

Supporting Information

Highly Efficient Blue Emitting One Dimensional Lead-Free Nanocrystals

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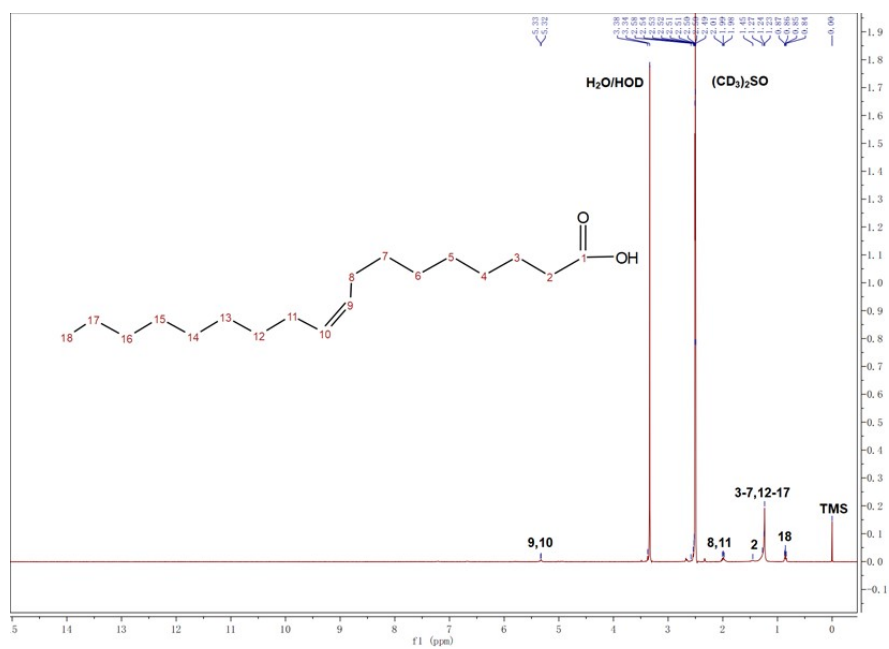


Figure S1. ^1H NMR spectra of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{l}_2$ NCs dissolved in $(\text{CD}_3)_2\text{SO}$.

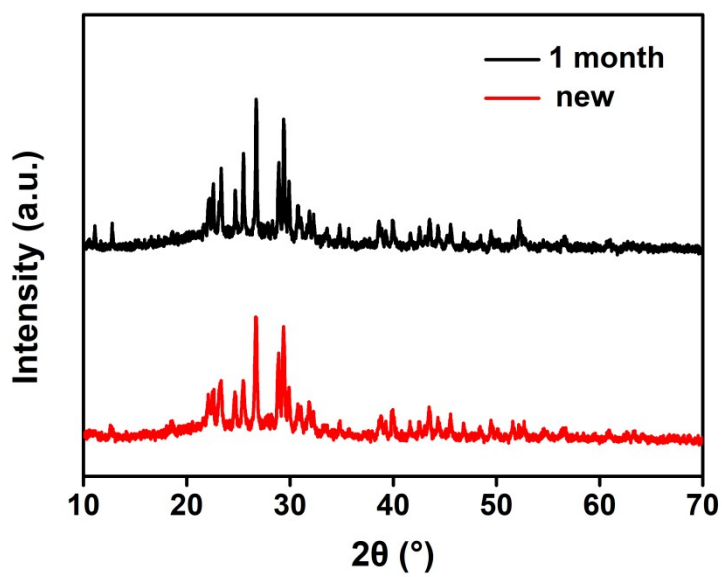


Figure S2. XRD patterns of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{l}_2$ powders stored in the air for 0 day and 1 month.

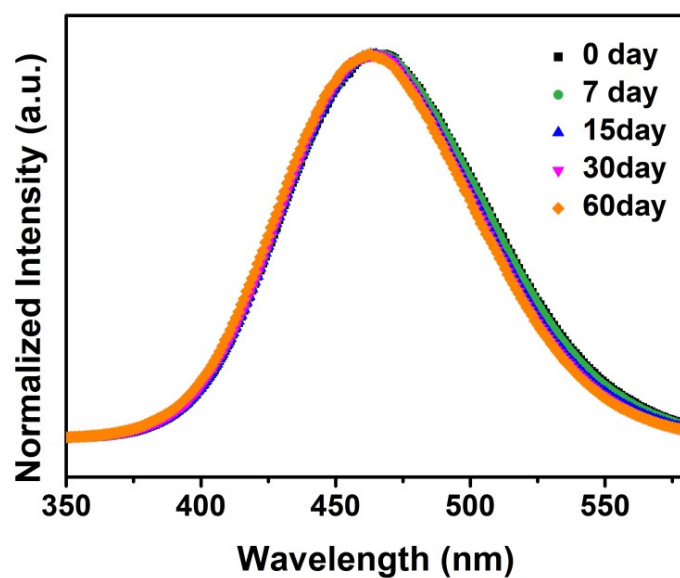


Figure S3. PL spectra of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ colloidal NCs stored in ambient condition for different time.

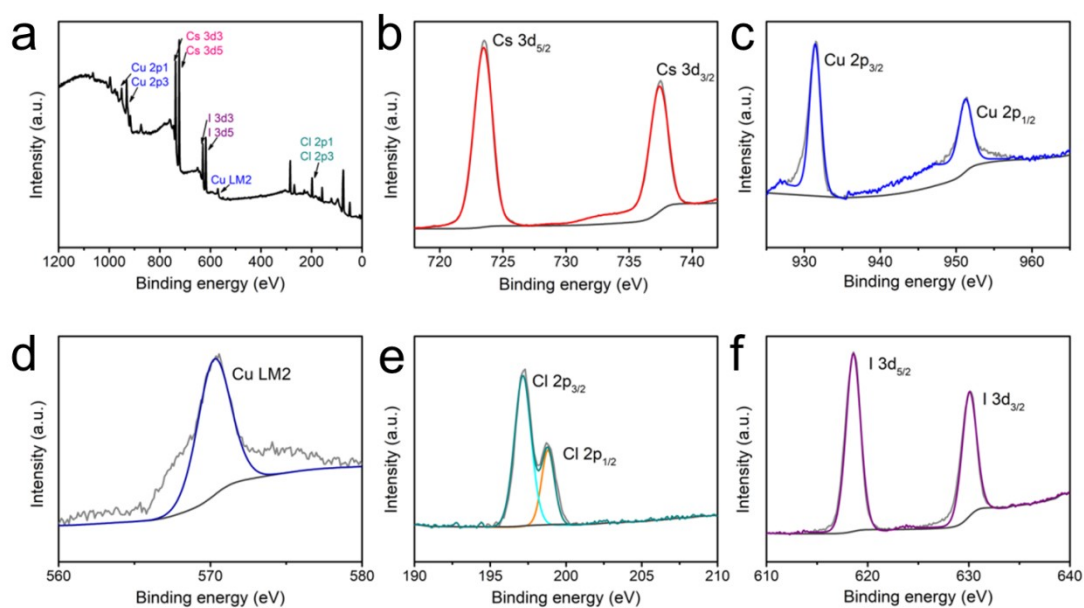


Figure S4. (a) X-ray photoelectron spectroscopic (XPS) analysis of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs powders and the high-resolution spectra of (b) Cs 3d; (c) Cu 2p; (d) Cu LM2; (e) Cl 2p; (f) I 3d.

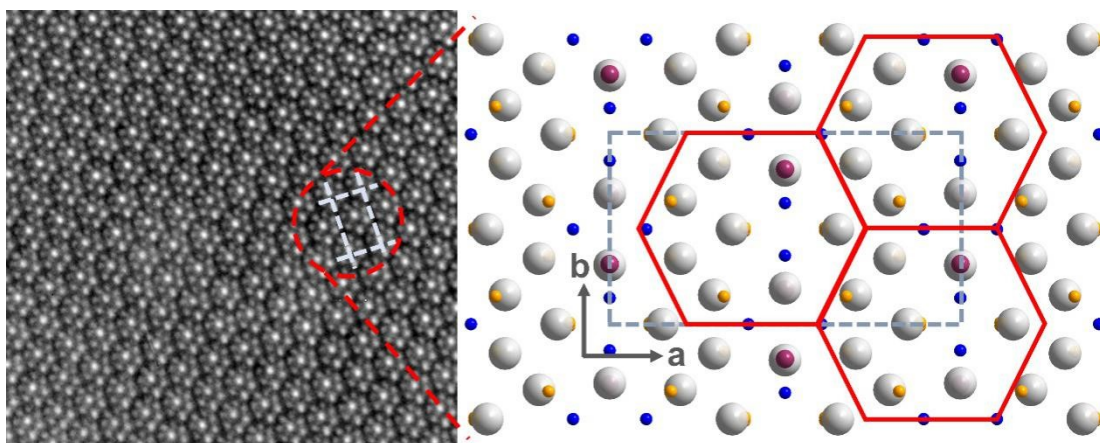


Figure S5. The transmission electron microscopy image of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs (right: projected view of (001) lattice planes; Gray, blue, yellow and purple spheres represent Cs, Cu, Cl and I atoms, respectively).

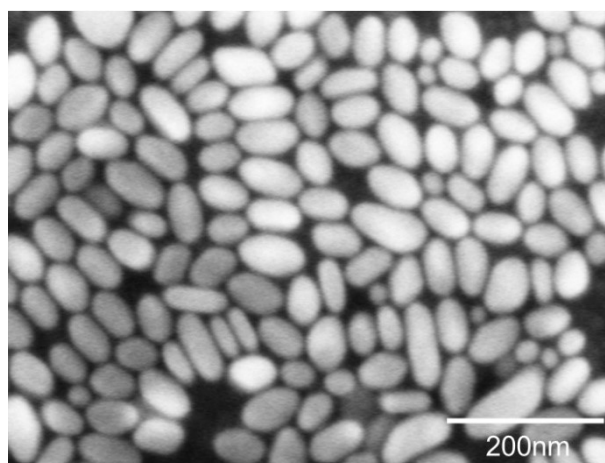


Figure S6. The scanning electron microscopy image of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs coated on ITO substrate.

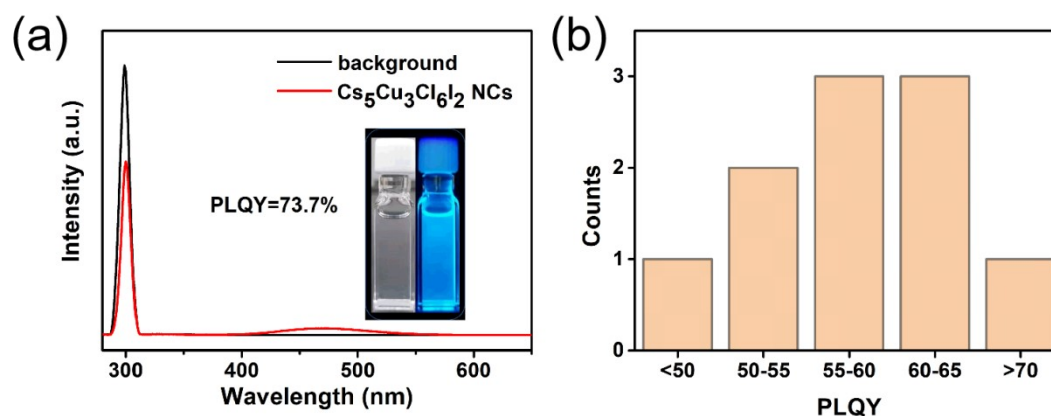


Figure S7. (a) PLQY and corresponding sample with and without excitation by UV light of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs. (b) The value of PLQY distribution histogram of as-prepared samples.

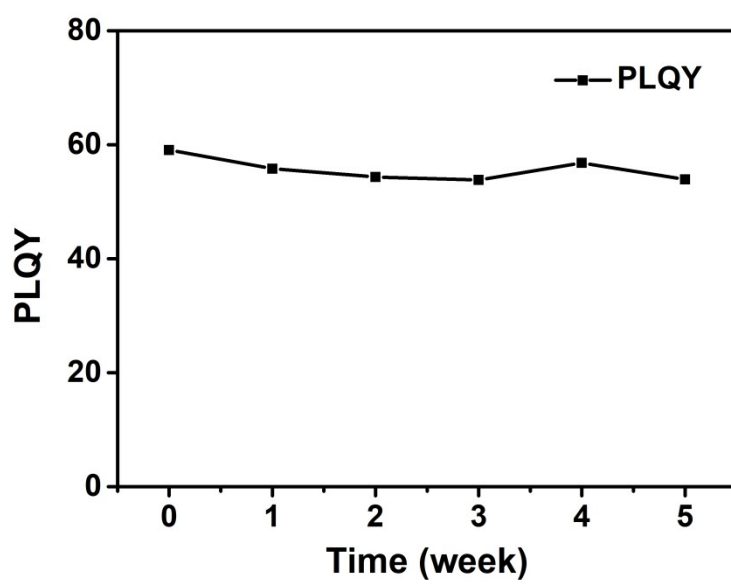


Figure S8. Plot of PLQYs over time for $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs.

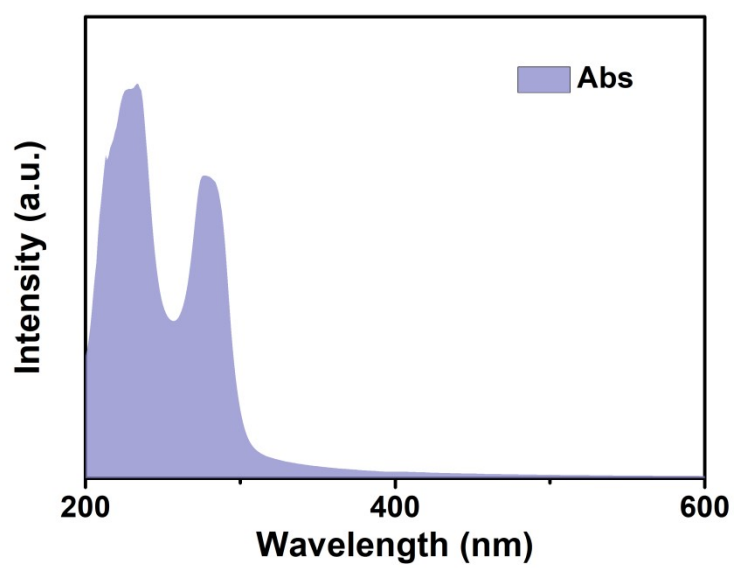


Figure S9. The UV/Vis absorption spectra of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs.

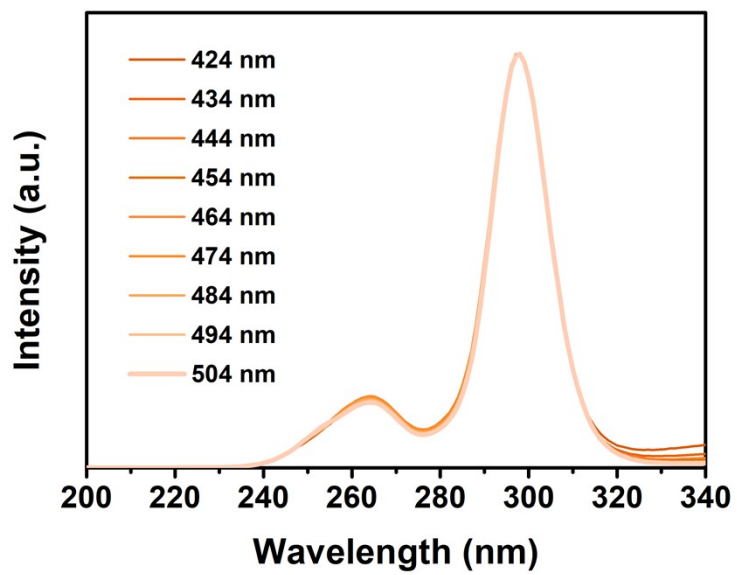


Figure S10. PL excitation (PLE) spectra measured at different emission wavelength.

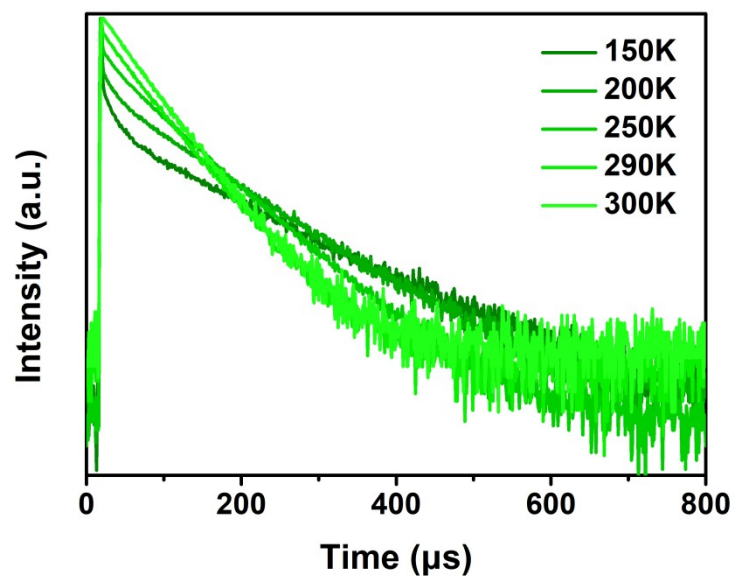


Figure S11. Time-resolved PL decay curves of $\text{Cs}_5\text{Cu}_3\text{Cl}_6\text{I}_2$ NCs.

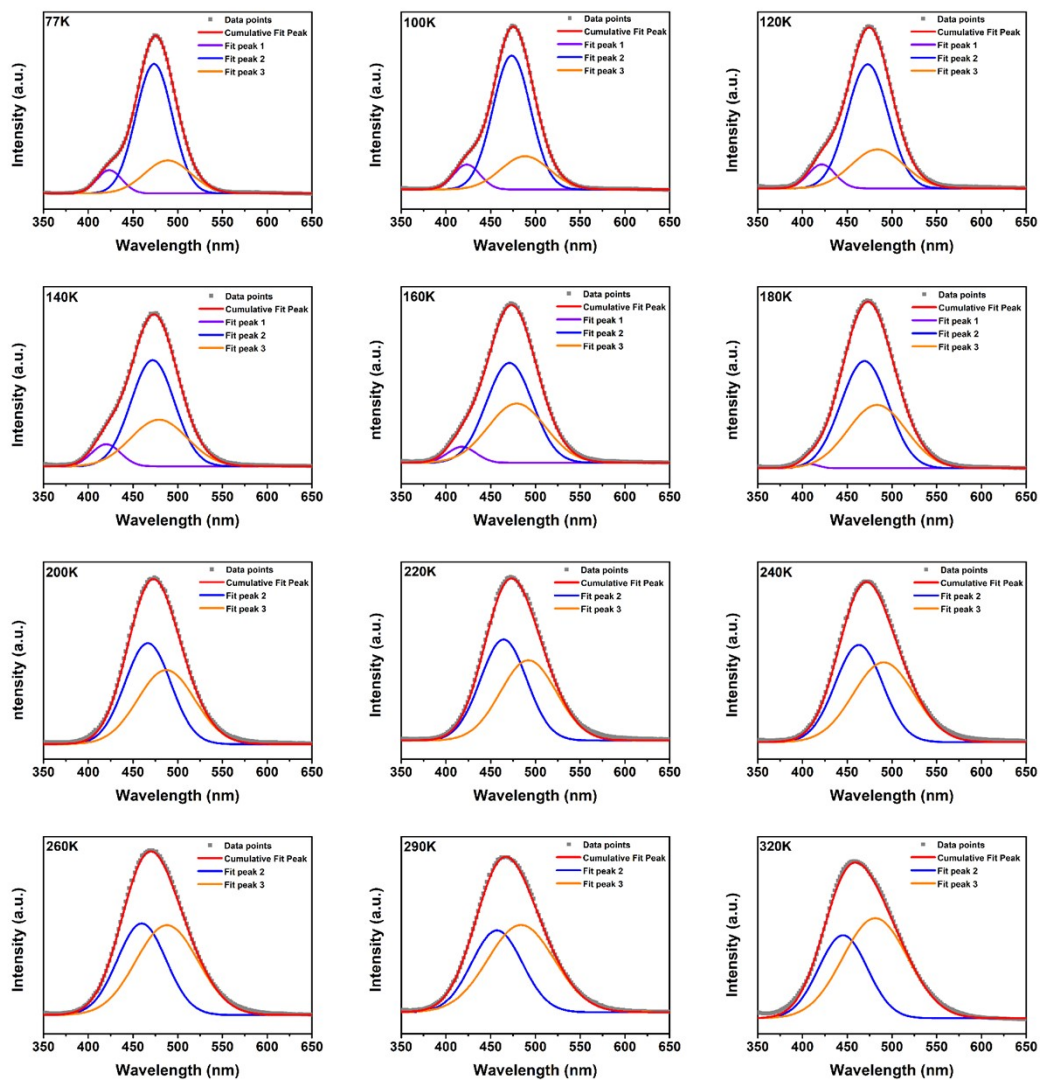


Figure S12. The fitted PL spectra by an exponentially modified Gaussian by multiple Gaussian distributions at different temperature.

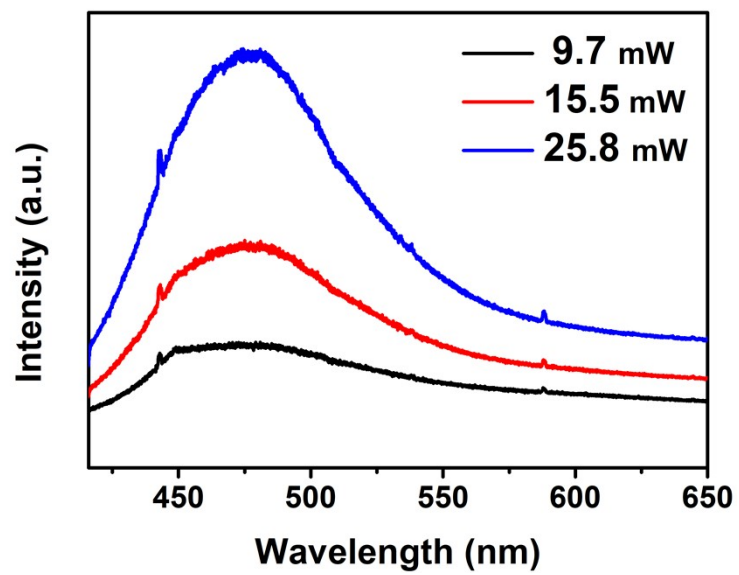
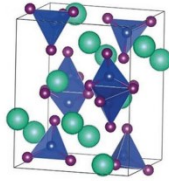
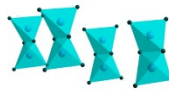
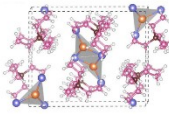
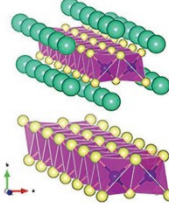
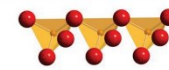
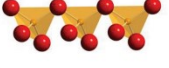
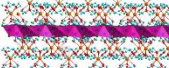


Figure S13. The PL spectra of Cs₅Cu₃Cl₆I₂ NCs at different power density.

Table S1. Structural and optical characteristics of copper halide perovskites-like materials.

Compound	Dimension	PL/ nm	Stokes shift/ nm	Polyhedral connection mode	Structure	Reference
$\text{Cs}_3\text{Cu}_2\text{I}_5$	0D	445	155	edge-sharing		1
$(\text{MA})_4\text{Cu}_2\text{Br}_6$		524	222			2
$(\text{C}_{16}\text{H}_{36}\text{N})\text{CuI}_2$		463	183			3
CsCu_2I_3	1D	568	240	edge-sharing		4
Rb_2CuCl_3		395	93	corner-sharing		5
Rb_2CuBr_3		385	85	corner-sharing		5
$[\text{KC}_2]_2[\text{Cu}_4\text{I}_6]$		545	145	edge-sharing and face-sharing		6

References

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