

Electronic Supplementary Information

Controlled molecular assemblies of chiral boron
dipyrromethene derivatives for circularly polarized
luminescence in the red and near-infrared regions

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Experimental

X-ray Diffraction Measurements. X-ray diffraction (XRD) measurements were carried out with a D8 ADVANCE (Bruker) using Ni-filtered Cu K α line radiation ($\lambda = 1.5418 \text{ \AA}$). The accelerating voltage was set at 40 kV with 20 mA current, and profiles were collected in the $5^\circ < 2\theta < 40^\circ$ range with a step size of 0.03° .

Small-angle X-ray scattering (SAXS) measurements. Small-angle X-ray scattering (SAXS) measurements were carried out with beam-line BL6A, Photon Factory, KEK, Japan (Proposal No. 21G524 and 22G027).

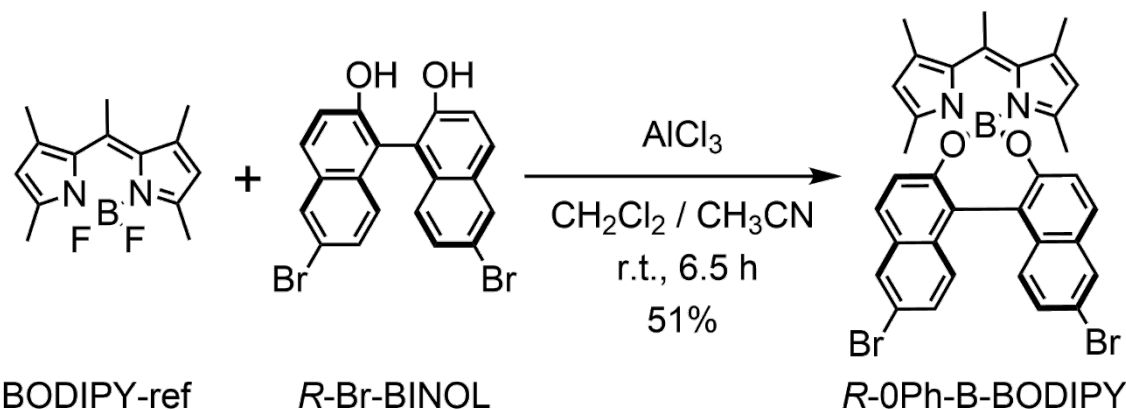
. The scattering vector, q , is defined as

$$q = \frac{4\pi}{\lambda} \sin \theta \quad (1),$$

where 2θ and λ are the scattering angle and wavelength (1.5 \AA), respectively. The camera length was 2410 mm. A PILATUS 2M detector (Dectris AG, Baden, Switzerland) detector was used with a q range of 0.07 to 2.0 nm^{-1} . Data processing, which included controlling the contrast of the 2D-patterns and the preparation of a 1D-profile from the obtained 2D-patterns, was performed using the FIT-2D software (Ver. 12.077, Andy Hammersley/ESRF, Grenoble, France).

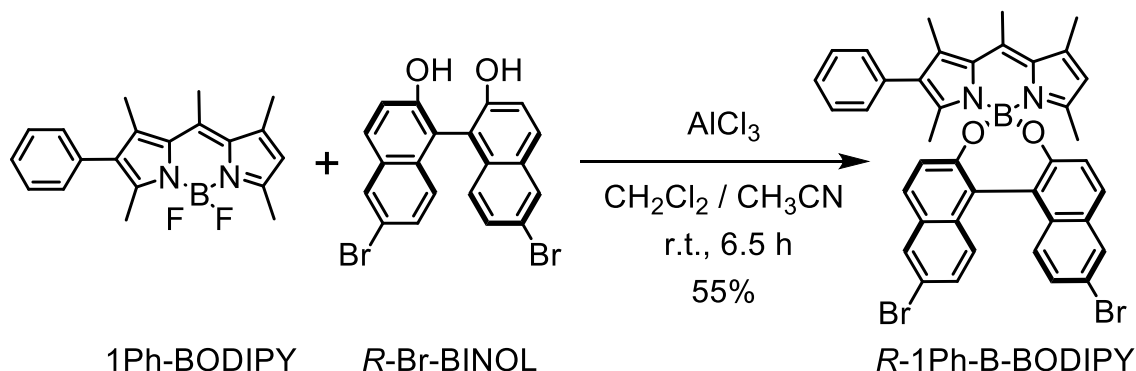
Synthesis

Scheme S1. Synthetic Scheme of *R*-0Ph-B-BODIPY.



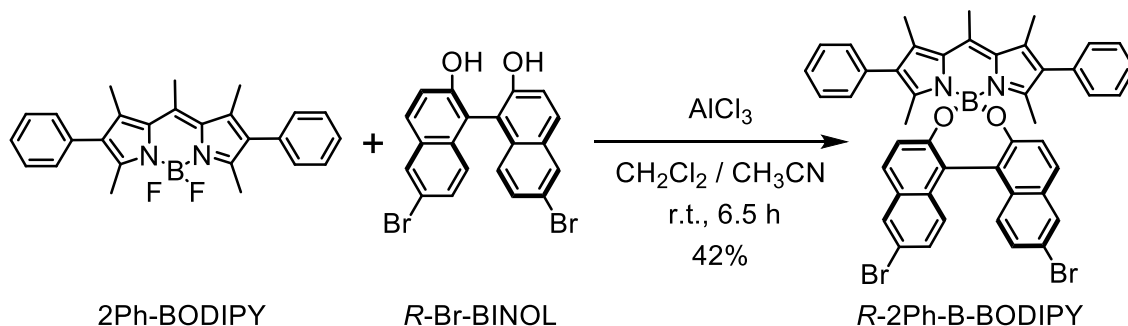
Synthesis of *R*-0Ph-B-BODIPY. BODIPY-ref (100 mg, 0.38 mmol), AlCl_3 (76 mg, 0.57 mmol) were dissolved in CH_2Cl_2 (10 ml) and refluxed for 30 min. After cooling down to room temperature, enantiopure *R*-Br-BINOL (337 mg, 0.76 mmol) in acetonitrile (5 ml) was added dropwise and stirred at room temperature for 6 h. Next, the solvent was washed with saturated NaCl aqueous solution, dried over anhydrous MgSO_4 and evaporated. Finally, the crude was purified by chromatography on silica gel eluting with hexane/ CH_2Cl_2 (1/1, v/v) and enantiopure *R*-0Ph-B-BODIPY (yield: 120 mg, 51%) was obtained. *S*-0Ph-B-BODIPY was synthesized by changing from enantiopure *R*-Br-BINOL to enantiopure *S*-Br-BINOL. ^1H NMR (CDCl_3) δ : 7.96 (2H, s), 7.65 (2H, d, $J = 10.0$ Hz), 7.21 (2H, d, $J = 8.6$ Hz), 7.14 (2H, d, $J = 8.6$ Hz), 7.02 (2H, d, $J = 8.1$ Hz), 5.86 (2H, s), 2.67 (3H, s), 2.43 (6H, s), 1.67 (6H, s). ^{13}C -NMR (CDCl_3) δ : 154.87 (s), 154.36 (s), 141.19 (s), 140.80 (s), 133.01 (s), 132.06 (s), 130.94 (s), 129.82 (s), 128.62 (s), 128.54 (s), 128.32 (s), 124.94 (s), 122.18 (s), 120.94 (s), 117.08 (s), 17.66 (s), 16.95 (s), 15.96 (s). High resolution MALDI-TOF MS: m/z calcd, 664.0532; found, 664.0515 [M].

Scheme S2. Synthetic Scheme of *R*-1Ph-BODIPY.



Synthesis of *R*-1Ph-B-BODIPY. 1Ph-BODIPY (100 mg, 0.30 mmol), AlCl_3 (59 mg, 0.44 mmol) were dissolved in CH_2Cl_2 (10 ml) and refluxed for 30 min. After cooling down to room temperature, enantiopure *R*-Br-BINOL (261 mg, 0.59 mmol) in acetonitrile (5 ml) was added dropwise and stirred at room temperature for 6 h. Next, the solvent was washed with saturated NaCl aqueous solution, dried over anhydrous MgSO_4 and evaporated. Finally, the crude was purified by chromatography on silica gel eluting with hexane/ CH_2Cl_2 (1/1, v/v) and enantiopure *R*-1Ph-B-BODIPY (yield: 115 mg, 55%) was obtained. *S*-1Ph-B-BODIPY was synthesized by changing from enantiopure *R*-Br-BINOL to enantiopure *S*-Br-BINOL. ^1H NMR (CDCl_3) δ : 7.96 (2H, s), 7.67 (1H, d, $J = 8.8$ Hz), 7.64 (1H, d, $J = 8.8$ Hz), 7.35-7.26 (3H, m), 7.23 (3H, d, $J = 8.8$ Hz), 7.21-7.17 (2H, m), 7.10 (1H, d, $J = 8.8$ Hz), 7.04-6.99 (4H, m), 5.85 (1H, s), 2.74 (3H, s), 2.45 (3H, s), 2.33 (3H, s), 1.71 (3H, s), 1.64 (3H, s). ^{13}C NMR (CDCl_3) δ : 154.82 (s), 154.75 (s), 152.51 (s), 141.38 (s), 140.87 (s), 137.08 (s), 133.91 (s), 133.42 (s), 132.82 (s), 132.10 (s), 132.01 (s), 131.07 (s), 130.80 (s), 130.35 (s), 129.84 (s), 129.69 (s), 129.49 (s), 128.65 (s), 128.57 (s), 128.54 (s), 128.50 (s), 128.40 (s), 128.16 (s), 127.80 (s), 127.70 (s), 126.91 (s), 125.05 (s), 124.72 (s), 122.21 (s), 121.31 (s), 120.48 (s), 117.16 (s), 117.08 (s), 17.71 (s), 17.47 (s), 15.79 (s), 15.73 (s), 14.94 (s). High resolution MALDI-TOF MS: m/z calcd, 740.0845; found, 740.0842 [M].

Scheme S3. Synthetic Scheme of *R*-2Ph-B-BODIPY



Synthesis of *R*-2Ph-B-BODIPY. 2Ph-BODIPY (100 mg, 0.24 mmol), AlCl₃ (50 mg, 0.37 mmol) were dissolved in CH₂Cl₂ (10 ml) and refluxed for 30 min. After cooling down to room temperature, enantiopure *R*-Br-BINOL (250 mg, 0.56 mmol) in acetonitrile (5 ml) was added dropwise and stirred at room temperature for 6 h. Next, the solvent was washed with saturated NaCl aqueous solution, dried over anhydrous MgSO₄ and evaporated. Finally, the crude was purified by chromatography on silica gel eluting with hexane/CH₂Cl₂ (1/1, v/v) and enantiopure *R*-2Ph-B-BODIPY (yield: 127 mg, 42%) was obtained. *S*-2Ph-B-BODIPY was synthesized by changing from enantiopure *R*-Br-BINOL to enantiopure *S*-Br-BINOL. ¹H NMR (CDCl₃) δ: 7.96 (2H, s), 7.66 (2H, d, *J* = 8.6 Hz), 7.33-7.31 (4H, m), 7.27-7.24 (3H, m), 7.19-7.17 (3H, m), 7.01 (2H, d, *J* = 9.1 Hz), 6.96 (4H, d, *J* = 7.2 Hz), 2.81 (3H, s), 2.35 (6H, s), 1.68 (6H, s). ¹³C NMR (CDCl₃) δ: 154.66 (s), 153.05 (s), 141.60 (s), 137.24 (s), 133.99 (s), 133.90 (s), 133.30 (s), 132.08 (s), 130.98 (s), 130.36 (s), 129.74 (s), 128.62 (s), 128.58 (s), 128.26 (s), 128.20 (s), 126.97 (s), 124.86 (s), 120.92 (s), 117.18 (s), 17.94 (s), 15.80 (s), 14.78 (s). High resolution MALDI-TOF MS: *m/z* calcd, 816.1158; found, 816.1168 [M].

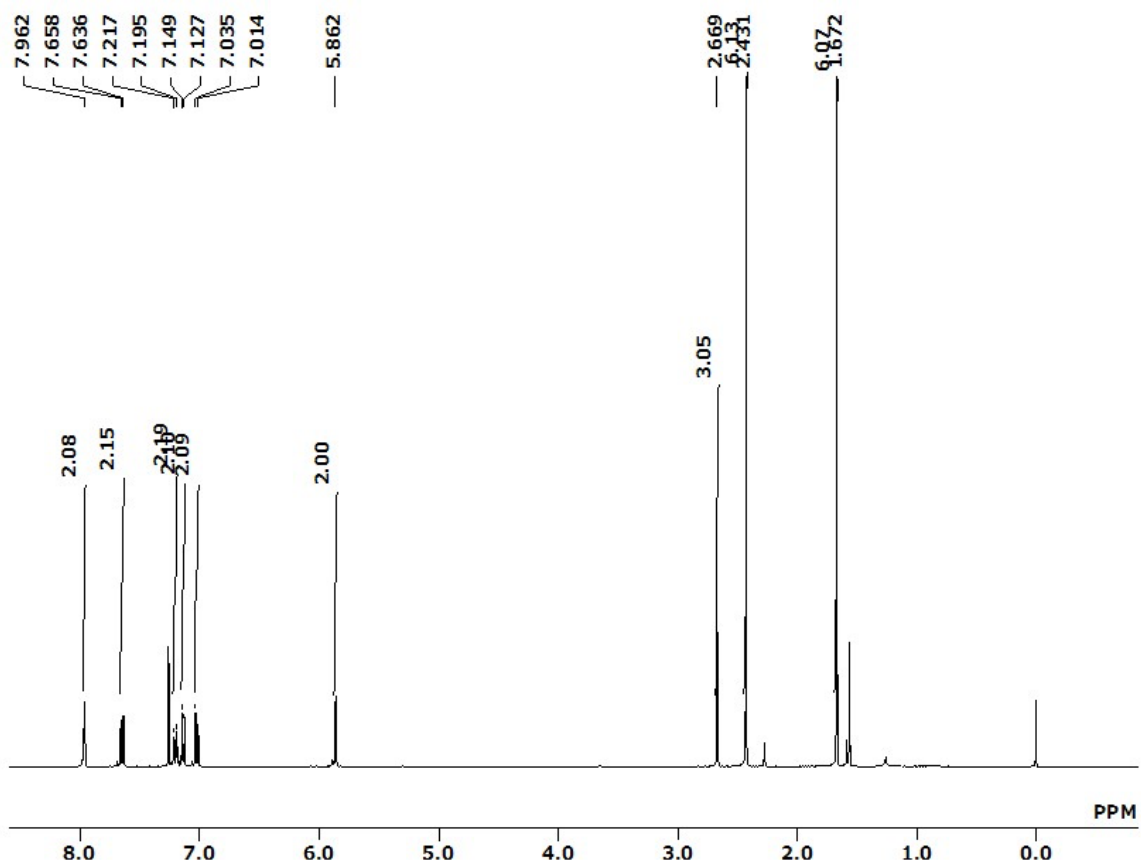


Fig. S1 ^1H NMR spectrum of *R*-0Ph-B-BODIPY.

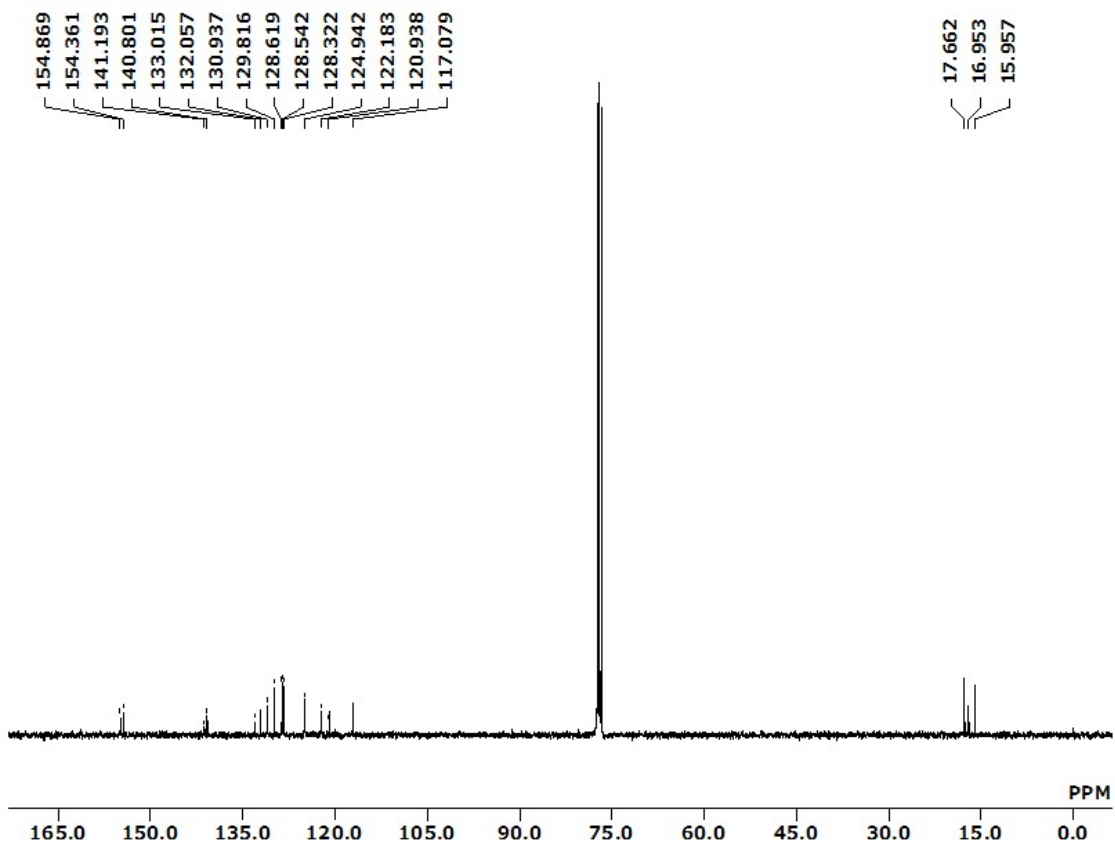


Fig. S2 ^{13}C NMR spectrum of *R*-0Ph-B-BODIPY.

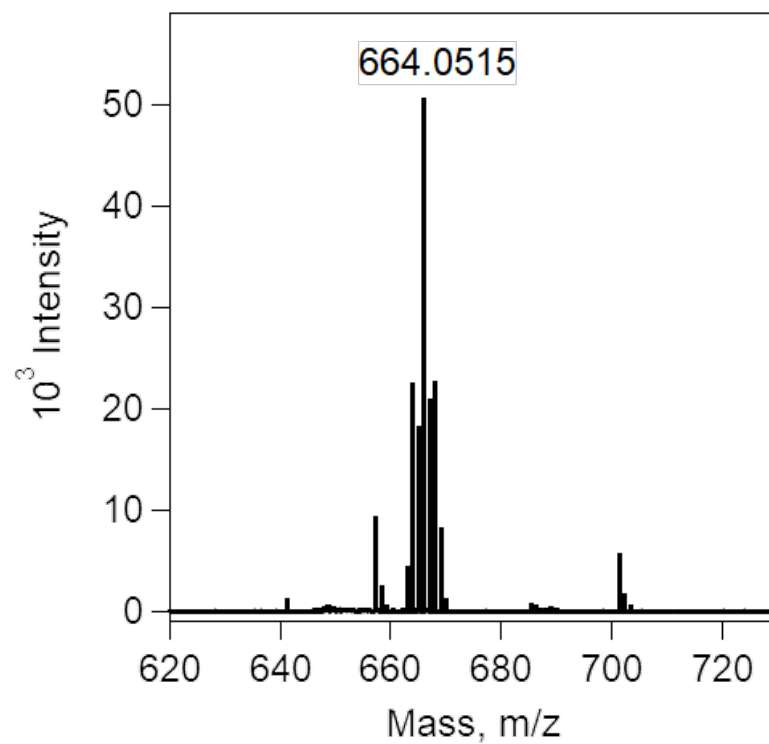


Fig. S3 High resolution MALDI-TOF MS spectrum of *R*-0Ph-B-BODIPY.

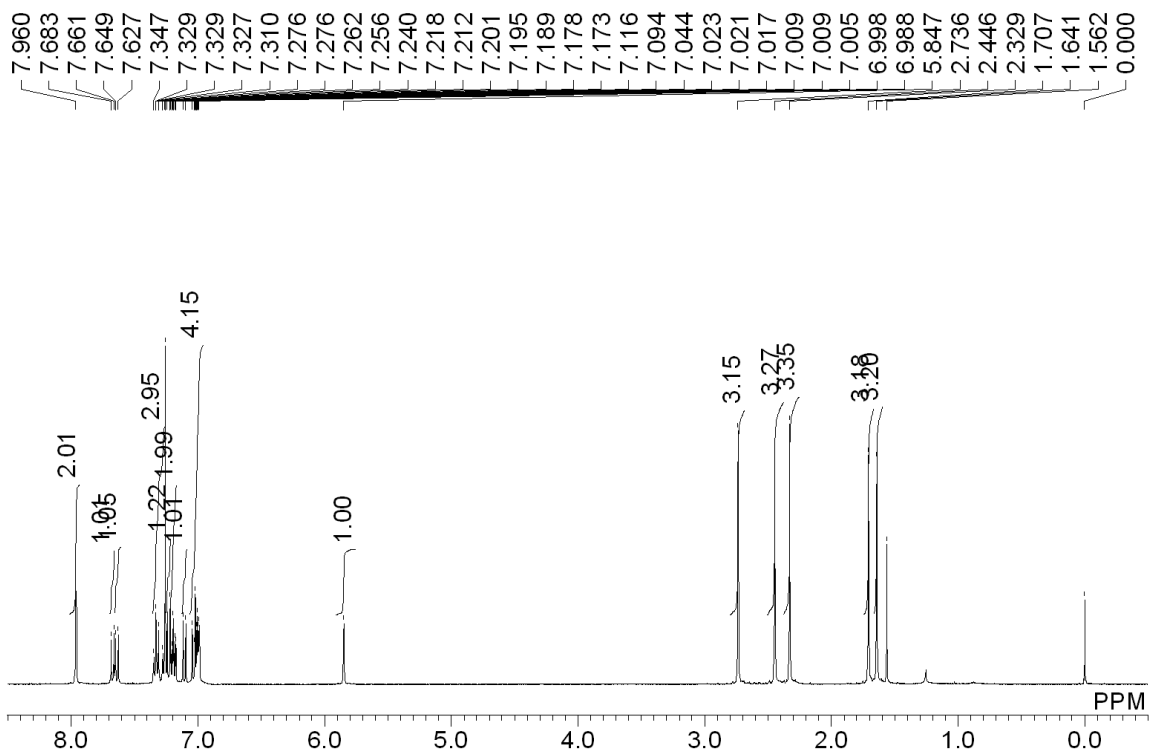


Fig. S4 ^1H NMR spectrum of *R*-1Ph-B-BODIPY.

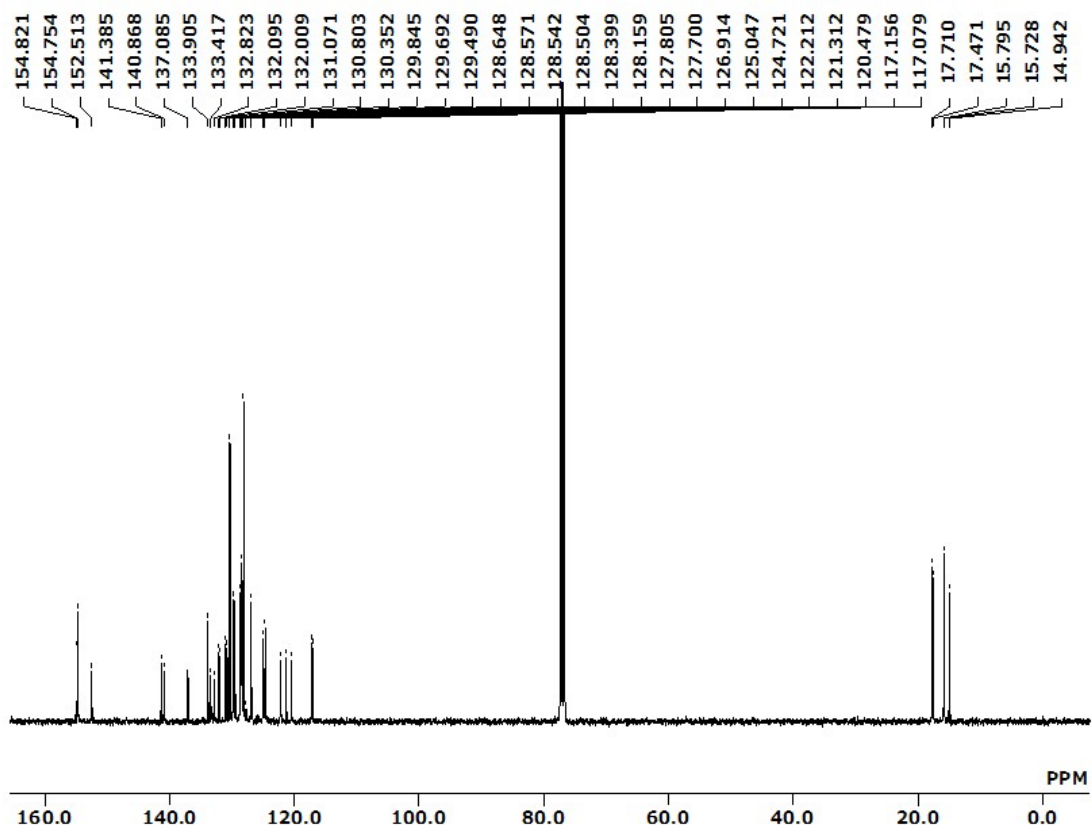


Fig. S5 ^{13}C NMR spectrum of *R*-1Ph-B-BODIPY.

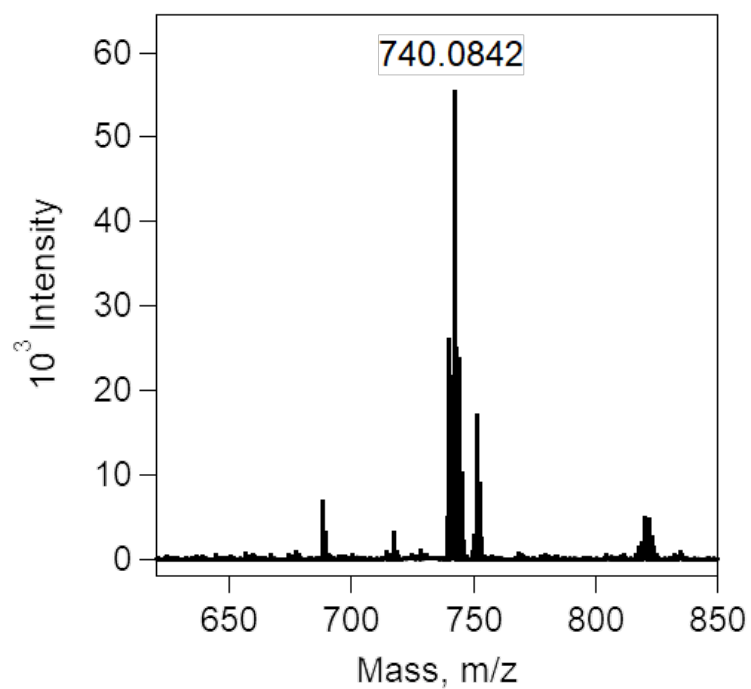


Fig. S6 High resolution MALDI-TOF MS spectrum of *R*-1Ph-B-BODIPY.

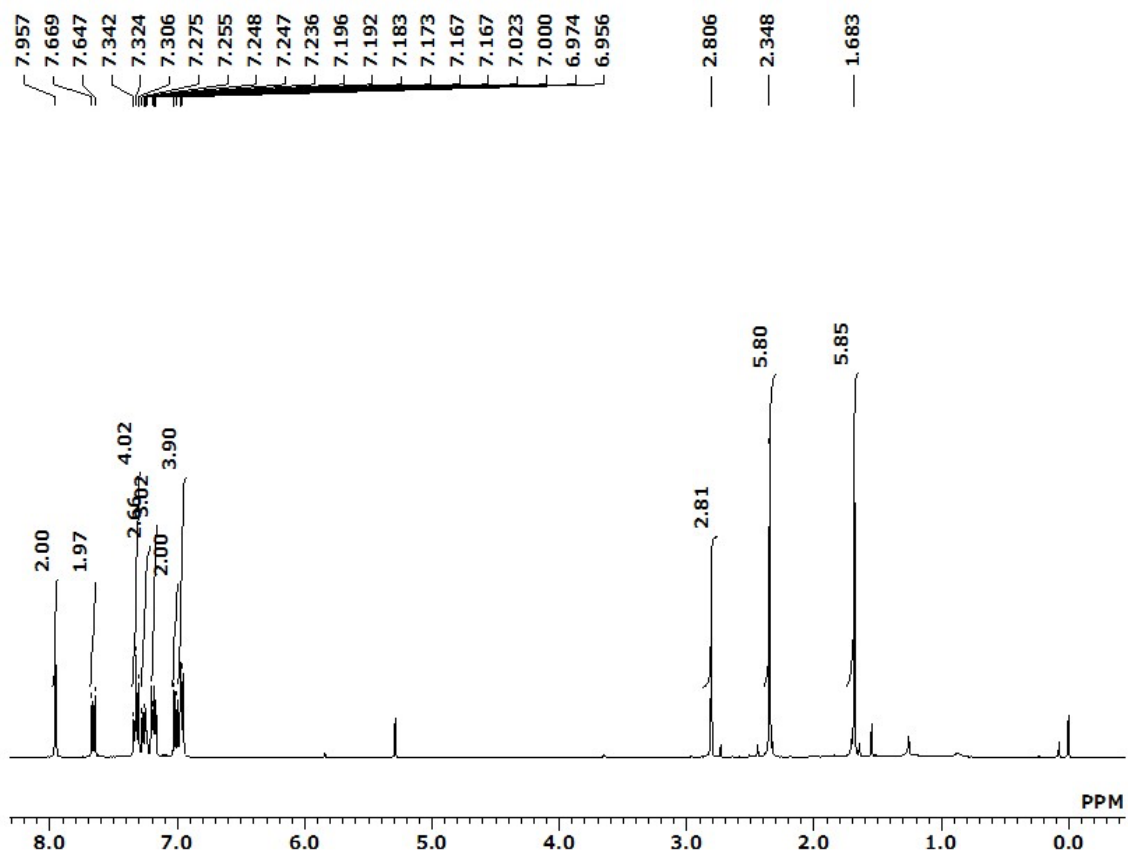


Fig. S7 ^1H NMR spectrum of *R*-2Ph-B-BODIPY.

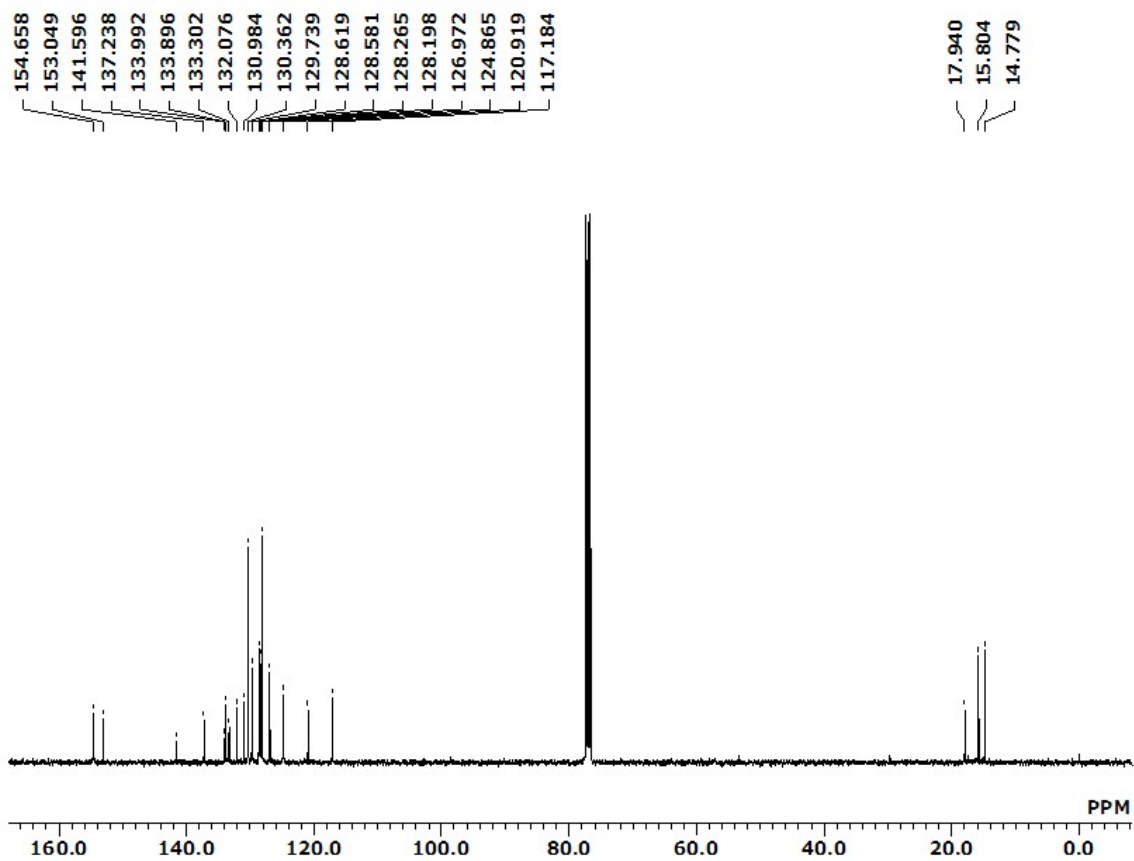


Fig. S8 ^{13}C NMR spectrum of *R*-2Ph-B-BODIPY.

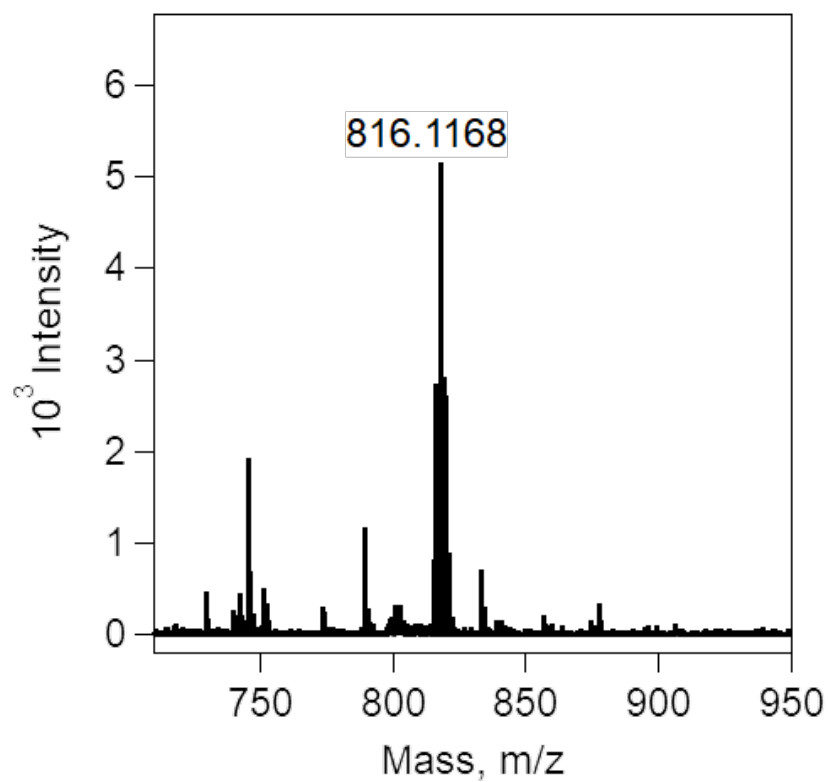


Fig. S9 High resolution MALDI-TOF MS spectrum of *R*-2Ph-B-BODIPY.

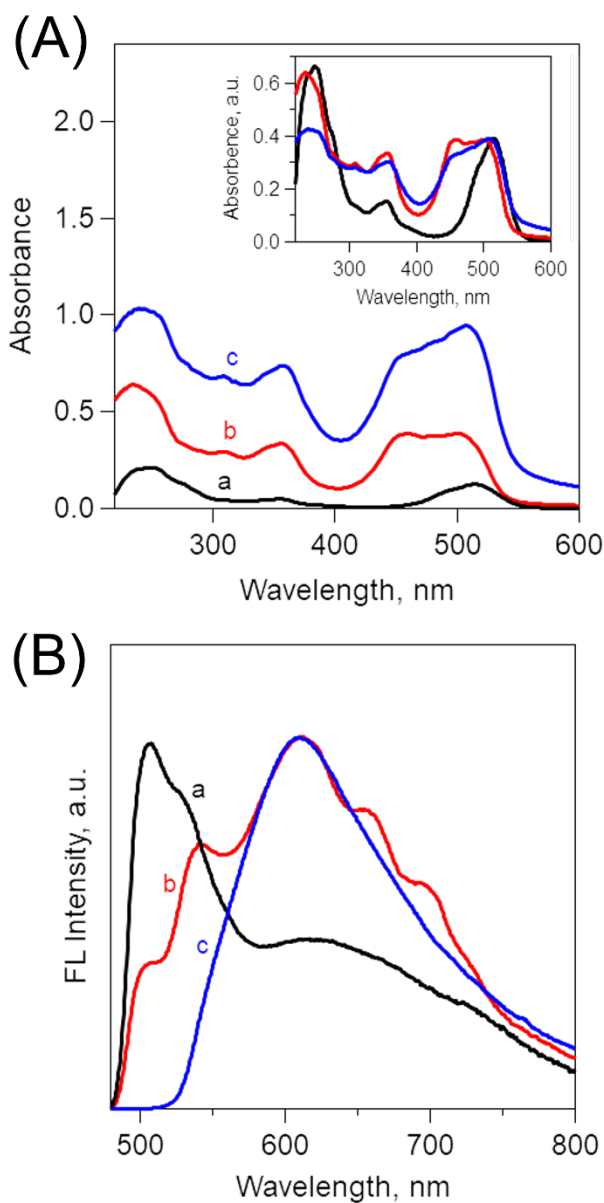


Fig. S10 (A) Concentration-dependent absorption spectra of (OPh-B-BODIPY)_m. (a) 3.0 μM, (b) 30 μM, and (c) 300 μM in H₂O/THF = 99/1 (v/v). The insertion figure indicated the normalized spectra for comparison. (B) Concentration-dependent fluorescence spectra of (OPh-B-BODIPY)_m, (a) 3.0 μM, (b) 30 μM, and (c) 300 μM. $\lambda_{\text{ex.}} = 470$ nm. We can see a broad fluorescence spectrum derived from molecular aggregation under the experimental condition of 30 μM in H₂O/THF = 99/1 (v/v).

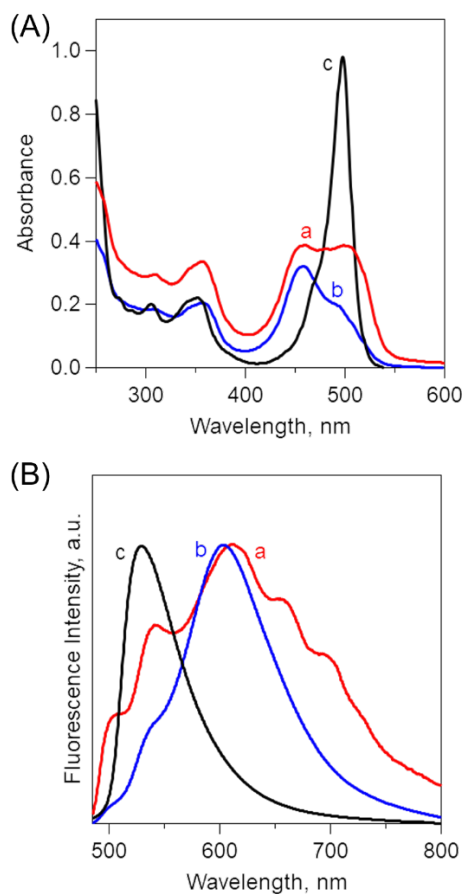


Fig. S11 (A) Absorption spectra of (0Ph-B-BODIPY)_m in a mixed solvent of H₂O and THF with different volume ratios. (a) H₂O/THF = 99/1 (v/v), (b) H₂O/THF = 70/30 (v/v), and (c) H₂O/THF = 50/50 (v/v). The final concentrations of 0Ph-B-BODIPY were fixed as constant (= 30 μM). (B) Absorption spectra of (0Ph-B-BODIPY)_m in a mixture solvent of THF and water with different ratios. (a) H₂O/THF = 99/1 (v/v), (b) H₂O/THF = 70/30 (v/v), and (c) H₂O/THF = 50/50 (v/v). The final concentration was fixed as constant (= 30 μM). $\lambda_{\text{ex.}} = 470$ nm.

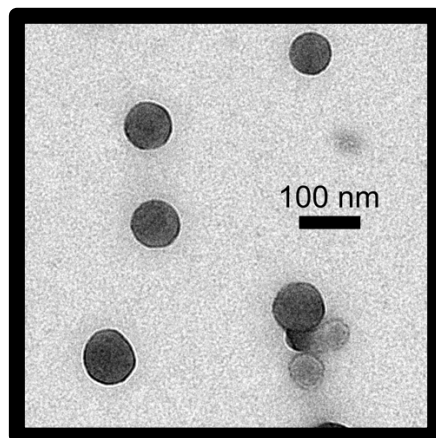


Fig. S12 TEM image of $(0\text{Ph-B-BODIPY})_m$ in $\text{H}_2\text{O}/\text{THF} = 70/30$ (v/v). The final concentration of 0Ph-B-BODIPY was $30 \mu\text{M}$ in $\text{H}_2\text{O}/\text{THF}$.

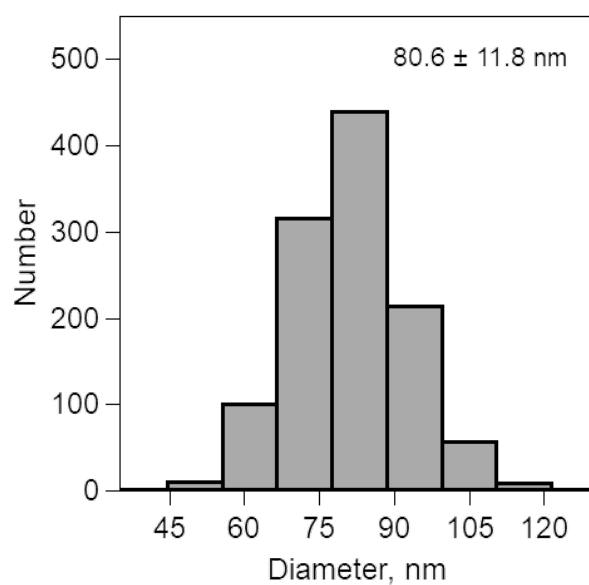


Fig. S13 Diameter-distributions of (0Ph-B-BODIPY)_m in H₂O/THF = 70/30 (v/v) analysed by TEM images. The final concentration of 0Ph-B-BODIPY is 30 μM in H₂O/THF.

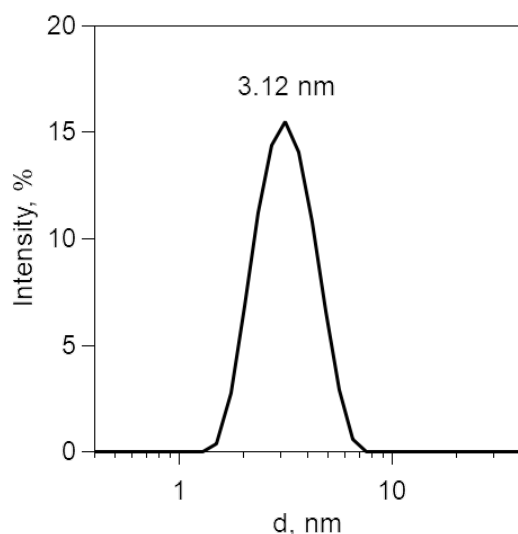


Fig. S14 Size-distributions of (0Ph-B-BODIPY)_m in H₂O/THF = 50/50 (v/v) (final concentration of 0Ph-B-BODIPY: [0Ph-B-BODIPY] = 30 μM in H₂O/THF) by dynamic light scattering (DLS) measurements.

To discuss the solvent ratio-dependent structural changes, the following two different (0Ph-B-BODIPY)_m systems with volume ratios such as H₂O/THF = 70/30 and 50/50 were prepared by maintaining the final concentrations (30 μM) in mixed solvents. Absorption and fluorescence spectra and TEM images of these systems prepared in H₂O/THF = 70/30 and 50/50 (v/v) are totally different from those prepared in H₂O/THF = 99/1 (v/v) (Figs. S11-S14 in ESI). Isotropic spherical assemblies prepared in H₂O/THF = 70/30 (v/v) (Figs. S12-S13) are in sharp contrast with the anisotropic fibrous assemblies in H₂O/THF = 99/1 (v/v) (Fig. 2A in the text). Furthermore, absorption and fluorescence spectra (Fig. S11) and dynamic light scattering (DLS) result (Fig. S14) in H₂O/THF = 50/50 (v/v) demonstrated a monomer-like behavior, but not aggregate formations (No specific aggregate structures were observed by TEM measurements). Thus, such fibrous assemblies could be fabricated only under the excess volume condition of H₂O relative to THF such as 30 μM 0Ph-B-BODIPY in H₂O/THF = 99/1 (v/v).

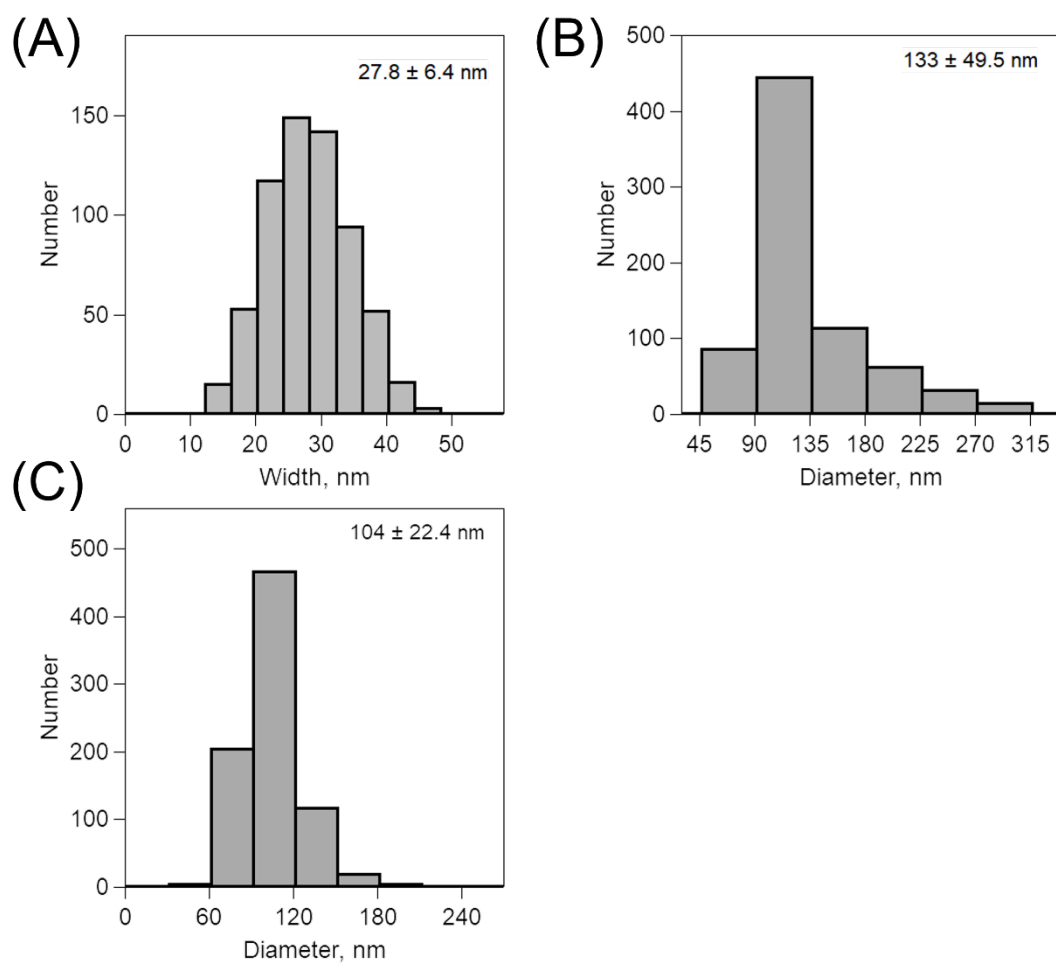


Fig. S15 The width and diameter-distributions of B-BODIPY assemblies. (A) The width-distribution of (0Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v) ([0Ph-B-BODIPY] = 30 μM), (B) the diameter-distribution of (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v) ([1Ph-B-BODIPY] = 30 μM), and (C) the diameter-distribution of (2Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v) ([2Ph-B-BODIPY] = 30 μM).

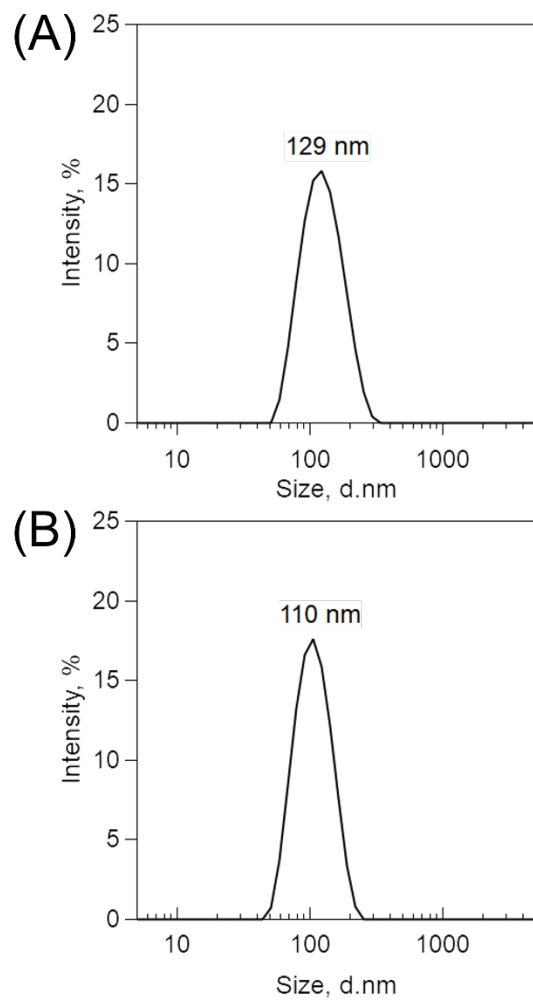


Fig. S16 Size-distributions of (A) (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v) ([1Ph-B-BODIPY] = 30 μM) and (B) (2Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v) ([2Ph-B-BODIPY] = 30 μM) by dynamic light scattering (DLS) measurements.

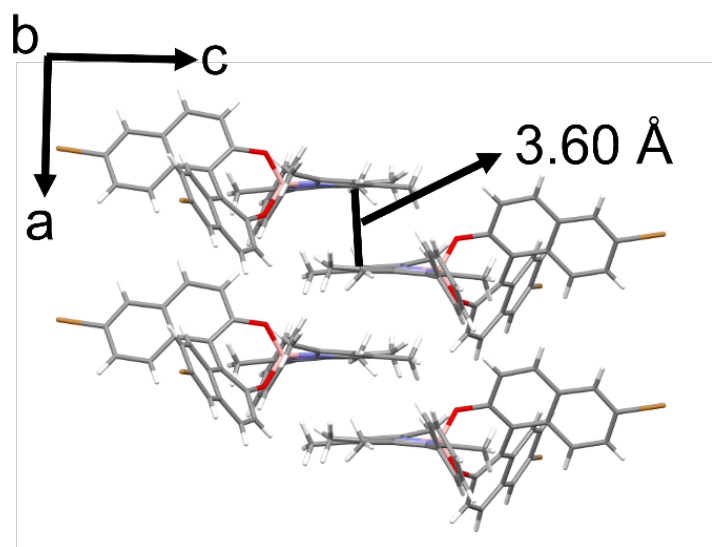


Fig. S17 Single-crystal structures of π -stacking between two neighboring *R*-1Ph-B-BODIPY. Black line: π -stacking distance: 3.60 Å.

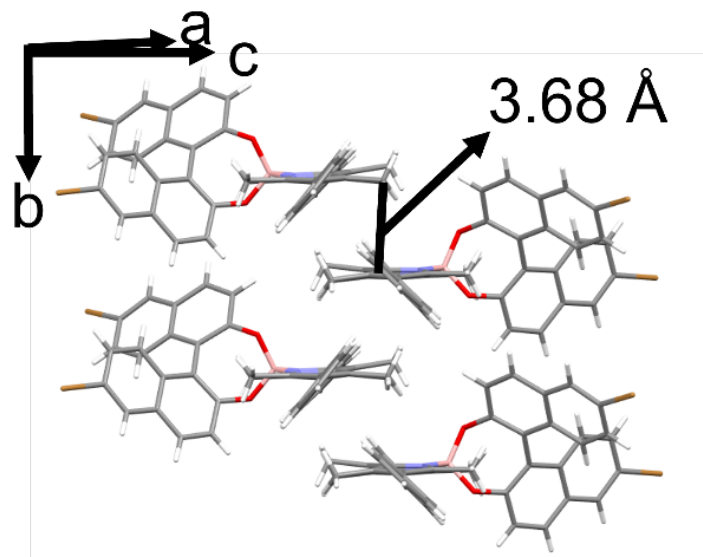


Fig. S18 Single-crystal structures of π -stacking between two neighboring *R*-2Ph-BODIPY. Black line: π -stacking distance: 3.68 Å.

Table S1. Single crystal structures and crystallographic data of *R*-0Ph-B-BODIPY.

Compound	<i>R</i> -0Ph-B-BODIPY
Formula	C ₃₄ H ₂₇ BBr ₂ N ₂ O ₂
Formula Weight	666.22
Crystal System	Orthorhombic
Space Group	P2 ₁ 2 ₁ 2 ₁ (#19)
a, Å	6.8301(16)
b, Å	21.461(5)
c, Å	22.842(5)
α , degree	90.0000
β , degree	90.018(6)
γ , degree	90.0000
V, Å ³	3348.2(14)
Z	4
D _{calc} , g cm ⁻³	1.322

Table S2. Single crystal structures and crystallographic data of *R*-1Ph-B-BODIPY.

Compound	<i>R</i> -1Ph-B-BODIPY
Formula	C ₄₆ H ₄₅ BBr ₂ N ₂ O ₂
Formula Weight	828.49
Crystal System	Orthorhombic
Space Group	P2 ₁ 2 ₁ 2 ₁ (#19)
a, Å	6.7759(3)
b, Å	22.5581(11)
c, Å	26.3420(13)
α , degree	90.0000
β , degree	90.0000
γ , degree	90.0000
V, Å ³	4026.4(3)
Z	4
D _{calc} , g cm ⁻³	1.367

Table S3. Single crystal structures and crystallographic data of *R*-2Ph-B-BODIPY.

Compound	<i>R</i> -2Ph-B-BODIPY
Formula	C ₄₆ H ₃₅ BBr ₂ N ₂ O ₂
Formula Weight	818.41
Crystal System	Monoclinic
Space Group	P2 ₁ (#4)
a, Å	14.0582(3)
b, Å	7.39498(13)
c, Å	17.5218(3)
α , degree	90.0000
β , degree	93.311(7)
γ , degree	90.0000
V, Å ³	1818.53(6)
Z	2
D _{calc} , g cm ⁻³	1.495

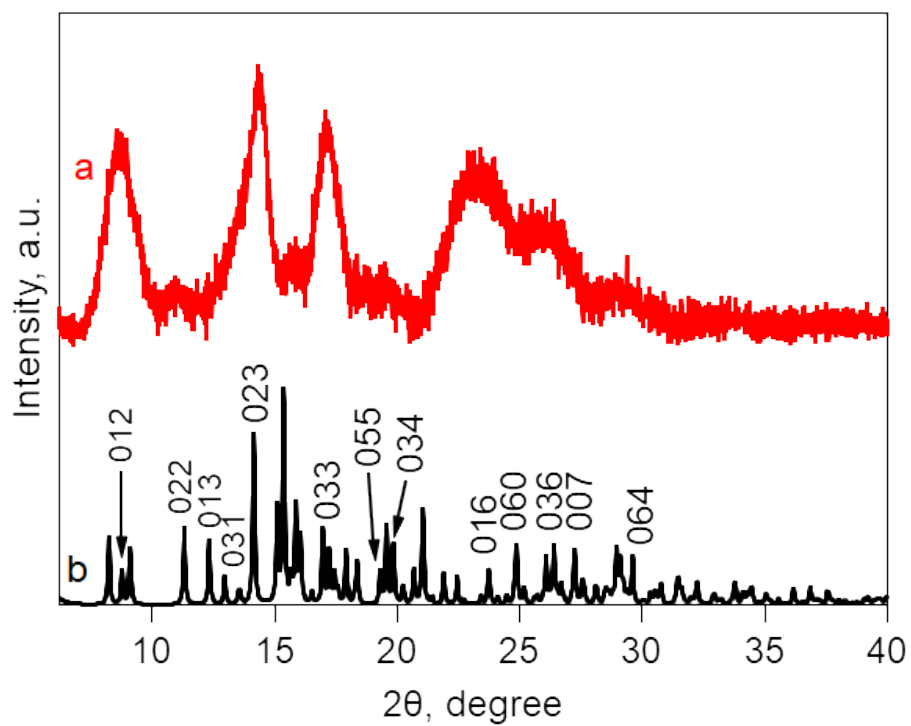


Fig. S19 XRD patterns of (a) $(0\text{Ph-B-BODIPY})_m$ prepared in $\text{H}_2\text{O}/\text{THF} = 99/1$ (v/v) and (b) a simulated pattern from the crystal structure of 0Ph-B-BODIPY.

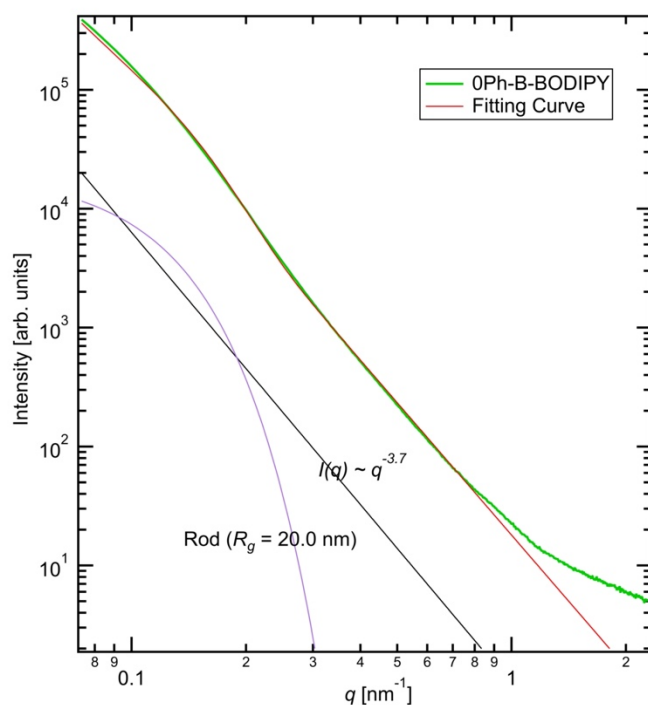


Fig. S20 Small-angle X-ray scattering (SAXS) profile of (0Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v).

In SAXS profile of (0Ph-B-BODIPY)_m, we fitted the profile with eq. (2), which is based on the cross-section plot, a method for analyzing fibrous aggregates.

$$I(q) = aq^{-3.7} + be^{-\frac{1}{2}q^2R_1^2} \quad (2)$$

Here, $I(q)$ is scattering intensity, a , and b are constant, q is scattering angle, and R_1 is radius of gyration for rods. The profile in the low- q region is described by $I(q) \sim q^{-3.7}$. The scattering profile originates from the smooth interface owing to the surface dimension of 2.3. This interface is believed to be derived from micron-scaled particles. Eq. (2) could be fitted to the profile, and SAXS data also suggested fibrous assemblies of (0Ph-B-BODIPY)_m. Consequently, R_1 was calculated to be 20.0 nm, which is comparable to the width value in TEM image (Fig.2A and Fig. S11A in ESI†).

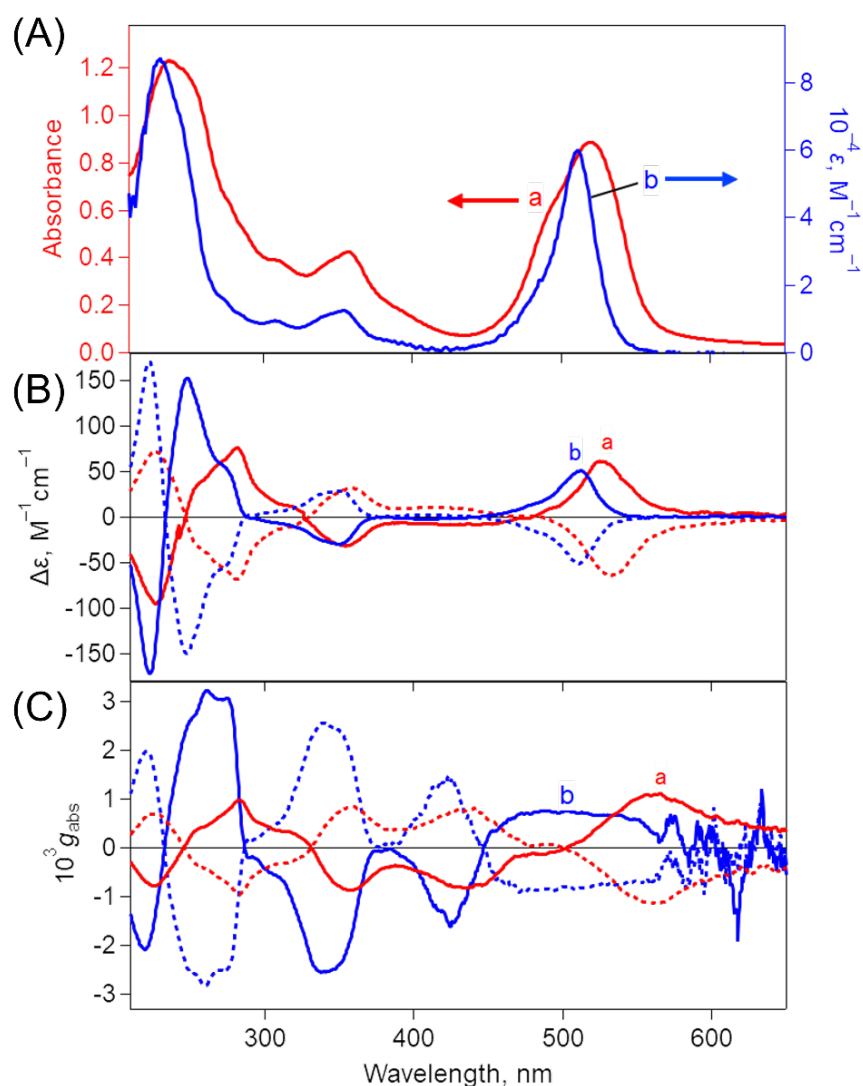


Fig. S21 (A) Absorption spectra of (a) (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [1Ph-B-BODIPY] = 30 μM and (b) 1Ph-B-BODIPY in THF. (B) CD spectra of (a) (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [1Ph-B-BODIPY] = 30 μM and (b) 1Ph-B-BODIPY in THF. Solid and dotted lines correspond to *S* and *R* forms, respectively. (C) Dissymmetry factor (g_{abs}) profiles of (a) (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [1Ph-B-BODIPY] = 30 μM and (b) 1Ph-B-BODIPY in THF. Solid and dotted lines correspond to *S* and *R* forms, respectively.

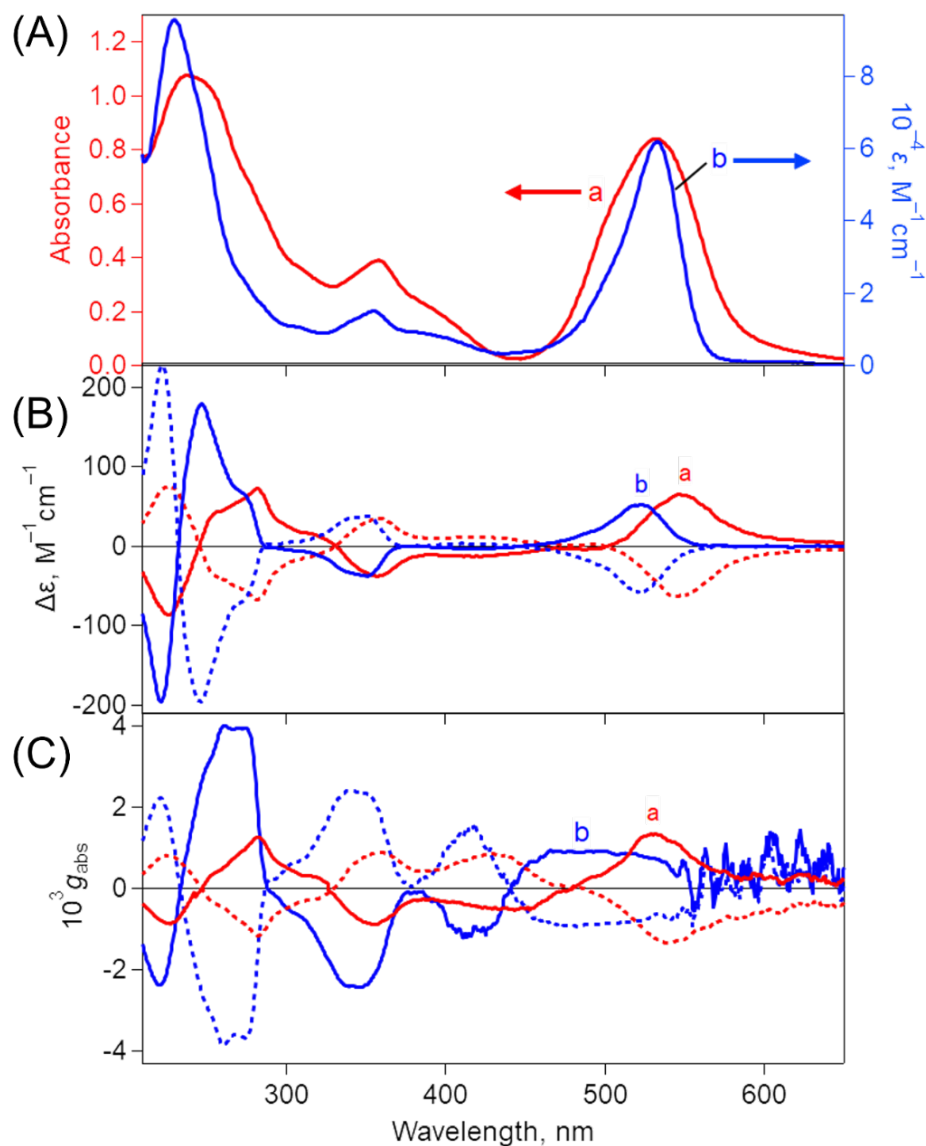


Fig. S22 (A) Absorption spectra of (a) (2Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [2Ph-B-BODIPY] = 30 μM and (b) 2Ph-B-BODIPY in THF. (B) CD spectra of (a) (2Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [1Ph-B-BODIPY] = 30 μM and (b) 2Ph-B-BODIPY in THF. Solid and dotted lines correspond to *S* and *R* forms, respectively. (C) Dissymmetry factor (g_{abs}) profiles of (a) (2Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v). [2Ph-B-BODIPY] = 30 μM and (b) 2Ph-B-BODIPY in THF. Solid and dotted lines correspond to *S* and *R* forms, respectively.

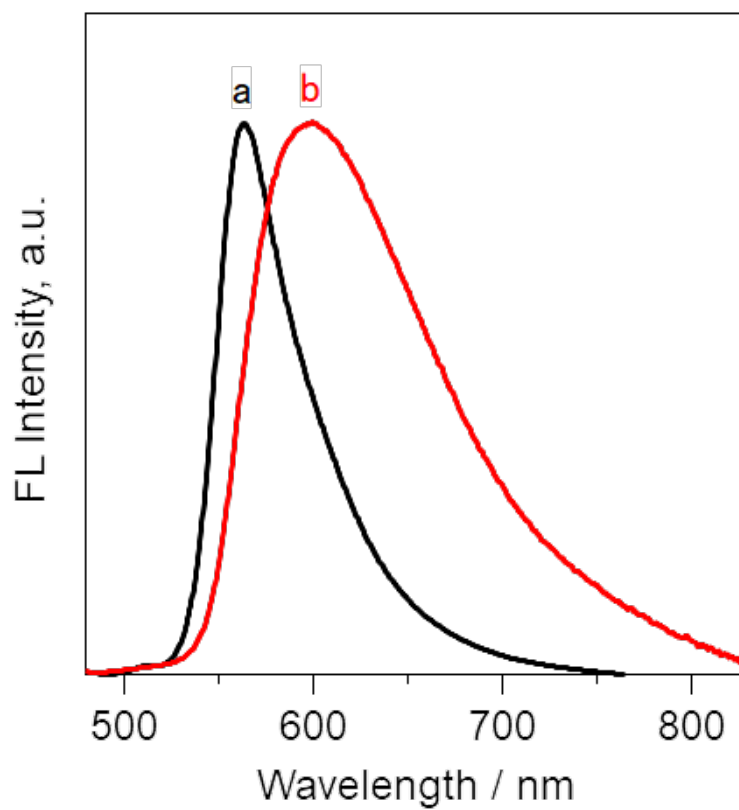


Fig. S23 Fluorescence spectra of (a) 2Ph-B-BODIPY (monomer: black line) in THF and (b) (2Ph-B-BODIPY)_m (red line) prepared in H₂O/THF = 99/1 (v/v). λ_{ex} : 470 nm.

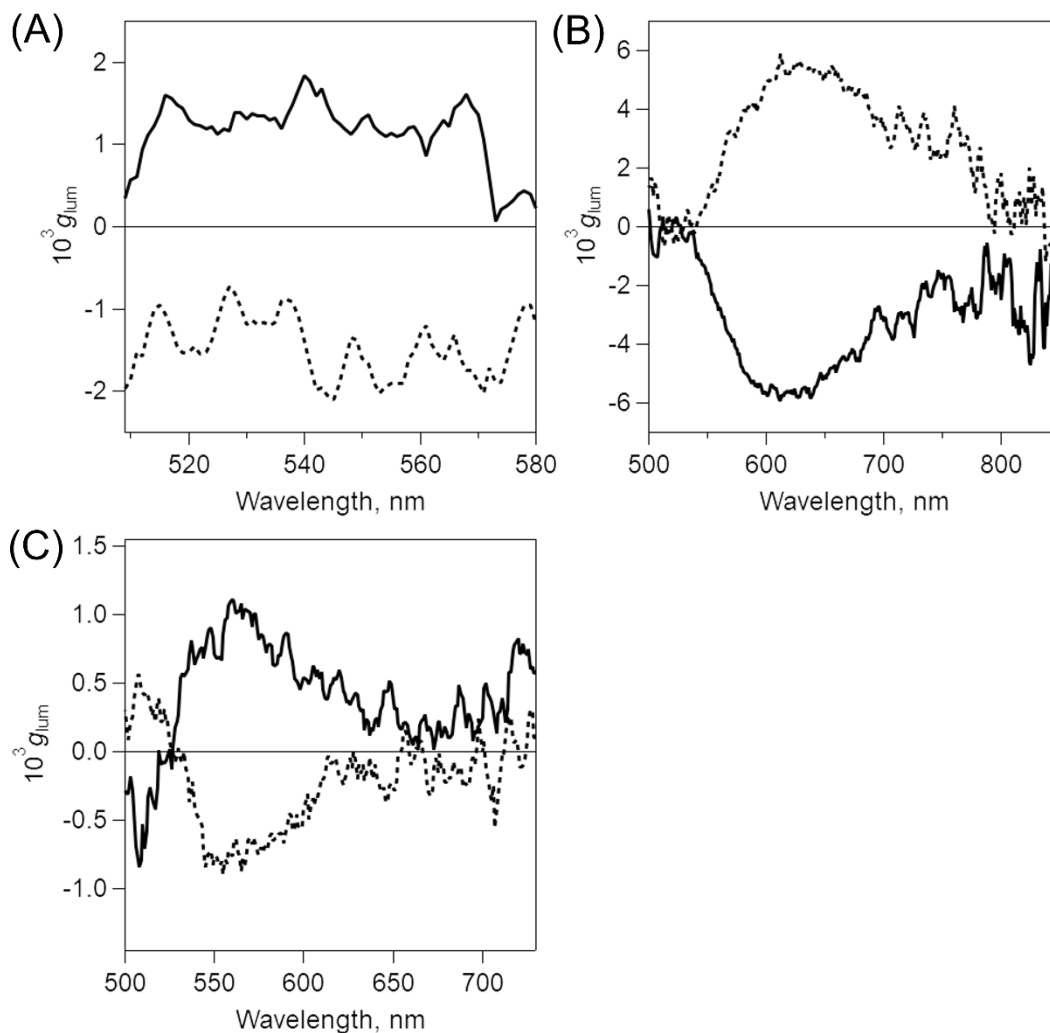


Fig. S24 Dissymmetry factor (g_{lum}) profiles corresponding to CPL spectra of (A) 0Ph-B-BODIPY (monomer) in THF, λ_{ex} : 350 nm. (B) (0Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v), λ_{ex} : 350 nm. (C) (1Ph-B-BODIPY)_m prepared in H₂O/THF = 99/1 (v/v), λ_{ex} : 350 nm. Solid and dotted lines correspond to *S* and *R* forms, respectively.

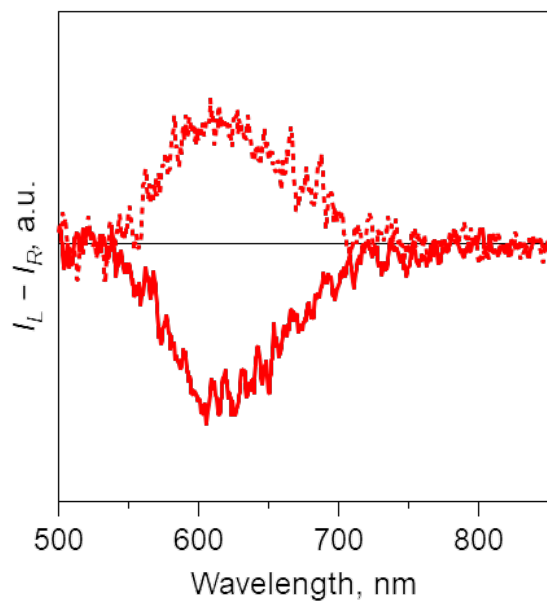


Fig. S25 CPL spectra of (OPh-B-BODIPY)_m prepared in H₂O/THF = 70/30 (v/v). [OPh-B-BODIPY] = 30 μM, λ_{ex}: 350 nm. Solid and dotted lines correspond to *S* and *R* forms, respectively.

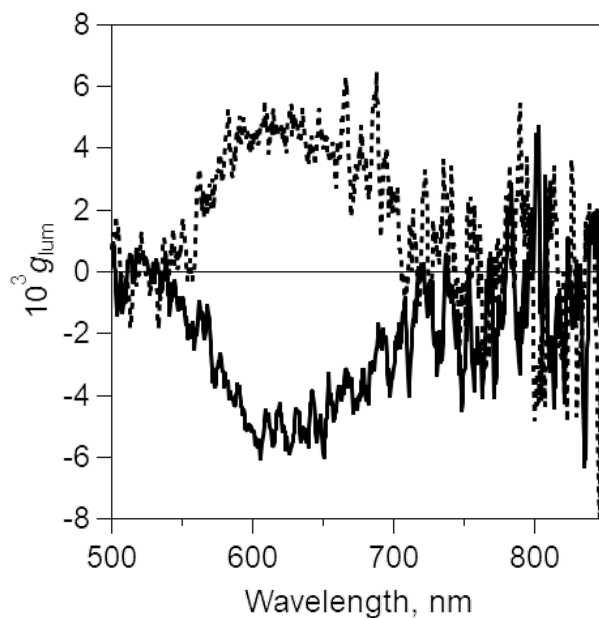


Fig. S26 Dissymmetry factor (g_{lum}) profiles corresponding to CPL spectra of $(0\text{Ph-B-BODIPY})_m$ prepared in $\text{H}_2\text{O}/\text{THF} = 70/30$ (v/v). $[0\text{Ph-B-BODIPY}] = 30 \mu\text{M}$, $\lambda_{ex}: 350 \text{ nm}$. Solid and dotted lines correspond to *S* and *R* forms, respectively.

$(0\text{Ph-B-BODIPY})_m$ prepared in $\text{H}_2\text{O}/\text{THF} = 70/30$ (v/v) demonstrated broader CPL spectra relative to the monomers, whereas the corresponding g_{lum} value slightly decreased as compared to that in $\text{H}_2\text{O}/\text{THF} = 99/1$ (v/v) (Figs.S25 and S26). The plausible reason is attributable due to the relative weak interaction between nearby 0Ph-B-BODIPY molecules. In contrast, in the case of $(0\text{Ph-B-BODIPY})_m$ prepared in $\text{H}_2\text{O}/\text{THF} = 50/50$ (v/v), unfortunately we could not observe appropriate CPL spectra despite the repeated measurements due to the unstable suspension of $(0\text{Ph-B-BODIPY})_m$ in $\text{H}_2\text{O}/\text{THF}$. This strongly suggested that the excess volume condition of H_2O (poor solvent) relative to THF (good solvent) (e.g., $\text{H}_2\text{O}/\text{THF} = 99/1$ (v/v)) play an important role for preparation of stable aggregate structures.