Supporting information

Regulation of multi-color fluorescence of carbonized polymer dots by multiple contributions of effective conjugate size, surface state, and molecular fluorescence

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Time of reaction		6 h		10 h
Temperature Solvent	120 °C	160 °C	200 °C	160 °C
H ₂ O	CDs-1	CDs-4	CDs-7	CDs-10
DMF	CDs-2	CDs-5	CDs-8	CDs-11
DMF+HNO ₃	CDs-3	CDs-6	CDs-9	CDs-12

Table S1. CDs prepared under different conditions

 Table S2. Elemental content at different CDs.

	С	Ν	0
CDs-1	50.34	11.73	37.92
CDs-2	<mark>63</mark>	12.53	24.47
CDs-3	62.73	12.56	24.71
CDs-4	62.62	8.22	29.16
CDs-5	<mark>62.64</mark>	15.22	<mark>22.14</mark>
CDs-6	<mark>63.46</mark>	<mark>15.41</mark>	<mark>21.13</mark>
CDs-7	70.41	<mark>7.41</mark>	22.18
CDs-8	<mark>64.59</mark>	17.91	17.49
CDs-9	<mark>62.38</mark>	19.76	17.85
CDs-10	58.51	11.53	<mark>29.96</mark>
CDs-11	60.28	17.79	21.93
CDs-12	<mark>61.97</mark>	17.35	<mark>20.69</mark>



Figure S1. UV–vis spectra of CDs synthesized with the same reaction solvent and different reaction temperatures and reaction times: (A) H₂O; (B) DMF; (C) DMF+HNO₃.



Figure S2. The 2D fluorescence emission spectra of A) CDs-1, B) CDs-2, C) CDs-3, D) CDs-4, E) CDs-5, F) CDs-6, G) CDs-7, H) CDs-8, I) CDs-9, J) CDs-10, K) CDs-11 and L) CDs-12.



Figure S3. The FTIR spectrum of CDs synthesized with the same reaction solvent and different reaction temperatures and reaction times: (A) H₂O; (B) DMF; (C) DMF, HNO₃.



Figure S4. XPS spectra (full survey) of A) CDs-1, B) CDs-2, C) CDs-3, D) CDs-4, E) CDs-5, F) CDs-6, G) CDs-7, H) CDs-8, I) CDs-9, J) CDs-10, K) CDs-11 and L) CDs-12.



Figure S5. The size of (A) CDs-2, (B) CDs-10, (C) CDs-5, (D) CDs-6, (E) CDs-11 and (F) CDs-12.



Figure S6. Multicolor fluorescent CPDs synthesized in different solvents.



Figure S7. UV-vis spectra of Multicolor fluorescent CPDs.



Figure S8. The luminescence emission spectra of Multicolor fluorescent CPDs: A) DMF; B) CH₃COOH; C) CH₃CH₂OH; D) H₂O.



Figure S9. The luminescence emission spectra of r-CD (after two weeks).