

## Supplementary Information

### Colorimetric detection of ozone in aqueous solution and imaging in living cells using a novel hemicyanine dye

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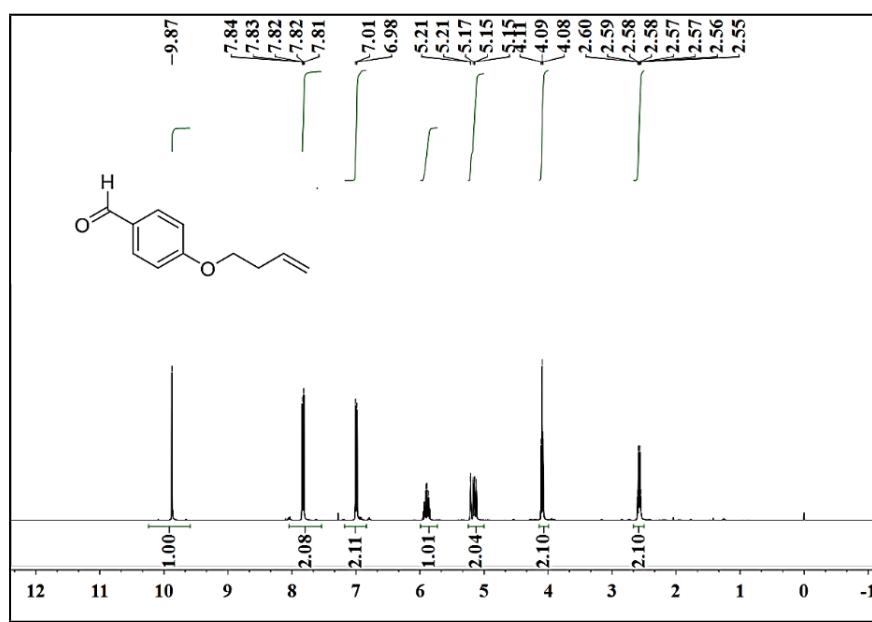
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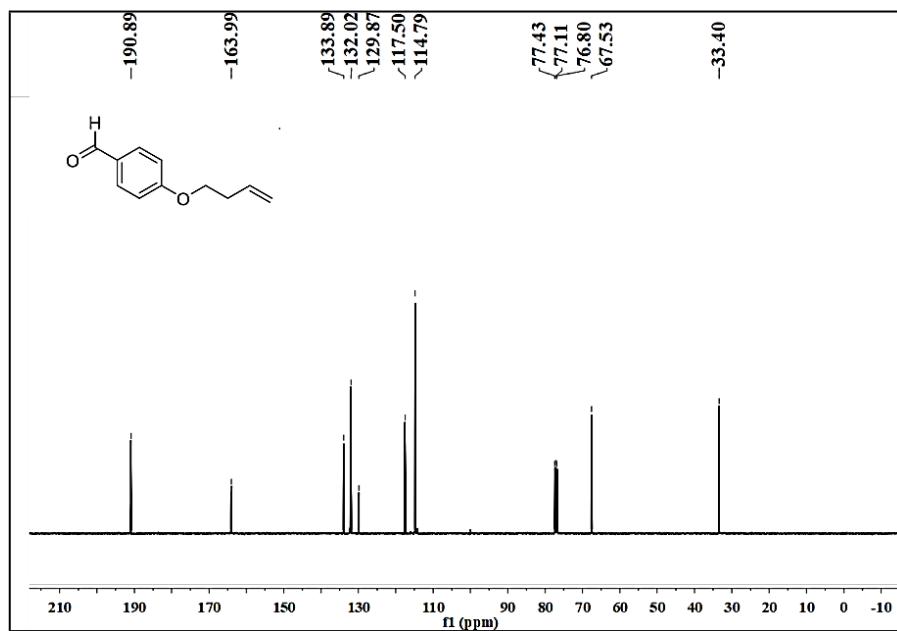
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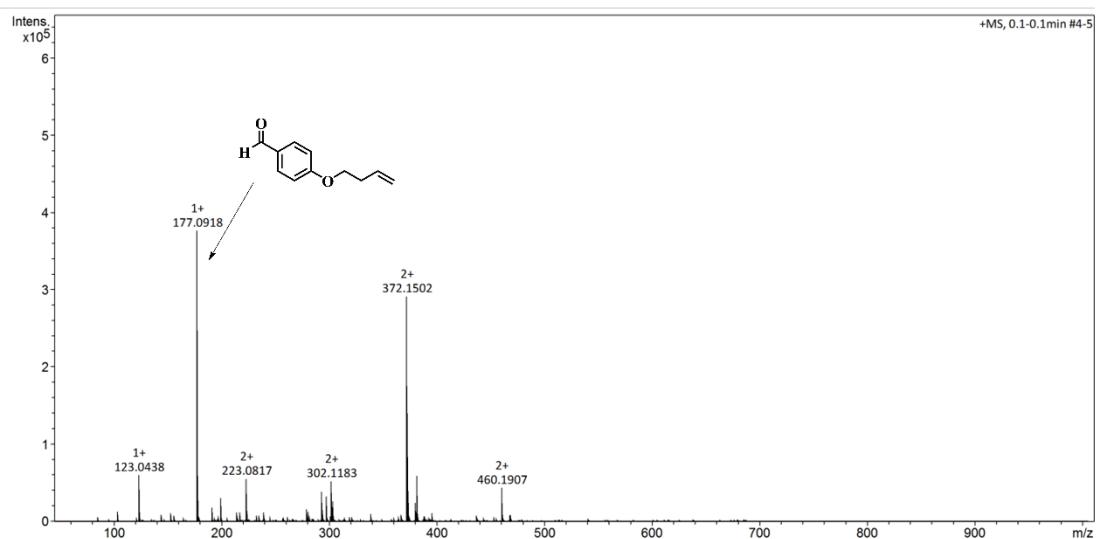
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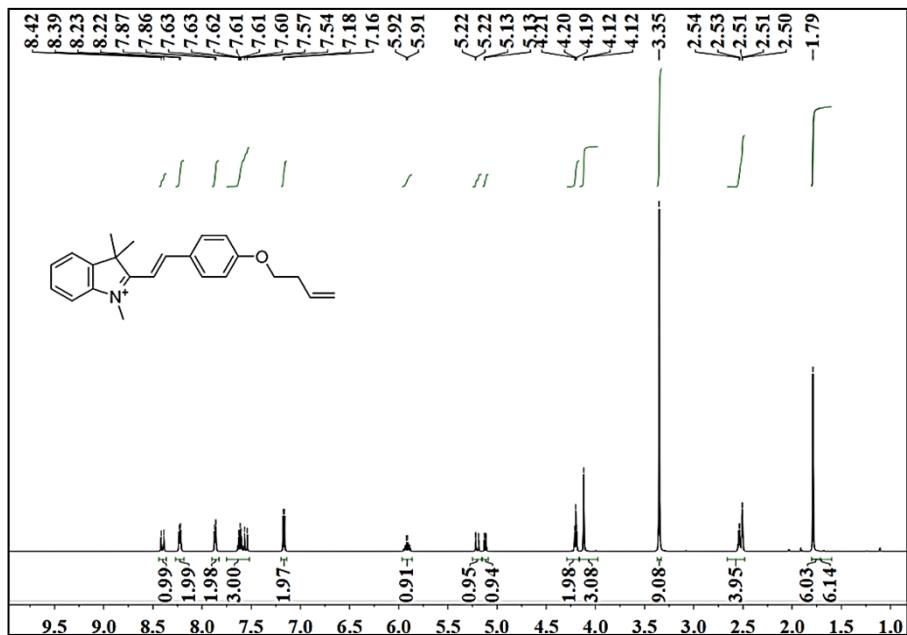
**Fig.S<sub>1</sub>** <sup>1</sup>H NMR spectroscopy of compound 1(CDCl<sub>3</sub>).



**Fig.S<sub>2</sub>** <sup>13</sup>C NMR spectroscopy of compound 1(CDCl<sub>3</sub>).



**Fig.S<sub>3</sub> MS spectroscopy of compound 1 in sheme1.**



**Fig.S<sub>4</sub> <sup>1</sup>H NMR spectroscopy of the probe HCB (DMSO) .**

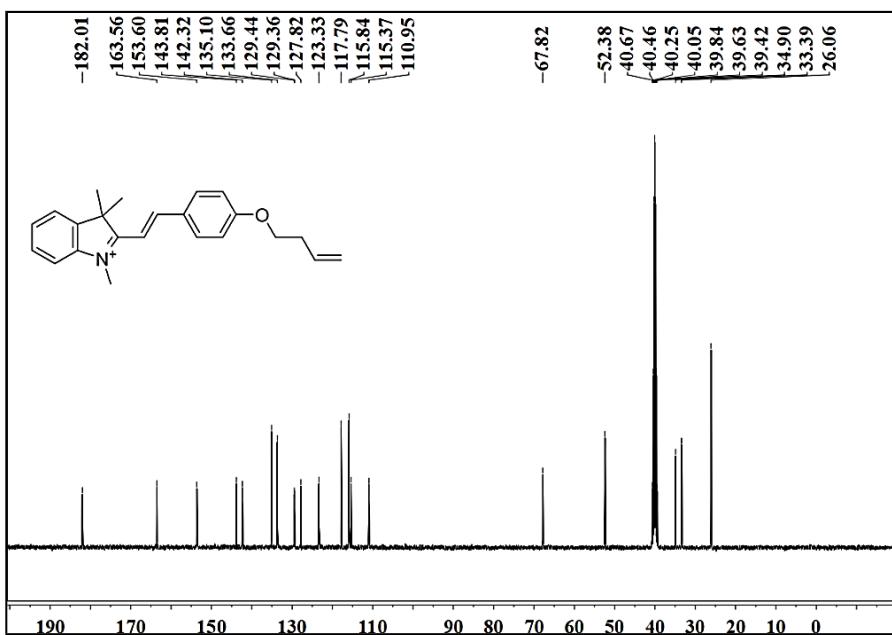


Fig.S<sub>5</sub>  $^{13}\text{C}$  NMR spectroscopy of the probe HCB (DMSO) .

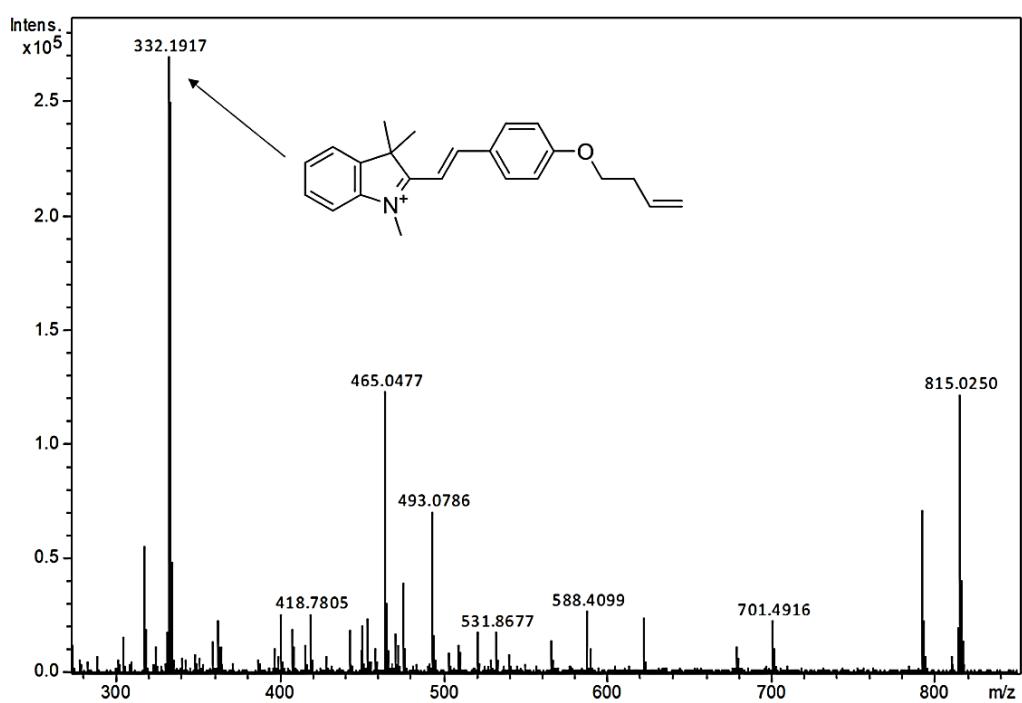
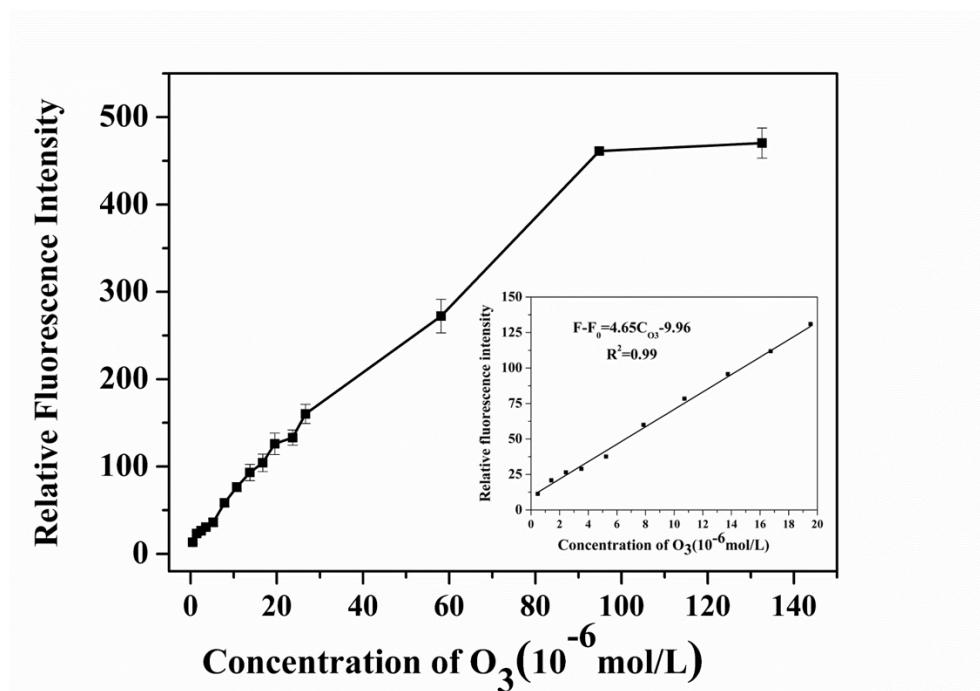
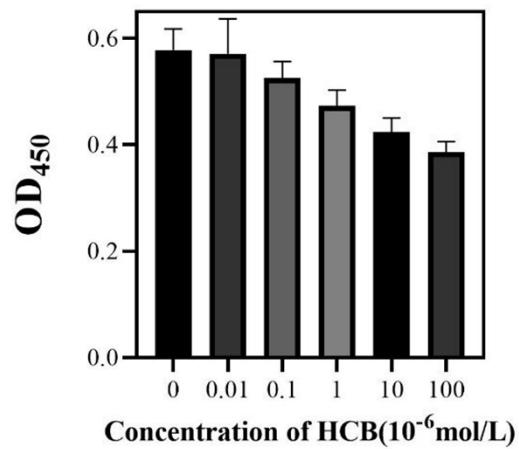


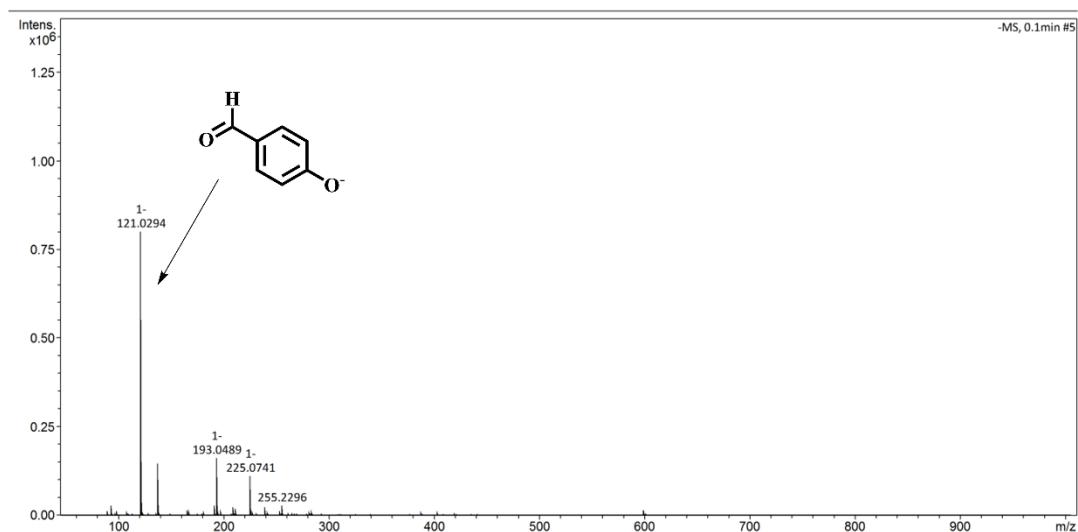
Fig.S<sub>6</sub> MS spectroscopy of the probe HCB.



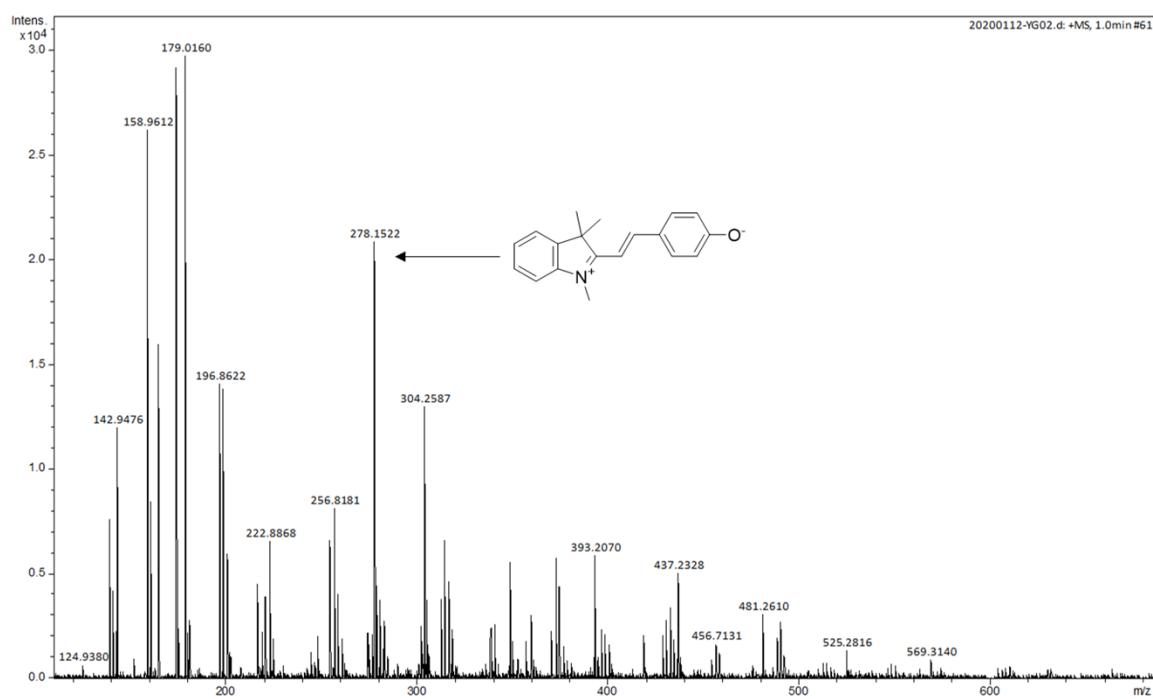
**Fig.S<sub>7</sub>** The fluorescence titration curves and calibration graphs for spectrophotometric determination of  $O_3$  in the simulated physiological media. ( $[HCB] = 5.00 \times 10^{-5} \text{ mol/L}$ ,  $[HCO_3^-] = 1.00 \times 10^{-3} \text{ mol/L}$ ,  $[HPO_4^{2-}] = 2.00 \times 10^{-1} \text{ mol/L}$ ,  $[H_2PO_4^-] = 2.00 \times 10^{-1} \text{ mol/L}$ ,  $[Na^+] = 1.00 \times 10^{-3} \text{ mol/L}$ ,  $[K^+] = 1.00 \times 10^{-3} \text{ mol/L}$ ,  $[Ca^{2+}] = 5.00 \times 10^{-4} \text{ mol/L}$ ,  $[Mg^{2+}] = 5.00 \times 10^{-4} \text{ mol/L}$ ,  $[Cl^-] = 2.00 \times 10^{-3} \text{ mol/L}$ , pH=7.40 PBS buffer, response at 37°C for 40 min).



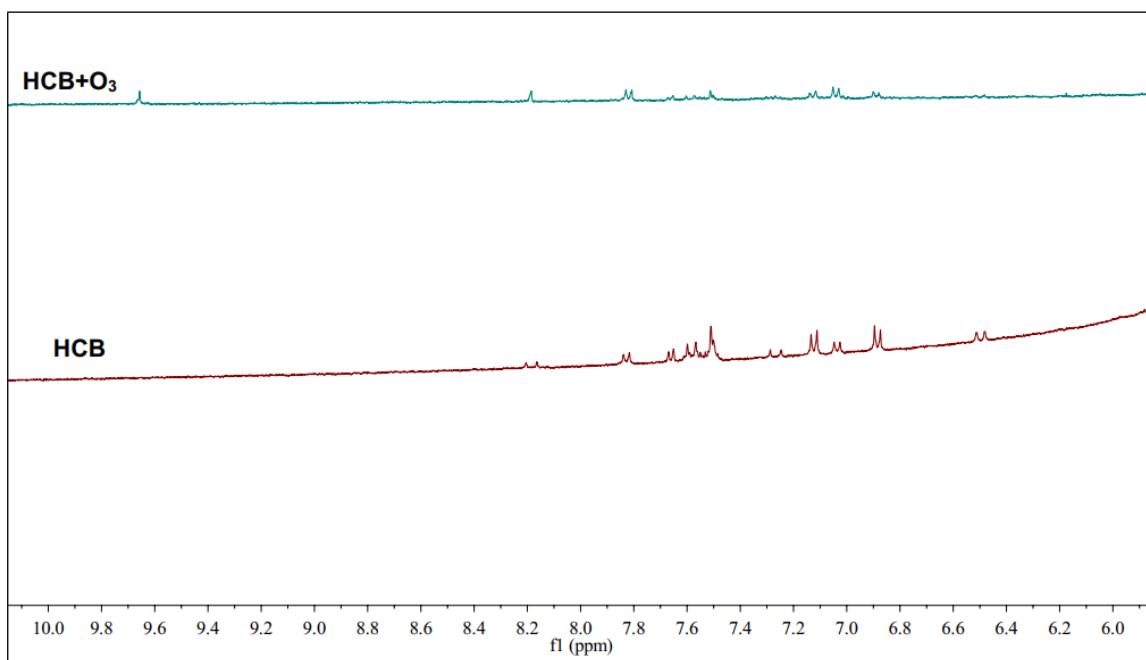
**Fig.S<sub>8</sub>** Results of the toxicity of HCB to RAW264.7 cells determined by CCK-8 method.



**Fig.S<sub>9</sub> MS spectroscopy of compound 1(sheme1)-O<sub>3</sub>.**



**Fig.S<sub>10</sub> MS spectroscopy of HCB-O<sub>3</sub>.**



**Fig.S<sub>11</sub>** <sup>1</sup>H NMR spectra of probes HCB in DMSO-d6 and D<sub>2</sub>O before and after the addition of O<sub>3</sub>.