

Supporting Information

Tumor-Triggered Targeting Ammonium Bicarbonate Liposomes for Tumor

Multimodal Therapy

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Materials

The cholesterol (Chol), 1,2-Dipalmitoyl-sn-glycero-3-phosphocholine (DPPC), and 1,2-distearoyl-sn-glycero-3-phosphateethanolamine-N-(amino-(polyethyleneglycol)-2000) (DSPE-PEG₂₀₀₀) were supplied by Avanti Polar Lipids (USA). The 1,2 - stearyl - sn - glycerol - 3 - phosphatethanolamine (DSPE) and 1,2 - distearoyl - sn - glycerol - 3 - phosphoethanolamine - N - [folate (polyethylene glycol) - 2000] (DSPE - PEG₂₀₀₀ - FA) were supplied by AVT (Shanghai, China). Doxorubicin hydrochloride (DOX) and indocyanine green (ICG) were supplied by Melonepharma (Dalian, China). Methoxy PEG₅₀₀₀ propionaldehyde (mPEG₅₀₀₀ - CHO) was supplied by JenKem Technology (Beijing, China), and other chemicals were supplied by Sinopharm Chemical Reagent Co. Ltd, China.

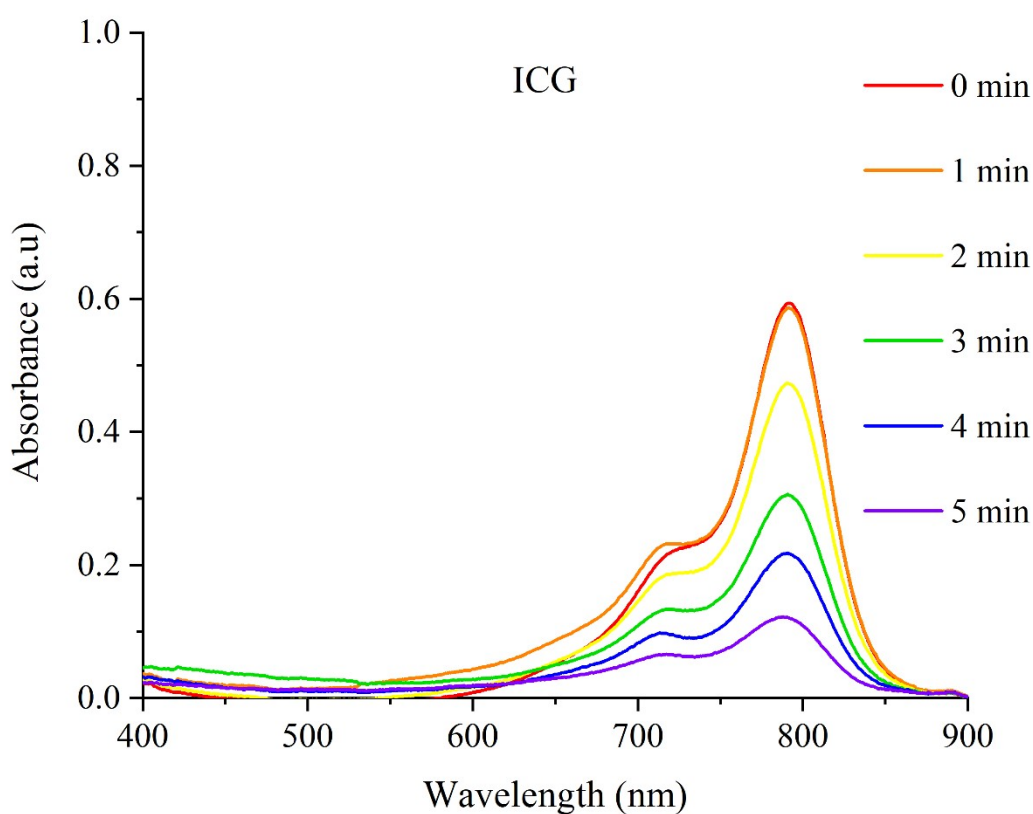


Figure S1 UV-visible absorption spectra of free ICG

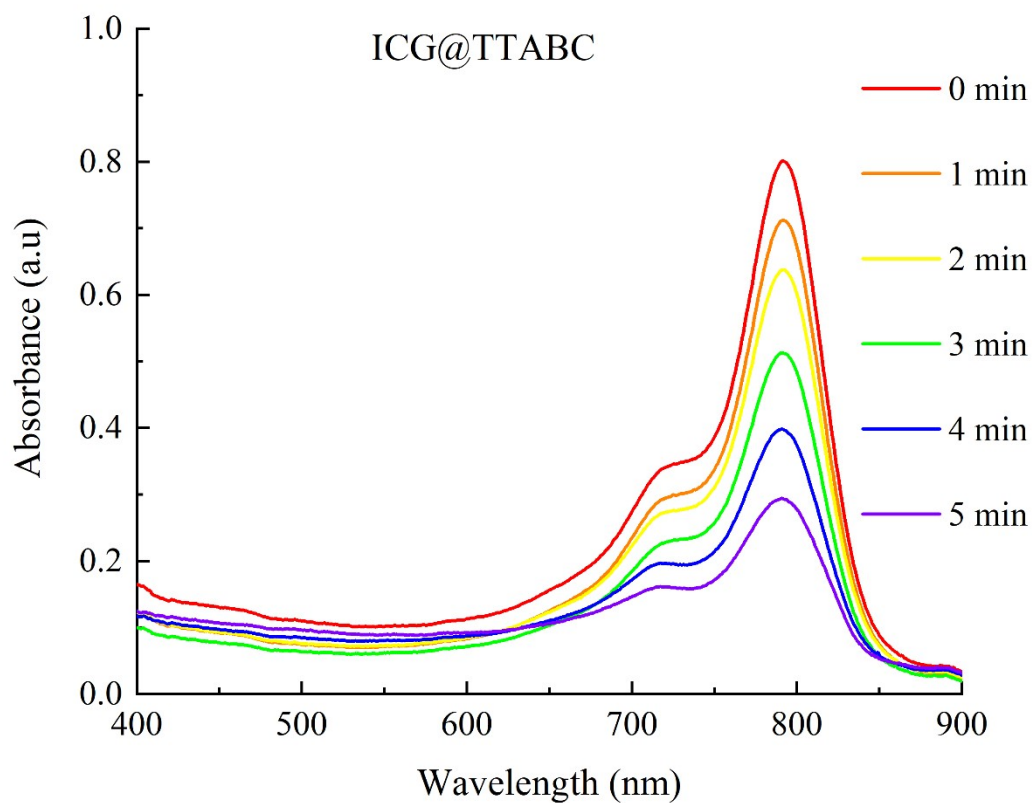


Figure S2 UV-visible absorption spectra of ICG@TTABC

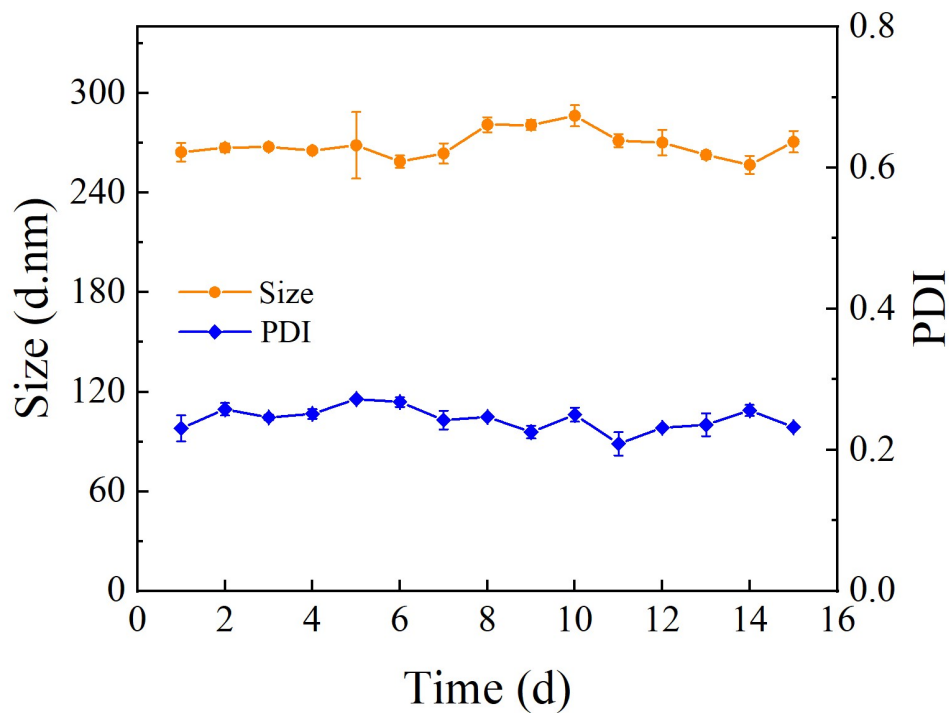


Figure S3 Hydrodynamic size of ICG&DOX@TTABC within 15 days

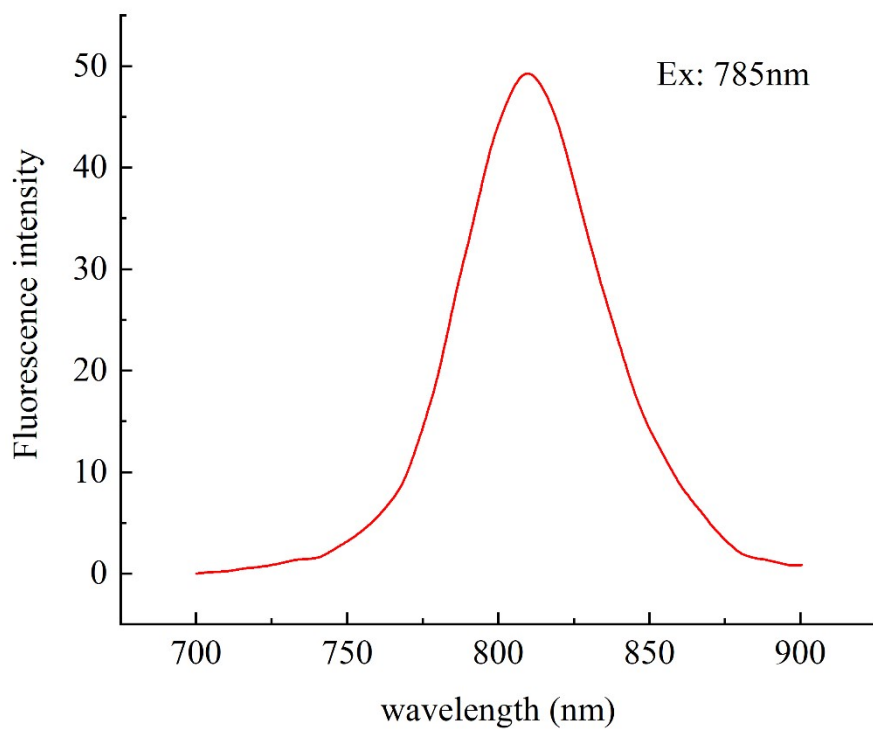


Figure S4 Fluorescence emission spectrum of ICG

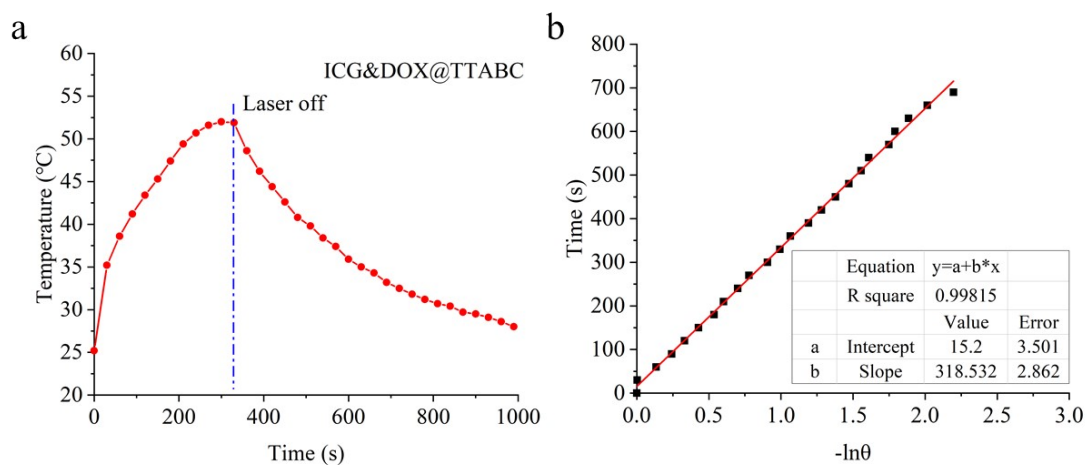


Figure S5 a) The photothermal response of the ICG&DOX@TTABC aqueous solution irradiated with NIR laser (808 nm, 1.5 W/cm²) for 330 s, and then the laser was shut off. b) Linear time data *versus* Ln θ obtained from the cooling period of panel.

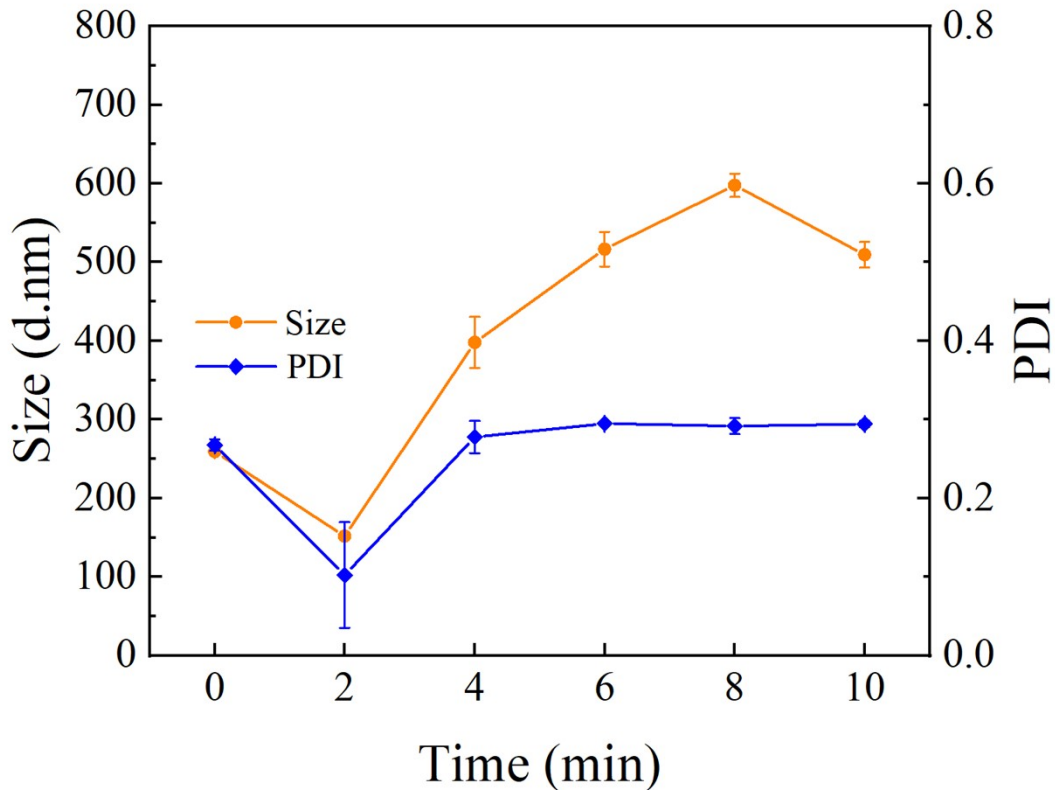


Figure S6 Hydrodynamic size of ICG&DOX@TTABC versus irradiation time (808 nm, 1.5 W/cm²)

Calculation of the photothermal conversion efficiency

The photothermal conversion efficiency of ICG&DOX@TTABC was determined according to previous method.¹⁻³

$$\sum_i m_i c_{i,p} \frac{dT}{dt} = Q_{NCs} + Q_{Dis} - Q_{Surr} \quad (1)$$

where m and $C_{i,p}$ are the mass and heat capacity of water, respectively. T is the solution temperature, Q_{NCs} is the energy inputted by ICG&DOX@TTABC, Q_{Dis} is the baseline energy inputted by the sample cell, and Q_{Surr} is heat conduction away from the system surface by air.

Q_{NCs} is the photothermal energy input by ICG&DOX@TTABC.

$$Q_{NCs} = I(1 - 10^{-A_\lambda})\eta \quad (2)$$

where I is incident laser power, η is the conversion efficiency from incident laser energy to thermal energy, and A_λ is the absorbance of the ICG&DOX@TTABC at wavelength of 808 nm. In addition, source term, Q_{Dis} is the heat associated with the light absorbance of the solvent, which is measured independently to be $Q_{Dis} = 18.9$ mW using pure water without ICG&DOX@TTABC. Furthermore, Q_{Surr} is thermal energy lost to the surroundings.

$$Q_{Surr} = hS(T - T_{Surr}) \quad (3)$$

where h is heat transfer coefficient, S is the surface area of the container, and T_{Surr} is ambient temperature of the surroundings.

Once the laser power is defined, the heat input ($Q_{NC} + Q_{Dis}$) will be finite. Since the heat output (Q_{Surr}) is increased along with the increase of the temperature according to the Eq.3, the system temperature will rise to a maximum when the heat input is equal to heat output:

$$Q_{NCs} + Q_{Dis} = Q_{Surr-max} = hS(T - T_{Surr}) \quad (4)$$

where the $Q_{Surr-Max}$ is heat conduction away from the system surface by air when the sample cell reaches the equilibrium temperature, and T_{max} is the equilibrium temperature. The 808 nm laser heat conversion efficiency (η) can be determined by substituting Eq.2 for Q_{NCs} into Eq. 4 and rearranging to get

$$\eta = \frac{hS(T - T_{Surr}) - Q_{Dis}}{I(1 - 10^{-A\lambda})} \quad (5)$$

where Q_{Dis} was measured independently to be 18.9 mW, the ($T_{max} - T_{Surr}$) was 27.0 °C according to Figure S5a, I is 1.5 W/cm², A_{808} is the absorbance (0.6390) of ICG&DOX@TTABC at 808 nm (Figure 2e). Thus, only the hS remains unknown for calculating η

In order to get the hS , a dimensionless driving force temperature, θ is introduced using the maximum system temperature, T_{max}

$$\theta = \frac{T - T_{Surr}}{T_{max} - T_{Surr}} \quad (6)$$

Substituting Eq.6 into Eq.1 and rearranging Eq.1

$$\frac{d\theta}{dt} = \frac{hS}{\sum_i m_i c_{i,p}} \left[\frac{Q_{NCs} + Q_{Dis}}{hS(T_{max} - T_{Surr})} - \theta \right] \quad (7)$$

When the laser was shut off, the $Q_{NCs} + Q_{Dis} = 0$, Eq.7 changed to

$$dt = \frac{\sum_i m_i c_{i,p}}{hS} \frac{d\theta}{\theta} \quad (8)$$

and integrating, giving the expression

$$t = - \frac{\sum_i m_i c_{i,p}}{hS} \ln \theta \quad (9)$$

$$\frac{\sum_i m_i c_{i,p}}{hS}$$

Therefore, time constant for heat transfer from the system is determined to be $\frac{\sum_i m_i c_{i,p}}{hS} = 318.532$ s by applying the linear time data from the cooling period (after 330 s) vs negative natural logarithm of driving force temperature (Figure S5b). In addition, the m is 1.0 g and the C is 4.2 J/(g °C). Thus, according to Eq. 9, the hS is deduced to be 13.18 mW/°C. Substituting 13.18 mW/°C of the hS into Eq.5, the 808 nm laser heat conversion efficiency (η) of ICG&DOX@TTABC can be calculated to be 29.16%.

References

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