

Supporting Information

GSH-Depleting Metal-Polyphenol-Network Nanoparticles with Dual Enzyme Activities Endow Enhanced Ferroptosis

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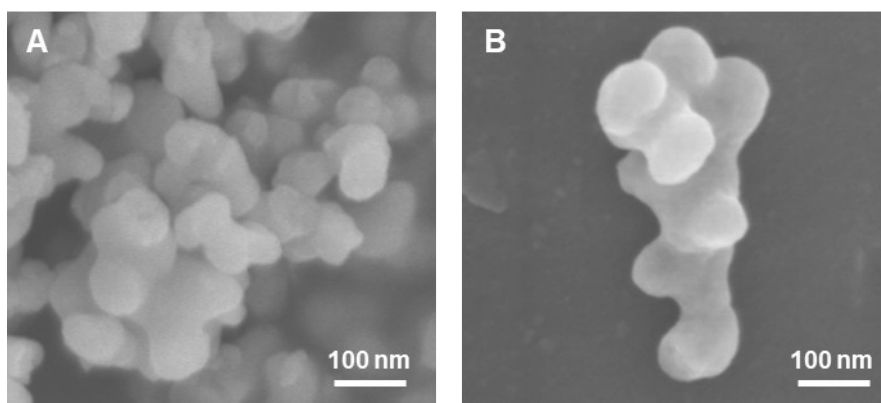


Figure S1. SEM images of Ce-aMOFs (A) and MEFs (B) nanoparticles, with a scale bar of 100 nm.

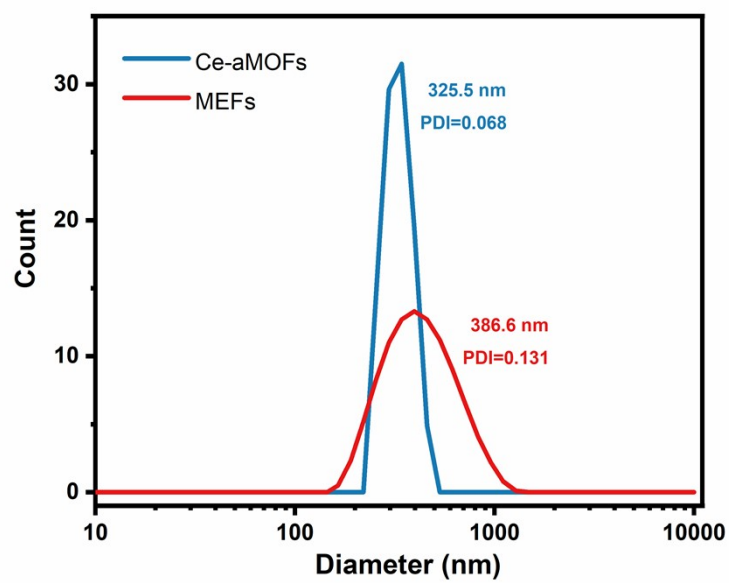


Figure S2. DLS analysis results for Ce-aMOFs and MEFs ($100 \mu\text{g mL}^{-1}$, $n=3$).

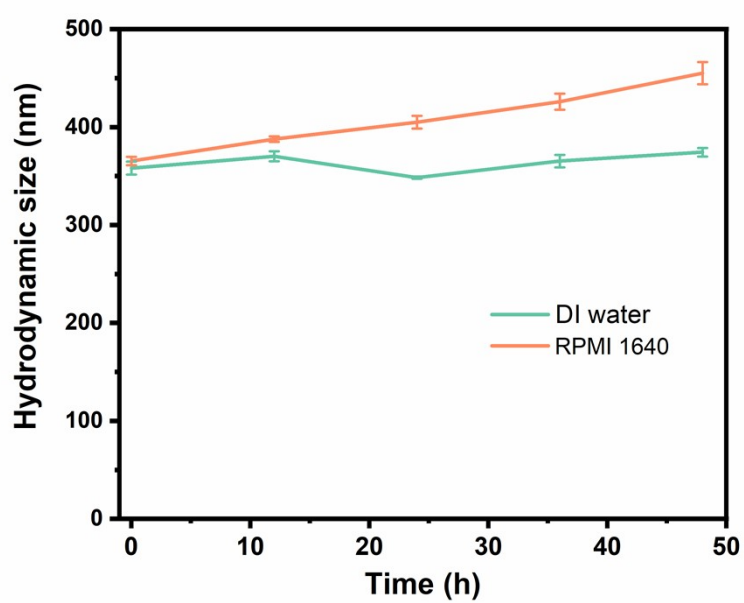


Figure S3. Stability of MEFs in DI water and RPMI 1640 basic medium at room temperature, (n=3).

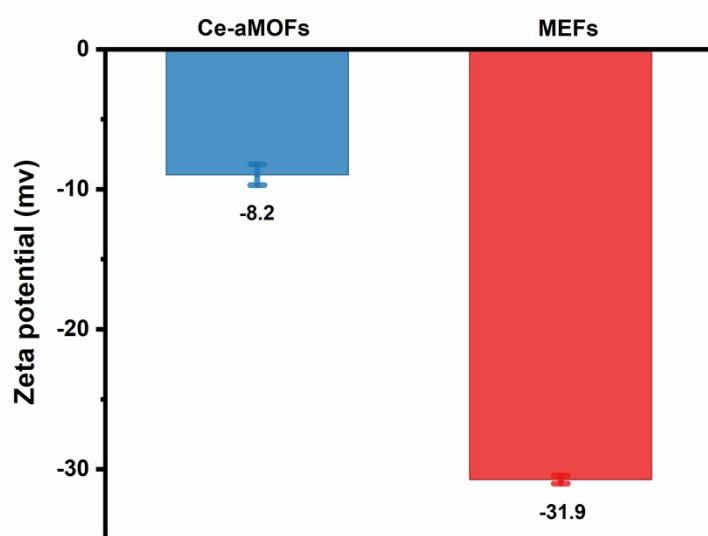


Figure S4. Zeta potentials of Ce-aMOFs and MEFs at pH 7.4 (n=3).

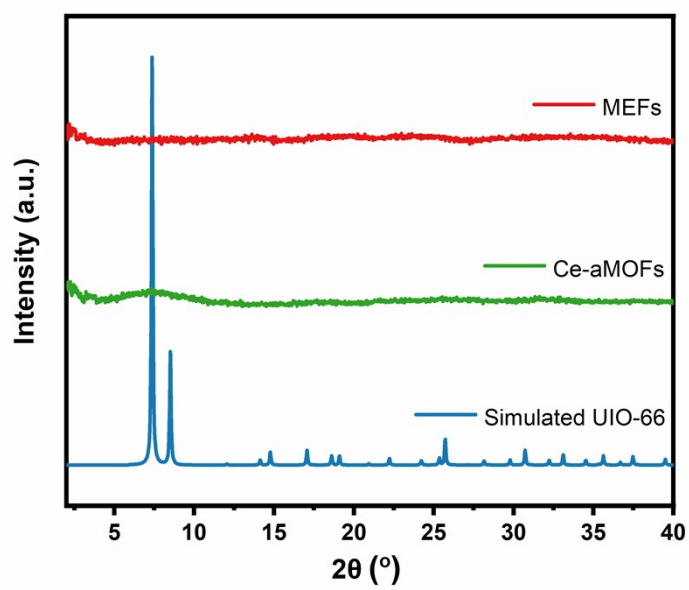


Figure S5. XRD patterns for Ce-aMOFs and MEFs.

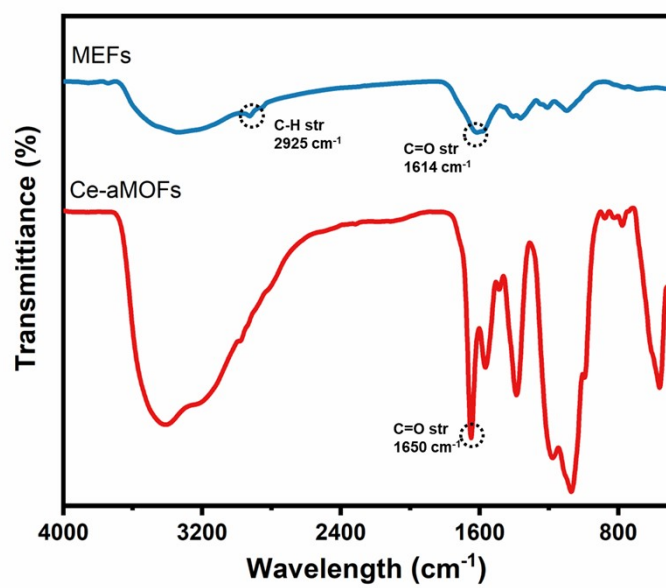


Figure S6. FT-IR spectra of Ce-aMOFs and MEFs.

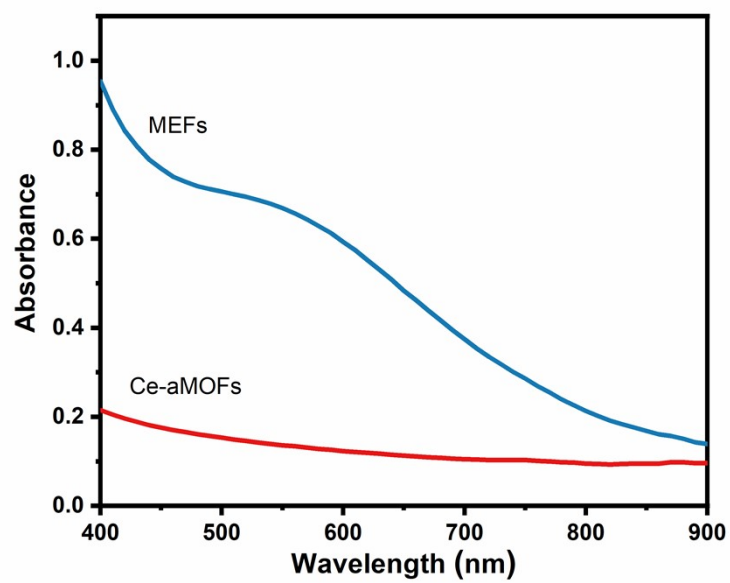


Figure S7. UV-vis-NIR absorption spectra of $100 \mu\text{g mL}^{-1}$ Ce-aMOFs and MEFs.

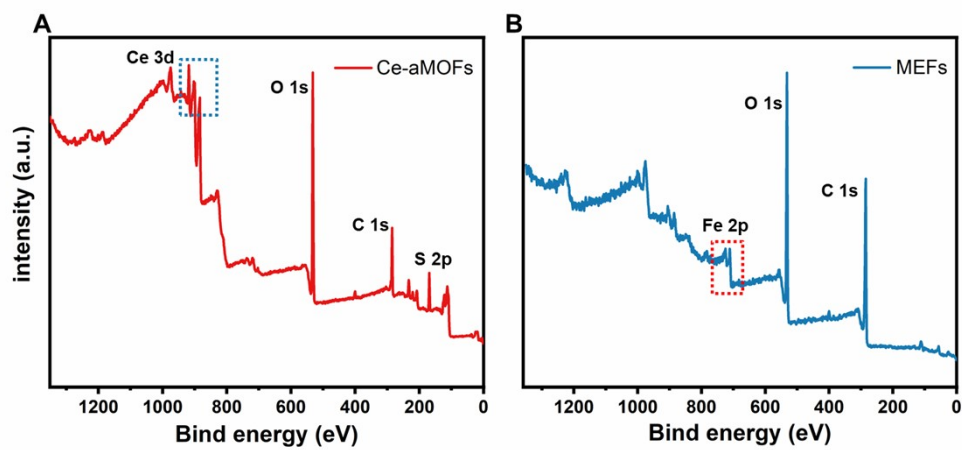


Figure S8. X-ray photoelectron spectroscopy survey spectrum of Ce-aMOFs and MEFs (Elements: C 1s, O 1s, S 2p, Cu 2p, Ce 3d).

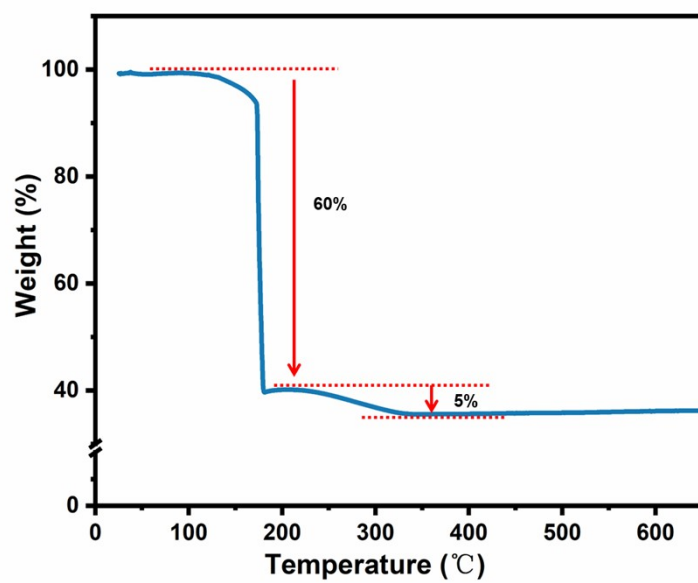


Figure S9. TGA analysis for MEFs in O₂ gas purging at a flow rate of 20 mL min⁻¹.

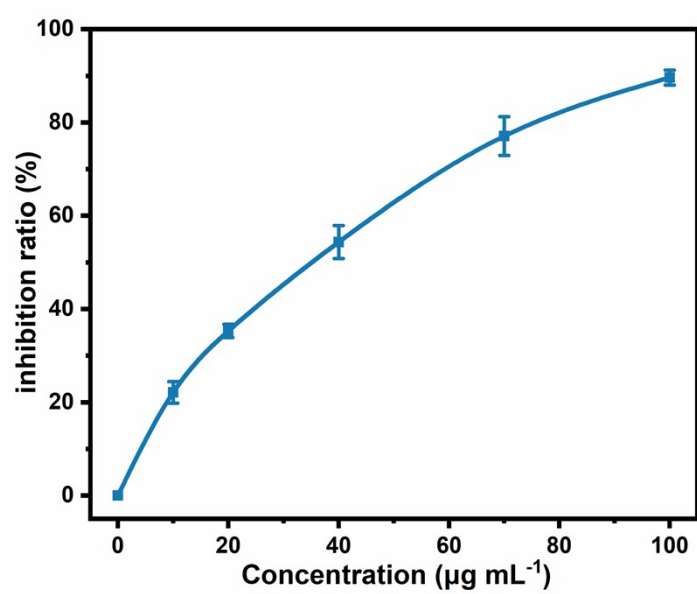


Figure S10. Inhibition rate of Ce-aMOFs (0-100 $\mu\text{g mL}^{-1}$) against the production of WST formazan.

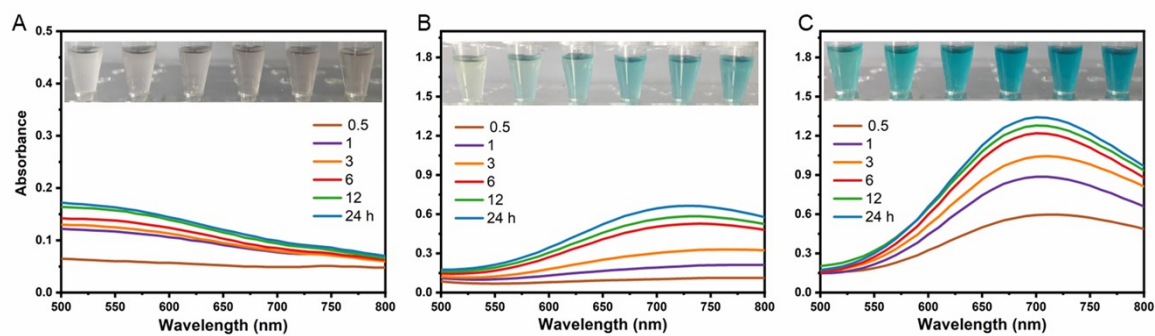


Figure S11. UV/vis-NIR absorption spectra of the MEFs solutions (200 $\mu\text{g mL}^{-1}$) after releasing Fe^{3+} at various pH values (A: pH 7.4, B: pH 6.5, C: pH 5.0). Inset: Prussian blue staining of the supernatant acquired at different incubation times.

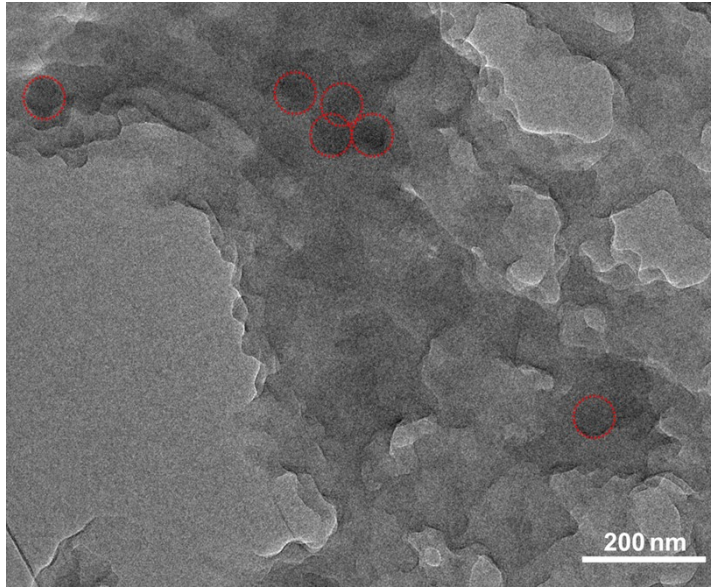


Figure S12. TEM image of MEFs degrading in pH 5.0 pbs for 12h, with a scale bar of 100 nm.

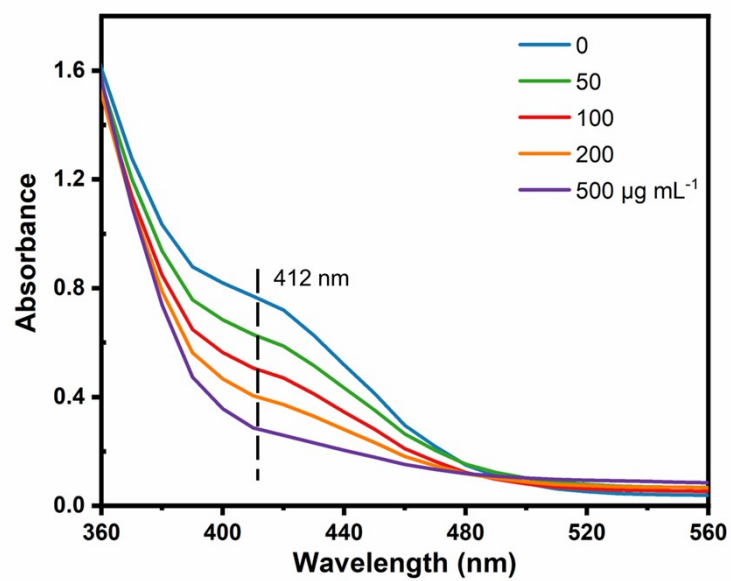


Figure S13. GSH depletion by MEFs with different concentration levels (0-500 $\mu\text{g mL}^{-1}$) in 8 h, as characterized by the absorbance at 412 nm.

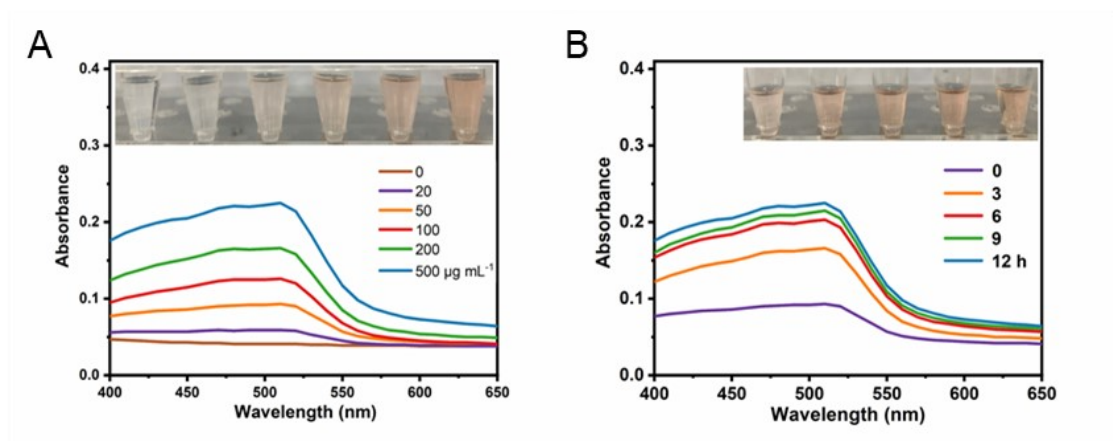


Figure S14. (A) The time-dependent production of Fe^{2+} from 500 $\mu\text{g/mL}$ MEFs in the presence of GSH (5 mM). (B) The variation of UV/vis absorption spectra of the 1,10-phenanthroline hydrate- Fe^{2+} complexes after treatment with MEFs (0-500 $\mu\text{g mL}^{-1}$) for 6 h in an environment containing GSH (5 mM).

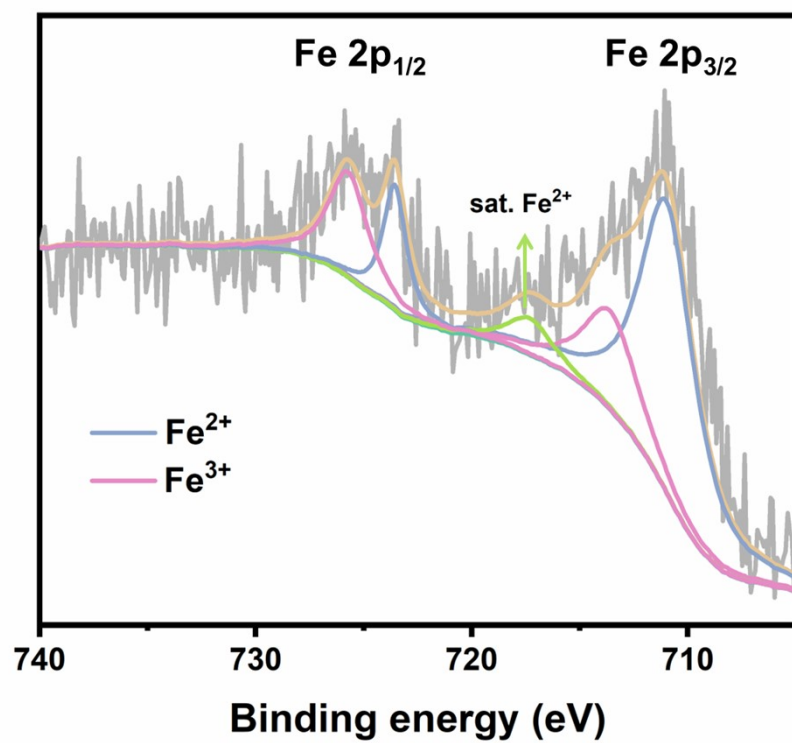


Figure S15. Valence status of Fe element in MEFs after reaction with GSH for 1h.

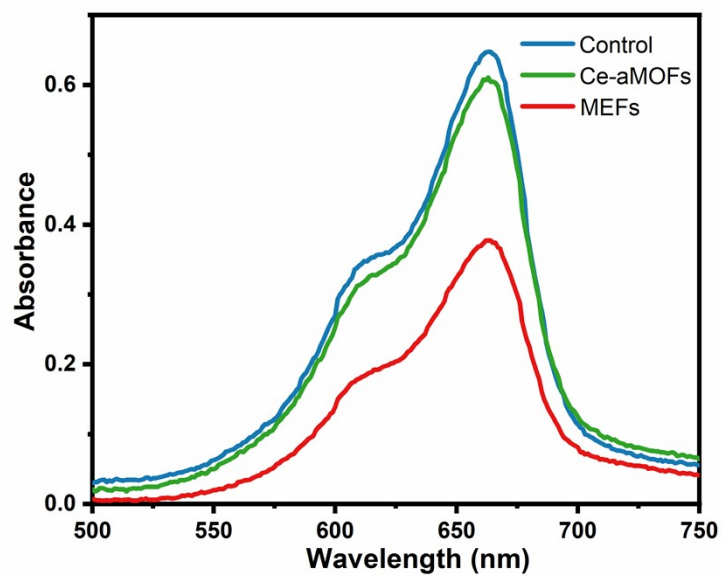


Figure S16. MB degradation by $\cdot\text{OH}$ generated under $100\text{ }\mu\text{g mL}^{-1}$ Ce-aMOFs and MEFs with an incubation time of 2 h. Control group: GSH ($100\text{ }\mu\text{L}$, 1 mM), H_2O_2 ($80\text{ }\mu\text{L}$, 100 mM) and MB ($10\text{ }\mu\text{L}$, $100\text{ }\mu\text{g mL}^{-1}$) in $500\text{ }\mu\text{L}$ aqueous solution.

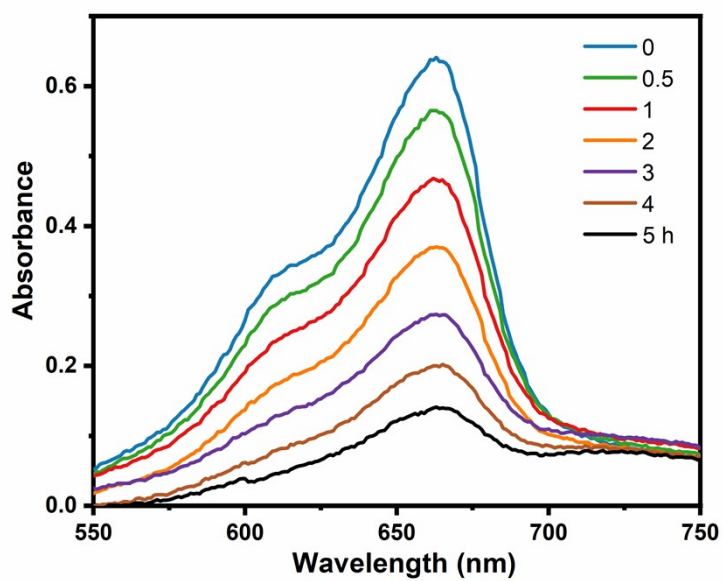


Figure S17. Degradation of MB by the MEFs-mediated Fenton-like reaction at different times with a concentration of $100 \mu\text{g mL}^{-1}$.

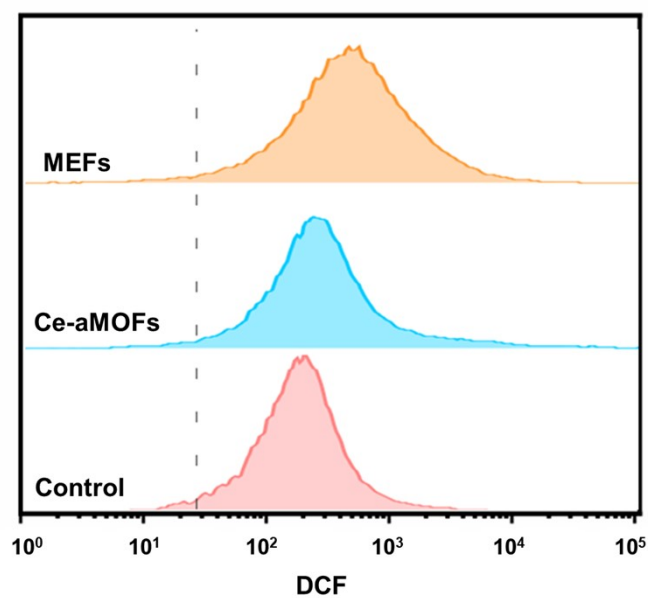


Figure S18. Flow cytometry analysis of ROS generation after treated with Ce-aMOFs and MEFs ($100 \mu\text{g mL}^{-1}$) for 8 h.

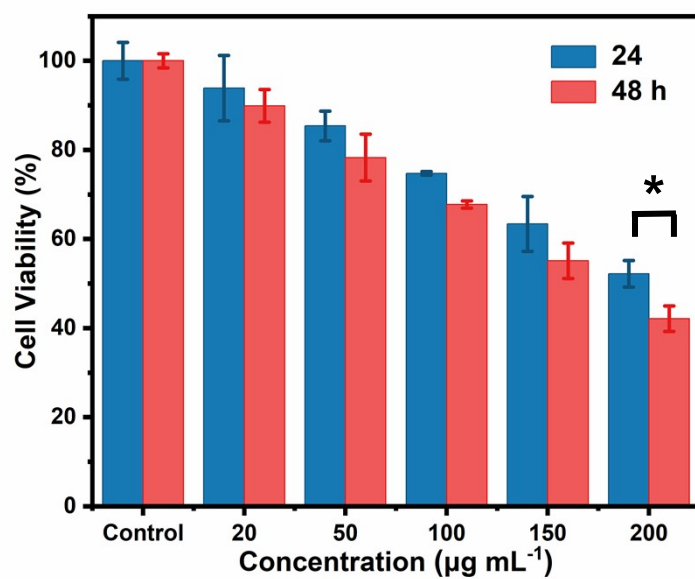


Figure S19. 4T1 cell viability after incubating with 20-200 $\mu\text{g mL}^{-1}$ of Ce-aMOFs for 24 h and 48h.

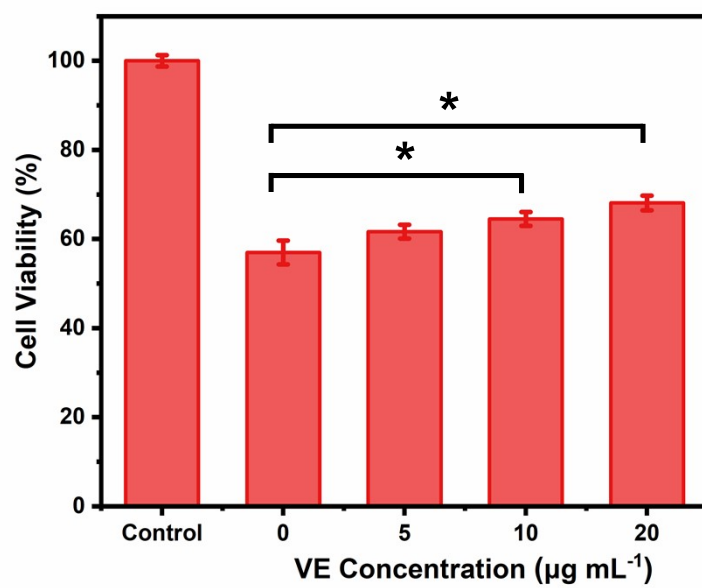


Figure S20. Relative cell viabilities of MEFs-treated 4T1 cells in the presence of vitamin E (VE).

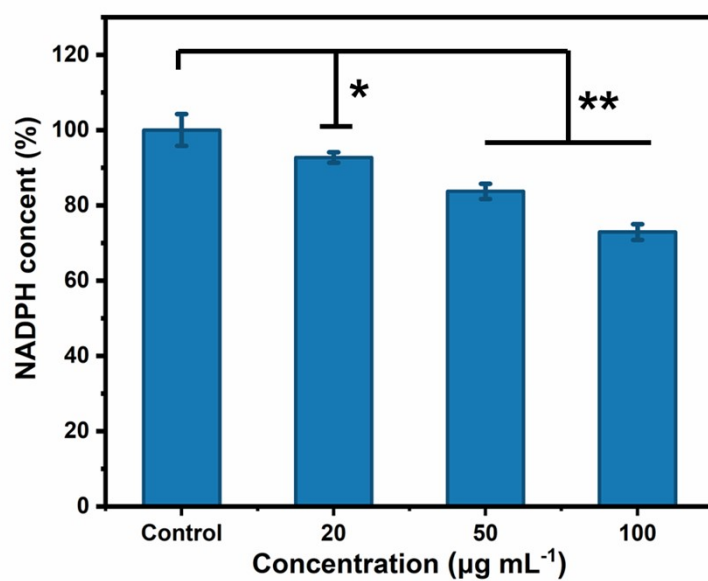


Figure S21. NADPH levels in 4T1 cells after treatment with Ce-aMOFs (20-100 $\mu\text{g mL}^{-1}$) for 6 h.

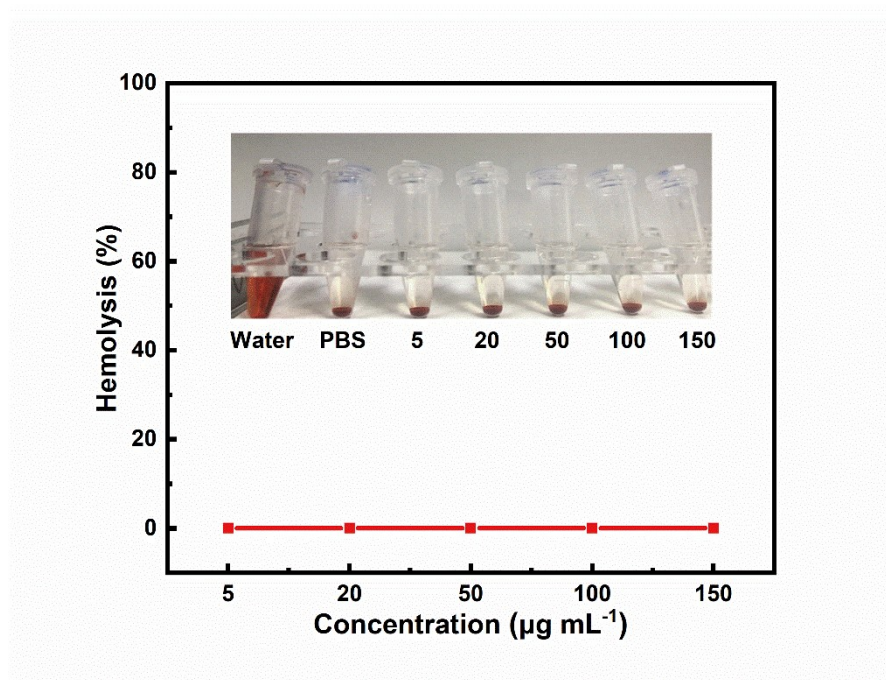


Figure S22. Hemolysis of MEFs ($5\text{-}150\ \mu\text{g mL}^{-1}$) after incubation with red blood cells. PBS: negative control, deionized water: positive control. Inset: hemolysis photographs after centrifugation.

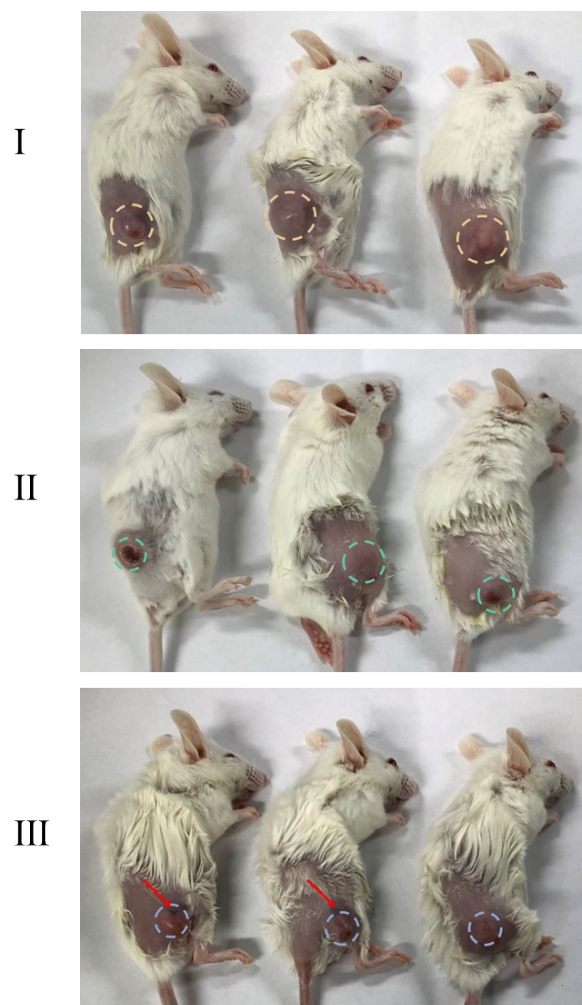


Figure S23. Photographs of 4T1 tumor-bearing mice from different treatment groups after 14 days of treatments. (I) PBS (pH 7.4, 10 mM, 50 μ L), (II) Ce-aMOFs solution (2 mg mL⁻¹, 50 μ L), (III) MEFs solution (2 mg mL⁻¹, 50 μ L).

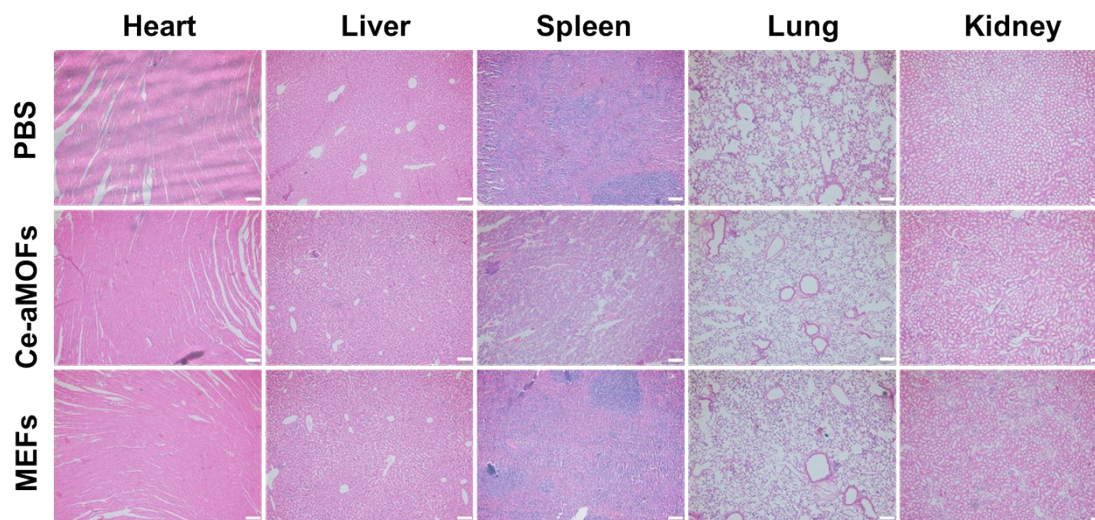


Figure S24. Images of H&E-stained major organ slices of mice from different treatment groups after 16 days of treatments. Scale bar 50 μm .