni-4.42% | 0.89 | 26.59 | 0.15 | 84.21 | 46.93 | 0.64 | 13.66 | 7.65 | 7.76 |
| Ni-5.28% | 0.84 | 28.62 | 0.19 | 94.78 | 57.34 | 0.59 | 10.24 | 7.20 | 10.03 |

1. Z.-J. Yong, S.-Q. Guo, J.-P. Ma, J.-Y. Zhang, Z.-Y. Li, Y.-M. Chen, B.-B. Zhang, Y. Zhou, J. Shu, J.-L. Gu, L.-R. Zheng, O. M. Bakr and H.-T. Sun, *J. Am. Chem. Soc.*, 2018, **140**, 9942-9951.