

Supporting Information

**Ratiometric and lifetime-based luminescent thermometer exploiting the
Co³⁺ luminescence in CaAl₂O₄:Co³⁺ and CaAl₂O₄:Co³⁺,Nd³⁺**

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luminescence thermometer

The average decay time of the broad-band luminescence of Co³⁺ was calculated with equation S1:

$$\tau_{avr} = \frac{A_1\tau_1^2 + A_2\tau_2^2}{A_1\tau_1 + A_2\tau_2} \quad (S1)$$

for a biexponential decay,

$$y = y_0 + A_1 \cdot \exp\left(-\frac{x}{\tau_1}\right) + A_2 \cdot \exp\left(-\frac{x}{\tau_2}\right) \quad (S2)$$

where: τ_1 , τ_2 – decay components of biexponential decay and
 A_1 , A_2 – amplitudes. y_0 denotes an experimental background of the decay curve.

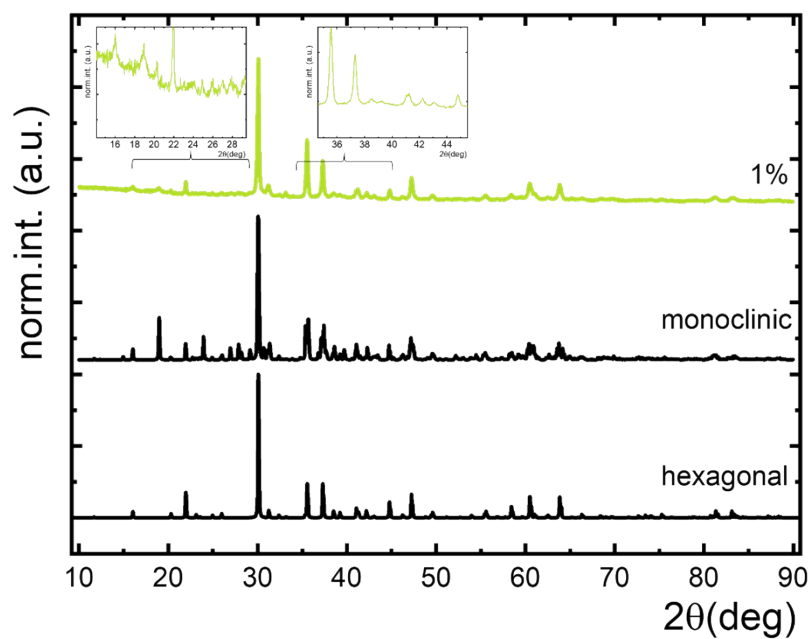


Figure S1. Comparison of the powder X-ray diffraction patterns of CaAl_2O_4 doped with 1% Co^{3+} annealed at 850 °C with reference patterns (ICSD 157457 -hexagonal CaAl_2O_4 and ICSD 260 – monoclinic CaAl_2O_4).

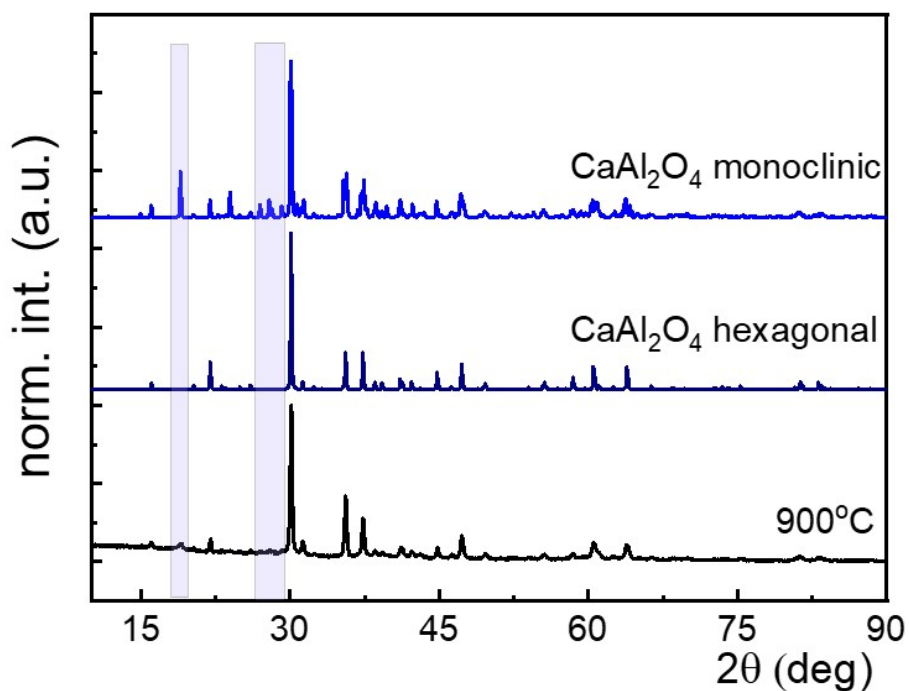


Figure S2. Comparison of powder X-ray diffraction patterns of $\text{CaAl}_2\text{O}_4: \text{Co}^{3+}$ annealed at 900 °C with reference patterns (ICSD 157457 -hexagonal CaAl_2O_4 and ICSD 260 – monoclinic CaAl_2O_4).

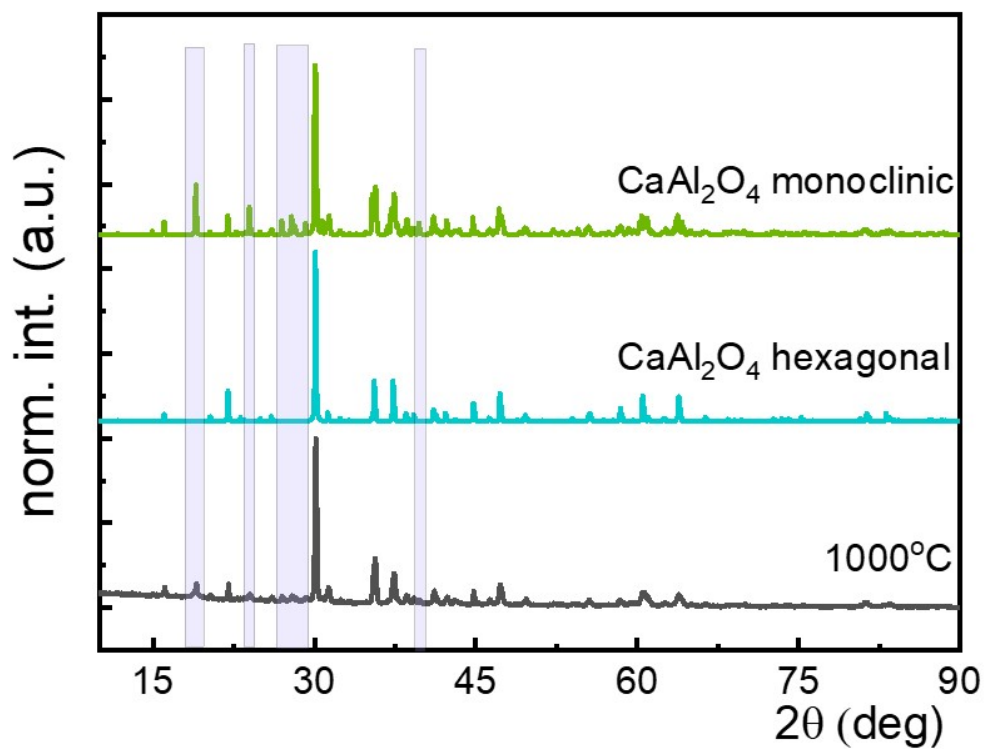


Figure S3. Comparison of powder X-ray diffraction patterns of CaAl₂O₄: Co³⁺ annealed at 1000 °C with reference patterns (ICSD 157457 -hexagonal CaAl₂O₄ and ICSD 260 – monoclinic CaAl₂O₄).

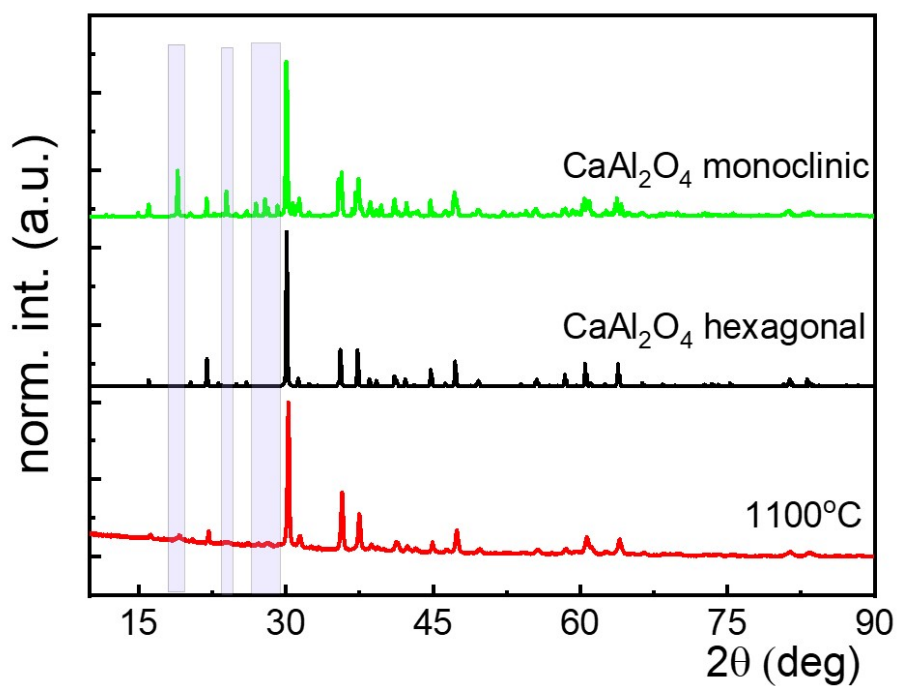
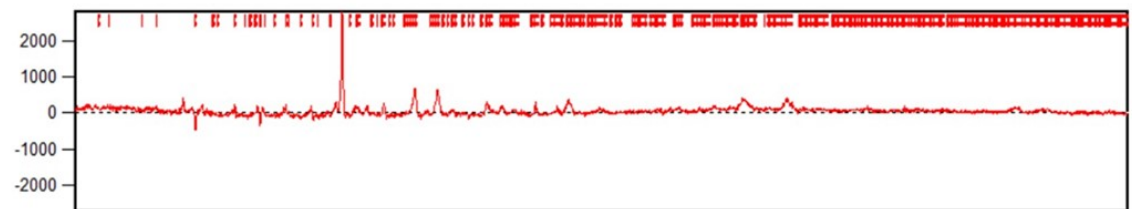
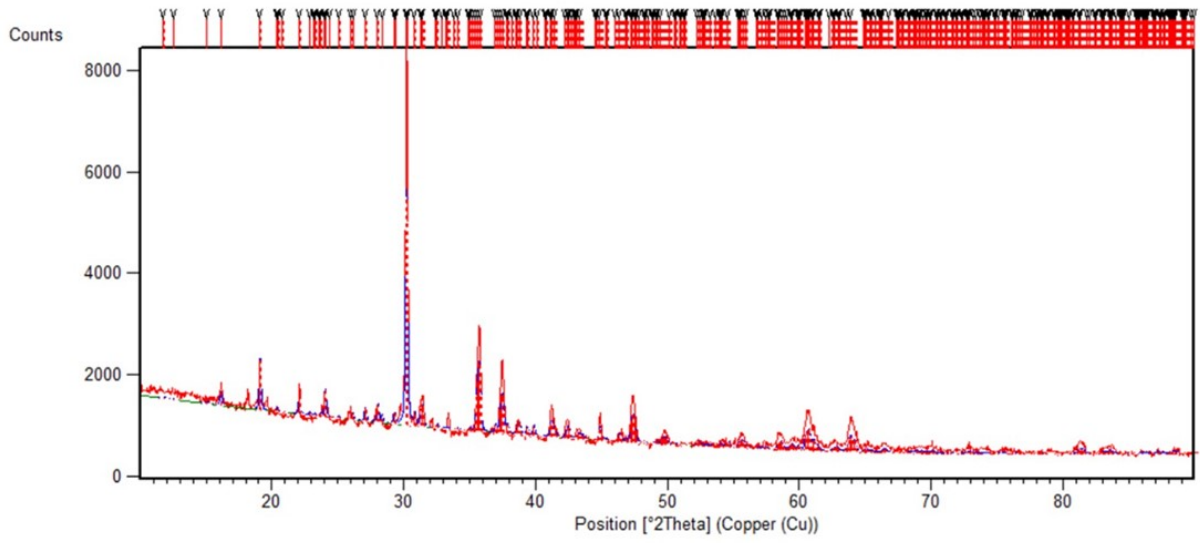
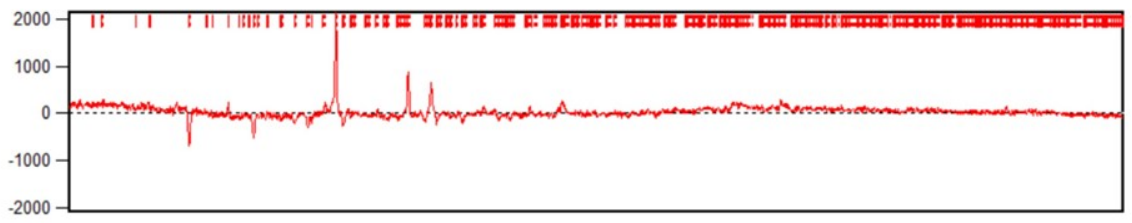
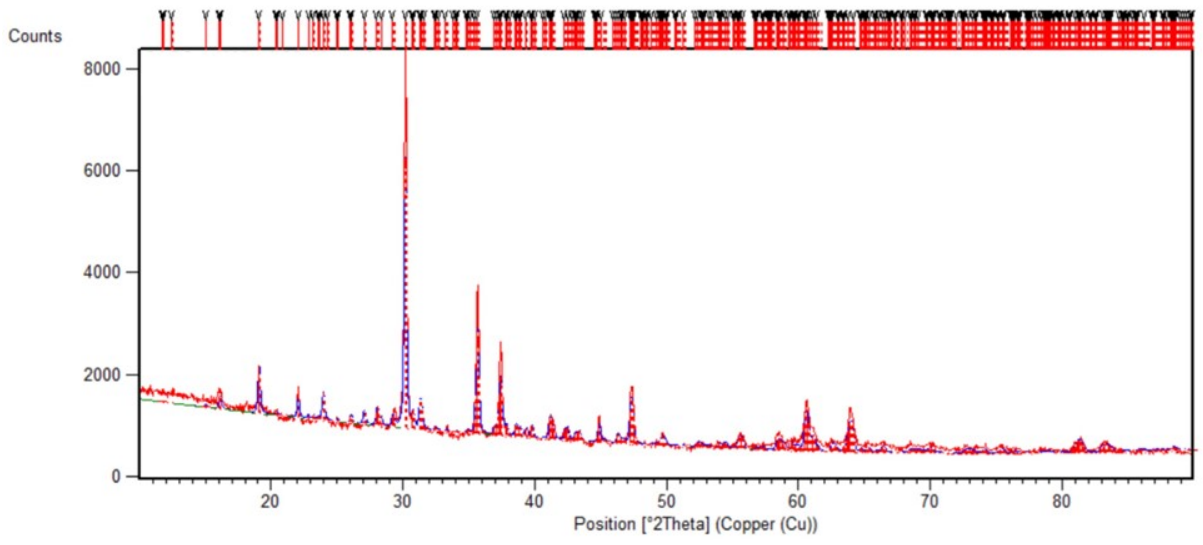


Figure S4. Comparison of powder X-ray diffraction patterns of CaAl₂O₄: Co³⁺ annealed at 1100°C with reference patterns (ICSD 157457 -hexagonal CaAl₂O₄ and ICSD 260 – monoclinic CaAl₂O₄).



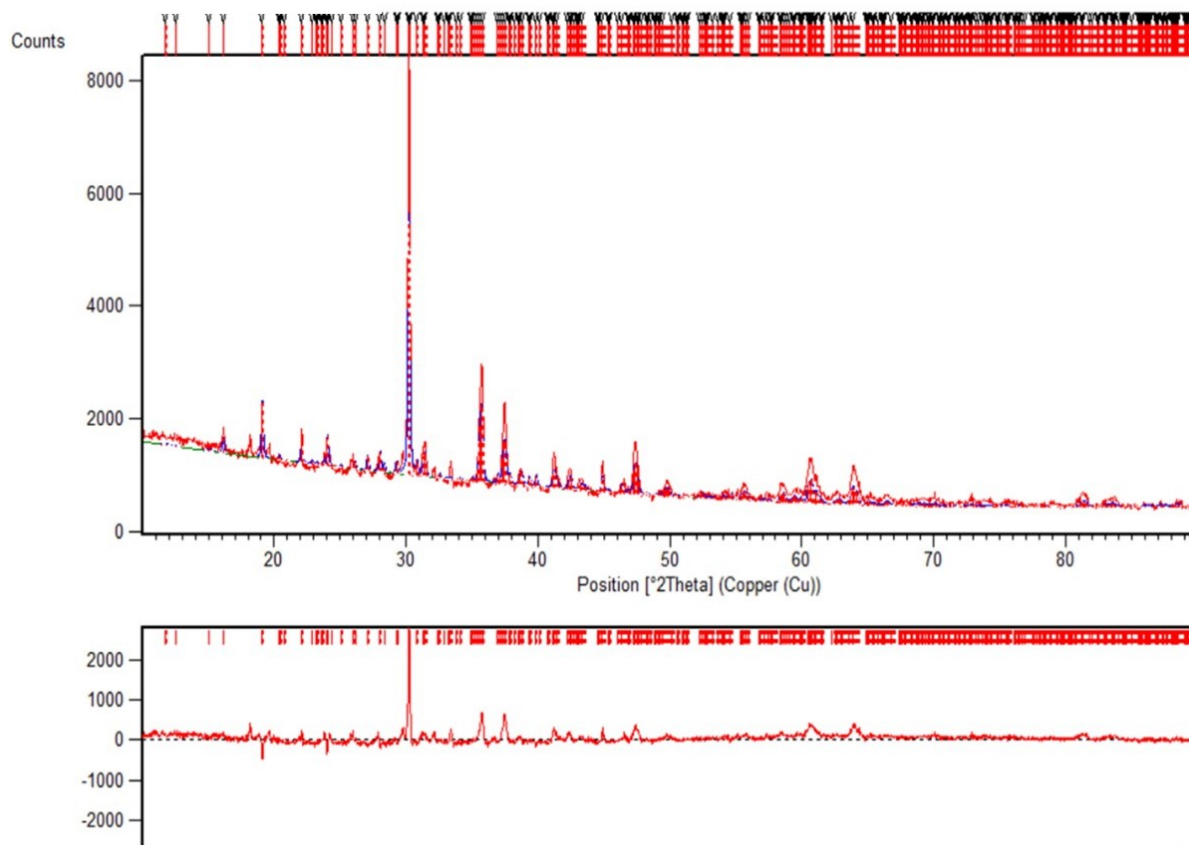


Figure S5. The representative results of the Rietveld refinement of the $\text{CaAl}_2\text{O}_4:\text{Co}^{3+}$ annealed at 1000°C -a); the $\text{CaAl}_2\text{O}_4:\text{Co}^{3+}$ annealed at 1100°C -b); and the $\text{CaAl}_2\text{O}_4:\text{Co}^{3+},\text{Nd}^{3+}$ annealed at 1100°C -c)

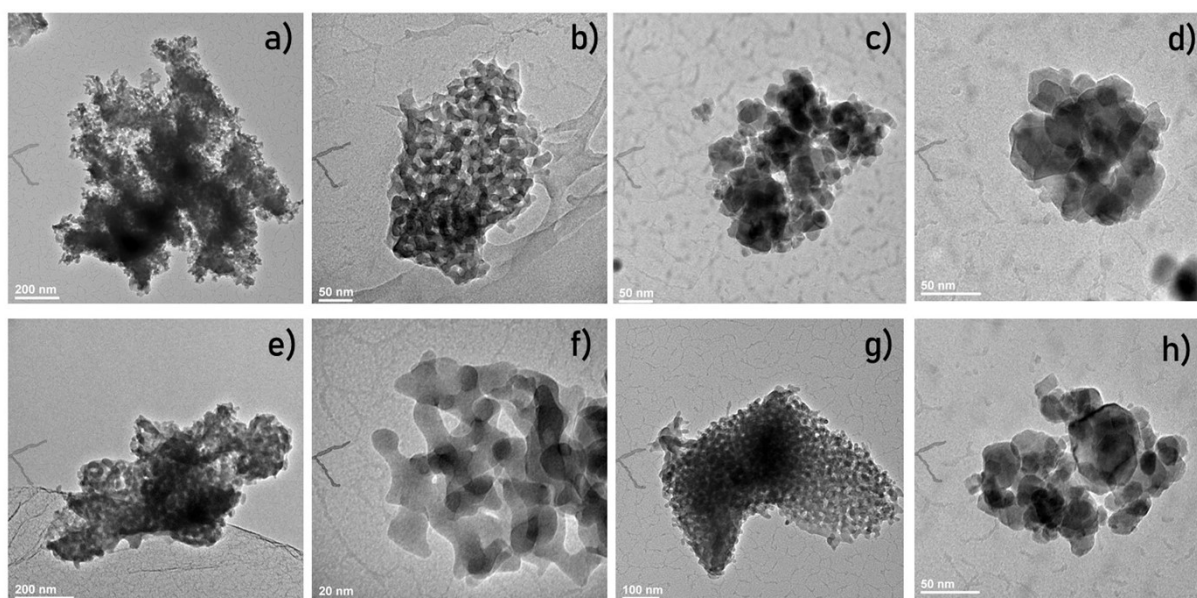


Figure S6. The representative TEM images of $\text{CaAl}_2\text{O}_4:\text{Co}^{3+}$ -a), b), c), d) and $\text{CaAl}_2\text{O}_4:\text{Co}^{3+},\text{Nd}^{3+}$ -e), f), g), h) annealed at 850°C -a), -e); 900°C -b), -f); 1000°C -c);-g) and 1100°C -d); -h).

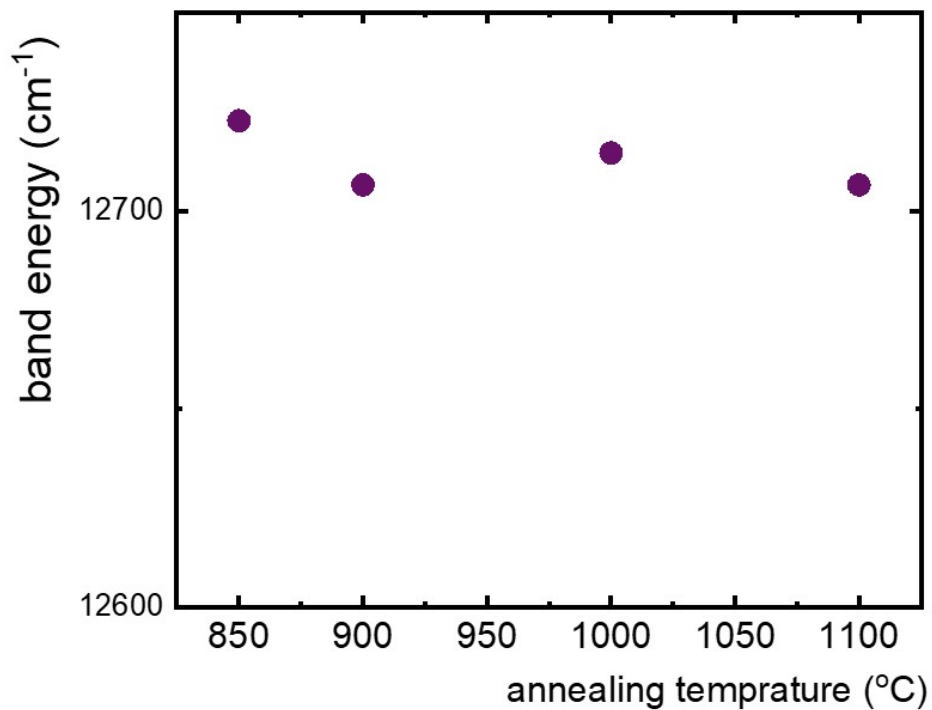


Figure S7. Energy of the maximum of the Co^{3+} -related broad-band emission as a function of annealing temperature of the prepared $\text{CaAl}_2\text{O}_4: 0.02\% \text{Co}^{3+}$ sample.

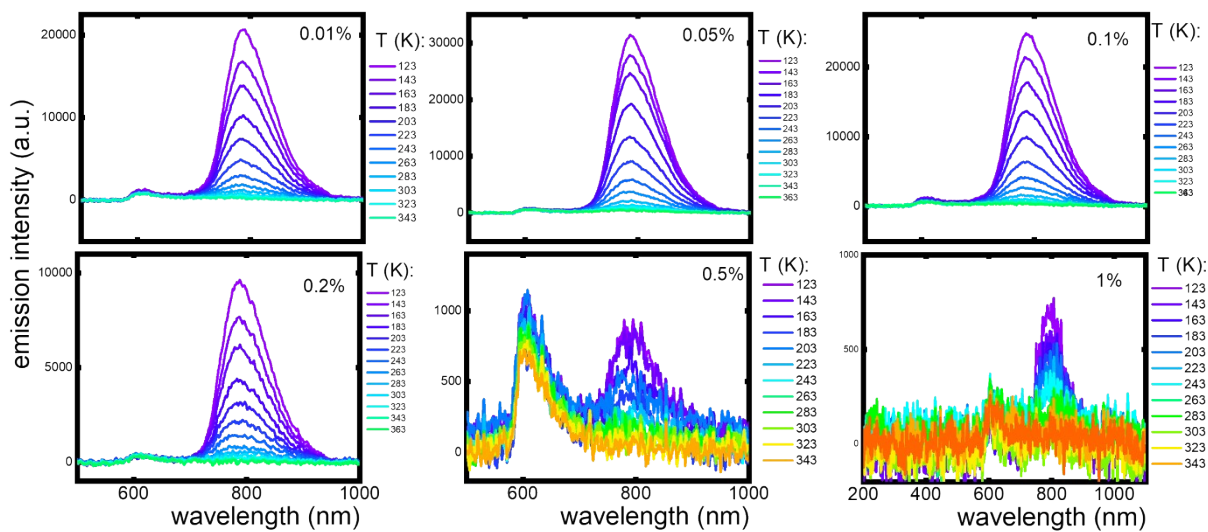


Figure S8. Temperature-dependent emission spectra of CaAl_2O_4 activated with Co^{3+} ions with varying doping fraction (emission spectra obtained upon $\lambda_{\text{exc}}=445$ nm using 590 longpass filter, artificial band at 600 nm results from experimental conditions).

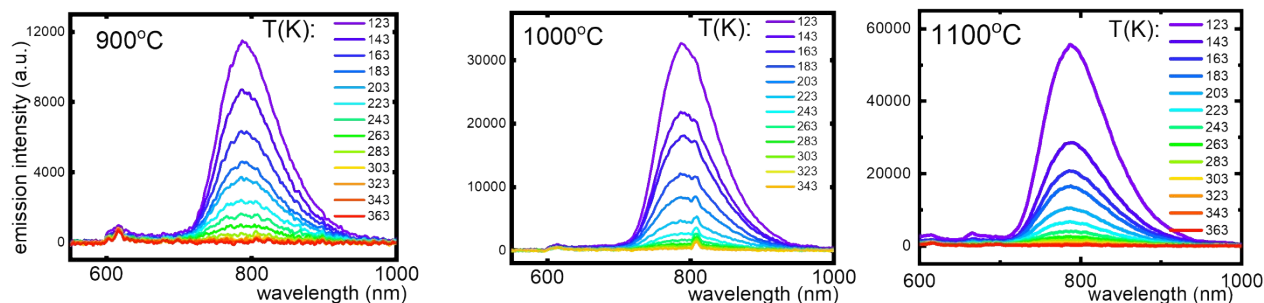


Figure S9. Temperature dependence of $\text{CaAl}_2\text{O}_4: 0.02\% \text{Co}^{3+}$ ions prepared by annealing at varying temperatures (emission spectra obtained upon $\lambda_{\text{exc}}=445$ nm using 590 longpass filter, artificial band at 600 nm and 800 nm results from experimental conditions).

$$\ln\left(\frac{I_0}{I_{em}} - 1\right) = -\Delta E_a \cdot \frac{1}{k_B T} \quad (\text{Eq. S3})$$

where I_0 corresponds to the initial emission intensity (at 123 K) and k_B is Boltzmann constant

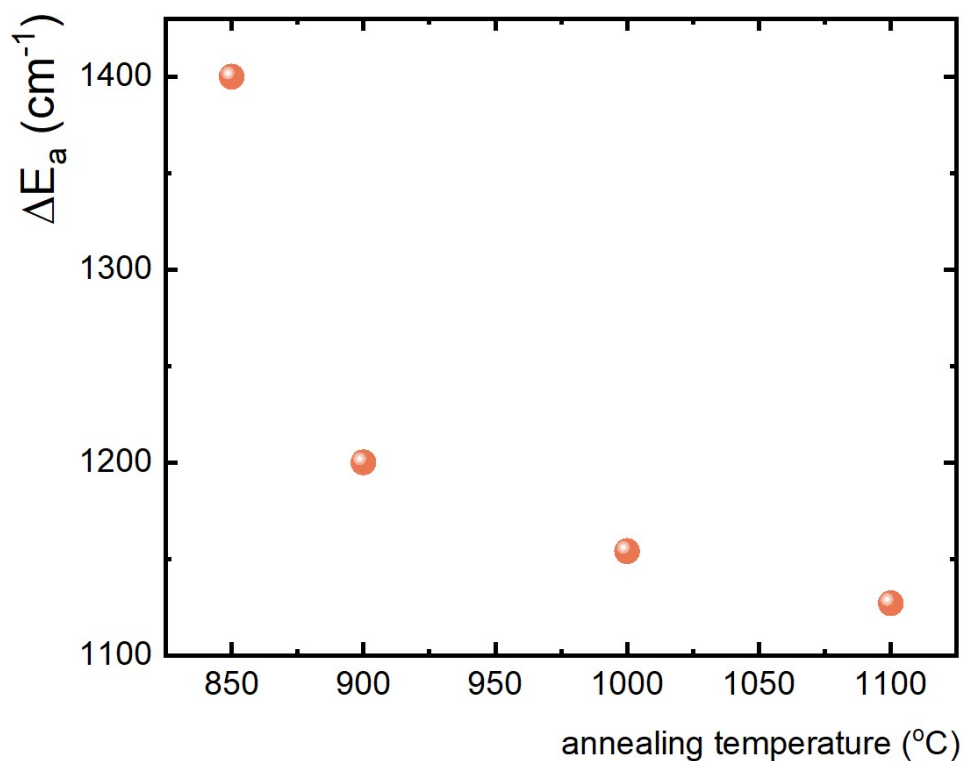


Figure S10. The dependence of the derived thermal crossover activation barrier on the annealing temperature of the prepared $\text{CaAl}_2\text{O}_4: 0.02\% \text{Co}^{3+}$ powders.

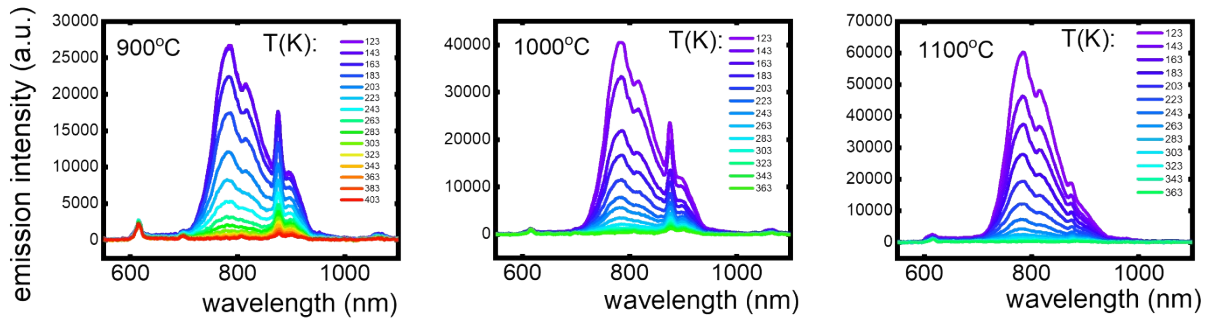


Figure S11. Temperature dependence of the emission spectra of $\text{CaAl}_2\text{O}_4:\text{Co}^{3+}, \text{Nd}^{3+}$ prepared by varying annealing temperatures (emission spectra obtained upon $\lambda_{\text{exc}}=445$ nm using 590 longpass filter, artificial band at 600 nm and 800 nm results from experimental conditions).

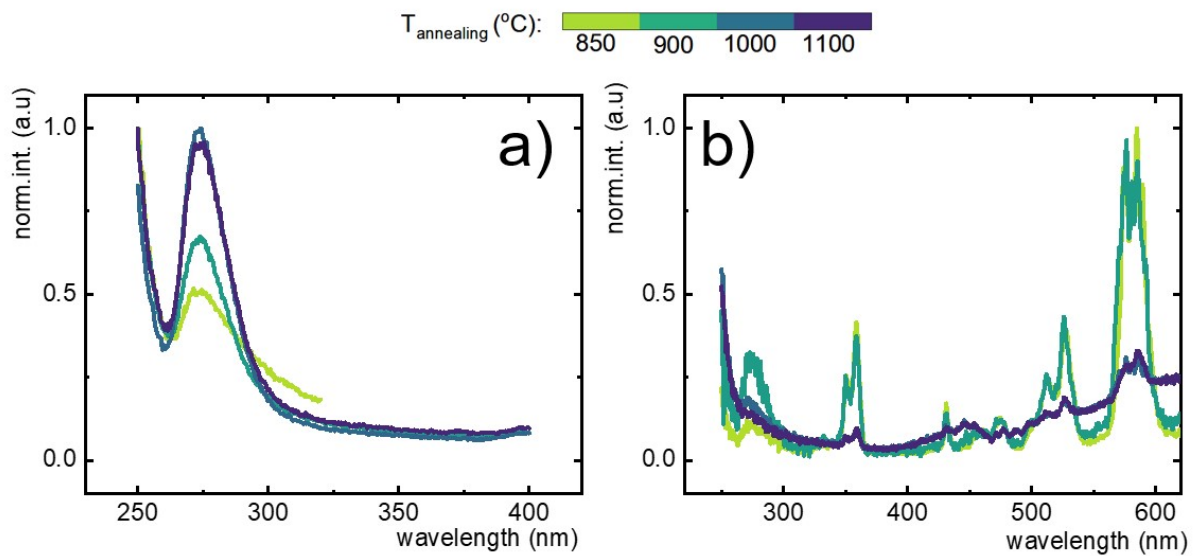


Figure S12. Excitation spectra upon detection of the Nd^{3+} -related emission ($\lambda_{\text{em}}=1064$ nm) of $\text{CaAl}_2\text{O}_4:0.02\%\text{Co}^{3+}, 1\%\text{Nd}^{3+}$ at 123 K and 303 K – a) and – b), respectively.

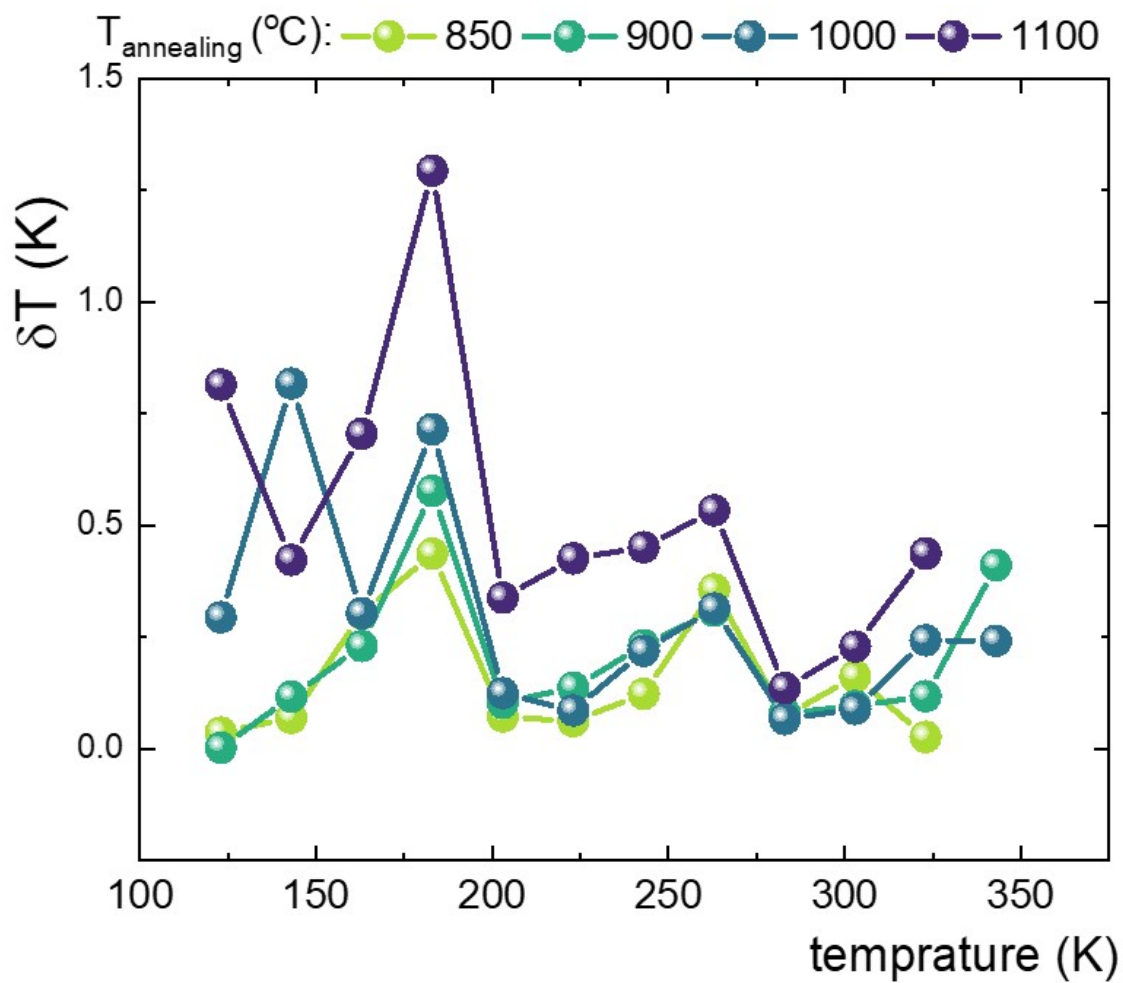


Figure S13. The statistical temperature uncertainty of the conceptualized ratiometric Co^{3+} , Nd^{3+} -co-doped CaAl_2O_4 luminescence thermometers.

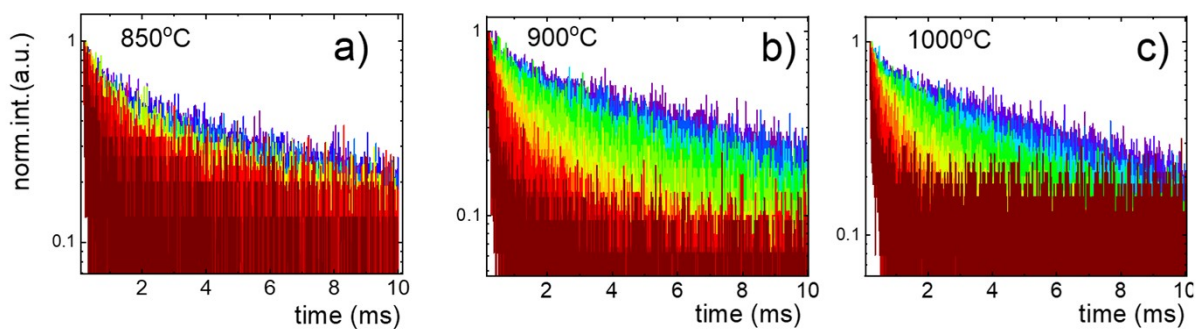


Figure S14. Temperature-dependent time-resolved luminescence of 0.02% Co^{3+} -doped CaAl_2O_4 ($\lambda_{\text{exc}}=266$ nm) $\lambda_{\text{em}}=800$ nm) annealed at 850°C -a), 900°C -b) and 1000°C -c).

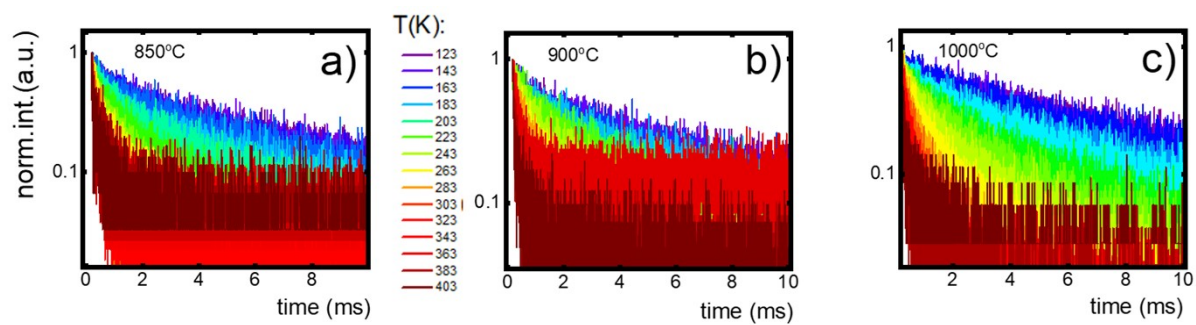


Figure S15. Temperature-dependent time-resolved luminescence of 0.02%Co³⁺, 1%Nd³⁺-co-doped CaAl₂O₄ ($\lambda_{exc}=266$ nm) $\lambda_{em}=800$ nm) annealed at 850°C -a), 900°C -b) and 1000°C -c).