## SUPPORTING INFORMATION

# "Lighting up" fluoride: cellular imaging and zebrafish model interrogations using a simple ESIPT based mycophenolic acid precursor-based probe <br> Neha Jain, *a Prasad M. Sonawane, *a Haoyan Liu, ${ }^{\text {b }}$ Arkaprava Roychaudhury, ${ }^{\text {c }}$ Youngseob Lee, ${ }^{\text {d }}$ Jongkeol An, ${ }^{\text {a }}$ Donghyeon Kim, ${ }^{\text {a }}$ Dongwook Kim, e Yunsu Kim, ${ }^{\text {f }}$ Yeu-Chun Kim, ${ }^{\text {b }}$ Kyung-Bin Cho, ${ }^{\text {d Hee- }}$ Sung Park, ${ }^{\text {f }}$ Cheol-Hee Kim, ${ }^{* c}$ David G. Churchill*a,b 

a. Department of Chemistry, Korea Advanced Institute of Science and Technology (KAIST), Daejeon 34141, Republic of Korea. E-mail: dchurchill@kaist.ac.kr
${ }^{\text {b. }}$ Department of Chemical \& Biomolecular Engineering, Korea Advanced Institute of Science and Technology (KAIST), Daejeon 34141, Republic of Korea c. Department of Biology, Chungnam National University, Daejeon 34134, Republic of Korea
d. Department of Chemistry, Jeonbuk National University and Research Institute for Physics and Chemistry, Jeonju 54896, Republic of Korea
e. Center for Catalytic Hydrocarbon Functionalizations, Institute for Basic Science. (IBS), Department of Chemistry, Korea Advanced Institute of Science and Technology (KAIST), Daejeon 34141, Republic of Korea.
f. KAIST Institute for Health Science and Technology (KIHST) (Therapeutic Bioengineering Section), Daejeon 34141, Republic of Korea

## Corresponding Authors

*Email: zebrakim@cnu.ac.kr (C. H. Kim)
*Email: dchurchill@kaist.ac.kr (D. G. Churchill)
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## General Information

All chemical reagents and solvents were purchased from the commercial suppliers (Aldrich, Tokyo Chemical Industry etc.), and used without further purification. Reactions were monitored by thin-layer chromatography (TLC). Column chromatography separations were conducted with silica gel (200-300 mesh). ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}, \mathrm{COSY}, \mathrm{HMBC}, \mathrm{HSQC}$, and NOESY NMR spectra were conducted on a Bruker Avance 400 MHz instrument. TMS was used as an external solvent. Chemical shifts are reported in the standard notation ( $\delta$ ) of ppm relative to the residual solvent peak at $7.26\left(\mathrm{CDCl}_{3}\right)$ for ${ }^{1} \mathrm{H}$ and $77.16\left(\mathrm{CDCl}_{3}\right)$ for ${ }^{13} \mathrm{C}$ as an internal solvent. The signal multiplicities were denoted by the abbreviations as s (singlet), d (doublet), t (triplet) and m (multiplet) etc. A Bruker Alpha ATR FTIR Spectrometer was used to collect infrared (IR) spectra. The frequency units are reciprocal centimeters ( $\mathrm{cm}^{-1}$ ). Bands can be depicted as broad (br), strong (s), medium (m), or weak (w). The electrospray ionization mass spectrometry (ESIMS) was performed on a BRUKER micrOTOF-Q II by the research support staff at KAIST. A Time-ofFlight mass spectrometer was measured at a resolution of 20000 . Absorption spectra were measured using a JASCO V-530 spectrophotometer. Fluorescence measurements were carried out with a Shimadzu RF-5301pc spectrofluorophotometer.

## Experimental Section

## Synthesis of Myco-F probe

The Myco-OH was prepared by published literature methods (see reference 25 in the main text).
Myco-OH ( $0.500 \mathrm{~g}, 1.85 \mathrm{mmol}$ ) was dissolved in DMF ( 15 mL ) in the $\mathrm{N}_{2}$ environment and stirred for 5 minutes. Then, this solution was treated with imidazole ( $0.277 \mathrm{~g}, 4.00 \mathrm{mmol}$ ) and stirred for an additional 15 minutes to afford a clear solution. After that time, tert-butyldimethylsilyl chloride ( 0.474 $\mathrm{g}, 3.00 \mathrm{mmol}$ ) was added slowly at $0^{\circ} \mathrm{C}$. This mixture was stirred for 6 h at room temperature. The reaction was monitored by TLC. After the reaction was determined to be finished by TLC, water was added into the reaction mixture; the organic layer was then extracted by diethyl ether ( $3 \times 25 \mathrm{~mL}$ ). The combined organic layer was washed with 25 mL saturated NaCl solution and was then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by silica gel flash chromatography using n hexane/EtOAc (16:1) as an eluent to afford Myco-F as a white solid ( $0.51 \mathrm{~g}, 71.7 \%$ yield, M. P. 135$136^{\circ} \mathrm{C}$ ). ${ }^{1 \mathrm{H}} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.28\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{6}\right), 3.11\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{H}_{2}\right), 2.68\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{H}_{10}\right), 1.05(\mathrm{~s}, 9 \mathrm{H}$, $\left.\mathrm{H}_{13}\right), 0.27\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{H}_{12}\right),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $168.18\left(\mathrm{C}_{3}\right), 166.13\left(\mathrm{C}_{1}\right), 150.94\left(\mathrm{C}_{7}\right), 133.14\left(\mathrm{C}_{5}\right)$, $131.14\left(\mathrm{C}_{9}\right), 130.87\left(\mathrm{C}_{8}\right), 130.40\left(\mathrm{C}_{6}\right), 119.65\left(\mathrm{C}_{4}\right), 30.95\left(\mathrm{C}_{12}\right), 25.51\left(\mathrm{C}_{13}\right), 23.74\left(\mathrm{C}_{2}\right), 18.31\left(\mathrm{C}_{10}\right), 16.39$ ( $\mathrm{C}_{11}$ ). MS/EI m/z C16H22BrNO3Si M+Na, 406.0449, found 406.0449.

## Spectral analysis measurements

A stock solution of Myco-F (10 mM) was prepared by dissolving the probe ( 3.8 mg ) in 1 mL of THF, and was further diluted in PBS ( pH 7.4 ) solution with various concentrations to allow for reduced probe concentrations for further studies. All the absorbance and emission spectra were examined in PBS (pH 7.4) solution at an excitation wavelength of 398 nm . Further, the stock solution of analytes was prepared in PBS ( 7.4 pH ). For determining the limit of detection, the fluorescent titration intensity of Myco-F ( 518 nm ) was used to obtain the value of the slope. The detection limit was calculated with the following equation

## Detection limit $=3 \sigma / k$

In which, $\sigma$ is the standard deviation of the 10 control (Myco-F without addition of the fluoride ion) and $k$ is the absolute value of the slope between the fluorescence intensity of Myco-F versus the fluoride ion concentration.

## DFT calculations

All calculations in this research used Density Functional Theory ${ }^{[a]}$ through the Gaussian 16 package ${ }^{[b]}$ with unrestricted B3LYP functional. ${ }^{[c]}$ The Def2-SVP basis set ${ }^{[d]}$ was used for structure optimizations and frequency calculations. The Def2-TZVPP basis set ${ }^{[d]}$ was used for energy calculations. Solvent (water) was included in the geometry optimizations through the CPCM scheme ${ }^{[\mathrm{ej}]}$ implemented in Gaussian 16. In the text, we use electronic energies rather than Gibb's free energies, as it is our repeated and consistent experience that the electronic energies are more in agreement with experimental results than Gibb's free energies, probably due to the fact that parametrization of B3LYP as well as the CPCM modelling is with respect to experiments, so certain free energy effects are already included.

## Cell cultures and cell imaging

HeLa cells were seeded in a cell culture flask with cell culture media (DMEM medium supplemented with $10 \% \mathrm{FBS}$ and $1 \% \mathrm{AAS}$ ), and were cultured in a humidified incubator at $37^{\circ} \mathrm{C}$ under $5 \% \mathrm{CO}_{2}$. Then, cells were trypsinized and subcultured after reaching $80 \%$ confluency. For cell imaging experiments, cells were seeded in the ibidi $\mu$-Slide 8 well and cultured overnight, then washed with DPBS once. After that, Myco-F ( $20 \mu \mathrm{M}$ ) in opti-mem was confined to the vessel containing cells for 1 h in an incubator at $37^{\circ} \mathrm{C}$. After the cells were washed with DPBS once, and fixed with $4 \%$ paraformaldehyde in DPBS for 15 min in $5 \% \mathrm{CO}_{2}$ incubator at $37^{\circ} \mathrm{C}$. Then, cells were washed with DPBS once again followed by 0 , $0.1,1,5,8 \mathrm{mM} \mathrm{NaF}$ in DPBS treatment for 15 min incubation in $5 \% \mathrm{CO}_{2}$ incubator at $37^{\circ} \mathrm{C}$. Finally, NaF solution with $70 \mu \mathrm{~L}$ of DPBS and observed cells under a confocal Laser Scanning Microscope instrument (Zeiss, LSM-880).

## Cell viability assay

1. Seeding of 6000 cells was carried out per a well in a 96-wells plate and incubated overnight under conditions of $5 \% \mathrm{CO}_{2}$ and $37^{\circ} \mathrm{C}$. 2. The cell culture media (DMEM+10\% FBS+ $1 \%$ AAS) was removed followed by rinsing of the cells by $100 \mu \mathrm{~L}$ PBS once. Then, the cells were treated with $100 \mu \mathrm{~L}$ sample $(997 \mu \mathrm{~L}$ DMEM $+3 \mu \mathrm{~L}$ THF to serve as a control; $5 \mu \mathrm{M}$ : $997 \mu \mathrm{~L}$ DMEM $+0.5 \mu \mathrm{~L}$ Myco-F $+2.5 \mu \mathrm{~L}$ THF; 10 $\mu \mathrm{M}: 997 \mu \mathrm{~L}$ DMEM $+1 \mu \mathrm{~L}$ Myco-F $+2 \mu \mathrm{~L}$ THF) and incubated in $5 \% \mathrm{CO}_{2}$ incubator at $37^{\circ} \mathrm{C}$ for 12 h and 24 h . 3. After that, add $20 \mu \mathrm{~L}$ of MTT (Thiazolyl Blue Tetrazolium Bromide) ( $5 \mathrm{mg} / \mathrm{ml}$, dissolved in PBS) in each well of media for the incubation of 4 h in $5 \% \mathrm{CO}_{2}$ incubator at $37^{\circ} \mathrm{C}$. 4 . Then, the media gently solubilized formazan crystals by $150 \mu \mathrm{~L}$ DMSO. 5 . The plate was shaken for 15 s prior to reading the absorbance at 570 nm .

## Animal Care

Adult wild types of zebrafish were housed in a zebrafish facility with optimum conditions of 14:10 hours light: dark in $28^{\circ} \mathrm{C}$. Wild type and transgenic males and females were crossed and their embryos were kept in egg water at $28.5^{\circ} \mathrm{C}$ until experiments were conducted.

## Exposure

To identify the lethality of the Myco-F probe, 6 dpf zebrafish were used. Five different concentrations ( $10,20,40,80$ and $100 \mu \mathrm{M}$ ) were prepared with $0.1 \%$ DMSO. Control ( $0.1 \%$ DMSO) was prepared separately. 6 dpf zebrafish larvae were exposed to control and five different concentrations in 24 wellplates; lethality was calculated using the Kruskal Wallis One Way ANOVA post-test. Similarly, $T g(h u c: e g f p)$ and $T g(f / k 1: e g f p)$ transgenic zebrafish lines were also exposed to three different concentrations ( 10,20 and $40 \mu \mathrm{M}$ ) to identify any abnormalities in neuronal development and vasculature development, respectively.

## Determination of morphological abnormalities:

Post 24-hour exposure, 6 dpf zebrafish were analyzed for morphological abnormalities. We checked the status of heartbeat rate, eye movement, eye size, pigment formation, pericardial edema, swim bladder development and overall morphological defects. Similarly, Tg(huc:egfp) and Tg(flk1:egfp) transgenic zebrafish were used to determine neuronal development and vasculature development in larval zebrafish. Each larva was then anaesthetized using Tricaine (MS-222) and used for imaging.

## Imaging

6 dpf zebrafish larvae were mounted using 3\% methyl cellulose (Sigma Aldrich, M0512); imaging was performed through the use of the Leica DFC50C instrument; fluorescence images were acquired using a Logos Biosystems Celena S instrument.

## Statistical Analysis

Lethal concentrations were made using a Kaplan-Meier estimate. Fluorescence intensity was measured using a Kruskal-Wallis Post hoc test.

CARBON_01CDCl3

Figure $\mathbf{S 1}$.
${ }^{1} \mathrm{H} \quad \mathrm{NMR}$
(top) and ${ }^{13} \mathrm{C}$ NMR
spectrum
of Myco F.


Figure S2. (top) ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and (bottom) ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY NMR spectrum of Myco-F.


Figure S3. (top) ${ }^{1} \mathrm{H}-$


Found: $\mathrm{M}+\mathrm{Na}^{+}=406.0449$
Calculated: $\mathrm{M}+\mathrm{Na}^{+}=406.0449$
Bromine isotope: 406.0449, 408.0424

| \# | $\mathrm{m} / \mathrm{z}$ | I |
| ---: | :--- | :--- |
|  |  |  |
| 177 | 399.2536 | 807 |
| 178 | 400.1944 | 354 |
| 179 | 400.3438 | 288 |
| 180 | 400.9878 | 718 |
| 181 | 401.2155 | 315 |
| 182 | 401.3147 | 281 |
| 183 | 401.9891 | 284 |
| 184 | 402.3440 | 219 |
| 185 | 403.0689 | 6776 |
| 186 | 404.0705 | 2767 |
| 187 | 405.0686 | 2009 |
| 188 | 405.2457 | 1318 |
| 189 | 406.0449 | 140007 |
| 190 | 407.0465 | 32041 |
| 191 | 408.0424 | 145972 |
| 192 | 408.3045 | 9430 |
| 193 | 409.0445 | 32540 |
| 194 | 409.3062 | 2567 |
| 195 | 409.7905 | 209 |
| 196 | 410.0459 | 10042 |
| 197 | 410.2901 | 529 |

Figure S4. ESI-MS spectrum of Myco-F.


Figure S5. ${ }^{1} \mathrm{H}$ NMR spectrum of Myco-F, Myco-OH and Myco-F and NaF.


Figure S6. ESI-MS spectrum of the reaction product of Myco-F and NaF.


NJ-Myco_OH_DMSO.61.fid -




Figure S7. ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{M y c o - O H}$ (top) and Myco-F (bottom) in DMSO-d ${ }_{6}$.




Figure S8. IR spectrum of $\mathbf{M y c o - O H}$ (top) and Myco-F (bottom).


Figure S9. UV- absorption spectra of Myco-F ( $10.0 \mu \mathrm{M}$ ) in the absence (black) and presence (red) of $\mathrm{F}^{-}$ $(8 \mathrm{mM})$ in the solution of PBS $(10 \mathrm{mM}, 7.4 \mathrm{pH})\left(\lambda_{\mathrm{ex}}=398 \mathrm{~nm}, \lambda_{\text {em }}=518 \mathrm{~nm}\right.$; slit width $=10 \mathrm{~nm} / 10 \mathrm{~nm}$ at RT.


Figure S10. Fluorescence intensity of probe Myco-F (10.0 $\mu \mathrm{M}$ ) in the absence (black) and presence (red) of $\mathrm{F}^{-}(8 \mathrm{mM})$ in the solution of PBS $(10 \mathrm{mM}, 7.4 \mathrm{pH})\left(\lambda_{\mathrm{ex}}=398 \mathrm{~nm}, \lambda_{\text {em }}=518 \mathrm{~nm}\right.$; slit width $=10$ $\mathrm{nm} / 10 \mathrm{~nm}$ at RT.

Figure S11. Plot of fluorescence intensity of Myco-F (10.0 $\mu \mathrm{M}$ ) with increasing concentration of $\mathrm{F}^{-}$( $0-$ $8 \mathrm{mM})$ in the PBS $(10 \mathrm{mM}, 7.4 \mathrm{pH})\left(\lambda_{\mathrm{ex}}=398 \mathrm{~nm}, \lambda_{\mathrm{em}}=518 \mathrm{~nm}\right.$; slit width $=10 \mathrm{~nm} / 10 \mathrm{~nm}$ at RT.


Figure S12.

Figure S13. Myco-F with at $\mathrm{pH} \quad 7.4$

Time dependent fluorescence
Myco-F (10.0 $\mu \mathrm{M}$ ) absence of $\mathrm{F}^{-}$at pH mM ) ( $\lambda_{\text {ex }}=398 \mathrm{~nm}$, nm ; slit width $=10$ at RT.

Selectivity study of different analytes (PBS 10mM).

Figure S14. Crystal structure of Myco- OH and Myco- F.

Table S1. Crystal data and structure refinement for A221080_169_Myco-OH.

| Empirical formula | $\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{BrN} \mathrm{O} 3$ |
| :---: | :---: |
| Formula weight | 270.08 |
| Temperature | 296(2) K |
| Wavelength | 0.71073 Å |
| Crystal system | Hexagonal |
| Space group | P61 |
| Unit cell dimensions | $a=15.9925(13) \AA$ A $\quad \alpha=90^{\circ}$ |
|  | $b=15.9925(13) \AA$ 風 $\quad \beta=90^{\circ}$ |
|  | $\mathrm{c}=20.794(3) \AA \quad \gamma=120^{\circ}$ |
| Volume | 4605.7(10) $\AA^{3}$ |
| Z | 18 |
| Density (calculated) | $1.753 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Absorption coefficient | $4.001 \mathrm{~mm}^{-1}$ |
| F(000) | 2412 |
| Crystal size | $0.326 \times 0.141 \times 0.124 \mathrm{~mm}^{3}$ |
| Theta range for data collection | 2.450 to $27.682^{\circ}$. |
| Index ranges | $-20<=\mathrm{h}<=20,-20<=\mathrm{k}<=20,-24<=1<=27$ |
| Reflections collected | 77336 |
| Independent reflections | $6998[\mathrm{R}$ (int) $=0.1125]$ |
| Completeness to theta $=25.242^{\circ}$ | 99.8 \% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 0.7456 and 0.2729 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Data / restraints / parameters | 6998 / 1 / 418 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 0.988 |
| Final R indices [ $1>2$ sigma( I ] | $\mathrm{R} 1=0.0367, w R 2=0.0779$ |
| R indices (all data) | $\mathrm{R} 1=0.0719, w R 2=0.0877$ |
| Absolute structure parameter | 0.051(14) |
| Largest diff. peak and hole | 0.344 and $-0.374 \mathrm{e} \cdot \AA^{-3}$ |

Table S2. Crystal data and structure refinement for A221450_012_Myco-F.

Empirical formula
Formula weight
Temperature
Wavelength
Crystal system
Space group
$\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{Br} \mathrm{N} \mathrm{O} 3 \mathrm{Si}$
384.34

296(2) K
0.71073 Å

Monoclinic
12/m

Unit cell dimensions

Volume
Z
Density (calculated)
Absorption coefficient
F(000)
Crystal size
Theta range for data collection Index ranges
Reflections collected
Independent reflections
Completeness to theta $=25.242^{\circ}$
Absorption correction
Max. and min. transmission
Refinement method
Data / restraints / parameters
Goodness-of-fit on $\mathrm{F}^{2}$
Final $R$ indices [ $1>2$ sigma( 1 )]
$R$ indices (all data)
Largest diff. peak and hole
$a=16.0619(14) \AA \quad \alpha=90^{\circ}$
$b=7.1033(7) \AA \quad \beta=90.135(2)^{\circ}$
$c=32.658(3) \AA \quad \gamma=90^{\circ}$
3726.0(6) $\AA^{3}$

8
$1.370 \mathrm{Mg} / \mathrm{m}^{3}$
$2.281 \mathrm{~mm}^{-1}$
1584
$0.125 \times 0.032 \times 0.029 \mathrm{~mm}^{3}$
2.495 to $25.353^{\circ}$.
$-19<=\mathrm{h}<=19,-8<=\mathrm{k}<=8,-38<=1<=39$
17887
3574 [R(int) $=0.0814]$
96.9 \%

Semi-empirical from equivalents
0.7452 and 0.6057

Full-matrix least-squares on $\mathrm{F}^{2}$
3574 / 218 / 308
1.026
$R 1=0.0553, w R 2=0.1070$
$R 1=0.1105, w R 2=0.1274$
0.288 and $-0.276 \mathrm{e} \cdot \AA^{-3}$

Table S3. Comparison of previously reported fluorescent probes for the detection of $\mathrm{F}^{-}$.

| No. | Probe Structure | Mechanism | Reaction medium | Detection limit |
| :---: | :---: | :---: | :---: | :---: |
| 1. |  | $(\mathrm{C}-\mathrm{H})^{+} \ldots$ <br> anion H bonding | $\begin{aligned} & \mathrm{CH}_{3} \mathrm{OH} / \mathrm{H}_{2} \mathrm{O} \\ & (9: 1, \mathrm{v} / \mathrm{v}) \end{aligned}$ | $0.039 \mu \mathrm{M}$ |
| 2. |  | H-bonding | $\mathrm{CH}_{3} \mathrm{CN}$ | $0.41 \mu \mathrm{M}$ |

3. 

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HeLa cells and HEK293 cells for 12 h and 24 h .

Figure S16. Investigation of cytotoxicity study on normal development of zebrafish larvae at 6 dpf after 24 h exposure of various concentrations (10, 20, 40, 80 and $100 \mu \mathrm{M}$ ) of Myco-F. Higher concentration of Myco-F induce body curvature phenotype 6dpf zebrafish larvae. Scale bar

(dashed line green and red) in $=100 \mu \mathrm{~m}$


Figure S17. (a) Effects of Myco-F with different concentrations ( $10 \mu \mathrm{M}, 20 \mu \mathrm{M}, 40 \mu \mathrm{M}$ ) on development of central nervous system in zebrafish larvae at $6 \mathrm{dpf}(\mathrm{n}=48)$. Fluorescent neurons are monitored in live transgenic zebrafish, $\operatorname{Tg}$ (huc:egfp). Midbrain region is marked with a red arrowhead. Scale bar $=200 \mu \mathrm{~m}$; (b) Effects of Myco-F with different concentrations (10, 20, $40 \mu \mathrm{M}$ ) on blood vessel development in transgenic zebrafish, Tg(flk1:egfp) at $6 \mathrm{dpf}(\mathrm{n}=48)$. Fluorescent blood vessels were monitored in live transgenic zebrafish, Tg(flk1:egfp). Intersegmental vessel (ISV) in the trunk region marked with yellow arrowheads. Scale bar $=200 \mu \mathrm{~m}$

Table S4. Relative energies of structures in $\mathrm{kcal} / \mathrm{mol}$

| $\mathrm{S}=0$ | Def2-SVP | $\begin{gathered} \text { Def2- } \\ \text { TZVPP } \end{gathered}$ | E(total) ${ }^{a}$ | Z0 | $\mathrm{E}\left(\right.$ thermal) ${ }^{\text {b }}$ | -T $\Delta \mathrm{S}{ }^{\text {b }}$ | $\Delta \mathrm{G}^{\mathbf{c}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| RC | 3.58 | -0.54 | 3.04 | -0.26 | +0.09 | -0.06 | 2.81 |
| TS ${ }_{1}$ | 8.79 | -1.15 | 7.64 | -0.99 | -0.10 | +0.20 | 6.76 |
| $\mathrm{IM}_{1}$ | 0.00 | $+0.00$ | 0.00 | +0.00 | $+0.00$ | $+0.00$ | 0.00 |
| $\mathrm{TS}_{2}$ | 34.00 | -1.27 | 32.73 | -1.52 | -0.04 | +0.09 | 31.26 |
| PC | 27.77 | -1.43 | 26.34 | -0.71 | +0.17 | -0.34 | 25.45 |
| $\mathrm{S}=1$ |  |  |  |  |  |  |  |
| RC | 66.56 | -0.97 | 65.59 | -2.30 | +0.29 | -1.11 | 62.47 |
| TS ${ }_{1}$ | 74.14 | -1.84 | 72.30 | -3.51 | +0.22 | -1.14 | 67.87 |
| $\mathrm{IM}_{1}$ | 61.43 | -0.22 | 61.21 | -2.35 | +0.22 | -1.02 | 58.06 |
| $\mathrm{TS}_{2}$ | 63.77 | +0.85 | 64.62 | -4.74 | -0.14 | -0.35 | 59.39 |
| $\mathrm{IM}_{2}$ | 53.09 | +0.49 | 53.58 | -1.72 | +0.16 | -0.78 | 51.24 |
| $\mathrm{TS}_{3}$ | 62.42 | -0.58 | 61.83 | -2.83 | +0.09 | -0.78 | 58.31 |
| PC | 58.92 | -0.51 | 58.41 | -2.19 | +0.37 | -1.37 | 55.22 |

${ }^{a}$ Electronic energy as sum of the two previous columns. ${ }^{b} \mathrm{~T}=298.15 \mathrm{~K} .{ }^{c}$ Sum of the four previous columns, but not used in the text due to added uncertainty (see Methods section).

Table S5. Relative energies of $S=1$ single point calculations on the $S=0$ optimized structure at B3LYP/Def2-TZVPP level, in kcal/mol

| $\mathrm{S}=1$ | E (total) |
| :---: | :---: |
| RC | 71.35 |
| $\mathrm{TS}_{1}$ | 80.89 |
| $\mathrm{IM}_{1}$ | 66.75 |


| $\mathrm{TS}_{2}$ | 67.43 |
| :--- | :--- |
| PC | 63.88 |

Table S6. Relative energies of $S=0$ single point calculations on the $S=1$ optimized structure at B3LYP/Def2-TZVPP level, in kcal/mol

| $\mathrm{S}=0$ | E (total) |
| :---: | :---: |
| RC | 10.89 |
| $\mathrm{TS}_{1}$ | 17.84 |
| $\mathrm{IM}_{1}$ | 7.68 |
| $\mathrm{TS}_{2}$ | 19.49 |
| $\mathrm{IM}_{2}$ | 24.54 |
| $\mathrm{TS}_{3}$ | 38.54 |
| PC | 32.08 |

Table S7. Mulliken spin density distributions of structures at B3LYP/Def2-TZVPP level
Note: There is no mulliken spin density distributions at $S=0$ Multiplicity.

| $\mathrm{S}=1$ | 6-ring | 5-ring | $\mathrm{O}_{\text {top }}$ | $\mathrm{O}_{\text {bottom }}$ | $\mathrm{O}_{\text {top-6ring }}$ | Br | Sum |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| RC | 1.42 | 0.08 | 0.19 | 0.13 | 0.11 | 0.08 | 2.00 |
| TS | 1.48 | 0.11 | 0.16 | 0.12 | 0.00 | 0.13 | 2.00 |
| $\mathrm{IM}_{1}$ | 1.33 | 0.13 | 0.17 | 0.17 | 0.15 | 0.06 | 2.01 |
| $\mathrm{TS}_{2}$ | 1.21 | 0.20 | 0.12 | 0.24 | 0.24 | 0.01 | 2.01 |
| $\mathrm{IM}_{2}$ | 1.01 | 0.31 | 0.07 | 0.26 | 0.35 | 0.00 | 2.00 |
| $\mathrm{TS}_{3}$ | 0.90 | 0.43 | 0.01 | 0.27 | 0.37 | 0.00 | 1.98 |
| PC | 0.98 | 0.33 | 0.07 | 0.26 | 0.37 | 0.00 | 2.00 |

Table S8. Mulliken spin density distributions of $S=1$ single point calculations on the $S=0$ optimized structures

| $\mathrm{S}=1$ | 6-ring | 5-ring | $\mathrm{O}_{\text {top }}$ | $\mathrm{O}_{\text {bottom }}$ | $\mathrm{O}_{\text {top-6ring }}$ | Br | Sum |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| RC | 1.39 | 0.11 | 0.16 | 0.14 | 0.14 | 0.06 | 2.00 |
| $\mathrm{TS}_{1}$ | 1.41 | 0.16 | 0.15 | 0.12 | 0.01 | 0.16 | 2.00 |
| $\mathrm{IM}_{1}$ | 1.36 | 0.13 | 0.16 | 0.15 | 0.15 | 0.05 | 2.00 |
| $\mathrm{TS}_{2}$ | 0.93 | 0.46 | 0.02 | 0.24 | 0.34 | 0.00 | 1.98 |
| PC | 1.00 | 0.36 | 0.08 | 0.22 | 0.34 | 0.00 | 2.00 |

Figure S17. DFT optimized structures. ${ }^{1}$ RC

${ }^{1} \mathrm{TS}_{1}$

${ }^{1} \mathrm{IM}_{1}$

${ }^{1} \mathrm{TS}_{2}$

${ }^{1} \mathrm{PC}$

${ }^{3} \mathrm{RC}$

${ }^{3} \mathrm{TS}_{1}$

${ }^{3} \mathrm{IM}_{1}$

${ }^{3} \mathrm{TS}_{2}$

${ }^{3} \mathrm{IM}_{2}$

${ }^{3} \mathrm{PC}$



Figure S18. Relative Energies of the reaction pathways (see below text).

Figure S18 shows an energy graph of the proposed chemical mechanism of tautomerization. $S=0$ singlet and $S=1$ triplet calculations show that the number of transition states in the different spin states were different. The gray states are triplet single-point calculations on the singlet optimized geometry; the light blue states are singlet single-point calculations on the triplet optimized geometry.

The $S=0$ singlet surface consists of only two steps, each with one transition state. The first step is rotation of the OH on the hexagonal ring to form a hydrogen bond with the opposite side O . The energy of this transition state $\left(\mathrm{TS}_{1}\right)$ was found to be $4.6 \mathrm{kcal} / \mathrm{mol}$ compared to the energy of the reactant. The second step is the H atom transfer in which the H -atom bonded to O on the hexagonal ring moves to the O atom of the pentagonal ring and the resulting OH concertedly rotates to point away from the hexagonal oxygen. The activation barrier of this $\mathrm{TS}_{2}$ was found to be $32.7 \mathrm{kcal} / \mathrm{mol}$. The Mulliken spin density distributions was 0 for all $S=0$ structures meaning that the surface depicts a closed shell surface.

The $S=1$ triplet surface consists of three steps. Like $S=0$, the first step is rotation of the OH of the hexagonal ring to form a hydrogen bond with the opposite side 0 . The $S=1$ reactant energy was 65.6 $\mathrm{kcal} / \mathrm{mol}$ compared to the $S=0$ lowest point, and the energy of the triplet $\mathrm{TS}_{1}$ was $72.3 \mathrm{kcal} / \mathrm{mol}$. The second transition state is the transfer of H from the hexagonal ring to the pentagonal ring O . The energy of this $\mathrm{TS}_{2}$ was $64.6 \mathrm{kcal} / \mathrm{mol}$. The third transition state is the rotation of OH on the pentagonal ring, and the energy of this $\mathrm{TS}_{3}$ was $61.8 \mathrm{kcal} / \mathrm{mol}$.

The $S=0$ lowest point Reactant ( $0.0 \mathrm{kcal} / \mathrm{mol}$ ) structure became $71.4 \mathrm{kcal} / \mathrm{mol}$ in energy when used in a $S=1$ single-point calculation. In addition, the two energies not found in the singlet surface ( $\left.{ }^{3} \mathrm{TS}_{2},{ }^{3} / \mathrm{M}_{2}\right)$, single-point $S=0$ calculations on the $S=1$ geometries resulted in energies of $19.5 \mathrm{kcal} / \mathrm{mol}$ and 24.5 $\mathrm{kcal} / \mathrm{mol}$, respectively, which are lower than the energy of ${ }^{1} \mathrm{TS}_{2}$, so it can be seen that they are on the downhill side of ${ }^{1} \mathrm{TS}_{2}$.

## Cartesian Coordinates of DFT geometries

## ${ }^{1} \mathrm{RC}$

C -0.251784 1.870930-0.000083 C $1.1187161 .539323-0.000103$ C $1.5299300 .207830-0.000064$ C $0.633831-0.891738-0.000232$ C - $0.713814-0.523694-0.000287$ C - 1.159162 0.809047-0.000282 H $1.8625212 .338588-0.000303$ Br 3.412491-0.119800 0.000170 C $1.103119-2.321128-0.000341$ O -0.691569 3.139609 0.000059 O -0.6915693 .1396090 .000059
H 0.0550753 .7579500 .000053 H $1.726846-2.5264400 .884691$ H $1.730486-2.525072-0.883054$ H $0.251662-3.009648-0.002360$ C -2.645740 0.816450-0.000222 C - $1.939520-1.397175-0.000098$ N - 3.032176-0.530511-0.000172 C - $4.415848-0.9620850 .000548$ $\begin{array}{lll}\text { C } \\ \text { H }-4.427669 & -2.058772 & -0.000091\end{array}$ H $-4.427669-2.058772-0.000091$
H $-4.939247-0.589367-0.892622$ $\begin{array}{llll}\text { H }-4.939247 & -0.589367 & -0.892622 \\ \text { H }-4.938057 & -0.590404 & 0.894859\end{array}$ H $-4.938057-0.5904040 .894859$
O -3.4233661 .7474770 .000025 O -2.021559-2.609188 0.000051
${ }^{1}$ TS 1
C - $0.2544311 .865603-0.015127$ C $1.1147071 .543296-0.011437$ C $1.5352790 .213085-0.001744$ C $0.640652-0.8870870 .000842$ C - $0.710797-0.528133-0.003579$ C - $1.1564870 .803828-0.008325$ H $1.8412092 .356791-0.023670$ Br 3.418758 - 0.1106250 .002328 C $1.113161-2.3147230 .006643$ O -0.655072 $3.166892-0.091653$ H - 0.7182623 .5542660 .794826 H $1.739649-2.5116370 .891485$ H $1.740882-2.518489-0.875800$ H $0.264577-3.0066950 .008581$ C - $2.6483350 .807261-0.002566$ C - $1.933250-1.404355-0.001445$ N -3.029417-0.538948-0.002714 C - $4.411992-0.9749820 .003209$ H -4.420583-2.071609 0.010606 H -4.937484-0.610124-0.891940 H -4.933210 -0.597712 0.895660 $\begin{array}{ll}\text { H } \\ \text { O } & -4.425719 \\ 1.737266 & 0.005987\end{array}$ O -2.014011-2.615788 0.001782
${ }^{1}$ IM1
C - 0.2707661 .8502470 .000025 C $1.1069691 .551070-0.000024$ C 1.5323100 .2220970 .000009 C $0.657576-0.9014660 .000062$ C -0.695955 -0.562069 0.000211 C - 1.1410900 .7646390 .000260 H 1.828990 2.368354-0.000111 Br 3.420057-0.079600-0.000020 C $1.150100-2.3222770 .000052$

O -0.695339 3.122448-0.000172 H-1.673385 $3.120489-0.000095$ H $1.776974-2.5194980 .884628$ H $1.778714-2.518954-0.883382$ H $0.307824-3.023159-0.000916$ C - 2.6107630 .7926400 .000129 C - 1.933142 -1.420958 0.000019 N-3.026637-0.534290 -0.000001 C - $4.415313-0.953116-0.000035$ H - $4.436434-2.049641-0.009008$ $\begin{array}{llll}\text { H }-4.436434 & -2.049641 & -0.009008 \\ \text { H }-4.935522 & -0.569768 & -0.890337\end{array}$ $\begin{array}{llll}\text { H }-4.935522 & -0.569768 & -0.89033 \\ \text { H }-4.931861 & -0.584882 & 0.898851\end{array}$ $\begin{array}{llll}\text { H } & -4.931861 & -0.584882 & 0.898851 \\ \text { O }-3.332959 & 1.778548 & 0.000023\end{array}$ O -2.035501-2.628713-0.000246

TS2
C 0.2781081 .9629890 .001305 C - $1.1453131 .568960-0.003597$ C - $1.5534060 .266835-0.000521$ C - $0.647695-0.8759040 .005276$ C $0.685245-0.5389880 .003451$ C $1.1679200 .814798-0.003378$ H - $1.8695792 .385361-0.006368$ $\mathrm{Br}-3.439686-0.081359-0.003078$ C - $1.150531-2.2889030 .011160$ O 0.6244243 .1497980 .012285 H 3.7725352 .0385140 .764713 H - $1.782139-2.480396-0.871782$ H - 1.784976 -2.472523 0.893697 $\begin{array}{llll}\text { H }-1.784976 & -2.472523 & 0.893697 \\ \text { H }-0.316955 & -2.999320 & 0.015377\end{array}$ C $2.5596770 .750724-0.007262$ C $1.887038-1.4289150 .005768$ N $2.996799-0.548485-0.001910$ C $4.382564-0.982739-0.012430$ H 4.386583-2.065787-0.184980 H $4.872728-0.7706590 .950082$ H $4.934885-0.477681-0.816794$ O $3.4372661 .743104-0.099228$ O $3.437266-1.743104-0.099228$
O $1.990398-2.6383620 .009259$ ${ }^{1} \mathrm{PC}$
C $0.2742551 .966576-0.000019$ C - 1.1450381 .5668950 .000066 C - 1.5513650 .2613880 .000045 C -0.649390-0.875969-0.000013 C $0.684770-0.530321-0.000020$ C $1.1600380 .821012-0.000010$ H -1.8729122 .3802280 .000113 H-1.872912 2.3802280 .000113 $\mathrm{Br}-3.440430-0.0864160 .000027$ C-1.146196-2.292424-0.000128 O $0.6208433 .156285-0.000034$ H 4.309468 1.539428-0.000105 H -1.777179-2.485094-0.883284 H-1.777878-2.485021 0.882531 H -0.309974-2.999868 0.000193 C $2.5615900 .760646-0.000020$ C $1.885254-1.416718-0.000013$ N $3.002333-0.534282-0.000014$ C $4.002333-0.534282-0.000014$ C $4.381818-0.9904230 .000184$ H $4.355171-2.087158-0.000588$

H $4.915326-0.6539160 .902551$
H $4.915917-0.652643-0.901338$
O $3.3700011 .789924-0.000112$
O $1.997448-2.623132-0.000024$

## RC

C - $0.2004551 .894861-0.000273$ C $1.1415971 .578961-0.000244$ C $1.5490810 .213655-0.000160$ C $0.619070-0.896319-0.000271$ C - $0.732568-0.567662-0.000307$ C - $1.1845000 .808247-0.000057$ H $1.8922122 .370291-0.000375$ Br 3.402711-0.127099 0.000266 C 1.089099-2.315447-0.000266 O -0.689774 $3.128008-0.000497$ H $0.0250243 .788116-0.001047$ H 1.721497 -2.510328 0.882022 H 1.723196-2.509909 -0.881397 H. $0.73106-2.500535-0.881397$ H $0.236270-3.002535-0.001140$ C -2.6342590 .8324330 .000309 C - $1.924670-1.418102-0.000228$ N - $3.010964-0.5246520 .000149$ C - $4.393341-0.9446010 .000346$ H - $4.414863-2.041205-0.001112$ Н-4.916697-0.562352-0.890264 H -4.915814-0.564851 0.892567 O - 3.4330361 .7703650 .000576 O - $2.045100-2.643666-0.000416$ ${ }^{3}$ TS1
C - 0.223372 1.903609-0.025935 C $1.1041211 .587180-0.026858$ C $1.5390860 .179458-0.009345$ C $0.638222-0.925678-0.007834$ C -0.706616 -0.589734-0.012494 C - 1.169613 0.819523-0.011045 H $1.8547682377008-0.047272$ 1.858419-0.100014 0.011208 C 3.388419 - 0.100014350 .011208 C $1.117355-2.349463-0.000966$ O -0.659655 $3.191592-0.100819$ H - 0.6855283 .5935040 .781081 H $1.740854-2.5442940 .886457$ H $1.747205-2.550750-0.882455$ H 0.265257-3.037662-0.001628 C - 2.6199980 .8269240 .004814 C $-1.899366-1.434722-0.007435$ - $2.886235-0.5363480 .001804$ $\mathrm{N}-2.986235-0.5363480 .001804$ C $-4.366377-0.9594300 .013995$ H - $4.388381-2.0556670 .008973$ H -4.896427-0.569371 -0.869421 H - $4.878326-0.5779990 .911769$ O -3.430656 1.7493460 .022213 O -2.026073-2.656669-0.008117

MI
C - $0.2387201 .850909-0.000146$ C $1.1268501 .588504-0.000122$ C $1.5505470 .235964-0.000161$ C $0.658417-0.904540-0.000290$

C - $0.721116-0.6208270 .000018$ C - 1.1546320 .7301270 .000082 H 1.846663 2.405768 -0.0000093 Br $3.416300-0.0729970 .000235$ C 1.157431-2.310419-0.000547 O -0.763283 $3.058964-0.000469$ H-1.757519 $2.959713-0.000099$ H $1.793900-2.4947000 .881574$ H 1.797096 -2.493593 -0.880508 H 0.317674-3.014621-0.002239 C -2.5835050 .8020800 .000052 C - $1.926476-1.446829-0.000107$ $\mathrm{N}-3.013369-0.5187240 .000080$ C - $4.403726-0.9173200 .000489$ H -4.439496-2.013702 -0.007014 H -4.922266 -0.525965 -0.888694 Н -4.918757-0.538824 0.897357 O -3.278269 1.841821 0.000119 O - $2.081523-2.668787-0.000237$ ${ }^{3}$ TS2
C -0.293377 1.816986-0.000002 C 1.1118101 .5783100 .000000 C 1.5467030 .2391290 .000000 C $0.696913-0.912700-0.000002$ C - $-0.724985-0.658814-0.000003$ C - 1.137591 0.670013-0.000005 H 1.8169642 .4083840 .000001 Br $3.434123-0.0386340 .000002$ C $1.413882-2.311623-0.000002$ C $1.213882-2.311623-0.000003$ - -0.9013752 .9545420 .000000 H - $2.0432752 .670945-0.000001$ H $1.854062-2.4915950 .880806$ H 1.854059-2.491595-0.880815 H 0.383362 -3.027640 -0.000002 C - 2.550058 0.779365-0.000002 C - 1.939172 -1.459582 0.000000 $\mathrm{N}-3.033546-0.4937120 .000000$ C - $4.430156-0.8781760 .000004$ H $-4.471211-1.9746390 .000010$ H -4.942729-0.493463-0.89496 H -4.942727-0.493453 0.894966 O -3.142098 1.912892 0.000001 O -2.135502-2.676961 0.000001

M2
C - $0.2819321 .893078-0.000012$ C $1.1267471 .575209-0.000001$ C $1.5584580 .238326-0.000003$ C $0.697140-0.876725-0.000010$ C - $-0.720098-0.5814030 .000005$ C-1.165985 0.7459840 .000024 H 1.839807 2.400046-0.000017 Br 3.453035-0.055745 0.000007 C 1.181719-2.292450 -0.000000 O -0.751185 3.057531-0.000043 H-2.818822 2.5909300 .000121 H 1.815809 -2.486902 0.881635 H $1.815870-2.486886-0.881613$ H 0.338326-2.992374 -0.000043

C - 2.5822340 .7479250 .000002 C - $1.899135-1.436533-0.000005$ $\mathrm{N}-3.020711-0.526076-0.000005$ C - $4.406232-0.9571790 .000006$ H - $4.404825-2.0539480 .000100$ H - $4.928572-0.590230-0.895922$ H -4.928603 -0.590070 0.895849
O -3.390883 1.791218 0.000019
O -2.044300 -2.660543-0.000015
TS3
C $0.2630831 .929263-0.000708$ C - $1.1405761 .573033-0.002129$ C - $1.5691770 .240935-0.000317$ C -0.691694-0.856013 0.003774 C $0.716718-0.5310160 .003235$ C $1.1836230 .792578-0.003505$ H-1.857016 $2.395379-0.001587$ $\mathrm{Br}-3.461280-0.069812-0.001625$ C - $1.159507-2.2785670 .006907$ O 0.6462363 .1168270 .007610 H 3.8256422 .0380320 .768441 H-1.787847-2.480883-0.877074 H - 1.796243 -2.474695 0.886148 H -0.311170-2.971283 0.012778 C $2.6099870 .756441-0.004102$ C $1.882440-1.4141430 .004803$ N $3.008081-0.541265-0.002407$ C $4.383140-1.005918-0.009433$ H 4.363788-2.089615-0.175434 H $4.874137-0.7963380 .953276$ H $4.947121-0.512599-0.813194$ O 3.484198 1.761139-0.099233 O 1.989765-2.642375 0.007776

## PC

C 0.2588721 .9334980 .000078 C - 1.1451591 .5740440 .000169 C - 1.5688610 .2383110 .000164 C - $0.690482-0.8551850 .000104$ C $0.721345-0.5304680 .000059$ C 1.1823410 .7998460 .000043 H - 1.8634392 .3945470 .000209 $\mathrm{Br}-3.461249-0.076590-0.000007$ C - $1.153733-2.279458-0.000197$ O 0.6377083 .1215150 .000032 H 4.3482031 . $561067-0.000138$ H-1.783641-2.481849-0.883055 H -1.783641 - $2.481849-0.883055$ H - $1.787167-2.4812620 .880206$ $\begin{array}{llllllllll}\text { H }-0.302619 & -2.968990 & 0.001500 \\ \text { C } 2.605992 & 0.764727 & -0.000135\end{array}$ C 2.605992 0.764727-0.000135 C $1.880184-1.408627-0.000094$ C 4.383903-1.004950 0.000309 H 4.337961 -2.101080 0.000092 H $4.922740-0.6725840 .901223$ H $4.923509-0.672282-0.900024$ O $3.4107811 .813989-0.000310$ O $2.002482-2.6359630 .000116$

