Sensitive photoelectric sensing for 5-HMF detection

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2.1 Characterization of materials

The structure of the material was analyzed by X-ray diffraction (XRD) by Cu-K α rays ($\lambda = 1.5406$ A) with a scanning range of 5° - 90° and a scanning speed of 5°/min. The surface morphology and microstructure of SiO₂@TiO₂ and h-TiO₂ nanospheres were observed using high-resolution thermal field emission scanning electron microscopy (FESEM) and H-7650 transmission electron microscopy (TEM). The elemental composition and chemical valence states of materials were determined with X-ray photoelectron spectroscopy (XPS). Brunauer-Emmett-Teller (BET) surface area test gave N₂ adsorption-desorption isotherm and the corresponding pore size distribution.



Fig. S1 XRD images of SiO_2 ($aTiO_2$ and $h-TiO_2$).



Fig. S2 (a) Full XPS spectra of SiO₂@TiO₂ and h-TiO₂; (b) O 1s, (c) Ti 2p, (d) Si 2p corresponding XPS spectra.



Fig. S3 (a) N_2 adsorption-desorption isotherms of SiO₂@TiO₂ and h-TiO₂; (b) BJH aperture profile.



Fig. S4 CV curves of SiO₂@TiO₂/ITO, h-TiO₂/ITO and blank ITO in

 $K_3[Fe(CN)_6]$ and $K_4[Fe(CN)_6]$ mixtures.



Fig. S5 h-TiO₂ photocurrent response obtained by etching SiO_2 with different concentrations of NaOH solution.



Fig. S6 Effect of concentration of photoactive material on photocurrent response.



Fig. S7 Effect of pH of electrolyte solution on photocurrent response of photoactive materials.

Analytical method	Linear range/µM	LOD/nM	References
PAD	0.200-396.5	237.89	S1
HPLC	216-2160	216002	S2
HPLC	6.20-92.5	18.5	S3
¹ H NMR	95.2-537.5	95160	S4
ELISA	0-158.6	142.7	10
MEKC	7.90-198.2	3409.7	S5
MEKC	79.30-634.4	713.7	11
LC-MS	7.90-79.3	2616.9	S6
FAAS	31.70-1903.1	10071.1	S7
h-TiO ₂ PEC	0.00001-0.100	0.001	this work

Table S1 Comparison of 5-HMF detection by different methods.

References

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