Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2023

Electronic Supplementary Information to

Solvent-assisted dispersive solid phase extraction of bisphenols using iron (III)

thenoyltrifluoroacetonate complex (Fe(TTA)₃) as new nanostructured sorbent: A proof of concept

- 5 Celestine Vubangsi Gemuh¹, Aleš Bezrouk², Rocío Pérez³, Caitlan Ayala³ Petr Solich¹, Burkhard Horstkotte¹,*
 - ¹ Department of Analytical Chemistry, Faculty of Pharmacy in Hradec Králové, Charles University, Akademika Heyrovského 1203, 500 05 Hradec Králové
 - ² Department of Medical Biophysics, Faculty of Medicine in Hradec Králové, Charles University,
- 10 Šimkova 870, Hradec Kralove, 500 03
 - ³ Chemistry Department, Louisiana State University, Baton Rouge, LA, 70803.
 - *Corresponding Author Contact Information:

Burkhard Horstkotte, Department of Analytical Chemistry, Faculty of Pharmacy, Charles University,

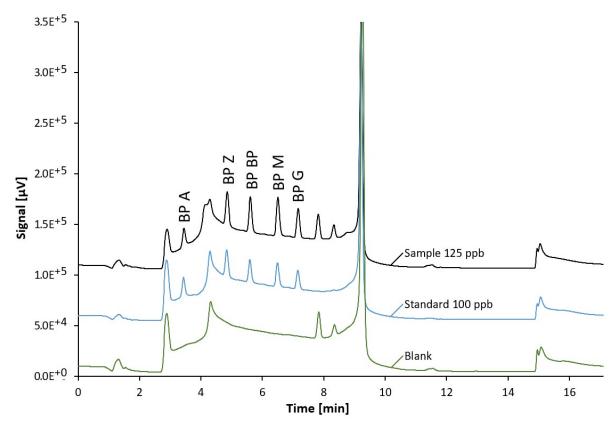
15 Akademika Heyrovského 1203/8, 500 05 Hradec Králové, Czech Republic

Tel.: +420 495 067 183

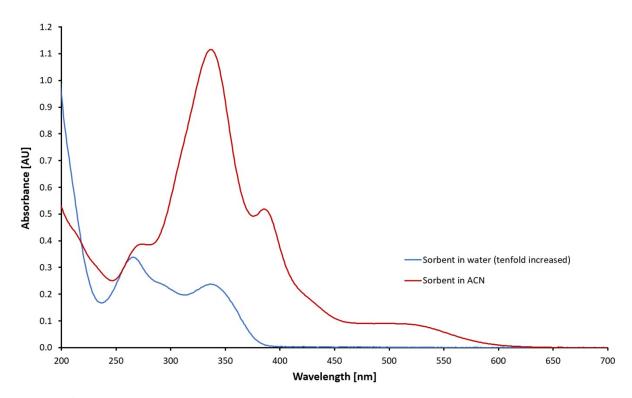
E-mail: horstkob@faf.cuni.cz

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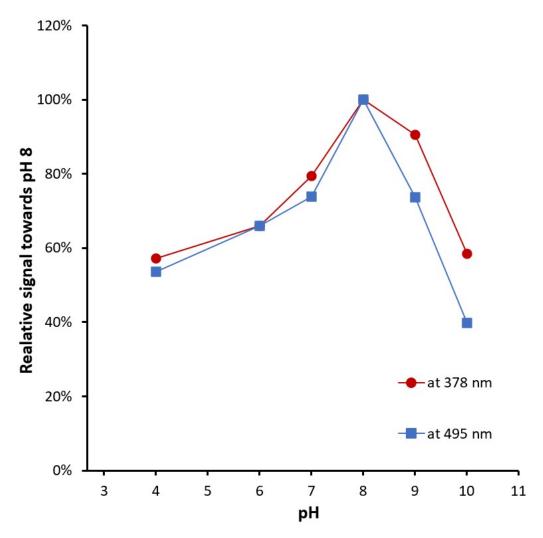
- 20 ESI-1: Representative chromatograms
 - ESI -2: UV/VIS absorption spectra of Fe(TTA)3 complex
 - ESI -3: Study of Fe(TTA)3 complex stability at different pH values
 - ESI -4: FT-IR spectra of 2-thenoyltrifluoroacetone



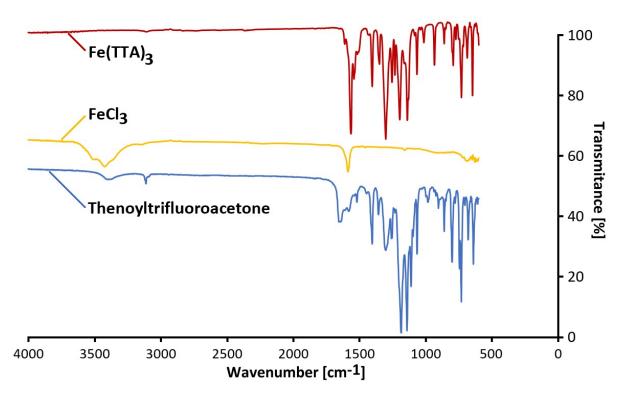
ESI-1: Representative chromatograms from MeOH extracts originating from Fe(TTA)₃-d μ SPE of blank WWTP effluent sample, 100 ppb mixed standard, and sample spiked at 125 μ g L⁻¹ concentration level. Conditions: flow rate of 1 mL/min, injection volume of 20 μ L, mobile phase A: water, B: acetonitrile, gradient elution, wavelength 230 nm, run time of 17.01 min. Peaks at 3 min, 4.5 min, and 8.7 min are related to the sorbent (partially complexed Fe³⁺, pure TTA, and complex, respectively).



ESI-2: UV/VIS absorption spectra of $Fe(TTA)_3$ complex dissolved in acetonitrile at 15.5 mg L⁻¹ (red) and distilled water, saturated with $Fe(TTA)_3$, tenfold increase (blue).



ESI-3: Study of Fe(TTA)₃ complex stability at different pH values as described in section 3.1. Where 15 mg L⁻¹ of Fe(TTA)₃ was prepared in saturated octanol and equal volume of buffer solutions of pH 4, 6, 7, 8, 9 10 was added and stirred for 10 mins and the octanol phase recovered and measured with UV/VIS spectrophotometer.



ESI-4: FT-IR spectra of 2-thenoyltrifluoroacetone represented by blue line, FeCl₃ by yellow line and Fe(TTA)₃ by red line measured in ATR mode.