Electronic Supplementary Material (ESI) for Journal of Materials Chemistry C. This journal is © The Royal Society of Chemistry 2022

## **Electronic Supplementary Information**

## Single band ratiometric luminescence thermometry based on Pr<sup>3+</sup>

## doped oxides containing charge transfer states

Jiawen Wang, ab Junyi Li, b Ruoshan Lei, \*b Shilong Zhao b and Shiqing Xu \*b

<sup>a</sup> College of Materials and Chemistry, China Jiliang University, Hangzhou, 310018, China

<sup>b</sup> Institute of Optoelectronic Materials and Devices, China Jiliang University, Hangzhou, 310018,

China



**Fig. S1.** Diffuse reflectance spectrum of the Pr<sup>3+</sup>: GZT phosphor.

Supplementary Note 1. The calculation of IVCT energy. According to previous studies, the position of the IVCT band will affect the electron distribution between the  ${}^{3}P_{0}$  and  ${}^{1}D_{2}$  energy levels. Boutinaud et al. have proposed an empirical equation to evaluate the energy of  $Pr^{3+}-Ti^{4+}$  IVCT as follows:<sup>1,2</sup>

$$IVCT(cm^{-1}) = 58800 - 49800 \left[ \frac{\chi(Ti^{4+})}{d(\Pr^{3+} - Ti^{4+})} \right]$$
(S1)

where  $\chi(Ti^{4+})$  is the optical electronegativity of Ti<sup>4+</sup> (2.05), and  $d(Pr^{3+}-Ti^{4+})$  is the average distance between  $Pr^{3+}$  and Ti<sup>4+</sup>. For  $Pr^{3+}$ : GZT, the mean distance between  $Gd^{3+}$ -Ti<sup>4+</sup> is ~3.37 Å based on the crystal structure data (ICSD-251934).<sup>3</sup> Accordingly, the calculated IVCT energy is ~351 nm (28520 cm<sup>-1</sup>), which is consistent with the experimental data detected in the excitation spectrum (340 nm, 29412 cm<sup>-1</sup>).



Fig. S2. (a) Experimentally measured and fitted plot of  $LIR=I_{617}/I_{491}$  versus temperature. (b) The corresponding  $S_a$ 

and S<sub>r</sub> as a function of temperature.



**Fig. S3.** Absolute error ( $\Delta T$ ) as a function of temperature for  $LIR_{_{3P_0 \rightarrow ^{3}H_4}}(I_{EX:f-f}/I_{EX:IVCT})$  data.



Fig. S4. Schematic diagram of the present SBR method in Pr<sup>3+</sup>: GZT.



**Fig. S5.** (a) XRD patterns of Gd<sub>2</sub>MgTiO<sub>6</sub> and Pr<sup>3+</sup>: Gd<sub>2</sub>MgTiO<sub>6</sub> samples. Due to the lack of structural information of Gd<sub>2</sub>MgTiO<sub>6</sub> in the crystal database, the standard data of Dy<sub>2</sub>MgTiO<sub>6</sub> with an analogous structure (mp-1213085) is used as a reference. (b) XRD patterns of LiLaMgWO<sub>6</sub> and Pr<sup>3+</sup>: LiLaMgWO<sub>6</sub> samples, and the standard data of LiLaMgWO<sub>6</sub> (PDF-370243) is used as a reference. (c) XRD patterns of La<sub>2</sub>ZnTiO<sub>6</sub> and Pr<sup>3+</sup>: La<sub>2</sub>ZnTiO<sub>6</sub> samples with the standard data of La<sub>2</sub>ZnTiO<sub>6</sub> (ICSD-172755) as a reference.



Fig. S6. (a) Temperature-dependent PL spectra of  $Pr^{3+}$ :  $Gd_2MgTiO_6$  upon the excitations of  $Pr^{3+}-Ti^{4+}$  IVCT ( $\lambda_{ex}$ =329 nm) and  $Pr^{3+}$ :  ${}^{3}H_4 \rightarrow {}^{3}P_2$  transition ( $\lambda_{ex}$ =451 nm). (b) Temperature-variational intensities of the 491 nm emission ( $Pr^{3+}$ :  ${}^{3}P_0 \rightarrow {}^{3}H_4$  transition) under two different excitation lines ( $\lambda_{ex}$ =329 and 451 nm).



Fig. S7. (a) Temperature-dependent PL spectra of  $Pr^{3+}$ : LiLaMgWO<sub>6</sub> upon the excitations of  $Pr^{3+}-W^{6+}$  IVCT ( $\lambda_{ex}$ =322 nm) and  $Pr^{3+}$ :  ${}^{3}H_{4}\rightarrow {}^{3}P_{2}$  transition ( $\lambda_{ex}$ =449 nm). (b) Temperature-variational intensities of the 491 nm emission ( $Pr^{3+}$ :  ${}^{3}P_{0}\rightarrow {}^{3}H_{4}$  transition) under two different excitation lines ( $\lambda_{ex}$ =322 and 449 nm).



Fig. S8. (a) Temperature-dependent PL spectra of  $Pr^{3+}$ : La<sub>2</sub>ZnTiO<sub>6</sub> upon the excitations of the  $Pr^{3+}$ -Ti<sup>4+</sup> IVCT state ( $\lambda_{ex}$ =321 nm) and  $Pr^{3+}$ :  ${}^{3}H_{4} \rightarrow {}^{3}P_{2}$  transition ( $\lambda_{ex}$ =449 nm). (b) Temperature-variational intensities of the 489 nm emission ( $Pr^{3+}$ :  ${}^{3}P_{0} \rightarrow {}^{3}H_{4}$  transition) under two different excitation lines ( $\lambda_{ex}$ =321 and 449 nm).

## References

- Y. Gao, F. Huang, H. Lin, J. Xu and Y. Wang, Sens. Actuators B Chem., 2017, 243, 137-143.
- 2 R. Shi, L. Lin, P. Dorenbos and H. Liang, J. Mater. Chem. C, 2017, 5, 10737-10745.
- 3 N. Das, M. A. Nath, G. S. Thakur, M. Thirumal and A. K. Ganguli, J. Solid State Chem., 2015, 229, 97-102.