

# **Biodegradable Covalent Organic Frameworks Achieving Tumor Micro-Environment Responsive Drug Release and Antitumor Treatment**

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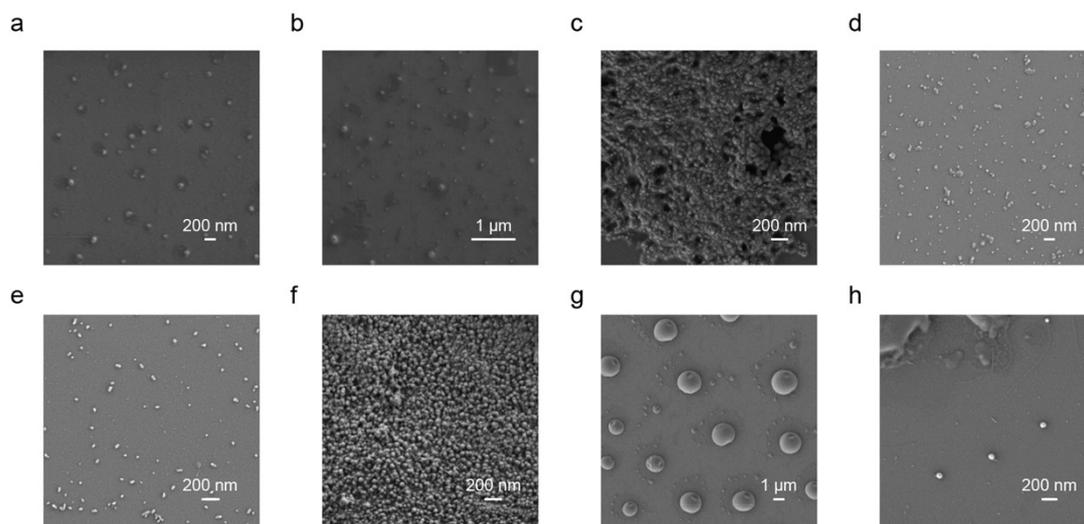
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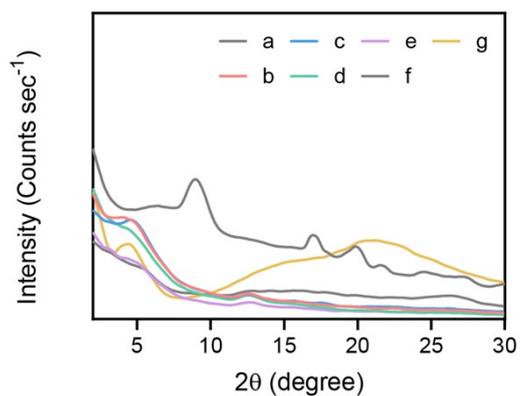
Dr. Kai Hao, State Key Laboratory of Physical Chemistry of Solid Surfaces, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, China, E-mail: kaihao@xmu.edu.cn.

**Table S1.** Different conditions for the exploration of synthetic COF-TpAzo NPs.

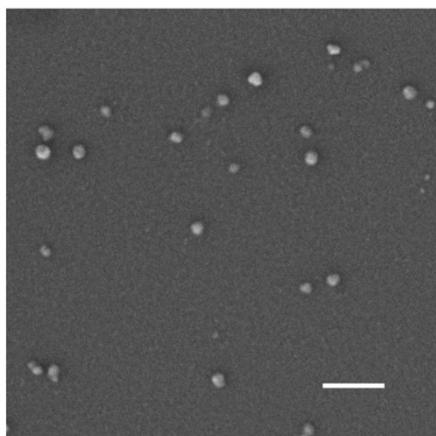
Entry	Solvent	Volume of TFA ( $\mu\text{L}$ )	Temperature ( $^{\circ}\text{C}$ )	Time (h)	Precipitate solvent	Wash solvent	SEM	XRD
1	DCM	20	70	15	ether	DMF	S1a	-
2	DCM	20	70	15	ether	DCM	S1b	S2a
3	DMF	20	120	15	ether and n-hexane	-	S1c	-
4	DCM	20	70	15	n-hexane	-	S1d	S2b
5	DCM	20	35	24	n-hexane	DMAc	S1e	S2c
6	DCM	20	70	24	n-hexane	DMAc	S1f	S2d
7	DCM	20	70	72	n-hexane	DMAc	S1g	S2e
8	Dry DCM	20	70	72	n-hexane	DMAc	S1h	S2f
9	DCM	10	70	72	n-hexane	DMAc	-	S2g



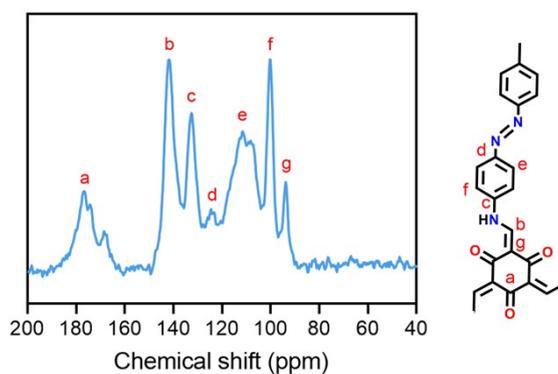
**Figure S1.** SEM images of the nanoparticles obtained by the corresponding synthesis method in **Table S1**.



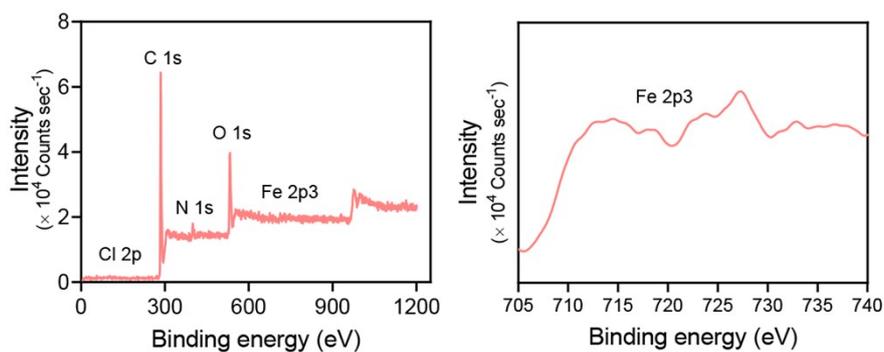
**Figure S2.** XRD spectrums of the nanoparticles obtained by the corresponding synthesis method in **Table S1**.



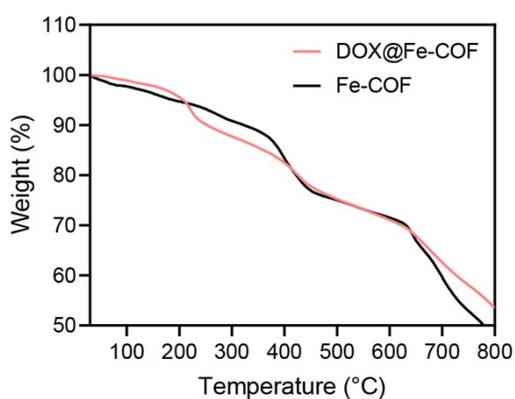
**Figure S3.** Larger magnification SEM image of the COF-TpAzo NPs. Scale bar: 200 nm.



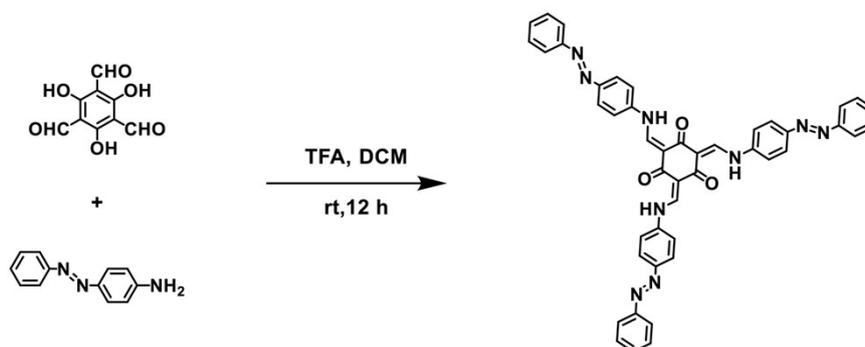
**Figure S4.** <sup>13</sup>C CP-MAS NMR spectra of the COF-TpAzo NPs.



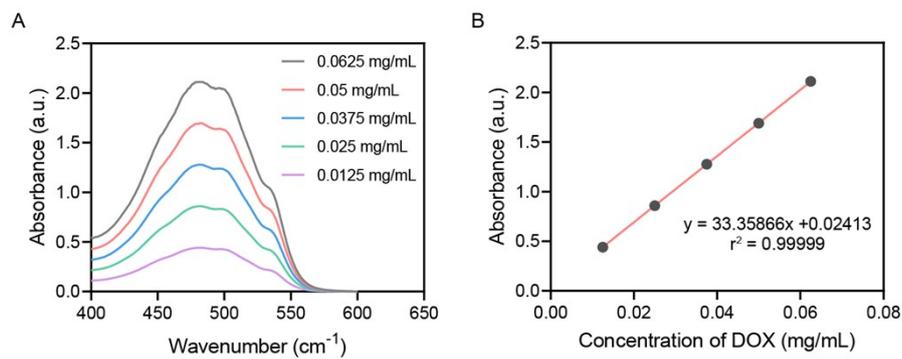
**Figure S5.** XPS spectra of Fe-COF.



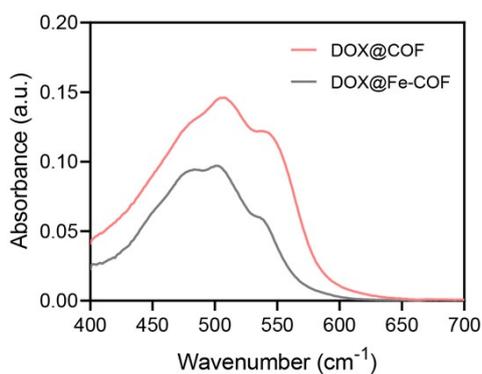
**Figure S6.** Thermogravimetric analyses of COF-TpAzo and Fe-COF under  $N_2$  atmosphere.



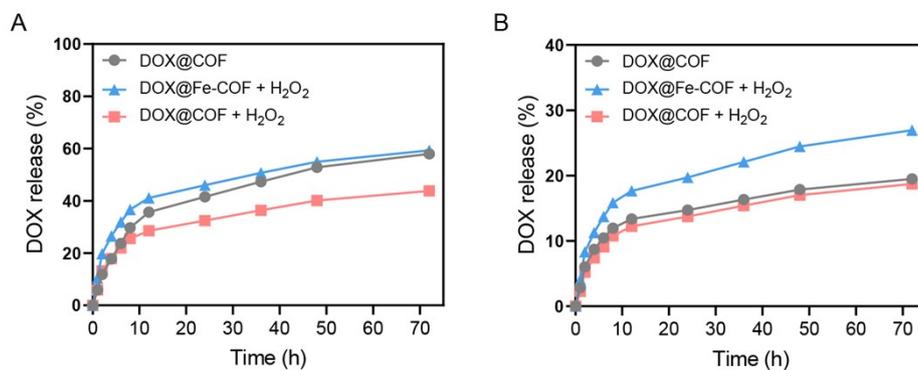
**Figure S7.** The synthesis route of the model compound.



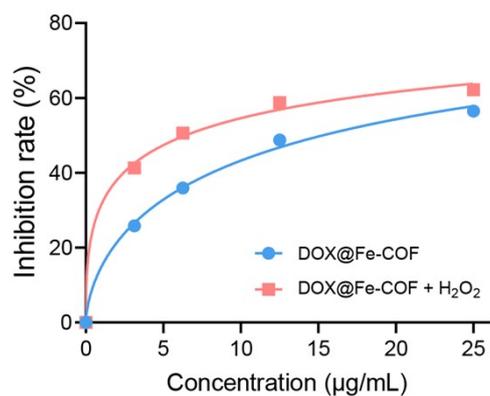
**Figure S8.** (A) UV-vis spectra of DOX·HCl of different concentrations in DMSO. (B) Standard curve of DOX·HCl in DMSO.



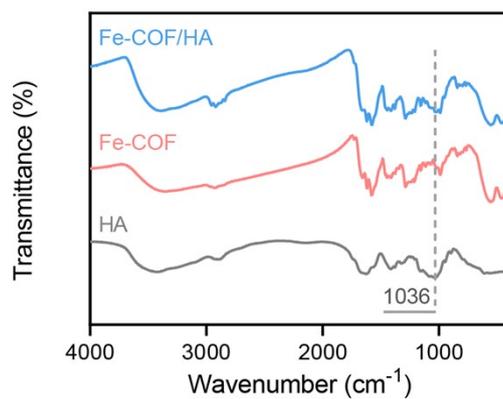
**Figure S9.** UV-vis spectra of DOX soaked from 1 mg DOX@COF NPs and 1 mg DOX@Fe-COF NPs in DMSO.



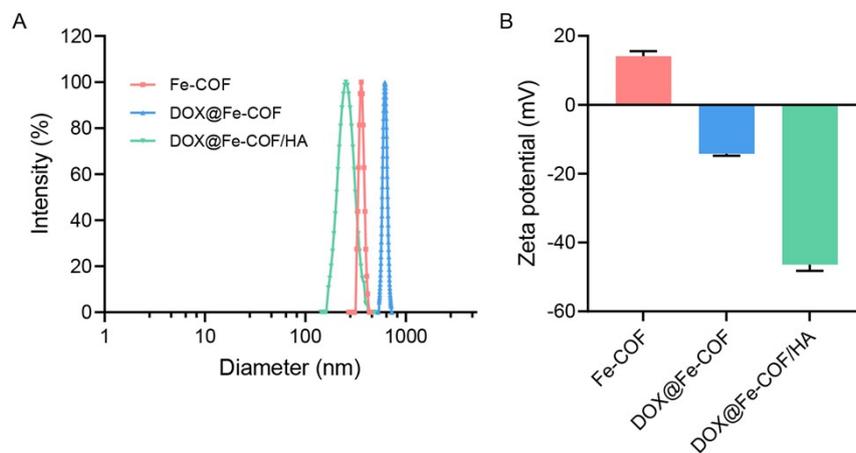
**Figure S10.** DOX release profiles from NPs in PB solution of different pH (A) pH = 5.5. (B) pH = 7.4.



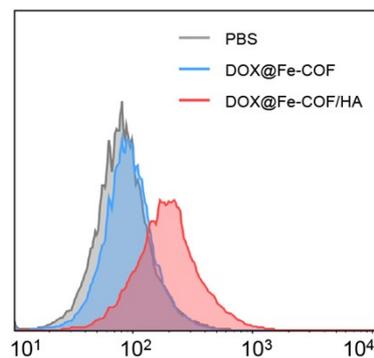
**Figure S11.** Calculation of IC<sub>50</sub> of DOX@Fe-COF with and without H<sub>2</sub>O<sub>2</sub>. IC<sub>50</sub> of DOX@Fe-COF without H<sub>2</sub>O<sub>2</sub> was 15.28 µg/mL; IC<sub>50</sub> of DOX@Fe-COF with H<sub>2</sub>O<sub>2</sub> was 6.43 µg/mL.



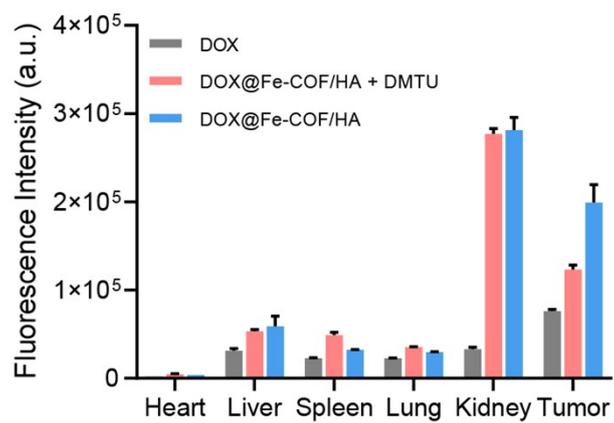
**Figure S12.** FT-IR spectrum of Fe-COF/HA, Fe-COF, and HA. The peak at 1036 cm<sup>-1</sup> was attributed to the C-O-C stretching of HA.



**Figure S13.** Hydrodynamic size and zeta potential of Fe-COF, DOX@Fe-COF, DOX@Fe-COF/HA. Fe-COF:  $333.6 \text{ nm} \pm 6.6 \text{ nm}$ ; DOX@Fe-COF:  $602.5 \text{ nm} \pm 7.6 \text{ nm}$ ; DOX@Fe-COF/HA:  $295.7 \text{ nm} \pm 3.2 \text{ nm}$ .



**Figure S14.** Flow cytometry of cellular uptake of DOX@Fe-COF and DOX@Fe-COF/HA.



**Figure S15.** The quantification of the normalized fluorescence signal of the organs and tumors of each group.