

Electronic Supplementary Information (ESI) Materials:

Half-metallic double perovskite oxides: recent developments and future perspectives

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Table S1 The fundamental structural and physical properties, synthesized methods, structural characterized methods and applications of half-metallic double perovskite oxides as well as the device's performance based on such family of materials

HM DP oxides	Description			Ref.	
Basic fundamental properties	Structure	Goldschmidt's tolerance factor (t)		S1	
		$t = \frac{\langle r_A \rangle + r_O}{\sqrt{2} \left(\frac{\langle r_B \rangle + \langle r_B'' \rangle}{2} + r_O \right)}$	$1.05 < t$	Hexagonal	S2, S3, S4
			$1.00 < t < 1.05$	Cubic	
			$0.97 < t < 1.00$	Tetragonal	
	$t < 0.97$		Orthorhombic (Pbnm) / monoclinic (P21/n)		
Physical properties	High Curie temperature (T_C), magnetization (M), magnetoresistance (MR), and high electron polarizability (P)		S5-S8		
Synthesis of HM DP oxides	0D HM DP oxides	Solid-state reaction (SSR) method combining with high-energy ball milling process		S9	
		Molten salt synthesis (MSS) method		S10	
		Sol-gel process		S11	
		Coprecipitation method		S12	
		Combustion method		S13	
		Hydrothermal method		S14	
	1D HM DP oxides	Hydrothermal method		S15	
	2D HM DP oxides	Pulsed laser deposition (PLD)		S16	
		Magnetron sputtering		S17	
		Chemical solution deposition (CSD)		S18	
	3D HM DP oxides	Solid-state reaction (SSR) method Spark plasma sintering method Floating-zone method		S19 S20 S21-S23	
Microstructural characterization techniques	X-ray diffraction (XRD)/neutron diffraction (ND) methods			S24	
	Scanning/transmission electron microscopy (S/TEM)			S25	
	Energy dispersive X-ray spectroscopy (EDS)			S26	
	Electron energy loss spectroscopy (EELS)			S27	
	X-ray photoelectron spectroscopy (XPS)			S28	
	X-ray magnetic circular dichroism (XMCD)			S29	
	Mössbauer spectroscopy			S30	
Applications	Typical devices based on HM DP oxides	Each device's performance		Ref.	
	Magnetic tunnel junctions (MTJs)	A SFMO/STO/Co tunnel nanojunction measured at 4 K and applied bias voltage of 10 mV; TMR ~ 50%, negative spin polarization (P) of SFMO film, $P \sim -85\%$, indicating the electron tunneling from SFMO through STO.		S31	
	Spin filtering devices (SFDs) based on tunnel junctions	(a) Metallic/magnetic insulator/magnetic insulator/metallic (M-MI-MI-M) junctions or (b) ferromagnetic metal/nonmagnetic metal/magnetic insulator/ferromagnetic metal (FM-NM-MIFM) junctions; SFDs can operate at room temperature and exhibit magnetic field sensitivity much larger than conventional tunnel junctions conventional MTJs; TMR as high as 10^5 predicted in a metal/spin filter 1/spin filter 2/metal structure (double spin-filter junction)		S32	

Field effect transistors (FETs)	A spin MOSFET consisting of a MOS structure and half-metallic-ferromagnet (HMF) contacts for the source and drain; exhibiting high (low) current drive capability in the parallel (antiparallel) magnetization; extremely large magnetocurrent ratio γ_{MC} (defined as $\frac{(I_D^P - I_D^{AP})}{I_D^{AP}}$) ($\gamma_{MC} > 1000\%$ @ $V_{DS} < 1.0$ V)	S33
Josephson junctions	HM ferromagnet-superconductor bilayers of $\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3/\text{Pr}_{1.85}\text{Ce}_{0.15}\text{CuO}_4$ (LCMO/PCCO) grown by PLD on NdGaO_3 (001) substrates; long-ranged S-F proximity effects observed in the LCMO film (up to a distance much larger than for Cooper pairs in a singlet spin state to exist), which can be attributed to induced triplet-pairing correlations mediated by spin-active interfacial regions.	S34
Applications in the field of electrocatalysts	Ir-based Ba_2MIrO_6 (e.g., M = Y, La, Ce, Pr, Nd and Tb) DP oxides synthesized by standard solid-state reactions, exhibiting tunable performance towards acidic OER. $\text{Ba}_2\text{PrIrO}_6$ is the best acidic OER electrocatalyst, surpassing the benchmark IrO_2 .	S35
Solid oxide fuel cells (SOFC)	La^{3+} -doped $\text{Sr}_2\text{FeMoO}_{6-\delta}$ (SLFM with $0 \leq x \leq 1$) double perovskites used as anode materials for solid oxide fuel cells; SOFCs with SLFM($x=0.2$) as anode have demonstrated excellent and stable performance under direct CH_4 ; the high catalytic activity for methane conversion of $>99\%$ at 800 °C.	S36

Table S2. The calculated methods used and the possible HM DP oxide systems as well as the experimentally verified examples.

Calculation methods	Material systems	Calculation results	Experimentally verified	Ref.
Full-potential linearized augmented Plane wave (FPLAPW)	$\text{Sr}_2\text{CrReO}_6$	Semiconductor to half-metal transition observed through 5% volume compression	Yes	S37
	$\text{Sr}_{2-x}\text{La}_x\text{Fe}_{1+y/2}\text{Mo}_{1-y/2}\text{O}_6$	Half-metal feature preserved in SLFMO for $x = 1/2$ and 1 contrary to the case for SFMO	Yes	S38, S39
	Sr_2MoBO_6 (B=W, Re, Os)	$\text{Sr}_2\text{MoOsO}_6$ is a compensated half metal	Yes	S40
	$\text{La}_2\text{NbMnO}_6$	HM DP oxide	No	S41
	$\text{Bi}_2\text{CuCrO}_6$	Doping with Pb, the system experiences a transition from A-type AFM phase to FiM phase	No	S42
Full-potential linear muffin-tin orbital method (FPLMTO)	$\text{Sr}_2\text{FeMoO}_6$, $\text{Sr}_2\text{FeReO}_6$, Sr_2CrWO_6	FiM HM DP oxides	Yes	S43
Linearized Augmented Plane Waves method (LAPW)	A_2FeMoO_6 (A = Ca, Sr, and Ba)	Ferromagnetic oxides	Yes	S44, S45
	$\text{LaMM}'\text{O}_3$, $\text{MM}' = \text{MnCo}$, CrFe , CrRu , CrNi , MnV , and VCu	La_2VMnO_6 , La_2VCuO_6 HM DP oxides	Yes, La_2VMnO_6 , Yes, La_2VCuO_6 ,	S46
	$(\text{Ba}_x\text{Sr}_{1-x})_2\text{CoWO}_6$ ($x=0.1, 0.2, 0.3, 0.5, 0.7, \text{ and } 0.9$)	Ba_2CoWO_6 and SrBaCoWO_6 are half-metals	Yes	S47
GGA+ U+SOC	$\text{Pb}_2\text{FeOsO}_6$	Tetragonal $I4/m$ structure and a C-type antiferromagnet	Yes	S48, S49
	$\text{Sr}_2\text{BB}'\text{O}_6$ (B= Cr, Mo, and B' = W, Re, Os)	HM DP oxide	Yes	S50
	$\text{Pr}_{2-x}\text{Sr}_x\text{MgIrO}_6$	$x = 0.5$ and 1.5 HM FiM DP oxide	No	S51
	Ca_2AOsO_6 (A = Cr, Mo)	when the volume is compressed to smaller than $0.9V_0$, $\text{Ca}_2\text{MoOsO}_6$ turns out to be a half metal	No	S52
	$\text{La}_2\text{NiCrO}_6$	FM HM DP oxide	No	S53

GGA + U	$(\text{Sr,Ca})_2\text{BRhO}_6$ (B = Cr, Mn, Fe)	Cr–Rh and Mn–Rh compounds predicted to be FM half-metals	No	S54
	$\text{Sr}_{2-y}\text{La}_y\text{FeMoO}_6$	FM HM DP oxides	Yes	S55
	$\text{Sr}_2\text{FeCoO}_6$	FM HM DP oxide	Yes	S56
	$\text{A}_2\text{CrRu}(\text{Os})\text{O}_6$ (A = Si, Ge, Sn, and Pb)	All its electronic structures convert HM-AF into unconventional AF-Is	No	S57
	Sr_2CoWO_6	FM and AFM phases calculated by GGA and GGA+U methods	Yes	S58
	$\text{La}_2\text{VCuO}_6, \text{La}_2\text{VTcO}_6$	HM AFM DP oxide	No	S59
	BaSrNiWO_6	FM or AFM states	Yes	S60
	$\text{Bi}_2\text{BB}'\text{O}_6$ (B, B' = 3d transitional metals)	$\text{Bi}_2\text{CrNiO}_6$ and $\text{Bi}_2\text{CrZnO}_6$, HM DP oxides	No	S61
	$\text{Lu}_2\text{NiIrO}_6$	FiM HM DP oxides	Yes	S62
	$\text{Pb}_2\text{XX}'\text{O}_6$ (X = Ti, Zr, Hf, V, Nb and Ta, X' = Tc, Ru, Os and Rh)	$\text{Pb}_2\text{NbTcO}_6, \text{Pb}_2\text{TaTcO}_6, \text{Pb}_2\text{TiRuO}_6, \text{Pb}_2\text{ZrRuO}_6, \text{Pb}_2\text{HfRuO}_6, \text{Pb}_2\text{VRuO}_6, \text{Pb}_2\text{NbRuO}_6, \text{Pb}_2\text{TadRuO}_6, \text{Pb}_2\text{ZrOsO}_6, \text{Pb}_2\text{HfOsO}_6, \text{Pb}_2\text{VOsO}_6, \text{Pb}_2\text{ZrRhO}_6$ and $\text{Pb}_2\text{HfRhO}_6$ FiM HM DP oxides	No	S63
	Y_2CrMnO_6	HM DP oxide	Yes	S64
	$\text{La}_{1.5}\text{CrFeO}_6$	$\text{La}_{1.5}\text{CrFeO}_6$ behaves as half-metal with the half-metallic gap of 0.42 eV	Yes	S65
Generalized gradient approximation (GGA)	BiPbVRuO_6 and BiPbVOsO_6	HM-AFM DP oxides	No	S66
GGA and GGA+U	$\text{Ba}_2\text{DySbO}_6$	HM DP oxide	Yes	S67, S68
GGA and GGA+U	$\text{Sr}_2\text{BB}'\text{O}_6$ (B, B' = 3d transition metal)	$\text{Sr}_2\text{ScCrO}_6, \text{Sr}_2\text{TiCrO}_6, \text{Sr}_2\text{MnCrO}_6, \text{Sr}_2\text{ZnMnO}_6,$ and $\text{Sr}_2\text{ZnFeO}_6$	$\text{Sr}_2\text{ScCrO}_6$, No $\text{Sr}_2\text{TiCrO}_6$, Yes $\text{Sr}_2\text{MnCrO}_6$, No $\text{Sr}_2\text{ZnMnO}_6$, No $\text{Sr}_2\text{ZnFeO}_6$, No	S69
GGA and GGA+U	$\text{Ba}_2\text{CdReO}_6$	FM HM DP oxide	Yes	S70
GGA, GGA+U, GGA+SOC, and GGA+SOC+U	$\text{Sr}_2\text{NiOsO}_6$	FM or AFM HM DP oxide	Yes	S71
GGA, GGA-SOC	$\text{Ba}_2\text{MnTeO}_6$	FM HM DP oxide	Yes	S72
GGA + mBJ	$\text{Ba}_2\text{FeNiO}_6$	100% spin polarization	No	S73
GGA-PBE	Ba_2MMoO_6 (M=Cr, Mn, Fe)	FM HM DP oxide	Yes	S74
LSDA+U	$\text{Ba}_2\text{CeCoO}_6$	FM HM DP oxide with space group of $\text{Fm}\bar{3}\text{m}$	No	S75
	$\text{Sr}_2\text{CoMoO}_6$	From an AFM semiconductor to a half-metal	Yes	S76
	$\text{Ba}_2\text{FeMoO}_6$	HM DP oxide	Yes	S77
LSDA+U	K_2MnRhO_6 and La_2CrWO_6	Compensated half metals	No	S78
LSDA and LSDA+U	$\text{Sr}_2\text{Fe}_{1-x}\text{Cr}_x\text{ReO}_6$ (x = 0.0, 0.25, 0.5, 0.75, 1.0)	Cr doping from $3 \mu_B$ for x = 0.0 down to $1.0 \mu_B$ for x = 1.0 per formula unit	Yes	S79
LDA	Ba_2MMoO_6 (M = Mn, Fe)	HM DP oxide	Yes	S80
DFT +U	$\text{Bi}_2\text{FeCrO}_6$	Doping to induce half-metal to insulator	Yes	S81
	$\text{Sr}_2\text{FeMoO}_6$	Antiferromagnets	Yes	S82
	$\text{La}_2\text{TiFeO}_6$	HM	Yes	S83
DFT	$\text{Ca}_2\text{Fe}_{1-x}\text{Ni}_x\text{OsO}_6$	x = 0.5, ferrimagnetism with $\mu_{\text{tot}} = 2 \mu_B/\text{f.u.}$	Yes	S84, S85
	$\text{La}_2\text{FeMnO}_6$	FM HM DP oxide	Yes	S86, S87

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