

## **Efficient pure near-infrared organic light-emitting diodes based on tris(2,4,6-trichlorophenyl) methyl radical derivatives**

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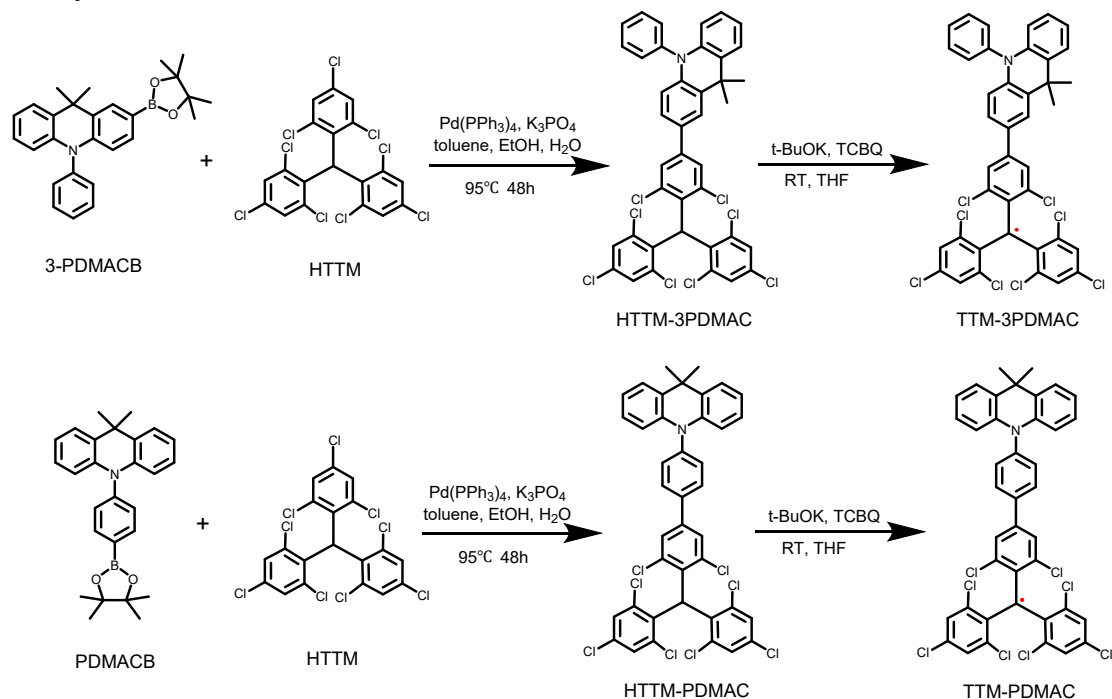
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## General information

All reagents and solvents required for synthesis and characterization are purchased from commercial suppliers and used directly without any treatment. A Bruker Avance-III 500 NMR spectrometer was used for the  $^1\text{H}$  collection in  $\text{CDCl}_3$  and  $\text{CH}_2\text{Cl}_2$  solvent with tetramethylsilane as the internal standard. Mass spectra of all compounds were recorded on Thermo Fisher ITQ1100 GC-MS mass detector. A Shimadzu UV-2550 spectrophotometer was applied to record the ultraviolet (UV)-visible spectra. Fluorescence spectra were recorded using a RF-5301 PC spectrophotometer and QE pro. All of PLQEs are determined with a calibrated integrating sphere system and using QE pro of Ocean Insight as fluorescence spectrometer. The electrochemical oxidation and reduction potentials were recorded using an electrochemical analyzer (CHI660C, CH Instruments, USA). The fluorescence decay spectra were recorded on an Edinburgh fluorescence spectrometer (FLS980), and the lifetime of the excited states was measured by the time-correlated single photon counting method under the excitation of a laser (375 nm). Thermal gravimetric analysis (TGA) were characterized by a TA INSTRUMENTS Q500 TGA analyzer. Ready-made indium tin oxide (ITO) glass substrates were purchased and cleaned. After dried with  $\text{N}_2$ , they were treated with UV irradiation for 20 min and next transferred to a vacuum deposition system with the pressure of  $4\text{-}6 \times 10^{-6}$  mbar. The  $\text{MoO}_3$  layer was deposited at a rate of  $0.3 \text{ \AA s}^{-1}$ . All the organic layers were deposited at  $0.4\text{-}0.6 \text{ \AA s}^{-1}$ . The evaporation rate of cathode LiF and Al metal layer were  $0.1 \text{ \AA s}^{-1}$  and  $0.8\text{-}1.4 \text{ \AA s}^{-1}$  respectively. The current-voltage characteristics were measured using a Keithley 2400 programmable electrometer. The EL spectra and EQEs were measured using QE pro spectroradiometer of Ocean Insight

together with a calibrated integrating sphere at room temperature in glove box.

## 1. Synthesis



Scheme 1. Synthetic routes and chemical structures of TTM-3PDMAC and TTM-PDMAC.

The HTTM<sup>1</sup> and 3-PDMACB<sup>2</sup> and PDMACB<sup>3</sup> were prepared as reported.

### (1) Synthesis of HTTM-3PDMAC

HTTM (1.0 g, 1.80 mmol) and the pinacol borane of 3-PDMAC (0.67 g, 1.80 mmol) was dissolved in a mixed solvent of toluene (12 ml), K<sub>3</sub>PO<sub>4</sub> aqueous solution (8 ml, 2 mol / L) and ethanol (4 ml), and catalyst Pd(PPh<sub>3</sub>)<sub>4</sub> (0.10 g, 0.09 mmol) was added under argon atmosphere. The mixture was stirred at 95°C for 48 h under argon atmosphere. After the reaction mixture cooling to room temperature, the solution was extracted with dichloromethane, organic layer was collected and dried. The solvent was removed under vacuum and the crude product was purified by silica gel column chromatography (using petroleum ether: dichloromethane = 10:1 v/v). HTTM-3PDMAC was obtained as a white solid. GC-MS (m/z): calculated for C<sub>40</sub>H<sub>25</sub>Cl<sub>8</sub>N,

802.94; found, 803.37;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$ 7.68 – 7.62 (m, 3H), 7.56 – 7.46 (m, 3H), 7.40 – 7.33 (m, 5H), 7.24 (t,  $J = 2.3$  Hz, 2H), 7.19 (dd,  $J = 8.6, 2.0$  Hz, 1H), 7.05 – 6.95 (m, 2H), 6.78 (s, 1H), 6.34 (d,  $J = 8.6$  Hz, 1H), 6.31 (dd,  $J = 7.9, 1.0$  Hz, 1H), 1.78 (s, 3H), 1.77 (s, 3H).

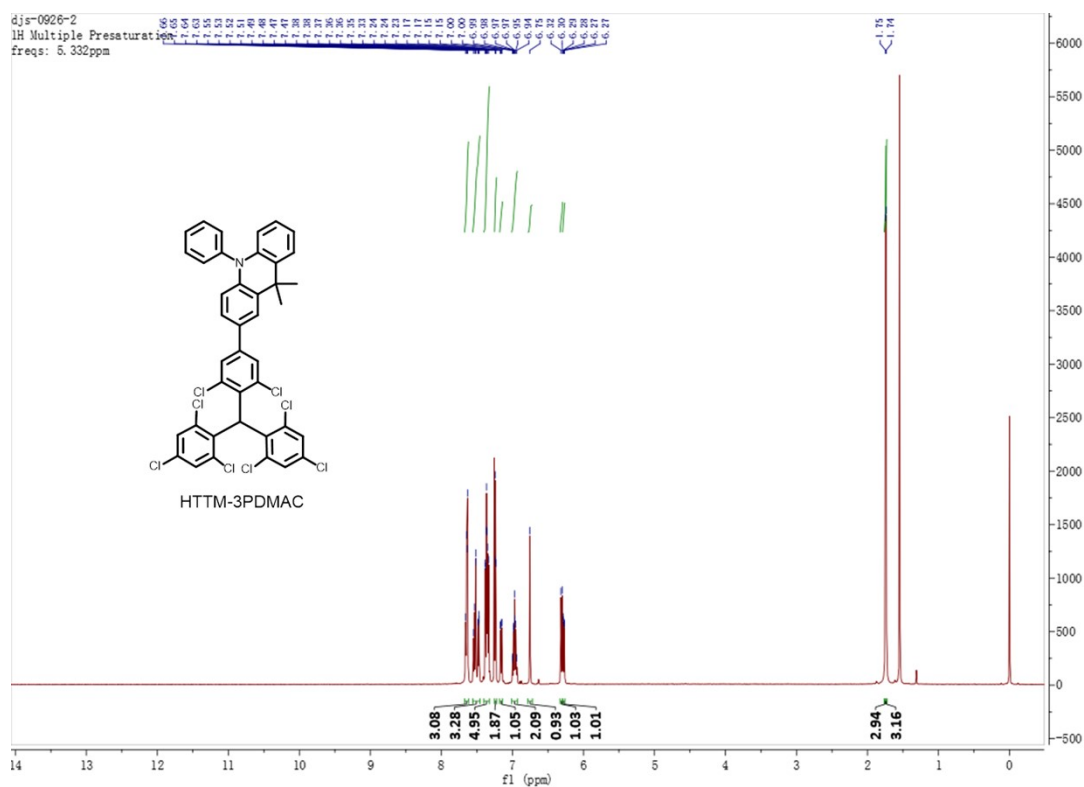


Figure S1.  $^1\text{H}$  NMR spectrum of HTTM-3PDMAC in  $\text{CDCl}_3$ .

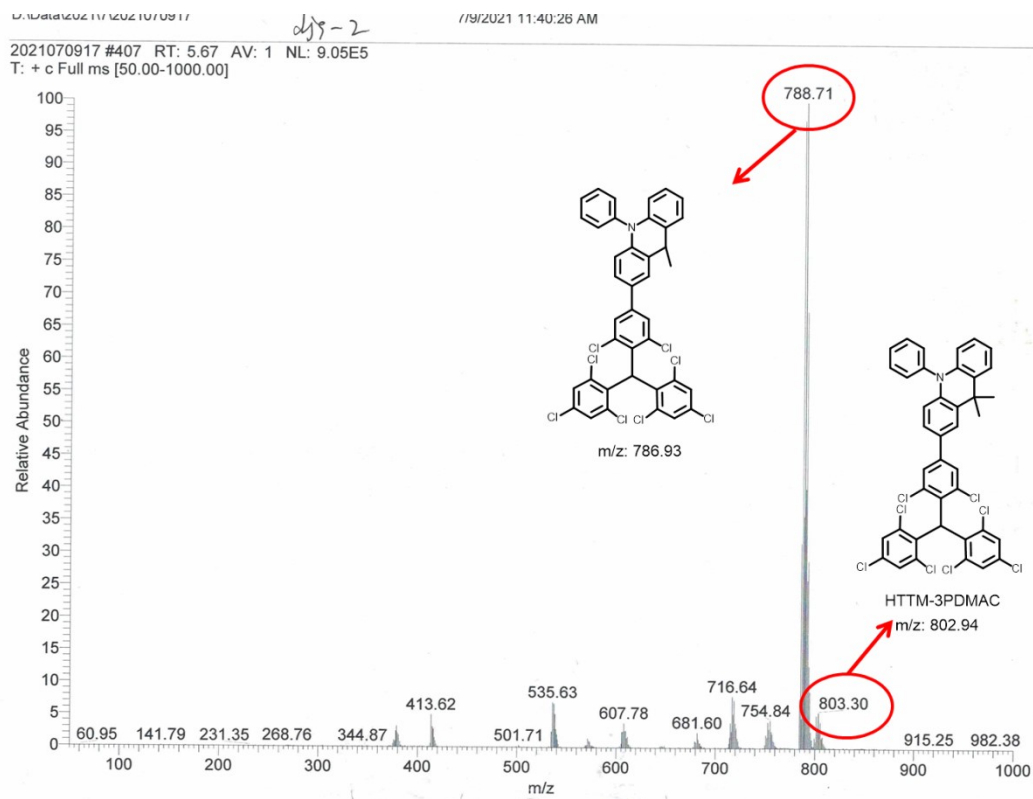


Figure S2. Mass Spectrum of HTTM-3PDMAC.

## (2) Synthesis of TTM-3PDMAC

Under argon atmosphere and in the dark, the HTTM-3PDMAC (1.00 equiv) was dissolved in dry THF (40 ml). Then KOtBu (4.00 equiv) was added, the solution become claret-colored immediately. The solution was stirred for 5 h in the dark at room temperature, and then p-Chloranil (5 equiv) was added. The solution was stirred for further 1 h. After the reaction finished, the solvent was removed under vacuum and the crude product was purified by silica gel column chromatography (using petroleum ether: dichloromethane = 10:1v/v). The crude product was recrystallized twice from dichloromethane and methanol and a green solid was obtained. GC-MS (m/z): calculated for  $C_{40}H_{25}Cl_8N$ , 801.94; found, 802.46;

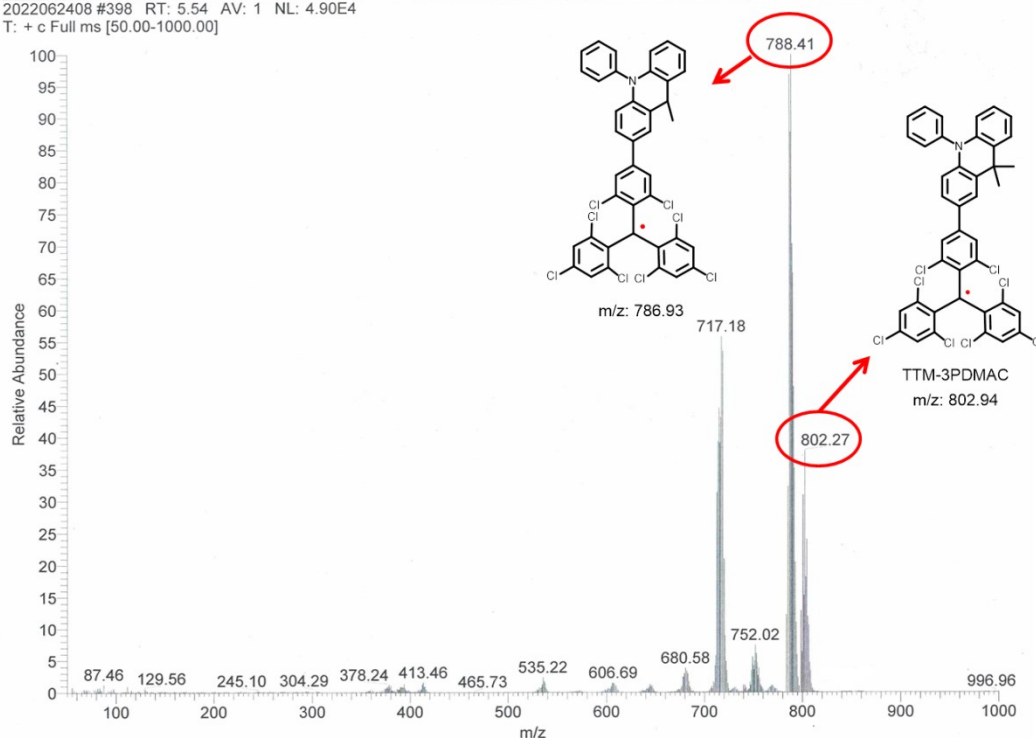
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Figure S3. Mass Spectrum of TTM-3PDMAC.

### (3) Synthesis of HTTM-PDMAC

HTTM (1.0 g, 1.80 mmol) and the pinacol borane of PDMAC (0.67 g, 1.80 mmol) was dissolved in a mixed solvent of toluene (12 ml),  $K_3PO_4$  aqueous solution (8 ml, 2 M) and ethanol (4 ml), and catalyst  $Pd(PPh_3)_4$  (0.10 g, 0.09 mmol) was added under argon atmosphere. The mixture was stirred at 95°C for 48 h under argon atmosphere and in the dark. After the reaction mixture cooling to room temperature, the solution was extracted with dichloromethane, organic layer was collected and dried. The solvent was removed under vacuum and the crude product was purified by silica gel column chromatography (using petroleum ether: dichloromethane = 10:1 v/v). HTTM-PDMAC was obtained as a white solid. GC-MS (m/z): calculated for  $C_{40}H_{25}Cl_8N$ , 802.94; found, 803.37;  $^1H$  NMR (500 MHz,  $CD_2Cl_2$ )  $\delta$  7.91 (d,  $J = 7.9$  Hz, 2H), 7.77 (d,  $J = 1.8$  Hz, 1H), 7.64 (d,  $J = 1.8$  Hz, 1H), 7.54 – 7.44 (m, 5H), 7.34 (d,  $J = 2.0$  Hz, 1H), 7.08 – 6.92

(m, 4H), 6.86 (s, 1H), 6.33 (d, J = 6.0 Hz, 2H), 1.72 (s, 6H).

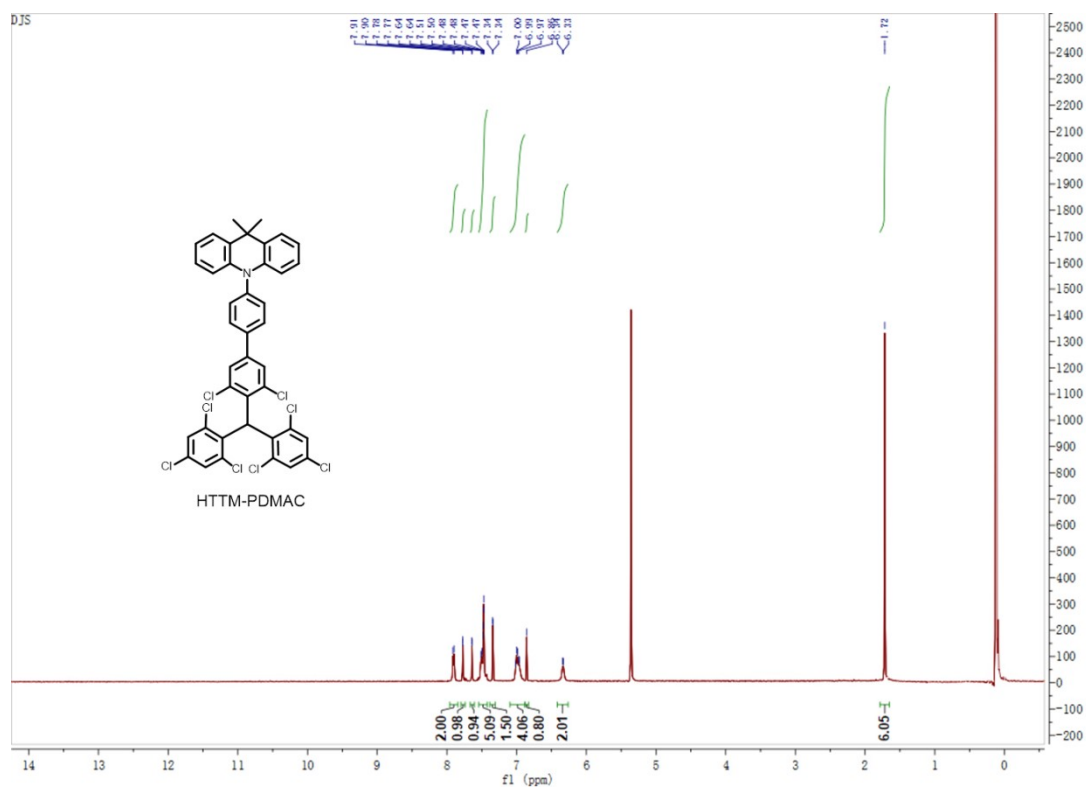


Figure S4.  $^1\text{H}$  NMR spectrum of HTTPM-PDMAC in  $\text{CD}_2\text{Cl}_2$ .

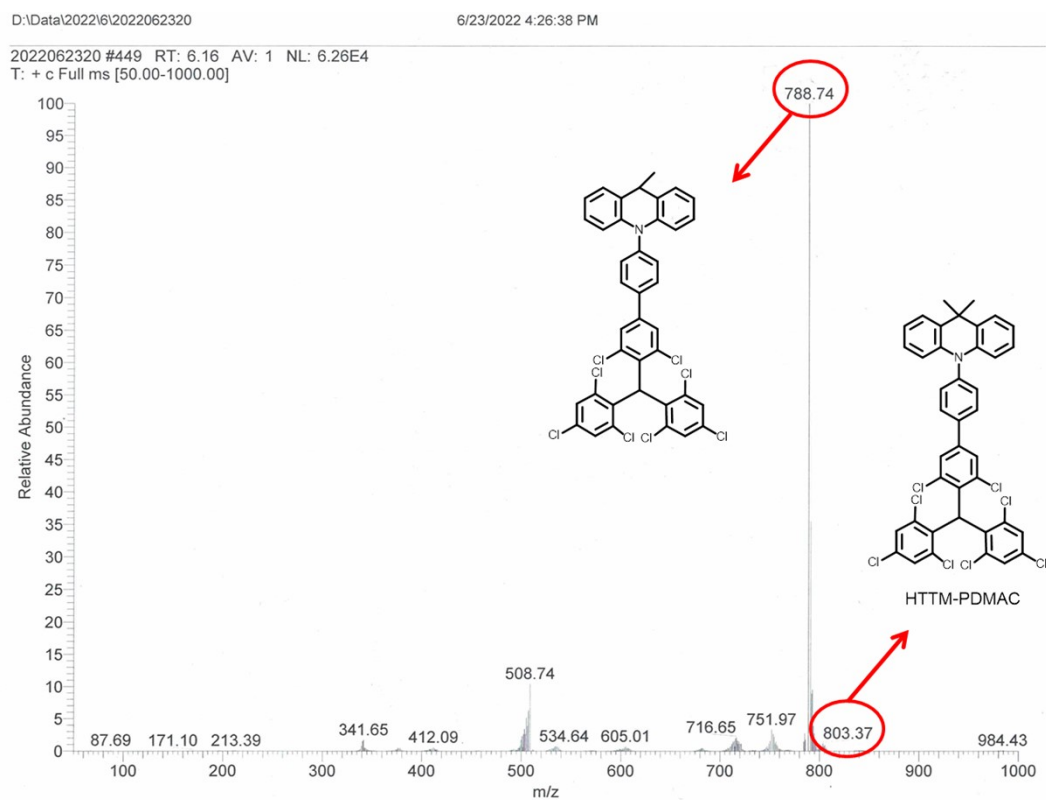


Figure S5. Mass Spectrum of HTTPM-PDMAC.

#### (4) Synthesis of TTM-PDMAC

Under argon atmosphere and in the dark, the TTM-PDMAC (1.00 equiv) was dissolved in dry THF (40 ml). Then KOtBu (10.00 equiv) was added, the solution become claret-colored immediately. The solution was stirred for 5 h in the dark at room temperature, and then p-Chloranil (5 equiv) was added. The solution was stirred for further 1 h. After the reaction finished, the solvent was removed under vacuum and the crude product was purified by silica gel column chromatography (using petroleum ether: dichloromethane = 10:1v/v). The crude product was recrystallized twice from dichloromethane and methanol and a gray solid was obtained. GC-MS (m/z): calculated for  $C_{40}H_{24}Cl_8N$ , 801.94; found, 802.46.

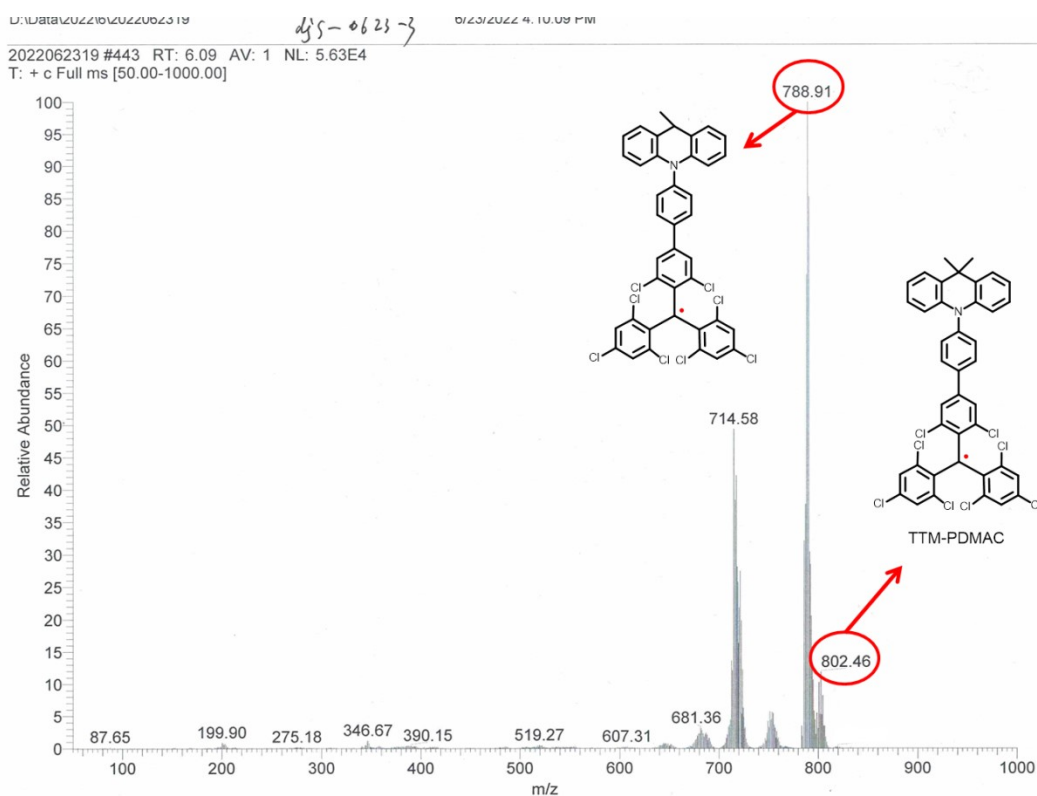


Figure S6. Mass Spectrum of TTM-PDMAC.



## 2. EPR spectra of TTM-3PDMAC and TTM-PDMAC

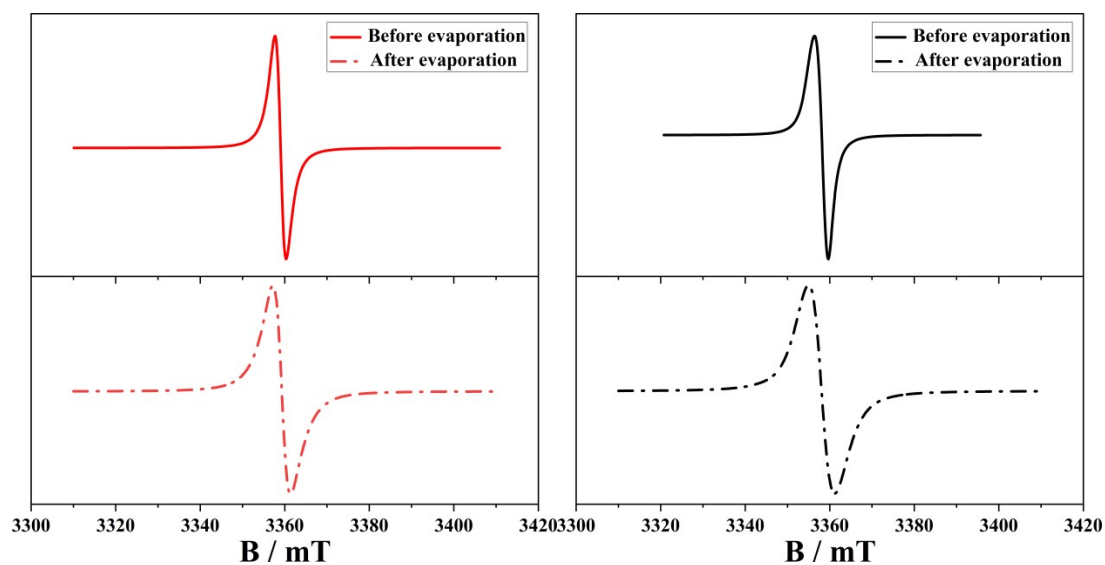
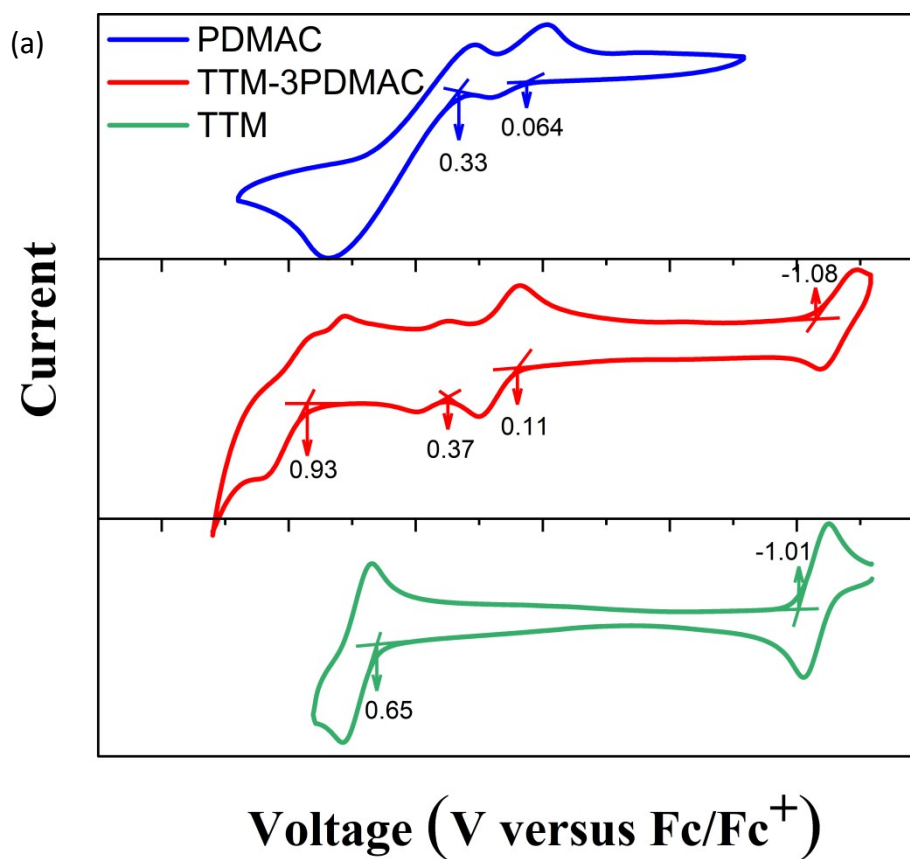
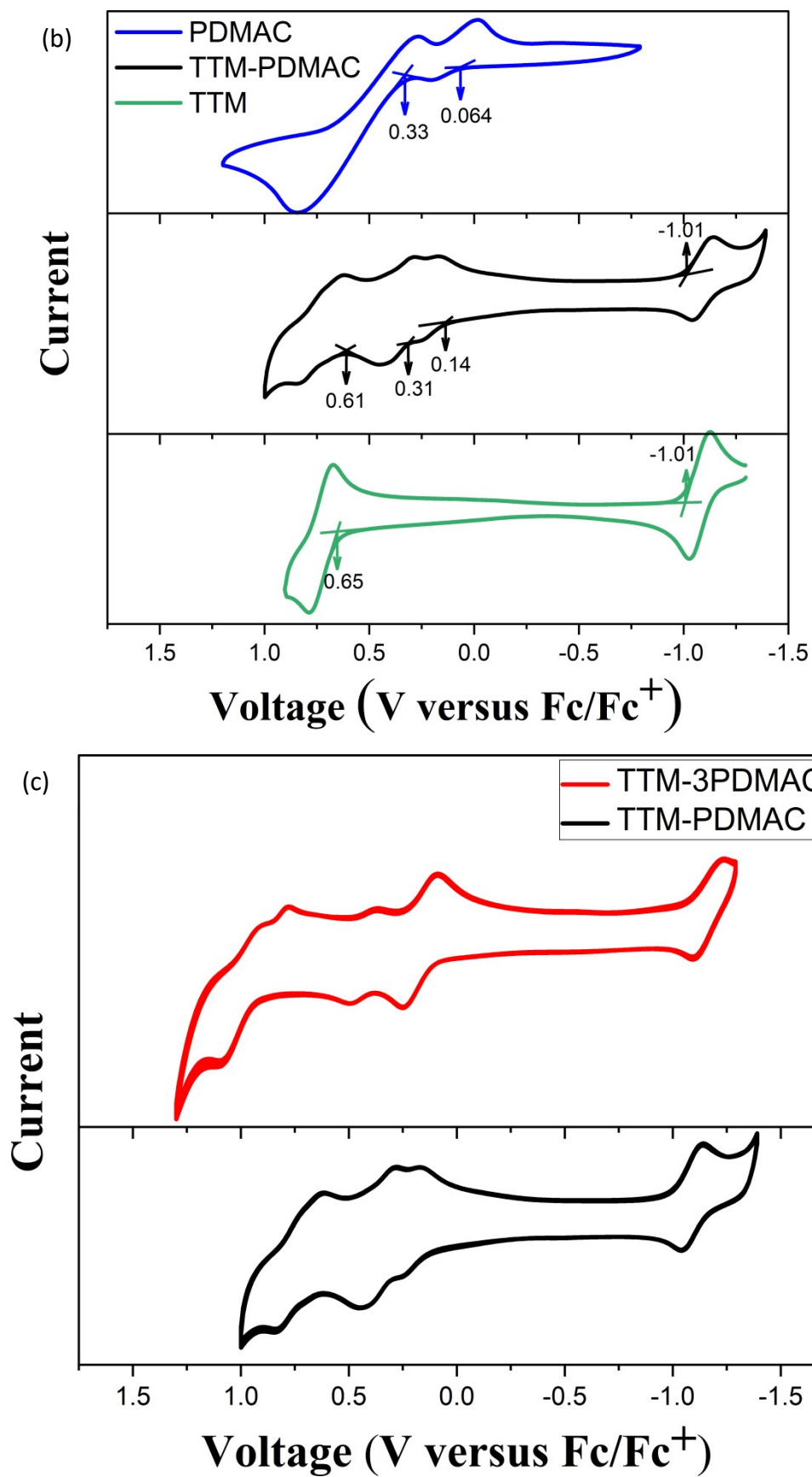


Figure S7. EPR spectra of TTM-3PDMAC (red) and TTM-PDMAC (black).

## 3. Electrochemical properties





#### 4. Quantum chemical calculations

Table S1. Cartesian coordinates of the optimized ground state geometries by DFT calculation (UB3LYP/6-31G(d, p))

TTM-3PDMAC:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.012310	1.220087	-0.080889
2	6	0	-1.373197	1.175807	0.073692
3	6	0	-1.960242	0.146539	0.809655
4	6	0	-1.161769	-0.838416	1.390867
5	6	0	0.223709	-0.794198	1.236214
6	6	0	0.810783	0.235132	0.500324
7	6	0	-3.447244	0.099023	0.975713
8	6	0	-4.067126	1.259658	0.261791
9	6	0	-3.972421	-1.178787	0.399174
10	17	0	-2.357233	2.389624	-0.642748
11	17	0	-1.885281	-2.106856	2.297813
12	6	0	-4.265085	2.466246	0.932944
13	6	0	-4.842788	3.547636	0.267731
14	6	0	-5.222370	3.422404	-1.068583
15	6	0	-5.024331	2.215870	-1.739760
16	6	0	-4.446707	1.134426	-1.074524
17	6	0	-4.959891	-1.895065	1.075443
18	6	0	-5.449127	-3.085755	0.538268
19	6	0	-4.950963	-3.560009	-0.675164
20	6	0	-3.963579	-2.843701	-1.351477
21	6	0	-3.474257	-1.653041	-0.814258
22	17	0	-3.797129	2.620592	2.579806
23	17	0	-4.202676	-0.352628	-1.901614
24	17	0	-2.257296	-0.770210	-1.647670
25	17	0	-5.573923	-1.310475	2.570830
26	17	0	-5.553942	-5.027376	-1.337139
27	17	0	-5.934286	4.755124	-1.888337
28	6	0	2.138861	0.277615	0.352095
29	6	0	2.863023	-0.905169	0.207189
30	6	0	4.256012	-0.867155	0.042447
31	6	0	4.919569	0.369841	0.009810
32	6	0	4.186449	1.548530	0.180542
33	6	0	2.800358	1.503530	0.342282
34	6	0	5.034779	-2.156714	-0.094926
35	6	0	6.521720	-2.002042	-0.326363

36	6	0	7.108158	-0.726316	-0.346457
37	7	0	6.317920	0.407234	-0.102219
38	6	0	7.321857	-3.138524	-0.518701
39	6	0	8.696111	-3.006767	-0.715361
40	6	0	9.280767	-1.742421	-0.712737
41	6	0	8.490763	-0.607466	-0.520244
42	6	0	6.898426	1.525194	0.024014
43	6	0	6.767982	2.249393	1.210153
44	6	0	8.306732	3.264903	-0.880337
45	6	0	7.666734	2.032236	-1.019911
46	6	0	4.455947	-2.929522	-1.272742
47	6	0	4.893888	-2.872442	1.242016
48	1	0	0.475293	2.031813	-0.661187
49	1	0	0.853380	-1.571016	1.694546
50	1	0	-4.999017	4.499191	0.797009
51	1	0	-5.323644	2.117159	-2.793642
52	1	0	-6.227815	-3.650728	1.071606
53	1	0	-3.570770	-3.217714	-2.308459
54	1	0	2.339800	-1.872648	0.222095
55	1	0	4.705690	2.518241	0.187454
56	1	0	2.230347	2.436669	0.462031
57	1	0	6.862898	-4.138193	-0.514336
58	1	0	9.317910	-3.900465	-0.872423
59	1	0	10.365535	-1.638137	-0.862416
60	1	0	8.959304	0.387643	-0.505162
61	1	0	6.159188	1.843047	2.031281
62	1	0	8.912491	3.660039	-1.709146
63	1	0	7.769600	1.461717	-1.954751
64	1	0	4.597667	-2.342731	-2.207815
65	1	0	4.978284	-3.907852	-1.366558
66	1	0	3.369417	-3.102659	-1.104677
67	1	0	4.380329	-3.847932	1.088900
68	1	0	5.903527	-3.049370	1.675714
69	1	0	4.295142	-2.243385	1.938118
70	6	0	8.185968	4.006303	0.306316
71	1	0	8.681479	4.949922	0.400935
72	6	0	7.405328	3.491079	1.366963
73	1	0	7.297727	4.036581	2.281156

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TTM-PDMAC:

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Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z

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1	6	0	4.247801	1.135347	0.516725
2	6	0	3.667047	-0.160854	0.989585
3	6	0	4.241756	-1.286491	0.187301
4	6	0	2.180211	-0.134611	0.817428
5	6	0	3.419779	-2.030204	-0.659302
6	6	0	3.955291	-3.079114	-1.406743
7	6	0	5.312725	-3.384153	-1.307623
8	6	0	6.134715	-2.640397	-0.461111
9	6	0	5.599190	-1.591530	0.286419
10	6	0	4.638850	1.279979	-0.814310
11	6	0	5.180090	2.487721	-1.254844
12	6	0	5.330126	3.550757	-0.364368
13	6	0	4.938998	3.406168	0.966623
14	6	0	4.397837	2.198382	1.407199
15	6	0	1.559851	0.994684	0.283259
16	6	0	0.174441	1.019146	0.122975
17	6	0	-0.590481	-0.085703	0.496747
18	6	0	0.029868	-1.215021	1.030819
19	6	0	1.415289	-1.239460	1.191200
20	17	0	2.179910	-2.631212	1.849488
21	17	0	4.453787	-0.030105	-1.911759
22	17	0	2.502607	2.356250	-0.177514
23	17	0	6.612160	-0.674893	1.329793
24	17	0	5.972633	-4.676795	-2.228797
25	17	0	5.997113	5.039139	-0.907330
26	6	0	-1.918413	-0.062226	0.343073
27	17	0	3.915846	2.020069	3.047580
28	17	0	1.746879	-1.654142	-0.781544
29	6	0	-2.566665	-1.125605	-0.285006
30	6	0	-3.952059	-1.101020	-0.445420
31	6	0	-4.689096	-0.013147	0.022349
32	6	0	-4.040873	1.050174	0.650504
33	6	0	-2.655449	1.025648	0.810843
34	7	0	-5.946499	0.009120	-0.123242
35	6	0	-6.530968	1.164541	-0.664285
36	6	0	-7.928997	1.293197	-0.686538
37	6	0	-8.808442	0.171969	-0.178861
38	6	0	-8.082405	-1.074879	0.275805
39	6	0	-6.679062	-1.122901	0.265271
40	6	0	-5.735106	2.229150	-1.099184
41	6	0	-6.321263	3.397414	-1.590111
42	6	0	-7.708513	3.516609	-1.633394
43	6	0	-8.511425	2.470808	-1.179505

44	6	0	-8.813269	-2.189507	0.714266
45	6	0	-8.152378	-3.336901	1.151660
46	6	0	-6.759980	-3.375840	1.162256
47	6	0	-6.026524	-2.269700	0.728922
48	6	0	-9.757942	-0.223026	-1.302230
49	6	0	-9.509076	0.719792	1.057476
50	1	0	3.307074	-3.665728	-2.074349
51	1	0	7.205243	-2.880953	-0.383007
52	1	0	5.488591	2.601806	-2.304516
53	1	0	5.057285	4.244555	1.668845
54	1	0	-0.314859	1.909763	-0.298198
55	1	0	-0.573401	-2.086349	1.325527
56	1	0	-1.985399	-1.983574	-0.653812
57	1	0	-4.463335	-1.939572	-0.940826
58	1	0	-4.622164	1.908071	1.019442
59	1	0	-2.144170	1.864237	1.306182
60	1	0	-4.639218	2.145472	-1.054138
61	1	0	-5.687252	4.224191	-1.942915
62	1	0	-8.170157	4.434925	-2.025287
63	1	0	-9.606623	2.569432	-1.208028
64	1	0	-9.912820	-2.158089	0.712968
65	1	0	-8.730083	-4.210468	1.488021
66	1	0	-6.237994	-4.278633	1.512248
67	1	0	-4.927202	-2.300429	0.752361
68	1	0	-9.171672	-0.611563	-2.164840
69	1	0	-10.452029	-1.013970	-0.939711
70	1	0	-10.343726	0.667354	-1.622932
71	1	0	-10.609417	0.723079	0.890125
72	1	0	-9.269614	0.077329	1.934214
73	1	0	-9.160387	1.758801	1.251503

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## 5. Electroluminescence Performances

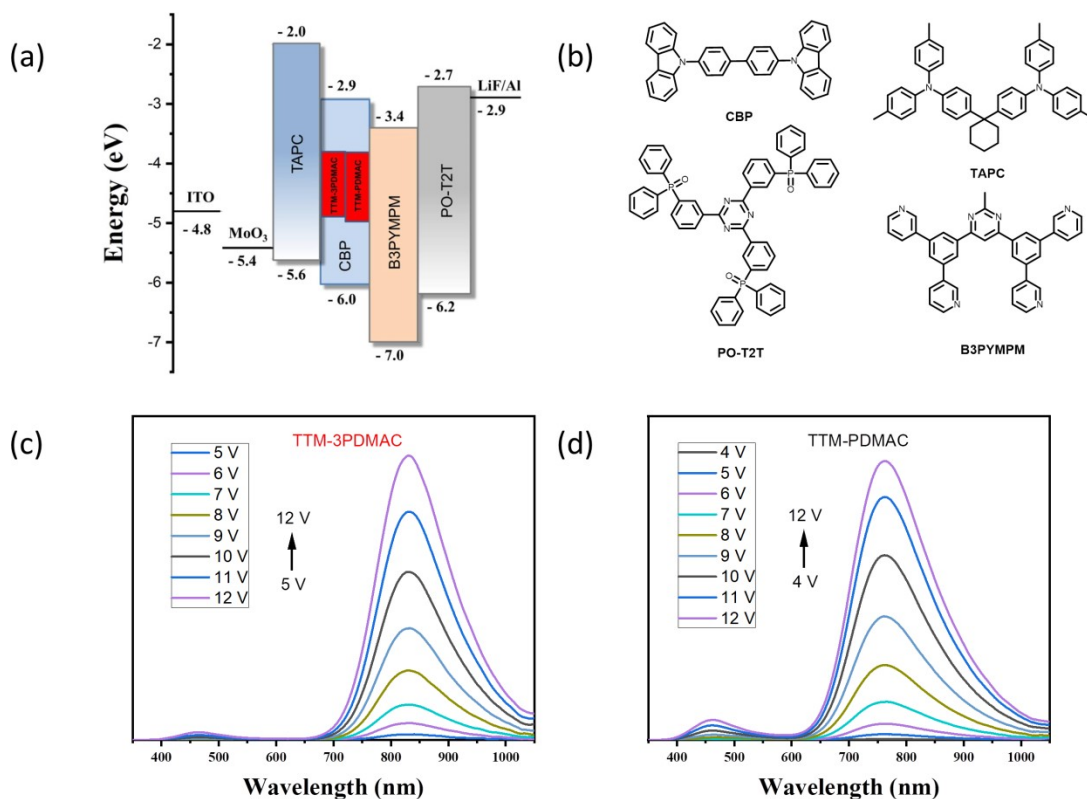


Figure S9. Energy-level diagram and electroluminescent properties of TTM-3PDMAC and TTM-PDMAC. a) Energy-level diagram of device; b) Materials used in this work; c) EL spectra of TTM-3PDMAC from 5-12V; d) EL spectra of TTM-PDMAC from 4-12V.

## 6. Summary of device performances of near infrared (NIR) OLED published to date

Table S2. Summary of the device performances of NIR OLED published to date with maximum electroluminescent (EL) wavelength over 800 nm.

$\lambda_{\text{EL}}^{\text{max}}$ (nm)	EQE (%)	Emitter material	[ref]	$\lambda_{\text{EL}}^{\text{max}}$ (nm)	EQE (%)	Emitter material	[ref]
800	1.9	metal free emitter	4	800	1	metal complex emitter	5
802	0.43	metal free emitter	6	803	9.58	metal complex emitter	7
804	2.2	metal free emitter	8	811	0.97	metal complex emitter	9
810	0.51	metal free emitter	10	814	1.5	metal complex emitter	11
814	0.5	metal free emitter	12	826	0.49	metal complex emitter	13
823	0.27	metal free emitter	14	847	0.19	metal complex emitter	15
824	0.16	metal free emitter	16	848	2.8	metal complex emitter	17
828	0.41	metal free emitter	10	846	1.5	metal complex emitter	17
830	2.47	metal free emitter	18	855	1	metal complex emitter	19
830	3.1	metal free emitter	this work	890	3.8	metal complex emitter	20
838	0.58	metal free emitter	21	900	3.8	metal complex emitter	22
840	1.12	metal free emitter	23	920	1.9	metal complex emitter	24
840	3.8	metal free emitter	25	1005	0.2	metal complex emitter	22
850	0.14	metal free emitter	14	1060	0.022	metal complex emitter	26
852	0.3	metal free emitter	10	1060	0.3	metal complex emitter	27
864	0.2	metal free emitter	6				

868	0.09	metal free emitter	14				
870	0.02	metal free emitter	14				
883	0.1	metal free emitter	28				
890	0.015	metal free emitter	29				
894	0.23	metal free emitter	10				
895	0.091	metal free emitter	30				
901	1.1	metal free emitter	31				
904	0.019	metal free emitter	32				
905	1.32	metal free emitter	33				
916	0.07	metal free emitter	21				
939	0.006	metal free emitter	30				
960	0.009	metal free emitter	28				
990	0.018	metal free emitter	30				
1010	0.003	metal free emitter	34				
1050	0.05	metal free emitter	35				
1050	0.16	metal free emitter	36				
1050	0.33	metal free emitter	36				
1080	0.73	metal free emitter	36				

## 7. References

1. Q. Peng, A. Obolda, M. Zhang and F. Li, *Angewandte Chemie*, 2015, **127**, 7091-7095.
2. T. Y. Sae Youn Lee, Hideaki Komiyama, Jiyoung Lee, and Chihaya Adachi *Advanced Materials*, 2016, **28**, 4019-4024.
3. J. Z. Shimin Hu, Xiangyu Zhu, Jingjing Guo, Shuming Chen, Zujin Zhao, and Ben Zhong Tang, *ACS applied materials & interfaces*, 2019, **11**, 27134-27144.
4. M. T. Sharbati, F. Panahi, A. Shourvarzi, S. Khademi and F. Emami, *Optik*, 2013, **124**, 52-54.
5. F. Nisic, A. Colombo, C. Dragonetti, D. Roberto, A. Valore, J. M. Malicka, M. Cocchi, G. R. Freeman and J. A. G. Williams, *Journal of Materials Chemistry C*, 2014, **2**, 1791-1800.
6. X. Du, J. Qi, Z. Zhang, D. Ma and Z. Y. Wang, *Chemistry of Materials*, 2012, **24**, 2178-2185.
7. S. F. Wang, Y. Yuan, Y. C. Wei, W. H. Chan, L. W. Fu, B. K. Su, I. Y. Chen, K. J. Chou, P. T. Chen, H. F. Hsu, C. L. Ko, W. Y. Hung, C. S. Lee, P. T. Chou and Y. Chi, *Advanced Functional Materials*, 2020, **30**.
8. Y. Yu, H. Xing, D. Liu, M. Zhao, H. H. Sung, I. D. Williams, J. W. Y. Lam, G. Xie, Z. Zhao and B. Z. Tang, *Angew Chem Int Ed Engl*, 2022, **61**, e202204279.
9. Z. L. Zhu, S. F. Wang, L. W. Fu, J. H. Tan, C. Cao, Y. Yuan, S. M. Yiu, Y. X. Zhang, Y. Chi and C. S. Lee, *Chemistry*, 2022, **28**, e202103202.
10. J.-F. Cheng, Z.-H. Pan, K. Zhang, Y. Zhao, C.-K. Wang, L. Ding, M.-K. Fung and J. Fan, *Chemical Engineering Journal*, 2022, **430**, 132744.
11. T.-C. Lee, J.-Y. Hung, Y. Chi, Y.-M. Cheng, G.-H. Lee, P.-T. Chou, C.-C. Chen, C.-H. Chang and C.-C. Wu, *Advanced Functional Materials*, 2009, **19**, 2639-2647.
12. R. T. F. Yixing Yang, Timothy T. Steckler, Sang Hyun Eom, John R. Reynolds, Kirk S. Schanze, and Jiangeng Xue, *APPLIED PHYSICS LETTERS*, 2008, **93**, 163305.
13. Y. Zhang, Z. Chen, X. Wang, J. He, J. Wu, H. Liu, J. Song, J. Qu, W. T. Chan and W. Y. Wong, *Inorg Chem*, 2018, **57**, 14208-14217.
14. Z. Z. Gang Qian, Min Luo, Dengbin Yu, Zhiqiang Zhang, Dongge Ma, and Zhi Yuan Wang, *J. Phys. Chem. C*, 2009, **113**, 1589-1595.
15. Y. Zhang, Q. Li, M. Cai, J. Xue and J. Qiao, *Journal of Materials Chemistry C*, 2020, **8**, 8484-8492.



16. X. Q. Wang, Y. Hu, Y. J. Yu, Q. S. Tian, W. S. Shen, W. Y. Yang, Z. Q. Jiang and L. S. Liao, *J Phys Chem Lett*, 2021, **12**, 6034-6040.
17. L. Huang, C. D. Park, T. Fleetham and J. Li, *Applied Physics Letters*, 2016, **109**, 23302.
18. L. Tejerina, A. G. Rapis, M. Rickhaus, P. Murto, Z. Genene, E. Wang, A. Minotto, H. L. Anderson and F. Cacialli, *J Mater Chem C Mater*, 2022, **10**, 5929-5933.
19. E. Rossi, A. Colombo, C. Dragonetti, D. Roberto, F. Demartin, M. Cocchi, P. Brulatti, V. Fattori and J. A. Williams, *Chem Commun (Camb)*, 2012, **48**, 3182-3184.
20. J. R. Sommer, R. T. Farley, K. R. Graham, Y. Yang, J. R. Reynolds, J. Xue and K. S. Schanze, *ACS Appl Mater Interfaces*, 2009, **1**, 274-278.
21. Y. J. Yu, Y. Hu, S. Y. Yang, W. Luo, Y. Yuan, C. C. Peng, J. F. Liu, A. Khan, Z. Q. Jiang and L. S. Liao, *Angew Chem Int Ed Engl*, 2020, **59**, 21578-21584.
22. K. R. Graham, Y. Yang, J. R. Sommer, A. H. Shelton, K. S. Schanze, J. Xue and J. R. Reynolds, *Chemistry of Materials*, 2011, **23**, 5305-5312.
23. A. Minotto, P. Murto, Z. Genene, A. Zampetti, G. Carnicella, W. Mammo, M. R. Andersson, E. Wang and F. Cacialli, *Adv Mater*, 2018, DOI: 10.1002/adma.201706584, e1706584.
24. J. L. Linyu Cao, Zhi-Qiang Zhu, Liang Huang, and Jian Li, *ACS Appl Mater Interfaces*, 2021, **13**, 60261-60268.
25. A. Shahalizad, A. Malinge, L. Hu, G. Laflamme, L. Haeberlé, D. M. Myers, J. Mao, W. G. Skene and S. Kéna - Cohen, *Advanced Functional Materials*, 2020, **31**, 2007119.
26. A. D. A. Afshin Shahalizad, Chantal Andraud, Muhammad Hasnan Sazzad, and Y. T. Dae-Hyeon Kim, Jean-Charles Ribierre, Jean-Michel Nunzi, Chihaya Adachi, *Organic Electronics*, 2017, **44**, 50-58.
27. F. D. Zhu-Qi Chen, Zu-Qiang Bian, Chun-Hui Huang, *Organic Electronics*, 2010, **11**, 369-376.
28. O. Fenwick, J. K. Sprafke, J. Binas, D. V. Kondratuk, F. Di Stasio, H. L. Anderson and F. Cacialli, *Nano Lett*, 2011, **11**, 2451-2456.
29. Y. Xuan, G. Qian, Z. Wang and D. Ma, *Thin Solid Films*, 2008, **516**, 7891-7893.
30. G. Tregnago, T. T. Steckler, O. Fenwick, M. R. Andersson and F. Cacialli, *Journal of Materials Chemistry C*, 2015, **3**, 2792-2797.
31. U. Balijapalli, R. Nagata, N. Yamada, H. Nakanotani, M. Tanaka, A. D'Aleo, V. Placide, M. Mamada, Y. Tsuchiya and C. Adachi, *Angew Chem Int Ed Engl*, 2021, **60**, 8477-8482.
32. D. G. Congrave, B. H. Drummond, P. J. Conaghan, H. Francis, S. T. E. Jones, C. P. Grey, N. C. Greenham, D. Credginton and H. Bronstein, *J Am Chem Soc*, 2019, **141**, 18390-18394.
33. M. T. Sharbati, F. Panahi and A. Gharavi, *IEEE Photonics Technology Letters*, 2010, **22**, 1695-1697.
34. Q. Liang, J. Xu, J. Xue and J. Qiao, *Chem Commun (Camb)*, 2020, **56**, 8988-8991.
35. B. D. Gang Qian, Min Luo, Dengbin Yu, Jie Zhan, Zhiqiang Zhang, and a. Z. Y. W. Dongge Ma, *Chem. Mater.*, 2008, **20**, 6208-6216.
36. G. Qian, Z. Zhong, M. Luo, D. Yu, Z. Zhang, Z. Y. Wang and D. Ma, *Advanced Materials*, 2009, **21**, 111-116.