

# Supporting information: Compositional Influence of Local and Long-Range Polarity in the Frustrated Pyrochlore System $\text{Bi}_{2-x}\text{RE}_x\text{Ti}_2\text{O}_7$ (RE = $\text{Y}^{3+}$ , $\text{Ho}^{3+}$ )

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All samples were heated at 600 °C to prevent volatilization of bismuth, then heated for 72 hours to final temperatures between 1050-1200 °C as follows. Bi<sub>1.5</sub>Y<sub>0.5</sub>Ti<sub>2</sub>O<sub>7</sub>: 9hr at 800 °C, 12hr at 1000 °C, 1050°C, 1100 °C, 1150 °C; BiYTi<sub>2</sub>O<sub>7</sub>: 6hr at 800 °C, 12hr at 1000 °C, 1050 °C, 1100 °C; Y<sub>2</sub>Ti<sub>2</sub>O<sub>7</sub>: 12hr at 1000 °C, 12hr at 1100 °C; Bi<sub>1.5</sub>Ho<sub>0.5</sub>Ti<sub>2</sub>O<sub>7</sub>: 6hr at 800 °C, 12hr at 1000 °C, 1050 °C; Bi<sub>0.5</sub>Ho<sub>1.5</sub>Ti<sub>2</sub>O<sub>7</sub>: 6hr at 800 °C, 12hr at 1150 °C; Ho<sub>2</sub>Ti<sub>2</sub>O<sub>7</sub>: 6hr at 800 °C, 12hr at 1150 °C, 1200 °C, 1200 °C.

Table 1: Refined crystallographic data at 300 K for the series Bi<sub>2-x</sub>RE<sub>x</sub>Ti<sub>2</sub>O<sub>7</sub> (RE<sup>3+</sup> = Y, Ho) with the A-site cation in the ideal 16*d* position.

Formula	Bi <sub>1.52(1)</sub> Y <sub>0.48(1)</sub> Ti <sub>2</sub> O <sub>7</sub>	Bi <sub>0.98(2)</sub> Y <sub>1.02(2)</sub> Ti <sub>2</sub> O <sub>7</sub>	Bi <sub>0.41(3)</sub> Y <sub>1.59(3)</sub> Ti <sub>2</sub> O <sub>7</sub>	Y <sub>2</sub> Ti <sub>2</sub> O <sub>7</sub>
Source	neutron synchrotron	neutron synchrotron	synchrotron	neutron synchrotron
Temperature (K)	300	300	300	300
Space Group	<i>Fd</i> $\bar{3}m$ (#227)	<i>Fd</i> $\bar{3}m$ (#227)	<i>Fd</i> $\bar{3}m$ (#227)	<i>Fd</i> $\bar{3}m$ (#227)
a (Å)	10.27890(6)	10.2200(5)	10.1505(2)	10.0982(2)
Volume a (Å <sup>3</sup> )	1086.028(4)	1067.45(20)	1045.85(10)	1029.747(5)
Z	8	8	8	8
Formula weight	567.74	503.52	435.22	385.16
A U <sub>xx</sub> (Å <sup>2</sup> )	0.054(1)	0.035(2)	0.016(1)	0.0112(4)
A U <sub>xy</sub> (Å <sup>2</sup> )	-0.021(1)	-0.014(2)	-0.003(1)	-0.0027(4)
Ti U <sub>iso</sub> (Å <sup>2</sup> )	0.0070(5)	0.0057(1)	0.006(2)	0.005(2)
O (48 <i>f</i> ) U <sub>iso</sub> (Å <sup>2</sup> )	0.014(2)	0.0126(4)	0.005(2)	0.010(2)
O (8 <i>b</i> ) U <sub>iso</sub> (Å <sup>2</sup> )	0.035(2)	0.0167(1)	0.001(2)	0.008(2)
O (48 <i>f</i> ) <i>x</i> (Å <sup>2</sup> )	0.3209(1)	0.3243(2)	0.3257(2)	0.3286(2)
11BM R <sub>wp</sub> (%)	13.4	16.9	16.7	14.5
NOMAD Bank 3 R <sub>wp</sub> (%)	5.9	10.1	–	8.9
NOMAD Bank 4 R <sub>wp</sub> (%)	6.0	6.5	–	8.9
NOMAD Bank 5 R <sub>wp</sub> (%)	5.8	7.3	–	6.1
Total R <sub>wp</sub> (%)	8.7	10.7	16.7	9.1
Formula	Bi <sub>1.49(1)</sub> Ho <sub>0.51(1)</sub> Ti <sub>2</sub> O <sub>7</sub>	Bi <sub>0.52(2)</sub> Ho <sub>1.48(2)</sub> Ti <sub>2</sub> O <sub>7</sub>		Ho <sub>2</sub> Ti <sub>2</sub> O <sub>7</sub>
Source	neutron	neutron		neutron
Temperature (K)	300	300		300
Space Group	<i>Fd</i> $\bar{3}m$ (#227)	<i>Fd</i> $\bar{3}m$ (#227)		<i>Fd</i> $\bar{3}m$ (#227)
a (Å)	10.282(3)	10.1561(7)		10.0852(2)
Volume a (Å <sup>3</sup> )	1087.1(4)	1047.6(3)		1025.8(1)
Z	8	8		8
Formula weight	603.73	560.56		537.65
A U <sub>xx</sub> (Å <sup>2</sup> )	0.061(1)	0.021(4)		0.010(2)
A U <sub>xy</sub> (Å <sup>2</sup> )	-0.020(1)	-0.006(3)		-0.003(2)
Ti U <sub>iso</sub> (Å <sup>2</sup> )	0.006(4)	0.006(2)		0.004(2)
O (48 <i>f</i> ) U <sub>iso</sub> (Å <sup>2</sup> )	0.017(2)	0.012(5)		0.0077(5)
O (8 <i>b</i> ) U <sub>iso</sub> (Å <sup>2</sup> )	0.033(2)	0.011(2)		0.0055(6)
O (48 <i>f</i> ) <i>x</i> (Å <sup>2</sup> )	0.32193(2)	0.3268(4)		0.3286(4)
NOMAD Bank 3 R <sub>wp</sub> (%)	8.5	5.5		6.7
NOMAD Bank 4 R <sub>wp</sub> (%)	9.3	6.4		8.3
NOMAD Bank 5 R <sub>wp</sub> (%)	8.2	6.0		6.4
Total R <sub>wp</sub> (%)	8.5	5.8		6.9

Table 2: Refined crystallographic data at 2 K for the series  $\text{Bi}_{2-x}\text{RE}_x\text{Ti}_2\text{O}_7$  ( $\text{RE}^{3+} = \text{Y}, \text{Ho}$ ) with the *A*-site cation in the ideal  $16d$  position.

Formula	$\text{Bi}_{1.52(1)}\text{Y}_{0.48(1)}\text{Ti}_2\text{O}_7$	$\text{Bi}_{0.98(2)}\text{Y}_{1.02(2)}\text{Ti}_2\text{O}_7$	$\text{Y}_2\text{Ti}_2\text{O}_7$
<b>Source</b>	neutron	neutron	neutron
<b>Temperature (K)</b>	2	2	2
<b>Space Group</b>	$Fd\bar{3}m$ (#227)	$Fd\bar{3}m$ (#227)	$Fd\bar{3}m$ (#227)
<b>a (Å)</b>	10.2690(3)	10.2196(5)	10.0671(2)
<b>Volume a (Å<sup>3</sup>)</b>	1082.89(4)	1067.30(20)	1020.27(5)
<b>Z</b>	8	8	8
<b>Formula weight</b>	567.74	503.52	385.16
<b>A U<sub>xx</sub> (Å<sup>2</sup>)</b>	0.068(1)	0.043(2)	0.006(4)
<b>A U<sub>xy</sub> (Å<sup>2</sup>)</b>	-0.030(1)	-0.018(2)	-0.0003(4)
<b>Ti U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.002(5)	0.0010(1)	0.003(2)
<b>O (48f) U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.013(2)	0.014(4)	0.006(2)
<b>O (8b) U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.032(2)	0.018(1)	0.005(2)
<b>O (48f) x (Å<sup>2</sup>)</b>	0.3209(1)	0.3246(2)	0.3290(2)
<b>NOMAD Bank 3 R<sub>wp</sub> (%)</b>	6.5	8.0	9.8
<b>NOMAD Bank 4 R<sub>wp</sub> (%)</b>	7.9	9.7	10.8
<b>NOMAD Bank 5 R<sub>wp</sub> (%)</b>	6.5	9.0	8.1
<b>Total R<sub>wp</sub> (%)</b>	6.8	8.7	9.3

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Formula	$\text{Bi}_{1.49(1)}\text{Ho}_{0.51(1)}\text{Ti}_2\text{O}_7$	$\text{Ho}_2\text{Ti}_2\text{O}_7$
<b>Source</b>	neutron	neutron
<b>Temperature (K)</b>	2	2
<b>Space Group</b>	$Fd\bar{3}m$ (#227)	$Fd\bar{3}m$ (#227)
<b>a (Å)</b>	10.2730(3)	10.0751(2)
<b>Volume a (Å<sup>3</sup>)</b>	1084.1(4)	1022.80(1)
<b>Z</b>	8	8
<b>Formula weight</b>	603.73	537.65
<b>A U<sub>xx</sub> (Å<sup>2</sup>)</b>	0.066(1)	0.010(2)
<b>A U<sub>xy</sub> (Å<sup>2</sup>)</b>	-0.028(1)	-0.003(2)
<b>Ti U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.004(4)	0.003(2)
<b>O (48f) U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.012(2)	0.009(5)
<b>O (8b) U<sub>iso</sub> (Å<sup>2</sup>)</b>	0.029(2)	0.0072(6)
<b>O (48f) x (Å<sup>2</sup>)</b>	0.32193(2)	0.3288(4)
<b>NOMAD Bank 3 R<sub>wp</sub> (%)</b>	8.7	6.0
<b>NOMAD Bank 4 R<sub>wp</sub> (%)</b>	8.6	9.4
<b>NOMAD Bank 5 R<sub>wp</sub> (%)</b>	10.8	7.0
<b>Total R<sub>wp</sub> (%)</b>	8.8	8.6

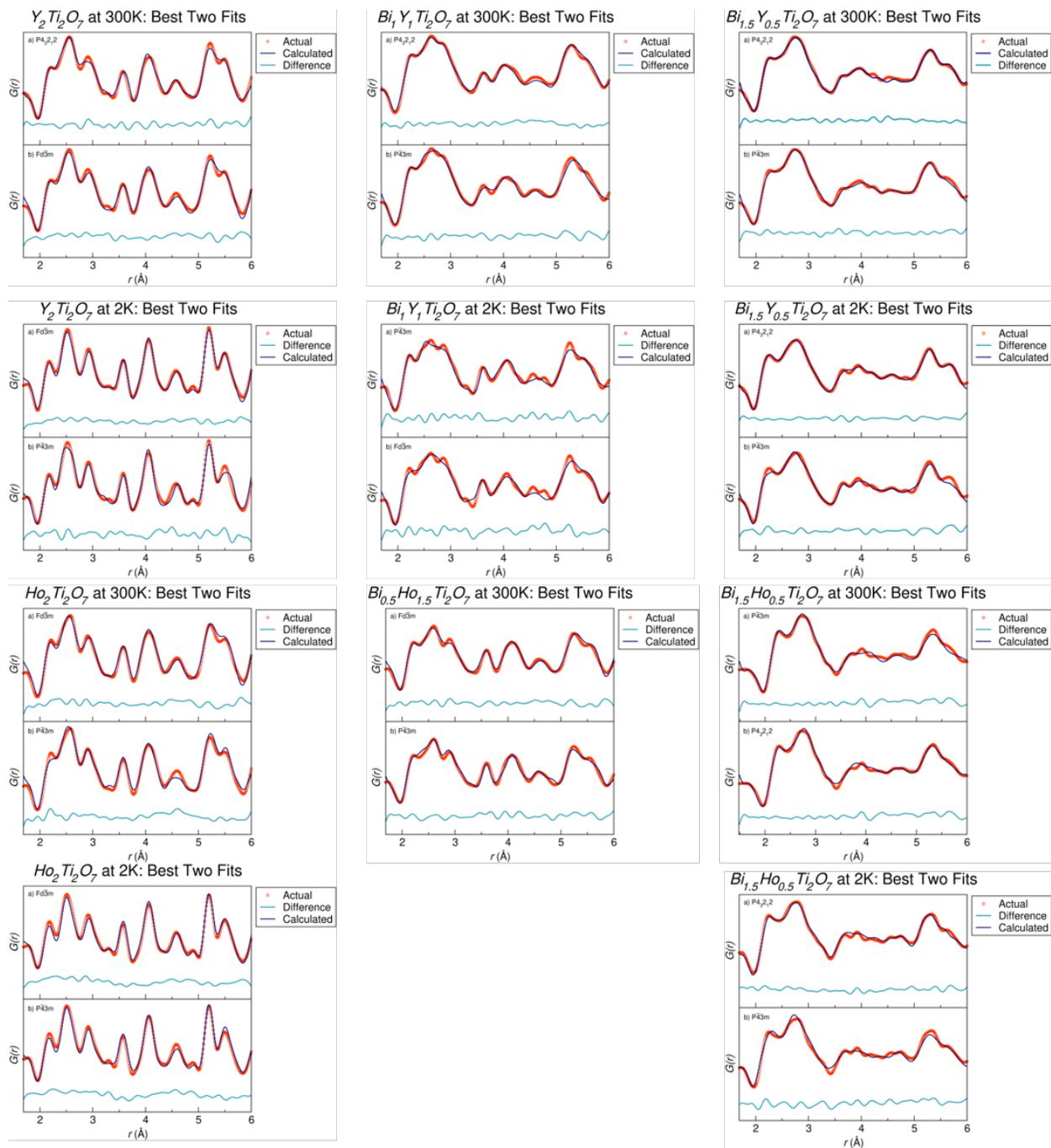


Figure 1: Fits of the neutron PDF (NOMAD, SNS) data from 1.5-6.0 Å against the lowest  $R_w$  (best fitting) structural models for each sample.

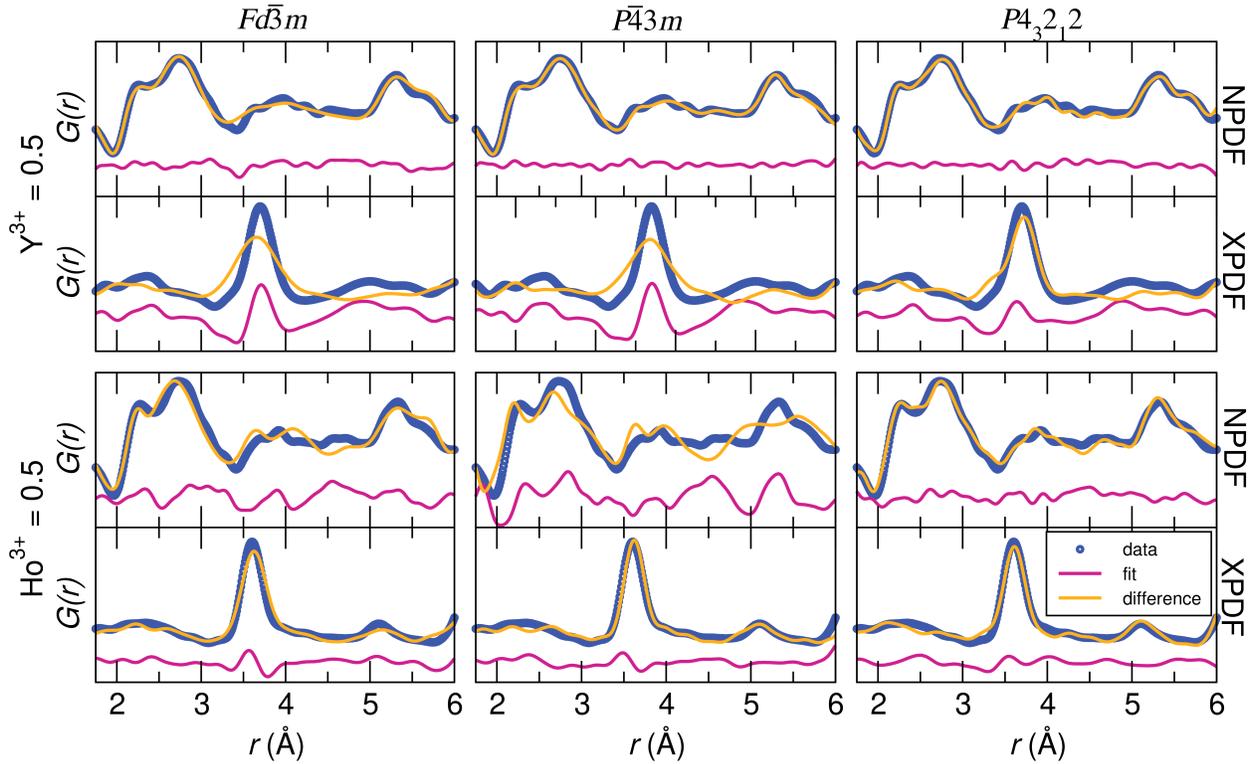


Figure 2: Fits (1.7-6.0 Å) from the joint analysis of the x-ray PDF (XPfDF, PDF, NSLS-II) and neutron PDF (NPfDF, NOMAD, SNS) data for  $x = 0.5$  in  $\text{Bi}_{2-x}\text{RE}_x\text{Ti}_2\text{O}_7$  ( $\text{RE} = \text{Y}^{3+}$ ,  $\text{Ho}^{3+}$ ) samples against the three candidate structural models.

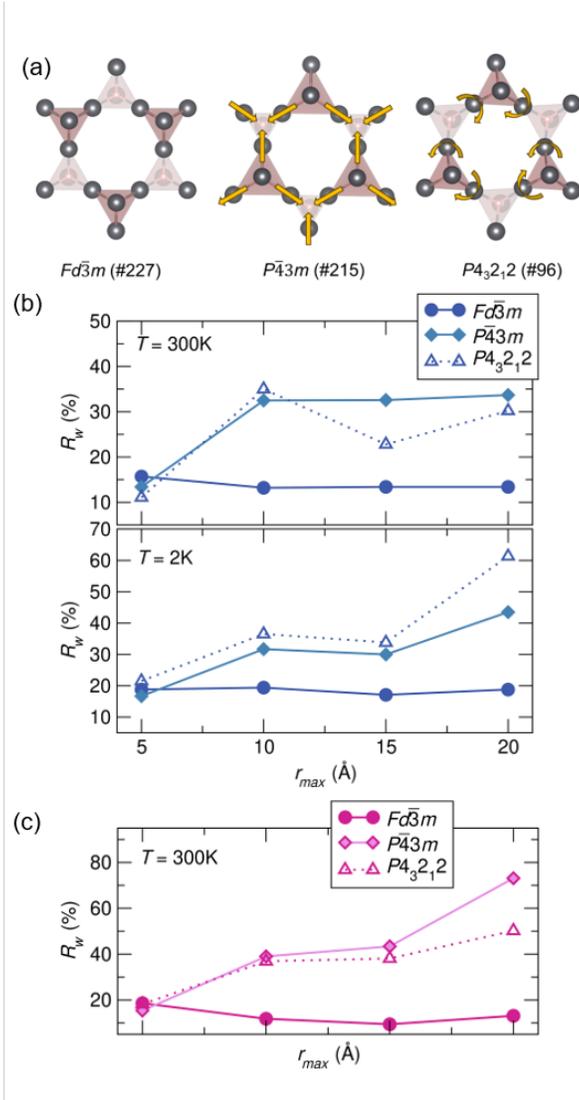


Figure 3: (a) Distortions of the  $A$ -site cations in the various structural models.  $R_w$  as a function of box-car  $r_{max}$  for (b)  $\text{Bi}_{1.0}\text{Y}_{1.0}\text{Ti}_2\text{O}_7$  at 300 and 2 K and (c)  $\text{Bi}_{0.5}\text{Ho}_{0.5}\text{Ti}_2\text{O}_7$  at 300 K (NPDF, NOMAD, SNS).

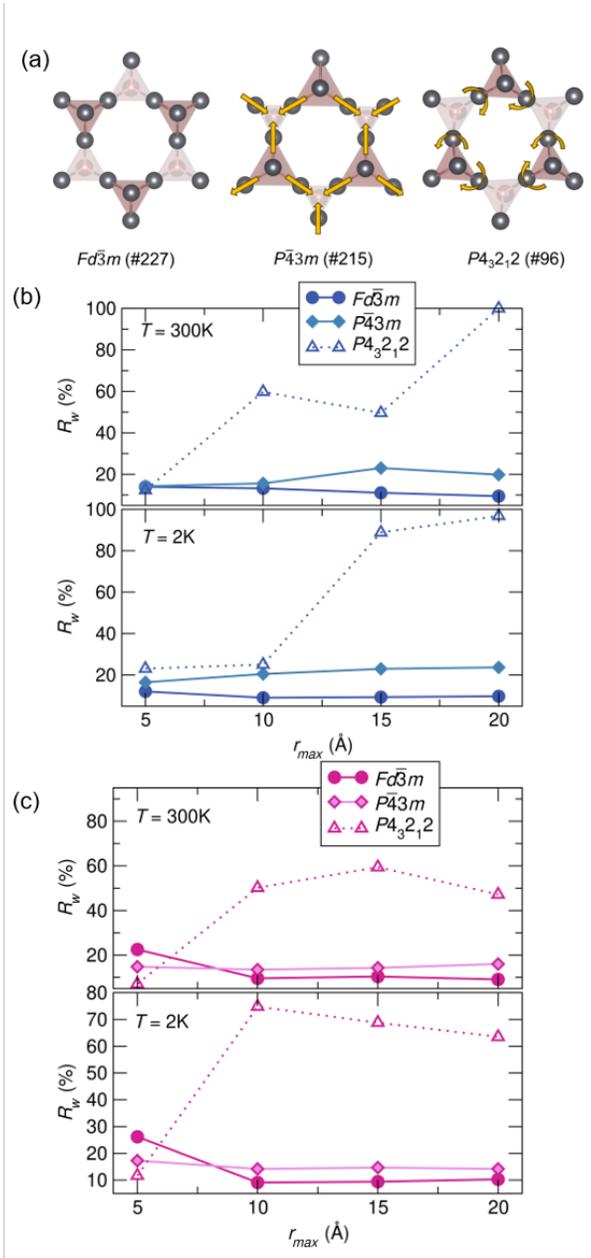


Figure 4: (a) Distortions of the A-site cations in the various structural models.  $R_w$  as a function of box-car  $r_{max}$  for (b)  $Y_2Ti_2O_7$  at 300 and 2K and (c)  $Ho_2Ti_2O_7$  at 300 and 2K (NPDF, NOMAD, SNS).

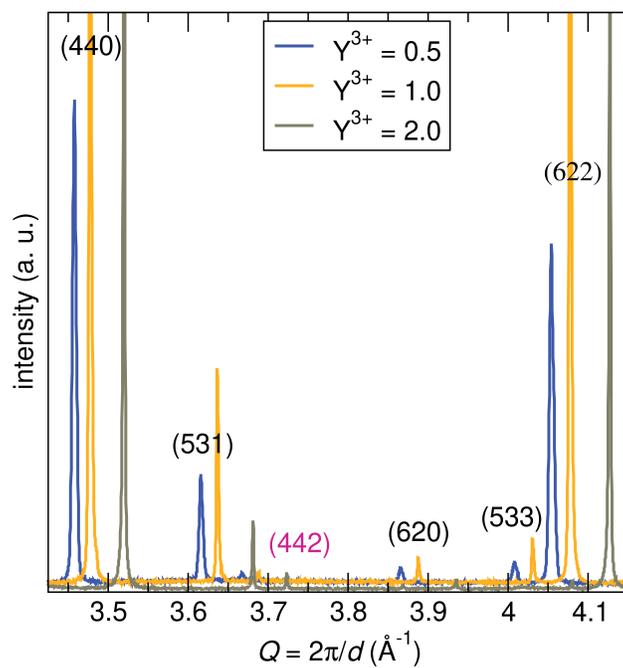


Figure 5: Section of the X-ray diffraction data (11BM, APS) showing the weak (442) reflection, hkl highlighted in pink.