High Voltage and Superior Cyclability of Indium Hexacyanoferrate Cathodes for Aqueous Na-ion Batteries Enabled by Superconcentrated NaClO₄ Electrolytes

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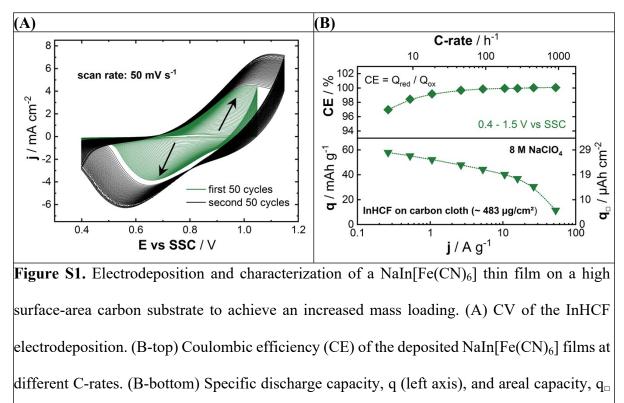
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Experimental

Electrodeposited NaIn[Fe(CN)₆] films with high mass loading:

The electrochemical deposition of NaIn[Fe(CN)₆] thin film on a circular piece (Φ = 12 mm) of conductive carbon cloth was performed similar to the method for Au quartz crystals in an electrochemical glass cell in three-electrode configuration under inert argon atmosphere. In short, the substrate was submerged in an aqueous solution containing 2 mM K₃[Fe(CN)₆] (99%, Sigma-Aldrich), 2 mM In(III)Cl₃ (\geq 99.9%, Carl Roth) and 0.25 M Na₂SO₄ (\geq 99%, Sigma-Aldrich). The deposition was performed by means of cyclic voltammetry with a scan rate of 50 mV/s in two different potential ranges (see **Figure** S1; first 50 cycles: 0.4 – 1.05 V *vs* SSC; followed by 50 cycles: 0.4 – 1.15 V *vs* SSC). After the synthesis, the sample was dried in the cell for 1 hour in argon atmosphere. The performance of the InHCF electrode, as shown in **Figure S1B**, was investigated by galvanostatic cycling in 8 M NaClO₄ within the potential range from 0.4 V to 1.5 V *vs* SSC.

An approximate mass loading of 483 μ g cm⁻² was calculated for the electrode using its measured maximum capacity (28 μ Ah cm⁻², see **Figure S1B**) and the experimentally determined specific capacity for InHCF of 58 mAh g⁻¹.



(right axis), of the NaIn[Fe(CN)₆] films at different current densities. An approximate mass loading of 483 μ g cm⁻² was calculated assuming a specific capacity of 58 mAh g⁻¹. The dotted lines in (B) serve as a guide to the eye.