

Electronic supplementary information

Mn²⁺-doped Cs₂ZnBr₄ scintillator for X-ray imaging

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Table S1. Main parameters of Rietveld refinement of Cs₂ZnBr₄:xMn²⁺.

x(Mn)	Space Group	Cell parameters (Å),	R _p , R _{wp} , R _{exp} , χ^2
0	<i>Pnma</i>	a = 10.19830 (7),	16.5, 11.7, 10.07, 1.34
		b = 7.73996 (8),	
		c = 13.53616 (7),	
		V = 1068.47 (3)	
25%	<i>Pnma</i>	a = 10.20406 (0),	17.0, 13.6, 9.17, 2.19
		b = 7.74563 (6),	
		c = 13.54518 (5),	
		V = 1070.57 (5)	

Table S2. Fractional atomic coordinates and isotropic displacement parameters (Å²) of Cs₂ZnBr₄:xMn²⁺

Atom	x	y	z	occ
0 %				
Cs1	0.6358 (9)	0.25	0.3962 (6)	0.5
Cs2	0.4756 (9)	0.25	0.8298 (4)	0.5
Zn	0.2283 (1)	0.25	0.4166 (1)	0.5
Br1	0.0033 (8)	0.25	0.3995 (4)	0.5
Br2	0.3117 (2)	0.25	0.5902 (5)	0.5
Br3	0.3265 (4)	0	0.3458 (4)	1.0
25 %				
Cs1	0.6362 (3)	0.25	0.3940 (9)	0.5
Cs2	0.4769 (1)	0.25	0.8317 (8)	0.5
Zn	0.2322 (7)	0.25	0.4144 (0)	0.37503
Mn	0.2322 (7)	0.25	0.4144 (0)	0.12497
Br1	0.0014 (3)	0.25	0.3977 (3)	0.5
Br2	0.3196 (1)	0.25	0.5918 (5)	0.5
Br3	0.3254 (1)	0	0.3458 (1)	1.0

Table S3. The result of ICP-OES measurements for Cs₂ZnBr₄:xMn²⁺

Mn/Zn feed ratio	molar	actual Zn concentration (mg/kg)	actual Mn concentration (mg/kg)	actual Mn-to-Zn molar ratio
0.05:0.95 (5%)		99023.1	636.7	0.64%
0.10:0.90 (10%)		98433.4	1268.5	1.29%
0.15:0.85 (15%)		97639.9	1480.1	1.52%
0.20:0.80 (20%)		95294.9	4258.3	4.47%
0.25:0.75 (25%)		88197.3	9490.8	10.76%
0.30:0.70 (30%)		91028.9	8324.5	9.14%

Table S4. PLQYs of Mn²⁺-doped Cs₂ZnBr₄ with different Mn²⁺ concentrations.

x Mn²⁺	PLQY
5%	12.66%
10%	19.57%
15%	21.75%
20%	26.73%
25%	58.82%
30%	45.05%

Table S5. PL lifetimes of the green emissions for Mn²⁺-doped Cs₂ZnBr₄ polycrystalline powders with different Mn²⁺ concentrations.

$\lambda_{\text{ex}} = 365 \text{ nm}, \lambda_{\text{em}} = 526 \text{ nm}$			
$x \text{ Mn}^{2+}$	$\tau_1 (\mu\text{s})$	$\tau_2 (\mu\text{s})$	τ_{average}
5%	25.41	151.58	93.62
10%	20.64	165.18	126.32
15%	15.06	165.25	131.41
20%	14.59	187.71	135.19
25%	14.17	287.80	255.90
30%	9.27	262.86	234.13

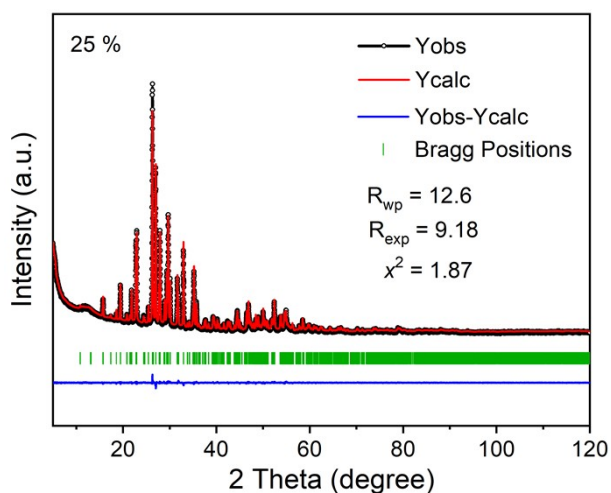
Table S6. PL lifetimes of the red emissions for Mn²⁺-doped Cs₂ZnBr₄ polycrystalline powders with different Mn²⁺ concentrations.

$x \text{ Mn}^{2+}$	$\tau (\mu\text{s})$
5%	158.30
10%	156.92
15%	156.15
20%	157.55
25%	163.63
30%	157.98

Table S7. Radioluminescence properties of recently reported metal halide scintillators.

Scintillators	Ligh yields (photons/MeV)	Detection limit (nGy _{air} /s)	Spatial resolution (lp/mm)	Imaging	Res
Cs ₃ Cu ₂ I ₅	79279	/	0.32 mm	Films	1
Cs ₃ Cu ₂ I ₅	30000	158		Single crystal	2
Rb ₂ CuCl ₃	16600	88.5	/	/	3
Rb ₂ CuBr ₃	91056	121.5	/	/	4
K ₂ CuBr ₃	23806	132.8	/	/	5
Cs ₂ AgI ₃ :Cu ⁺	82900	77.8	16.2	PDMS composite film	6
Rb ₂ AgBr ₃	25600	19	10.2	Film	7
(C ₃₈ H ₃₄ P ₂)MnBr ₄	66256	461.1	0.322 mm	PDMS composite film	8
TPP ₂ MnBr ₄	78000 ± 2000	8.8	15.7	Ceramic wafer	9
(ETP) ₂ MnBr ₄	35000 ± 2000	103	13.4 l	Transparent glass	10
(C ₂₄ H ₂₀ P ₂) ₂ MnBr ₄	/	608	14.5	TPU composite film	11
C ₄ H ₁₂ NMnCl ₃	50500	36.9	/	SO composite film	12
(C ₈ H ₂₀ N) ₂ MnBr ₄	24400	24.2	5	SO composite film	12
(Bmpip) ₂ Cu ₂ Br ₄	16000	710	/	/	13
(DIET) ₃ Cu ₃ Br ₃	20000 ± 700	189	11.71	PDMS composite film	14
(TBA)CuCl ₂	23373	/	/	PVDF composite film	15
(TBA)CuBr ₂	24134	/	/	PVDF composite film	15
Cs ₂ ZnBr ₄ :Mn ²⁺	15600	1160	5.06	PDMS composite film	This work

Note: “PMMA”, polymethyl methacrylate; “PDMS”, polydimethylsiloxane; “TPU”, thermoplastic polyurethane; “SO”, sucrose octaacetate; “PVDF”, polyvinylidene fluoride.

**Figure S1.** Rietveld refinement of 25%Mn²⁺-doped Cs₂ZnBr₄ sample.

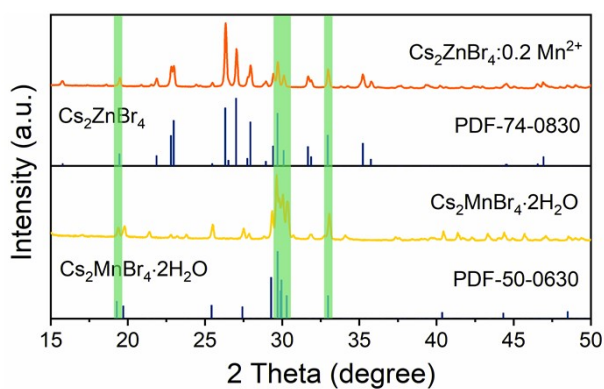


Figure S2. XRD patterns of $\text{Cs}_2\text{ZnBr}_4:25\%\text{Mn}^{2+}$ and $\text{Cs}_2\text{MnBr}_4\cdot 2\text{H}_2\text{O}$ samples.

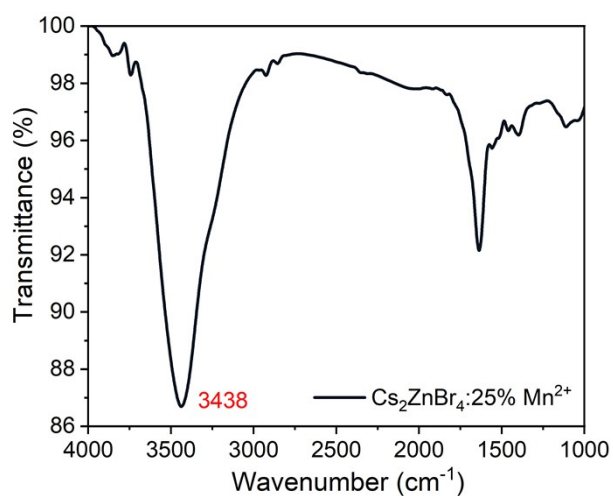


Figure S3. Fourier transform infrared spectrum (FTIR) of $\text{Cs}_2\text{ZnBr}_4:25\%\text{Mn}^{2+}$ sample.

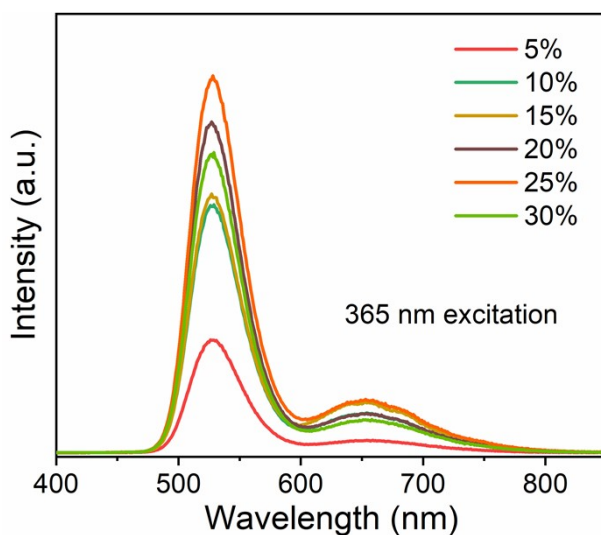


Figure S4. PL spectra of Mn^{2+} -doped Cs_2ZnBr_4 samples with different Mn^{2+} concentrations at room temperature ($\lambda_{\text{ex}} = 365 \text{ nm}$).

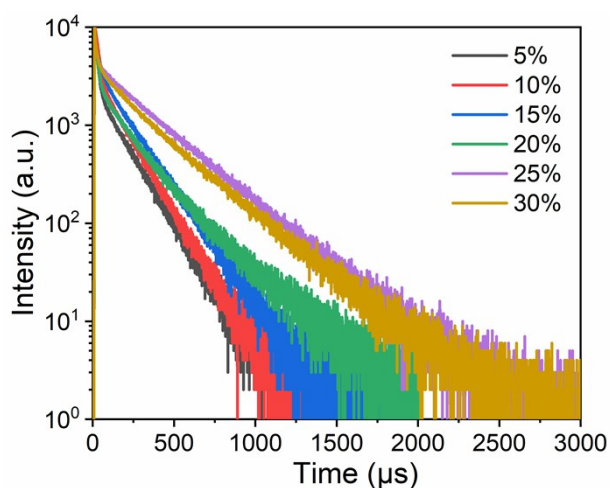


Figure S5. PL decay curves of Mn^{2+} -doped Cs_2ZnBr_4 samples with different Mn^{2+} concentrations at room temperature ($\lambda_{\text{ex}} = 365 \text{ nm}$, $\lambda_{\text{em}} = 526 \text{ nm}$).

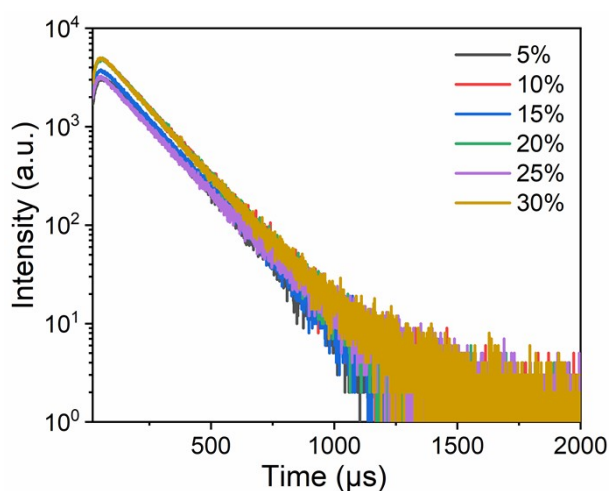


Figure S6. PL decay curves of Mn^{2+} -doped Cs_2ZnBr_4 samples with different Mn^{2+} concentrations at room temperature ($\lambda_{\text{ex}} = 365 \text{ nm}$, $\lambda_{\text{em}} = 655 \text{ nm}$).

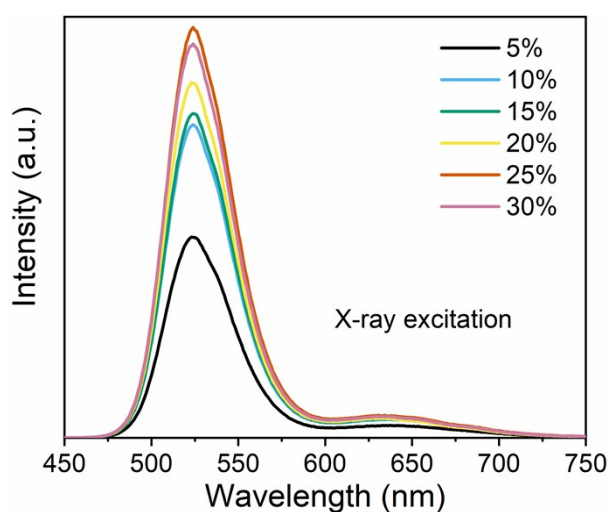


Figure S7. RL spectra of Mn^{2+} -doped Cs_2ZnBr_4 samples with different Mn^{2+} concentrations under X-ray excitation at room temperature (tube voltage: 50 kV; dose rate: $96.09 \mu\text{Gy}_{\text{air}} \text{ s}^{-1}$).

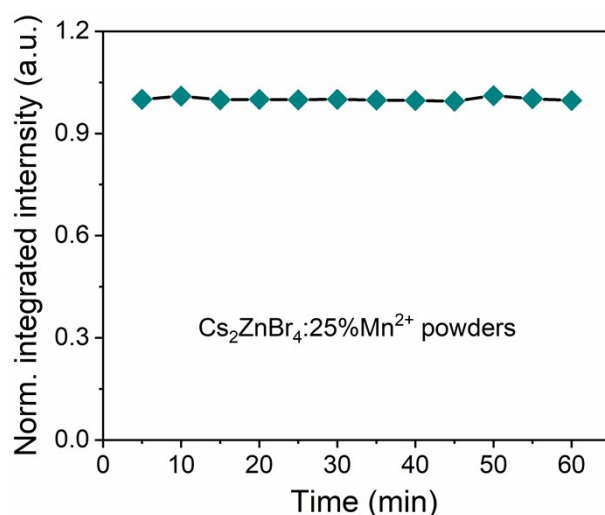


Figure S8. PL stabilities of $\text{Cs}_2\text{ZnBr}_4:25\%\text{Mn}^{2+}$ under continuous UV illumination ($\lambda_{\text{ex}} = 365 \text{ nm}$).

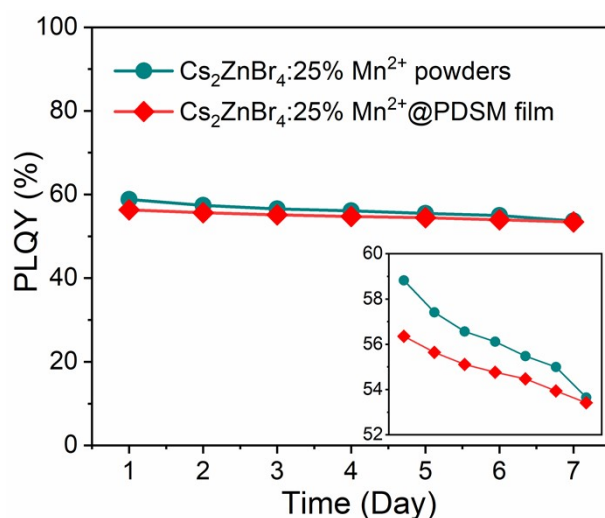


Figure S9. The stabilities of PLQYs of $\text{Cs}_2\text{ZnBr}_4:25\%\text{Mn}^{2+}$ polycrystalline powders and $\text{Cs}_2\text{ZnBr}_4:25\%\text{Mn}^{2+}$ @PDSM thin film, inset shows the enlarged pattern.

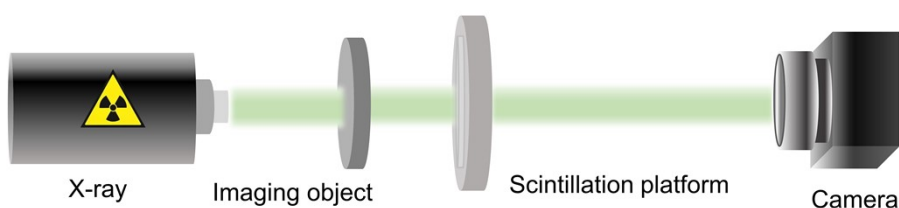


Figure S10. The home-made X-ray imaging system.

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