## **Supporting Information**

# A chemodosimetric approach for visual detection of nerve agent simulant diethyl chlorophosphate (DCP) in liquid and vapour phase

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Fig. S1. Linear response curve of BIPQ at 468 nm depending on the DCP concentration



Fig. S2. <sup>1</sup>H NMR (300 MHz) spectrum of BIPQ in DMSO-d<sub>6</sub>



Fig. S3: <sup>13</sup>C NMR (75 MHz) spectrum of BIPQ in DMSO-d<sub>6</sub>



Fig. S4. HRMS of BIPQ



Fig.S5. <sup>1</sup>H NMR (300 MHz) spectrum of the adduct BIPQ-DCP in DMSO- $d_6$ 





Fig. S6: <sup>13</sup>C NMR (75 MHz) spectrum of BIPQ-DCP adduct in DMSO-d<sub>6</sub>

Fig. S7. <sup>31</sup>P NMR (100 MHz) spectrum of the adduct BIPQ-DCP in CDCl<sub>3</sub>



#### Fig. S8. HRMS of BIPQ-DCP product



Fig. S9. <sup>1</sup>H NMR (300 MHz) spectrum of 6-((quinolin-8-yloxy)methyl)picolinaldehyde in DMSO-d<sub>6</sub>



Fig. S10. <sup>13</sup>C NMR (75 MHz) spectrum of 6-((quinolin-8-yloxy)methyl)picolinaldehyde in DMSO-d<sub>6</sub>



Fig. S11. HRMS of 6-((quinolin-8-yloxy)methyl)picolinaldehyde



Fig. S12. Lifetime decay profile of BIPQ and BIPQ-DCP



Fig. S13. Mole ratio plot of BIPQ for DCP



Fig. S14. Contour plots of selected molecular orbitals of receptor BIPQ



Fig. S15. Contour plots of selected molecular orbitals of BIPQ-DCP adduct

#### **Determination of Quantum yield**

For measurement of the quantum yields of BIPQ and the adduct BIPQ-DCP, we recorded the absorbance of the compounds in CHCl<sub>3</sub> solution. The emission spectra were recorded using the maximal excitation wavelengths and the integrated areas of the fluorescence-corrected spectra were measured. The quantum yields were then calculated by comparison with coumarin 153 ( $\Phi$ s= 0.544 in CHCl<sub>3</sub>) as reference using the following equation:

$$\mathbf{\Phi}_{\mathbf{x}} = \mathbf{\Phi}_{\mathbf{s}} \times \left(\frac{I_x}{I_s}\right) \times \left(\frac{A_s}{A_x}\right) \times \left(\frac{n_x}{n_s}\right)^2$$

Where, x & s indicate the unknown and standard solution respectively,  $\Phi$  is the quantum yield, I is the integrated area under the fluorescence spectra, A is the absorbance and n is the refractive index of the solvent. We calculated the quantum yields of BIPQ and BIPQ-DCP using the above equation and the values are 0.21 and 0.35, respectively.

CHCl <sub>3</sub> (solvent)	Quantum yield $(\varphi)$	$\tau$ (ns)	$k_{\rm r} (10^8 \times { m s}^{-1})$	$k_{\rm nr} (10^8 \times { m s}^{-1})$
BIPQ	0.21	1.71	1.228	4.619
BIPQ-DCP	0.35	2.31	1.515	2.814

Table S1: Fluorescence lifetimes, quantum yields and radiative and non-radiative rate constants

Table S2. Vertical electronic transitions of BIPQ and BIPQ-DCP adduct calculated by TDDFT/CPCM method

Compds.	$\lambda$ (nm)	E (eV)	Osc.	Key excitations
			Strength (f)	
BIPQ	327.07	3.7907	0.2264	(86%)HOMO→LUMO
	317.76	3.9018	0.1604	(73%)HOMO→LUMO+1
	316.42	3.9184	0.4604	(92%)HOMO-1→LUMO
	236.65	5.2391	0.1650	(65%)HOMO-2→LUMO+3
	236.58	5.2406	0.3821	(30%)HOMO-3→LUMO+1
				(29%)HOMO-2→LUMO+3
	234.94	5.2772	0.3931	(35%)HOMO-6→LUMO
	208.88	5.9357	0.1484	(45%)HOMO-8→LUMO+1
BIPQ-	568.57	2.1806	0.0184	(99%)HOMO→LUMO
DCP			0.00	
	425.72	2.9123	0.3260	(33%)HOMO-2→LUMO
				(66%)HOMO-1→LUMO
	417.74	2.9680	0.1445	(65%)HOMO-2→LUMO
				(33%)HOMO-1→LUMO
	310.00	3.9995	0.1738	(97%)HOMO-1→LUMO+1
	309.11	4.0110	0.1178	(95%)HOMO→LUMO+2
	259.50	4.7777	0.1252	(91%)HOMO-7→LUMO
	232.98	5.3217	0.1931	(75%)HOMO-4→LUMO+3
	232.14	5.3409	0.6233	(32%)HOMO-3→LUMO+2



Fig. S16. Change in emission spectra of BIPQ (10  $\mu M)$  in presence of various metal ions (20  $\ \mu M)$  in CHCl3

Table S3. Comparison of the present receptor (BIPQ) with the some reported receptors for the	he
selective detection of DCP	

Receptor	Solvent system	Detection limit	Reference
R <sub>2</sub> R <sub>1</sub> N R <sup>1=</sup> H, n-Bu R <sup>2=</sup> Ac, Boc	CH <sub>3</sub> CN	21 ×10 <sup>-9</sup> M	[1]
	THF/H <sub>2</sub> O (4/1, v/v)	8.45×10⁻ <sup>8</sup> M	[2]

	H <sub>2</sub> O-CH <sub>3</sub> CN (10:1, v/v)	5.6×10 <sup>-9</sup> M	[3]
	THF	1.5×10⁻ <sup>8</sup> M	[4]
	DMF (Fluorescence turn off chemosensor)	0.065 μM	[5]
	CH <sub>3</sub> CN/H <sub>2</sub> O (1/1, v/v)	2.1×10⁻ <sup>8</sup> M	[6]
HO S	CH3CN	0.186 µM	[7]
	CH <sub>3</sub> CN/H <sub>2</sub> O (1/1, v/v)	15.8×10 <sup>-9</sup> M	[8]



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